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Defense Logistics Agency

Defense Supply Center Richmond

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FINAL

REMEDIAL ACTION WORK PLAN ADDENDUM OPERABLE UNIT 7 DEFENSE SUPPLY CENTER RICHMOND

Prepared For



Defense Logistics Agency

And



United States Army Corps of Engineers
Baltimore District

Prepared by

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Executive Summary

This Remedial Action Work Plan (RAWP) Addendum describes the proposed follow-up in situ bioremediation (ISB) actions at Operable Unit (OU) 7 at Defense Supply Center Richmond (DSCR). OU 7 consists of impacted groundwater beneath and downgradient of OU 4 (Fire Training Area Source Area) along the central southern boundary of the installation. Contaminants from the OU 13 area (polycyclic aromatic hydrocarbon area) near OU 4 have also potentially impacted OU 7 groundwater. Three dissolved phase plumes of volatile organic compounds (VOC) are associated with former Fire Training Pit Areas 1, 2, and 3. Primary constituents in groundwater are tetrachloroethene (PCE), trichloroethene (TCE), cis-1,2-dichloroethene (cDCE), and vinyl chloride (VC) present at concentrations greater than cleanup levels established in the 2012 Record of Decision (ROD) for OU 7. The primary zone of groundwater impacted at OU 7 is the surficial aquifer, which is the target of follow-up ISB actions.

Follow-up ISB actions at OU 7 will target accelerated treatment of remaining hot spot areas in the surficial aquifer. Addressing these hot spot areas will reduce contaminant mass and contaminant flux to downgradient areas. ISB actions proposed in this work plan will target additional treatment of diffuse VOC plumes farther downgradient of source zones with the objective of reducing overall plume areas.

Proposed ISB actions for the Pit 1 plume will target additional treatment of two hot spot areas with the highest concentrations of PCE, TCE, cDCE, and VC in groundwater at OU 7. A third treatment area will target additional treatment of the diffuse VC plume in the farthest downgradient plume area in the Pit 1 monitored area. Proposed ISB actions for the Pit 2 plume will target additional treatment of two hot spot areas for degradation products cDCE and VC in previous 2021 treatment areas. A third area will target additional treatment of a diffuse VC plume in the farthest downgradient plume area in the Pit 2 monitored area. The proposed ISB action for the Pit 3 plume will address a remaining degradation product hot spot in the mid-plume area as a follow-up to 2021 ISB actions.

The process option of the ISB design in this work plan follows the remedial design/remedial action work plan for OU 7 (AECOM 2013) using metabolic anaerobic reductive dechlorination as the targeted degradation process to treat the chlorinated solvents. In this reaction, microorganisms gain energy as one or more chlorine atoms on a chlorinated ethene or ethane compound molecule are replaced with hydrogen atoms in an anaerobic environment. The chlorinated compound serves as the electron acceptor and molecular hydrogen usually serves as the electron donor (source of energy). Hydrogen used in this reaction is supplied by fermentation of organic substrates or a direct electron donor. Biodegradation of an organic substrate depletes the aquifer of dissolved oxygen, and sequentially reduces native electron acceptors nitrate, manganese, iron, sulfate, and carbon dioxide. In general, metabolic anaerobic reductive dechlorination occurs by sequential removal of chlorine atoms with the sequential reaction consisting of PCE \rightarrow TCE \rightarrow cDCE \rightarrow VC \rightarrow ethene.

For hot spot and source area treatment, emulsified vegetable oil (EVO) is the selected ISB substrate comprised of food-grade soybean oil, emulsifiers, and amendments with demonstrated effectiveness to support enhanced reductive dechlorination (ERD). Evidence of complete ERD pathways to ethene and methane is apparent for previous EVO injections at OU 7 and treatability studies. The low solubility of EVO provides for a long-lasting carbon source due to its slow rate of chemical dissolution into groundwater. EVO can also help sequester chlorinated VOC compounds, which will further reduce their mobility in the aquifer.

The hot spot and source area treatment, the injection process option selected for enhanced ISB for the surficial aquifer at OU 7 is direct push technology (DPT) using a pressure activated injection probe. Previous investigations, pre-design investigations and testing performed at OU 7 indicate that the high density of the mid-to-lower interval of the surficial aquifer will require injection pressures greater than 100 pounds per square inch to distribute reagents in this zone. The high density and variability of the surficial aquifer has limited the effectiveness of previous ISB actions using injection wells in this zone. The optimized reagent mixture will include EVO, sodium bicarbonate for pH buffering, and sodium ascorbate to create anaerobic water for bioaugmentation cultures to enhance and accelerate biodegradation processes. Designs for the treatment areas include 51 DPT injection points and five (5) injection wells. The ISB design period is three (3) years for EVO injections and 1.5 years for sodium lactate.

Sodium lactate is the selected ISB substrate for additional ISB treatment of diffuse vinyl chloride plumes at Pits 1 and 2 in the farthest downgradient areas. This will involve gravity feed injections at one injection well in the Pit 1 plume area and four injection wells in the Pit 2 plume area.

The proposed ISB actions will include remedy verification and performance monitoring. Injection process monitoring will track injection progress relative to the design and include field measurements during the injections to evaluate reagent distribution relative to the treatment design. Performance monitoring will include a baseline monitoring event corresponding to the annual monitoring event scheduled for March 2025. ISB implementation is expected to occur late in the second quarter of 2025 after completion of the annual sampling. The annual monitoring event at OU 7 includes sampling of 88 monitoring wells screened in the surficial aquifer. For ISB performance monitoring, the post-injection monitoring program for 2025-2026 includes two performance monitoring events and one annual event (March 2026). The performance monitoring network includes 14 monitoring wells in the plume areas targeted for treatment. Analytical parameters for each location will include field water quality parameters, volatile organic compounds, total organic carbon, geochemical parameters and select locations for microbial parameters.

ISB performance evaluations will: 1) evaluate reagent distribution and persistence relative to the design, 2) evaluate parameter trends along groundwater flow path across treatment areas and at each performance well, 3) evaluate reduction of contaminant mass using chemical and geochemical data, 4) evaluate changes in contaminant flux across treatment areas using well transects by integrating concentration and flow data, 4) evaluate changes in plume extent (area) by comparing pre-and post-ISB modeled plumes, and 5) evaluate changes in biodegradation rates.

