Core Content – Chemistry 2022-2023

Student Name:	School:
Teacher:	
Atoms and the Periodic Table	Energy in Chemical Reactions
☐ Historical Development of Atomic Theory	☐ Energy
☐ Modern Atomic Theory	☐ Heat
☐ The Structure of the Atom	☐ Calorimetry
☐ Elements, Compounds, and Mixtures	☐ Lab: Calorimetry and Specific Heat
☐ Atomic Numbers and Electron Configurations	☐ Thermochemical Equations
☐ The History and Arrangement of the Periodic	☐ Enthalpy of Reaction
Table	Mixtures, Solutions, and Acid-Base Reactions
☐ Periodic Trends	☐ Mixtures and Solutions
States and Changes of Matter	☐ Solutions and Solubility
☐ Gases	☐ Lab: Solubility
☐ Liquids	☐ Measures of Concentration: Molarity
☐ Solids and Plasmas	☐ Properties of Acids and Bases
☐ Phase Changes	☐ Arrhenius, Bronsted-Lowry, and Lewis Acids
☐ Changes in Matter	and Bases
☐ Lab: Physical and Chemical Changes	□ pH
Chemical Bonding and Reactions	☐ Lab: Measuring pH
☐ Ionic Bonding	☐ Neutralization Reactions
☐ Covalent Bonding	Organic Chemistry
☐ Lab: Ionic and Covalent Bonds	☐ Organic Compounds
☐ Evidence of Chemical Reactions	☐ Properties and Uses of Saturated Hydrocarbons
☐ Writing and Balancing Chemical Equations	☐ Properties and Uses of Unsaturated
☐ Types of Chemical Reactions	Hydrocarbons
☐ Lab: Types of Reactions	☐ Functional Groups
☐ Oxidation-Reduction	☐ Organic Reactions
☐ Balancing Oxidation-Reduction Equations	Nuclear Chemistry
Stoichiometry and the Gas Laws	☐ Types of Radioactive Decay
☐ Molar Masses	☐ Balancing Nuclear Reactions
☐ Introduction to Stoichiometry	☐ Half-Life
☐ Stoichiometric Calculations	☐ Lab: Half-Life
☐ Limiting Reactant and Percent Yield	☐ Nuclear Fission and Nuclear Fusion
☐ Gas Laws	☐ Nuclear Energy
☐ Lab: Charles' Law	
☐ Lab: Boyles Law	
☐ The Ideal Gas Law	
Reaction Rates and Equilibrium	
☐ Reaction Rate	
☐ Lab: Reaction Rate	
☐ Reversible Reactions and Equilibrium	

☐ Shifts in Equilibrium