Lesson Plans: T. Coble Date: 11/11-11/15/2019

Subject: Chemistry I (Pre-AP) Periods: 1 - 4

Use the periodic table as a model to predict the relative properties of elements based on the patterns of electrons in the outermost energy level and the composition of the nucleus of atoms.

	Daily Objectives	Activities
Monday	VETERAN'S DAY	NO SCHOOL
Tuesday	 The student will be able to draw the noble gas configuration of any element on the periodic table. The student will be able to identify an element given any of the three notations. 	 Bell-Ringer Notes, Chapter 4, Section 2 a. Noble-gas Configurations b. Using the 3 Configurations HW: Chapter 4, All Sections
Wednesday ROAR PERIOD	The teacher will demonstrate the proper use of lighting & extinguishing a bunsen burner and safety concerns associated with it.	 Bell-Ringer Lab: Flame Test Demonstrate Bunsen Burner Pre-Lab questions
Thursday	The student will calculate the energy release by the color of the flame produced by an element.	 Bell-Ringer Lab: Flame Test
Friday	The student will calculate the energy release by the color of the flame produced by an element.	 Bell-Ringer Lab: Flame Test

Formative Assessment: - Review bell ringers, procedures and results, Class Discussion, Method of checking:

teacher observation of student practice **Summative Assessment**: – Flame Test

Accommodations: as per IAP, IEP

Materials: text, Powerpoint slides, Wks., smart board, Lab material, Cellular devices