A project technical memorandum will summarize completed remedial action installation activities. Annual reports for OU 7 will report the results of remedy implementation, performance monitoring, monitored natural attenuation, and long-term monitoring components and will include data evaluations and an integrated analysis of remedy performance. Periodic updates of remedy performance and progress will occur during regulatory planning team meetings and for semi-annual restoration advisory board meetings.

Acronyms and Abbreviations

1,4-DCB1,4-dichlorobenzene1,1-DCE1,1-dichloroethene1,1,1-TCA1,1,1-trichloroethene3Dthree dimensional

AEHA United States Army Environmental Health Agency

BGS below ground surface

BTAG Biological Technical Assistance Group

BVC BAV1 Vinyl Chloride CB chlorobenzene

CDCE cis-1,2-dichloroethene
COC constituent of concern
CSM conceptual site model
CV coefficient of variation

DEQ Virginia Department of Environmental Quality

DHC Dehalococcoides

DLA Defense Logistics Agency
DO dissolved oxygen
DPT direct push technology

DSCR Defense Supply Center Richmond

EPA United States Environmental Protection Agency

ERD enhanced reductive dechlorination

EVO emulsified vegetable oil FFS focused feasibility study FTA Fire Training Area

ft. feet gal gallons

gpm gallons per minute > greater than

greater than or equal toHHRA human health risk assessmentHPT hydraulic profile tooling

HRSC high resolution site characterization

HAS hollow-stem auger IC institutional control

IDM investigative derived material

ISB in situ bioremediation

< less than

≤ less than or equal to

Law Engineering and Environmental Services

lbs pounds

LOD limit of detection
LOQ limit of quantitation
LTM long term monitoring
Meadows Meadows CMPG, Inc.
MCL maximum contaminant level
MIP membrane interface probe
MNA monitored natural attenuation

meq milliequivalents µg/L micrograms per liter

μS/cm microsiemens per centimeter

mg/L milligrams per Liter

MIP membrane interface probe

M-K Mann Kendall

MNA monitored natural attenuation

mV millivolt

NAVD88 North American Vertical Datum of 1988

NPL National Priorities List
ORP oxidation reduction potential

OU operable unit percent

PAH polycyclic aromatic aydrocarbon

PCB polychlorinated biphenyls

PCE tetrachloroethene
psi pounds per square inch
p-value probability value

qPCR quantitative polymerase chain reaction

RA remedial action

RAO remedial action objective RAWP remedial action work plan

RD remedial design
RI remedial investigation
ROD Record of Decision
SC specific conductance

SOP standard operating procedure
SPT standard penetration test
SS statistically significant
SSA singular spectrum analysis
SVOC semi-volatile organic compound

tceA tceA Reductase TCE trichloroethene

tDCE trans-1,2-dichloroethene TOC total organic carbon

USACE United States Army Corps of Engineers

USGS United States Geological Survey

VC vinyl chloride

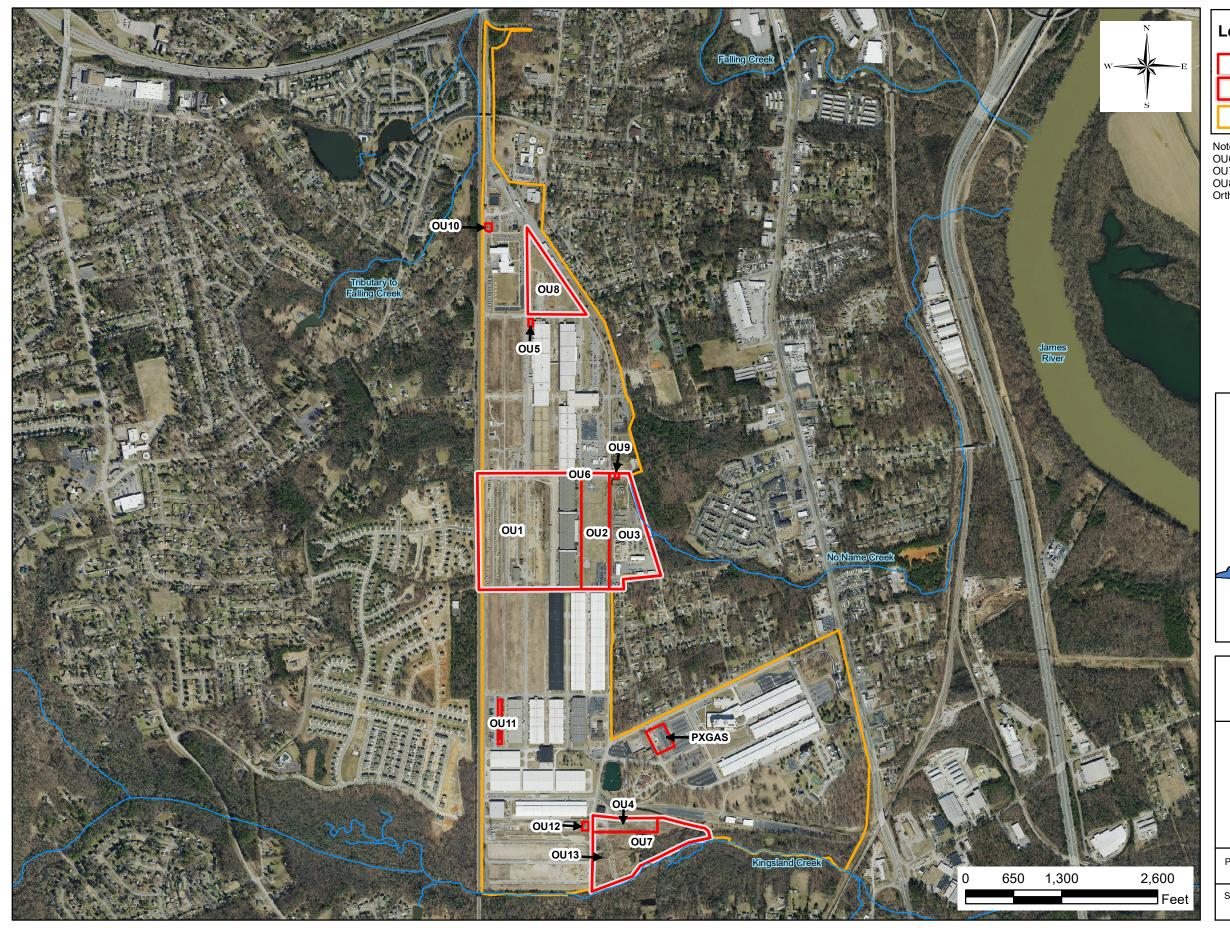
VCR Vinyl Chloride Reductase
VMP vapor monitoring point
VOC volatile organic compound

