# **Water Testing Activity**

#### Word Bank:

**Turbidity-** Measures the clarity of the water. Turbid water is caused by soil erosion, runoff, algae, and bottom sediment. Ideal turbidity is clear.

**pH-** Measures how acidic of basic the water is. The range is 0-14, 0 being the most acidic and 14 being the most basic. An example of a basic is Baking Soda. An example of an acid is Vinegar. Ideal pH is between 6.5 and 8.

**Dissolved Oxygen-** Dissolved Oxygen is important to the livelihood of aquatic life. The saturation % of dissolved oxygen shows water quality. Cold water can hold more oxygen than warm water. High levels of bacteria or rotting plants can cause % Saturation to decrease.



#### Secchi Disk-

**% of Saturation-** The level of Dissolved Oxygen in the water. If water is cold it can hold more oxygen and therefore will most likely have a lower level of a % saturation.

#### **Procedure:**

## **Step One: Identify the temperature**

- 1. Fill your jar with water (fill to the fill line).
- 2. Look to the side of the jar and identify what the temperature says in Celsius.
- 3. Record your results on the data sheet below.

#### **Step Two: Test the Turbidity\***

- 1. Look to the bottom of the container to see the Secchi Disk\* icon.
- 2. Compare the appearance of the Secchi Disk Icon at the bottom of the jar to the measurement of the turbidity on the provided chart.
- 3. Record your results on the data sheet below.

# **Step Three: Test the pH\***

- 1. Fill the large test tube to the 10 mL mark using water from your jar.
- 2. Add **one** pH tablet to the test tube (be sure you are using the **pH tablet**).
- 3. Cap and mix the test tube until the tablet has disappeared completely.
- 4. Compare the color of the sample to the pH color on the provided chart.
- 5. Record your results on the data sheet below.

## **Step Four: Test the Dissolved Oxygen\***

- 1. Fill the small vial with water all the way to the top using water from your jar.
- 2. Drop <u>two</u> Dissolved Oxygen tabs into vial-causing the water to overflow (be sure you are using the Dissolved Oxygen tablet).
- 3. Tighten the cap and shake for approximately 4 minutes until the tablet has completely disappeared.
- 4. Wait 5 minutes for the color to fully develop and compare to the sample Dissolved Oxygen on the provided chart.
- 5. Record your results on the data sheet below.

#### **Step Five: Identify the % of Saturation\***

- 1. Use the provided chart to identify the % of oxygen saturation of your water sample using your data for the temperature and the Dissolved Oxygen results.
- 2. Record your results on the data sheet below.

# Earth Echo Monitoring Data Sheet: Water Sample Set: A or B

Parameter	Sample 1	Sample 2
Water Temp Celsius		
Turbidity		
рН		
Dissolved Oxygen		
% Oxygen Saturation		

1. Write one statement about your results (something you learned or was really interesting or unexpected):
2. Write one question you now have that might lead to further study (think about HOW or WHY questions):