

Free their potential - teach



Future
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The whiteboard contains the following content:

Chemical equation: $2\text{H}_2(\text{g}) + \text{O}_2(\text{g}) \rightleftharpoons 2\text{H}_2\text{O}(\text{g}) + \text{Heat}$ with $\Delta H = -24$

CHANGE	SHIFT	EFFECT ON QUANTITY	EFFECT	INC DEC NONE
INC TEMP	→	$\text{H}_2\text{O}(\text{g})^P$	DE	
INC PR	←	$\text{H}_2(\text{g})^R$		
INC VOL	→	TEMP		

Other notes on the board include: $R \Rightarrow P$, $E_a = \text{ACTIVATION ENERGY}$, $\Delta H = \text{ENTHALPY CHANGE}$, and an energy profile diagram showing a reaction with activation energy E_a and enthalpy change ΔH .

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