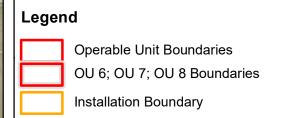
yr. year

1. Introduction

This document is a Remedial Action Work Plan (RAWP) Addendum for Operable Unit 7 (OU 7) at Defense Supply Center Richmond (DSCR) prepared under Contract W912DR22C0045 awarded by the United States Army Corps of Engineers (USACE) Baltimore District on September 19, 2022, to Meadows CMPG, Inc. (Meadows). Meadows and teaming partner AECOM have prepared this RAWP Addendum following the contract Performance Work Statement and requirements of Contract Line-Item Number 0025. This document describes the proposed follow-up in situ bioremediation (ISB) actions at OU 7 that target the surficial aquifer. Proposed actions will occur in the three plume areas associated with the former Fire Training Area (FTA) (OU 4).

DSCR is the headquarters of the Defense Logistics Agency (DLA) Aviation and is home to various other DLA, Department of Defense, and other federal organizations. The installation is eight miles south of the City of Richmond in Chesterfield County, Virginia. The United States Environmental Protection Agency (EPA) placed DSCR on the National Priorities List (NPL) in 1987. Since 1990, DLA has implemented an environmental restoration program at DSCR under a Federal Facility Agreement with the United States Environmental Protection Agency (EPA) Region 3 and the Virginia Department of Environmental Quality (DEQ). OU 7 designation is the impacted groundwater beneath and downgradient of the former FTA (OU 4) and the polycyclic aromatic hydrocarbon (PAH) Area (OU 13). Figure 1-1 has the layout of DSCR and OU locations.





OU6 Includes Groundwater Impacts Beneath OU's 1, 2, & 3 OU7 Includes Groundwater Impacts Beneath OU 4 OU8 Includes Groundwater Impacts Beneath OU 5 Orthoimagery Source: VGIN (2021)

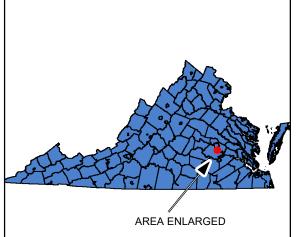




Figure 1-1DSCR Operable Units

Defense Supply Center Richmond Richmond, VA

Prepared By: DBC	Reviewed By: KL
Scale: 1" = 1,300'	Date: January 17, 2025

2. Background

Section 2 has background information for OU 7 including a site description, site history, and environmental setting.

2.1 OU 7 Description

OU 7 consists of impacted groundwater beneath and downgradient of OU 4 (FTA Source Area) along the central southern boundary of the installation (Figure 1-1). Contaminants from the OU 13 area (PAH Area) near OU 4 have also potentially impacted OU 7 groundwater. Figure 2-1 shows the layout and features of OU 7 area. The northern and western portions of OU 7 consist of open areas with grass and gravel cover with Building 72 located near G Road that borders the OU area to the west. A wooded area in the eastern area of OU 7 slopes down to the floodplain area of Kingsland Creek and the installation boundary road and fence line.

2.2 OU 7 Site History

The FTA had three separate, unlined pits used for firefighting training. Materials dumped into the pits, ignited, and extinguished during firefighting training included flammable liquid chemicals and petroleum products. Potential materials used for fuel included oils, solvents, pesticides, and herbicides (Law 1996a).

FTA Pit 1 in the eastern section of OU 4 consisted of a circular area with a diameter of 50 feet (ft.) and depth of three ft. Pit use occurred from the mid-1970s to 1979 with former uses potentially including drum storage or trash burning. Filling and grading of the pit area with soil occurred in 1983 (Dames & Moore 1989). FTA Pit 2 in the middle portion of OU 4 consisted of a 20 ft. x 40 ft. rectangular area with an unknown depth. Pit 2 use occurred from the late 1960s until the 1970s when Pit 1 replaced Pit 2. Filling and grading of the Pit 2 area with soil reportedly occurred between 1973 and 1975.

FTA Pit 3 in the western portion of OU 4 is a suspect fire training pit identified in 1969 aerial photographs. Construction of a 300,000-gallon aboveground storage tank (fuel oil) occurred over a portion of the Pit 3 area in 1975 (Law 1996a). A No. 4 fuel oil release of approximately 10,000 gallons occurred in 1978 from a cracked valve. Heavy rain at the time of the release caused the oil to overflow the tank containment berm and flow across OUs 4 and 13 before discharging into Kingsland Creek. The area of impacted soil southwest of the former FTA is designated OU 13. Removal of the tank occurred in 1997 and a separate record of decision (ROD) addressed OU 13 soil.

2.2.1 1981 Installation Assessment

The U.S. Army Toxic and Hazardous Materials Agency completed the first environmental assessment at DSCR, and their installation assessment report indicated possible groundwater impacts from the former FTA (OU 4, Army Chemical Systems Laboratory 1981).

2.2.2 1982 Investigations

Remedial investigations began in the OU 7 area before DSCR final listing on the NPL in 1987. Initial investigations completed by the United States Army Environmental Health Agency (AEHA) in March 1982 occurred in the FTA Pit 1 area. At the time of the investigation the pit had a 1-inch layer of petroleum product apparently mixed with fire extinguishing material floating on water, with a bottom 1-4-inch layer of petroleum-based sludge (Army Chemical Systems Laboratory 1981). Initial groundwater sampling of four monitoring wells in the Pit 1 area indicated the presence of volatile organic compounds (VOCs) in shallow groundwater in the parts per million range for chlorinated ethenes and chlorinated benzenes.

AEHA conducted follow-up sampling in October and November 1982 that included sampling surface water of Kingsland Creek at two stations. The results of the sampling did not indicate detection of VOCs in samples at concentrations substantially higher than laboratory quantitation limits (LOQ).

