Species - Variety:
**Musgravite (MagnesioTaffite)**

Locality (as stated in Dr. Gübelin's records):
**Not stated**

**Description**
- **Shape:** Pear
- **Weight:** 3.58 ct
- **Dimensions:** 12.34 x 6.23 x 6.85 mm
- **Diaphaneity:** Transparent
- **Color:** Medium light, grayish, bluish Violet

**Gemological Properties**
- **Optical Character:** Uniaxial negative
- **Refractive Index:** 1.715 / 1.722
- **Birefringence:** 0.007
- **Specific Gravity:** 3.61
- **Pleochroism:** None
- **Luster:** Vitreous
- **Phenomenon:** None
- **Fluorescence:**
  - LWUV: None
  - SWUV: None
- **Phosphorescence:**
  - LWUV: None
  - SWUV: None
- **Absorption Spectrum:**
  - Broad absorption below 425 nm and between 460-475 nm

**Photomicrograph**
This musgravite contains numerous small solid and fluid inclusions.

A group of large rounded inclusions thought to be apatite. The field of view is 4.35 mm across.

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Transmission infrared absorption spectra were collected in the mid-IR range (400–6000 cm⁻¹) using a Thermo-Nicolet 6700 FTIR spectrometer (KBr beamsplitter, MCT-B detector, 1 to 4 cm⁻¹ resolution, and 128 scans). A 6x beam condenser was used to focus light through the girdle region of the gemstone to obtain a maximum signal.

Raman spectra were collected over the range 100–2000 cm⁻¹ Raman shift using a Renishaw InVia Raman microscope (3 additive scans, and excitation produced by 514.5 nm Ar-ion or 633 nm He-Ne lasers). Analysis was done using a focused beam on the table facet of the gemstone unless otherwise noted.

Photoluminescence spectra were collected in the 517–1000 nm range using a Renishaw InVia Raman microscope (single scan, and excitation produced by 514.5 nm Ar-ion laser). Analysis was done using a focused beam on the table facet of the gemstone unless otherwise noted, and the laser intensity was adjusted to avoid detector saturation caused by sample fluorescence.

Transmission absorption spectra in the UV to near-IR range (250–1000 nm) were collected using a Hitachi U4001 spectrometer (2.0 nm slit, 1.0 nm data interval, and 120 nm/min scan speed). Unless otherwise noted, an unpolarized light beam was focused on the culet area of the optically unoriented sample.
Chemistry | EDXRF

**Qualitative energy-dispersive X-ray fluorescence (EDXRF) data** were collected using a Thermo ARL QuantX EDXRF analyzer (15 kV, 0.02–1.98 mA (auto adjusted based on deadtime), 100 seconds livetime, vacuum atmosphere, aluminum filter). These conditions were selected to optimize the detection of the transition metals (such as iron or chromium) that are responsible for the coloration of most gemstones. Only the peaks for chemical elements that were clearly identifiable are labeled by element symbols in the spectrum. Most other peaks are diffraction artifacts. Our EDXRF method can detect most elements with higher atomic numbers than sodium ($Z=11$) in the Periodic Table; depending on the element, the lower limit of detection is between 1 and 100 parts per million (0.0001 to 0.01%).