2.2.3 1989 Remedial Investigation

Dames & Moore completed a remedial investigation (RI) of the FTA from 1985 to 1988 that included completion of soil borings, installation of monitoring wells, and sampling of soil, soil gas, groundwater, sediment, and surface water. A 1989 RI Report (Dames & Moore 1989) summarized the results of these investigations along with a benthic macroinvertebrate survey, and human health risk assessment (HHRA). RI findings indicated impacts to groundwater within the FTA and downgradient to the area of Kingsland Creek. Primary constituents detected in groundwater included VOCs. Soils in the area of Pit 1 contained VOCs, semi-volatile organic compounds (SVOCs), pesticides, petroleum hydrocarbons, and cyanides at concentrations higher than Pit 2. The highest contamination levels occurred at a depth of two (2) ft. in Pit 1 and at the surface at Pit 2. The RI indicated a limited extent of soil contamination in the Pit 1 and 2 areas and minimal impacts at Pit 3 and a former drum storage area. Surface water sampling for the RI did not indicate a significant impact to Kingsland Creek.

2.2.4 1992 Field Investigations

In 1992, Law Engineering and Environmental Services (Law) completed additional investigations in 1992 including installation and sampling of monitoring wells throughout OU 7 to further define the extent of VOCs in groundwater. The 1996 RI addendum reports the results of these investigations (Law 1996a).

2.2.5 1994 Field Investigation Report for Fire Training Area

Engineering-Science, Inc. conducted field investigations in August through October 1993 to provide supplemental data for the RI addendum and focused feasibility study (FFS) report for OU 7 (Engineering-Science Inc. 1994). The scope of investigations included shallow soil borings, collection of hydro punch samples, installation of monitoring wells and pumping test wells, performance of pumping tests, geophysical logging of well borings, groundwater sampling, slug tests, and surveys. The investigation report refined determinations of the extent of VOCs in shallower and deeper groundwater downgradient of the pit areas into and beyond (south) of the Kingsland Creek area.

2.2.6 1996 Final Remedial Investigation Report Addendum

Law prepared a RI addendum report that presented results of investigations and sampling conducted after the 1989 RI. Soil samples showed higher concentrations of VOCs in the soils at Pits 1 and 3 with minimal concentrations detected near Kingsland Creek. Groundwater sample result interpretations identified three distinguishable plumes potentially corresponding to the FTA pits with chlorinated VOCs identified as the primary constituents in deeper groundwater. Surface water sampling of Kingsland Creek indicated low levels of VOCs.

The RI addendum also included a human health baseline risk assessment for OU 4 and OU 7. This risk assessment identified cancer risks higher than the EPA range of 1E-06 to 1E-04 for occupational exposure to surface soil in the FTA and a hypothetical scenario involving future use of groundwater as a potable water supply (Law 1996a).

2.2.7 1996 Focused Feasibility Study

An FFS issued in 1996 identified remedial action objectives (RAOs) for shallow groundwater that involved engineered remedies for source control and contaminant reduction and natural attenuation processes for deeper groundwater. Recommended alternatives included groundwater extraction/soil vapor extraction for shallow groundwater and institutional controls (ICs) for deeper groundwater. The FFS recommended performance of a pilot test for design of systems for shallow groundwater (Law 1996b).

2.2.8 1997 Pilot Study Report for Fire Training Area

A pilot test conducted in 1996 evaluated the feasibility of using dual phase extraction as a remedial technology to address contaminated groundwater at OU 7. The pilot test report recommended consideration of dual phase extraction as an element of an overall remediation plan addressing shallow groundwater contamination. This report also recommended preliminary design parameters for design of a full-scale system (Law 1997).

2.2.9 1989 Letter Addendum to OU 7 Feasibility Study

A 1999 letter addendum to the OU 7 FFS evaluated data obtained from a density-driven convection pilot test (completed from December 1998 to January 1999) as an additional alternative to the FFS. The pilot test occurred within the shallow groundwater zone near Kingsland Creek. This letter addendum recommended the inclusion and evaluation of the density-driven convection technology in the revised FFS based on the pilot test results (Law 1999).

2.2.10 Removal Action

A time-critical removal action occurred at OU 4 in 2004. Excavation areas at FTA Pits 1 and 3 extended to a depth of 6 ft. with removal quantities of approximately 484 and 498 cubic yards, respectively. A limited excavation of five (5) cubic yards occurred at FTA Pit 2. Confirmatory sampling results at Pits 1 and 3 indicated that residual chlorinated VOCs remained below the excavation depth (MACTEC 2005).

2.2.11 2006 Supplemental Feasibility Study

MACTEC presented the results of supplemental natural attenuation evaluations (2001-2002) and additional investigations conducted in 2004 in a supplemental feasibility study (MACTEC 2006a) and evaluated the data in the 2006 installation-wide conceptual site model (MACTEC 2006b). The conceptual site model (CSM) presented the nature and extent of contamination of groundwater, described concentration trends, and evaluated natural attenuation processes in groundwater.

2.2.12 Creeks Monitoring Program

MACTEC implemented a three-year creek monitoring program (2001-2004) that included an assessment of Kingsland Creek. This included performance of a HHRA in 2006 that concluded no further action required for Kingsland Creek for human health-based risks. A screening level ecological risk assessment completed in 2001 for Kingsland Creek and subsequent Kingsland Creek monitoring supported the conclusion that detected VOCs did not pose unacceptable risk to potential ecological receptors.

2.2.13 2010 Treatability Study Report for OU 6 and OU 7

AECOM performed treatability studies for VOC plumes at OU 7 in 2007-2009 (AECOM 2010a). These studies performed from April 2007 to February 2009 included completion of direct push technology (DPT) borings and vertical aquifer profiling across the OU 7 study area. Follow-up monitoring well installation included shallow aquifer wells. Six well clusters (CMT) and five monitoring wells installations characterized conditions in the confined aquifer below the surficial aquifer depths characterized by the DPT sampling.

A second phase of treatability studies included construction of two biowalls in the Pit 1 VOC plume area and two biowalls in the Pit 2 VOC plume area. Biowalls had installation depths of 20 ft. in the Pit 1 area, 22 ft. for the northern biowall in the Pit 2 area, and 25 ft. for the southern biowall in the Pit 2 area. Pit 1 biowalls had uncomposted mulch as the substrate amended with dolomite gypsum. Pit 2 biowalls had composted mulch/poultry as the substrate amended with dolomite.

A third phase of treatability study performed at Pit 3 in 2009 injected emulsified vegetable oil (EVO) with amendments (6% emulsion) into the surficial aquifer. The injection emulsion totaling 2,125 gallons into three injection wells included fluorescent dye to evaluate distribution of EVO into the subsurface. Detection of emulsion occurred at monitoring well OU7-MW-94 (20 ft. from injection wells) after injection of 200 gallons of EVO.

Treatability findings for the biowalls at OU 7 showed effective treatment of chlorinated VOCs in groundwater as groundwater passed through the wall areas. Results of the EVO injections at Pit 3 indicated a radius of influence (ROI) greater than 20 ft. and significant reductions of chlorinated VOCs (at residual dense non-aqueous phase liquid levels in groundwater) within five months of injections.

2.2.14 2010 Revised Focused Feasibility Study

A revised HHRA completed as part of the Revised FFS in 2010 evaluated risks for the onsite OU 7 area, Kingsland Creek, and offsite areas (AECOM 2010b). The revised FFS evaluated remedial alternatives in the context of the revised HHRA and in-situ bioremediation treatability tests performed at Pits 1 and 2 in 2007 and at Pit 3 in 2009.

RAOs established based on the HHRA included preventing unacceptable risks to human health and the environment from exposure to constituents of concern (COCs) in groundwater and reduction of COC concentrations to Federal maximum contaminant levels (MCLs). The FFS evaluated various remedial alternatives and recommended enhanced ISB, monitored natural attenuation (MNA), long-term monitoring (LTM), and ICs based on results of pilot testing and natural attenuation evaluations (AECOM 2010b).

2.2.15 Record of Decision

The ROD document for OU 7 finalized in May 2012 identifies OU 7 as groundwater underlying and downgradient from the former FTA (OU 4). Separate remedial actions have addressed OU 4 identified as source material that impacted OU 7 groundwater and also addressed OU 13 that potentially impacted groundwater (DSCR 2012).

The selected remedy in the ROD consists of the following elements:

- ISB to treat COCs in groundwater source areas (Pits 1, 2, and 3) and downgradient portions of the groundwater plumes.
- MNA that involves monitoring of COCs and geochemical conditions in groundwater to document that MNA is reducing chemical mass and concentrations over time.
- Annual LTM of the uppermost groundwater zone for a minimum of five years to monitor for potential leaching of SVOCs, PAHs, polychlorinated biphenyls (PCBs), pesticides, metals, and 2,3,7,8tetrachlorodibenzo-p-dioxin from OU 4 and OU 13 soils to groundwater, and also includes annual monitoring of surface water of Kingsland Creek until COCs in shallow groundwater have reached cleanup level established in the ROD.
- ICs described in the ROD including groundwater use restrictions, land use restrictions, control
 exposure to contaminated groundwater and implementation of ICs for future buildings within
 groundwater plume areas.

Table 2-1 presents the cleanup levels established in the ROD for the COCs in OU 7 groundwater. As described in the ROD, the selected remedy satisfies the statutory requirements of the Comprehensive Environmental Response, Compensation, and Liability Act. Since this remedy will result in hazardous substances, pollutants, or contaminants remaining on site above levels that allow for unlimited use and unrestricted exposure, a statutory review will be conducted within 5 years after initiation of the remedial action (RA), and at a subsequent frequency of at least once every 5 years, to ensure that the remedy is, or will be protective of human health and the environment. Protectiveness reviews will continue until site conditions enable unlimited use and unrestricted exposure.

Maximum Contaminant Level (µg/L)

Table 2-1 OU 7 COCs and Cleanup Levels

Contaminants of Concern

	(1.3. –)
Carbon Tetrachloride	5
Chlorobenzene	100
1,4-Dichlorobenzene	75
1,1-Dichloroethene	7
cis-1,2-Dichloroethene	70
trans-1,2-Dichloroethene	100
Tetrachloroethene	5
Trichloroethene	5
Vinyl Chloride	2

Contaminants of Concern

Maximum Contaminant Level (µg/L)

Chloroform 8

Notes: Maximum Contaminant Level from 40 CFR § 141.61 Maximum contaminant levels for organic contaminants and § 141.64 Maximum contaminant levels for disinfection byproducts including Total trihalomethanes (includes chloroform), µg/L = micrograms per liter.

2.2.16 2013 Remedial Actions

AECOM prepared a Remedial Design (RD)/RAWP in 2013 to implement the remedy components identified in the ROD (AECOM 2013). AECOM implemented remedial actions for the selected remedy at OU 7 in 2013, as generally detailed in in the OU 7 Injection Technical Memorandum that documented completed activities (AECOM 2015).

ISB treatment targeted eight (8) treatment areas in the Pit 1, 2, and 3 plumes at locations and in lithologic units not previously treated by the biowalls (Pits 1 and 2) and previous EVO injections (Pit 3). The ISB process options implemented included EVO as the carbon substrate pressure injected into injection wells designed for the specific treatment areas. In general, injections performed did not achieve design volumes in the treatment areas due to flow rates and pressures. An adaptive approach used increased EVO solution concentrations to achieve design loading rates where possible.

Remedy verification performed as part of the remedial actions included groundwater sampling three months after completion of injection activities followed by annual performance, MNA, and LTM groundwater monitoring. Annual LTM of surface water has occurred at three sampling stations in Kingsland Creek since remedy implementation in accordance with the ROD. Vapor monitoring of exterior vapor monitoring points (VMPs) at Building 72 began in October 2013 and at interior sub-slab VMP 116 in 2016. The EPA and DEQ approved discontinuation of monitoring of exterior VMPs in October 2016 due to persistent infiltration of water into the VMPs. EPA and DEQ approved a change in monitoring frequency at VMP-116 from quarterly to annual in September 2017. The vapor monitoring evaluates for potential accumulation of VOCs and methane beneath Building 72 related to remedy implementation in the OU 7 area. ICs implemented at OU 7 comply with the site-wide, land use control remedial design.

2.2.17 2016 Remedial Action

AECOM developed a RAWP Addendum in December 2015 to perform additional ISB actions as a followup to actions completed in 2013 (AECOM 2015). AECOM implemented ISB DPT injections (EVO) in January 2016 at the distal end of the identified Pit 1 plume in the installation fence line area and offsite area south of Kingsland Creek. High injection pressures and dense subsurface conditions required adaptive approaches to achieve targeted injection volumes and EVO loading rates.

2.2.18 2017 Remedial Actions

In 2017, USACE Baltimore District implemented follow-up remedial actions to the ISB injections performed in 2013, which included the installation of 12 additional injection wells (USACE 2017). These efforts involved ISB injections of Lactoil® in high concentration source zone areas at Pits 1, 2, and 3 for long-term treatment and injection of sodium lactate in lower concentration areas (mid plume at Pits 1 and 2). Post injection monitoring occurred three months after the completion of injections followed by annual site-wide groundwater monitoring. ISB injections occurred in December 2017 and January 2018. Figure 2-2 shows the ISB injection wells/areas.

2.2.19 2019 Pre-Design Investigations

Pre-design data collection activities occurred at OU 7 in 2019 to update the CSM and obtain data to support remedy optimization, as detailed in the 2019 Annual Report for OU 7 (AECOM-Meadows, 2021a). The scope of these activities included:

- Expansion of the 2019 annual sampling event from 78 to 168 monitoring well locations.
- Completion of vertical profile sampling at 14 stations in Kingsland Creek to evaluate the groundwater to surface water migration pathway including VOC sampling of surface water, sediment porewater, and groundwater beneath the creek.

Completion of high-resolution site characterization (HRSC) investigations at OU 4, OU 13, and OU 7
areas including around and within VOC plume areas. HRSC investigations included integrated
assessment of contaminant profiling (membrane interface probe [MIP]) and vertical aquifer profiling
and hydraulic properties determination (hydraulic profiling tool [HPT]).

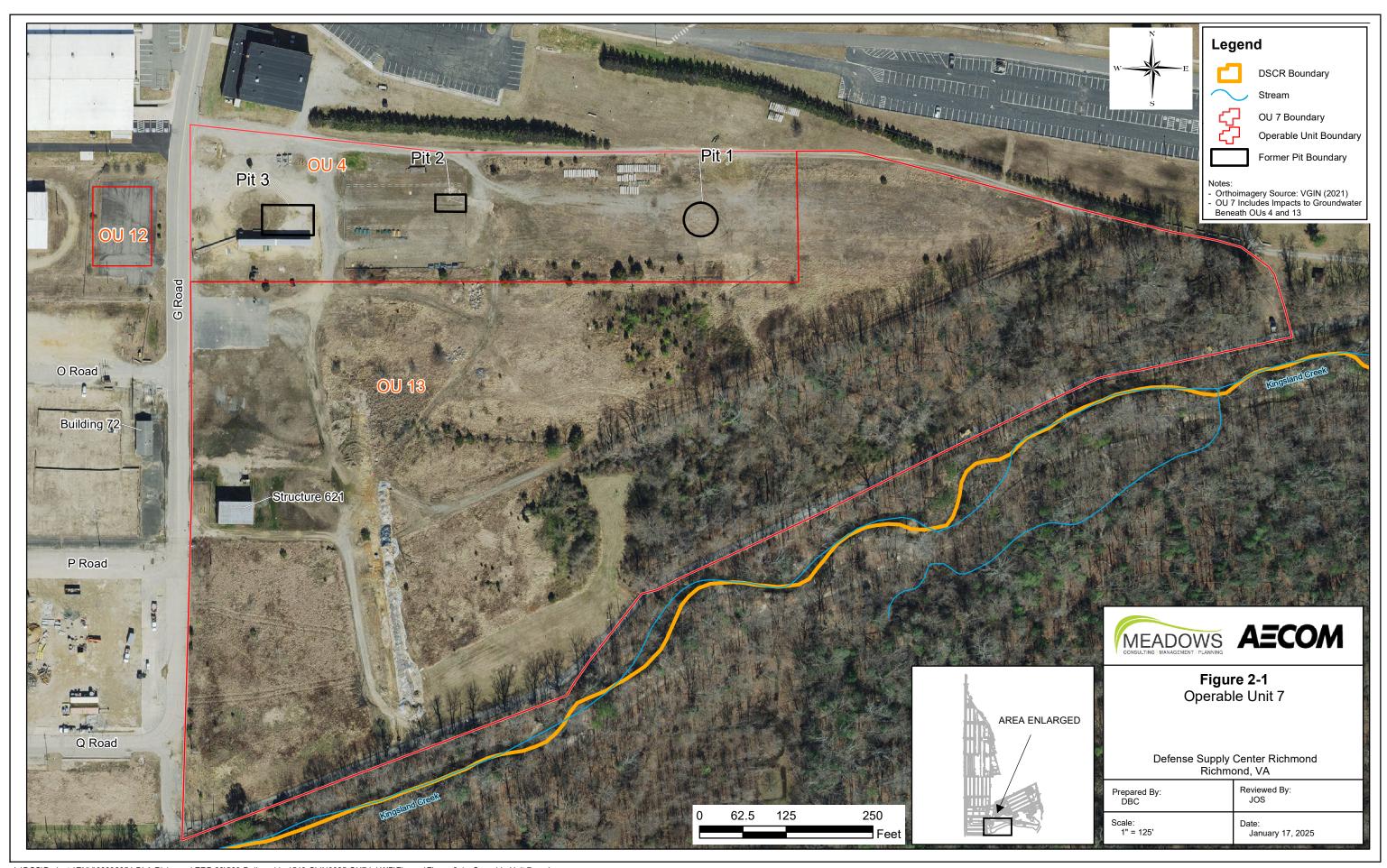
The 2019 Annual Report presents a detailed update of the CSM that includes a digital three-dimensional (3D) model of the OU 7 site area.

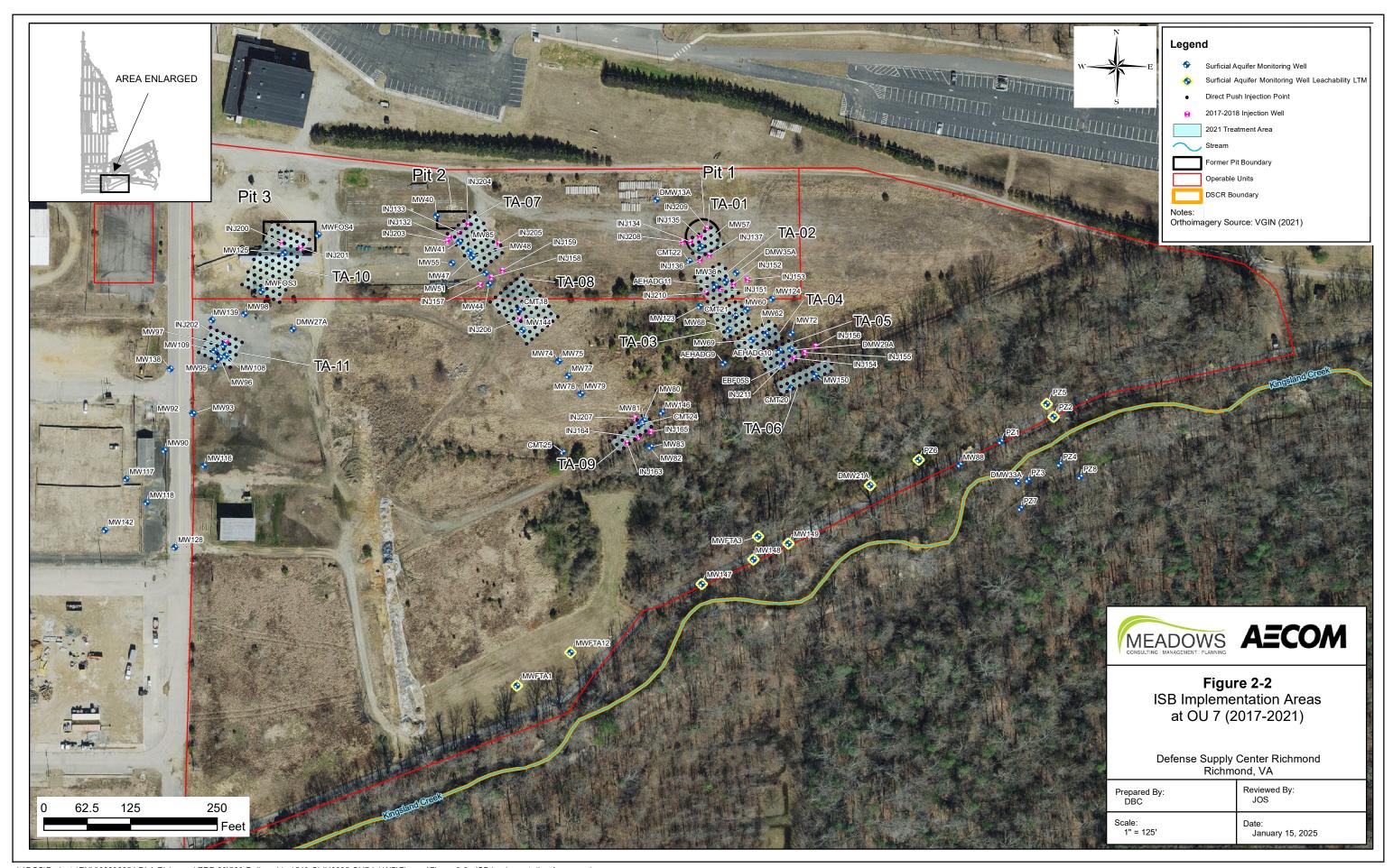
2.2.20 2021 Remedial Actions

A RAWP Addendum prepared in 2021 described proposed remedy optimizations for OU 7 (AECOM-Meadows, 2021b). The RAWP contained the following proposed actions:

- Targeted ISB injections with in-situ chemical reduction (ISCR) enhancement in 11 grid treatment areas within FTA Pit 1, 2, and 3 plume areas.
- Remedy performance monitoring and MNA actions with an optimized monitoring network.

Figure 2-2 on page 2-8 shows the ISB treatment areas for injections completed in December 2017/January 2018 and November/December 2021.





2.3 Environmental Setting

Section 2.3 describes the environmental setting for OU 7.

2.3.1 Topography

Historical site activities and development have altered the land surface and topography of the OU 7 area as part of the development of DSCR. Figure 2-3 (page 2-10) has a digital elevation model for OU 7 showing site topography. Overall topographic slope is toward the southeast with elevations ranging from a maximum of approximately 102 ft. North American Vertical Datum of 1988 (NAVD88) in the former Pit 3 area to a minimum of approximately 83 ft. NAVD88 along the fence line area near Kingsland Creek.

A perimeter road encircles the OU 7 area with a gravel road extending east-west across the site within a former railroad spur line area. The area south of this gravel road slopes to a low area in the alluvial flood plain of Kingsland Creek, the eastern portion of this area remains forested.

2.3.2 Surface Water and Wetlands

perimeter road in the eastern portion of OU 7.

Kingsland Creek borders OU 7 to the south along the installation boundary with the creek flowing in a northeasterly direction in this area and then flows east approximately 2.2 miles to the James River (Figure 1-1). The creek has a cobbly sediment substrate in the site area. Two storm drains extend north to south across the OU 7 with outfalls into the floodplain area of Kingsland Creek where ditches convey water to the creek. A freshwater forested/shrub wetland area exists in the Kingsland Creek floodplain area near and along the southern border of the installation according to the United States Fish and Wildlife Service's National Wetlands Inventory wetlands mapper accessed at https://www.fws.gov/wetlands/data/mapper.html. A portion of the mapped area extends north across the

2.3.3 Soils

The United States Department of Agriculture web survey accessed at https://websoilsurvey.sc.egov.usda.gov/App/WebSoilSurvey.aspx identifies the soil mapping units at OU 7 as made land reflecting disturbance and development of the area and Fluvaquents (map unit 1A) in the flood plain area of Kingsland Creek. The disturbed or reworked soil material varies in consistency, and ranges from loamy sand to clay. Soils in Map Unit 1A consist of the following:

0 to 8 inches: silt loam.

8 to 40 inches: silty clay loam.

40 to 50 inches: sand.

2.3.4 Site Geology

DSCR lies near the western edge of the Virginia Coastal Plain. General stratigraphy found beneath OU 7 from top down is the Eastover Formation, Aquia Formation, and Potomac Formation underlain by Petersburg Granite bedrock. Alluvial sediments overlie the Aquia Formation in the Kingsland Creek flood plain. Surficial fill material associated with site development overlies the majority of the OU 7 area. Table 2-2 (page 2-11) provides general stratigraphy information for the OU 7 area.

The digital 3D CSM developed for the OU 7 area includes a geologic model developed using multiple lines of data and geostatistical methods (kriging). This involved review of original data sources collected for the period 1982 to 2019 and selection of the highest resolution/data quality available for model development. Data types reviewed for model inputs included geological and hydrogeological publications, site and DSCR boring data, lithologic logs, geophysical logs and surveys, physical test data from soil cores, cone penetrometer testing data, and HPT data collected in 2019.

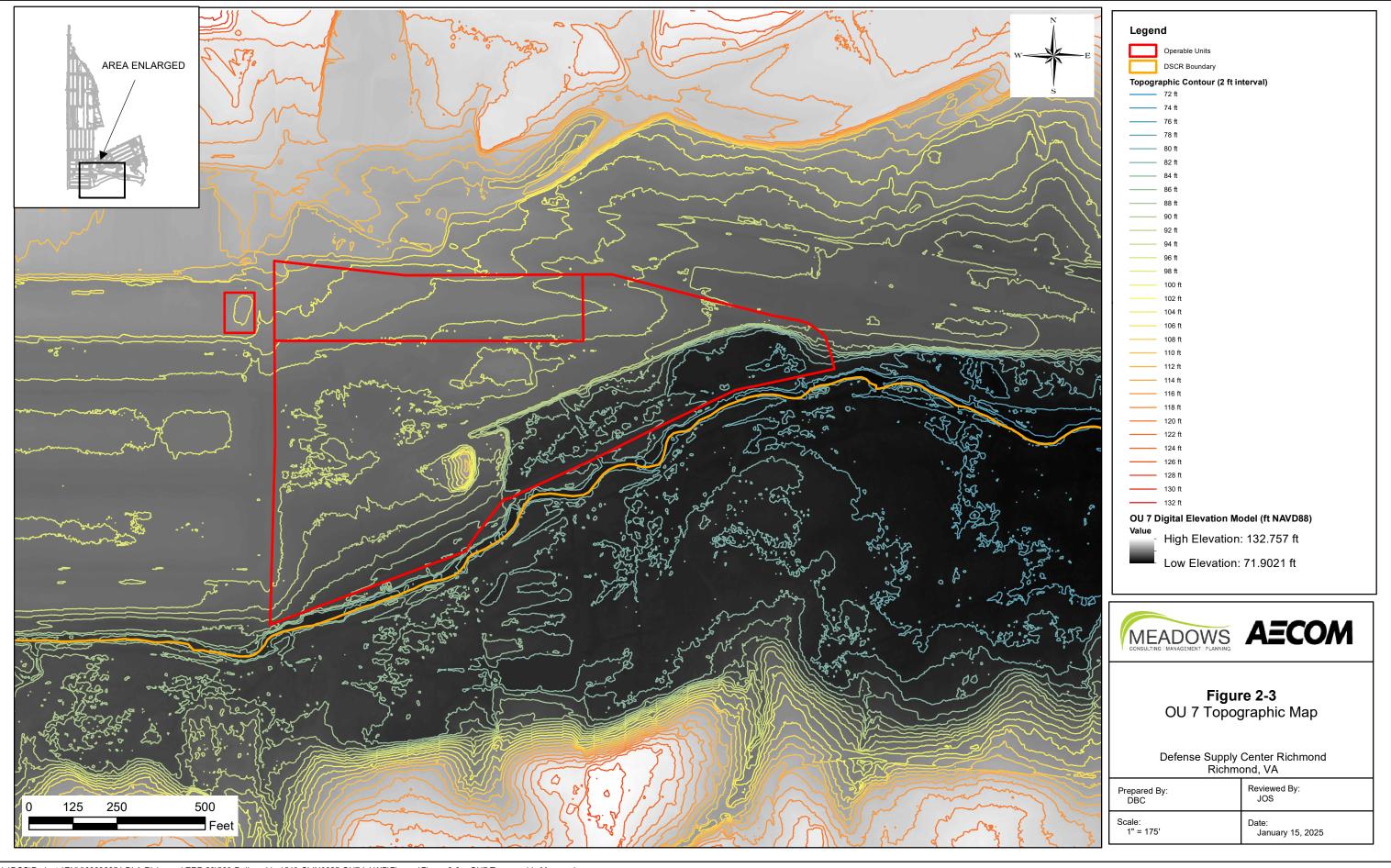


Figure 2-4 (page 2-12) has a north-south oriented geologic cross section for the southern half of DSCR from the OU 6 area south across the western OU 7 area and Kingsland Creek. This cross section illustrates changes in topography across the southern DSCR and OU 7 area where the ground surface slopes to the valley of Kingsland Creek. It also illustrates lateral variations in geology where the Calvert Formation pinches out in the OU 7 area and the Aquia Formation thickens and is unconformably overlain by Eastover Formation and underlain by the Potomac Formation. In the Kingsland Creek area, flood plain sediments (alluvium) overlie the Aquia Formation. Figure 2-5 (page 2-13) has an east-west oriented geologic cross section across OU 7 and illustrates a topographic slope toward the east and the Kingsland Creek flood plain. It also illustrates an increase in the thickness of the Aquia toward the east.

2.3.4.1 Alluvium

Alluvium occurs within the flood plain of Kingsland Creek in the southern lowland portion of OU 7. These floodplain deposits have a loose consistency and consist of variable silty sand (SM), clayey sand (SC), poorly sorted sand with gravel (SP), silty gravel with sand (GM), and clayey gravel with sand (GC). The estimated thickness of alluvium is 0 to 8 ft.

2.3.4.2 Eastover Formation

Eastover Formation sediments have variable mottling and color (brown, yellowish red, and reddish brown). Primary lithologies vary from silty sand (SM) to well graded sand (SW) with variable gravel and silty gravel (GM). Reworked surficial material generally consists of Eastover sediments. The thickness of Eastover Formation sediments below the fill material ranges from approximately 6 to 10 ft.

Table 2-2 OU 7 Stratigraphy

Geologic Formation	Age	Origin	Approx. THK (ft)	Primary Lithology Types
Alluvium	Holocene	Alluvial Floodplain Kingsland Creek	0-8	Silty sand (SM), clayey sand (SC), poorly graded sand with gravel (SP), silty gravel with sand (GM), clayey gravel with sand (GC)
Eastover	Pliocene	Alluvial	6-10	Silty sand (SM), well graded sand (SW) with variable gravel, silty gravel (GM), fine grained interbeds of clay and silt
Aquia	Paleocene Early Eocene	Marine	15-22	Silty sand (SM) with basal silty gravel (GM), fine grained interbeds of clay and silt
Potomac	Cretaceous	Alluvial	36-52	Silty sand (SM), poorly graded sand with silt (SP), silty gravel (GM), clay interbeds, clayey sand with gravel (SC)
Petersburg Granite	Mississippian	Bedrock		Granite to Granodiorite Bedrock

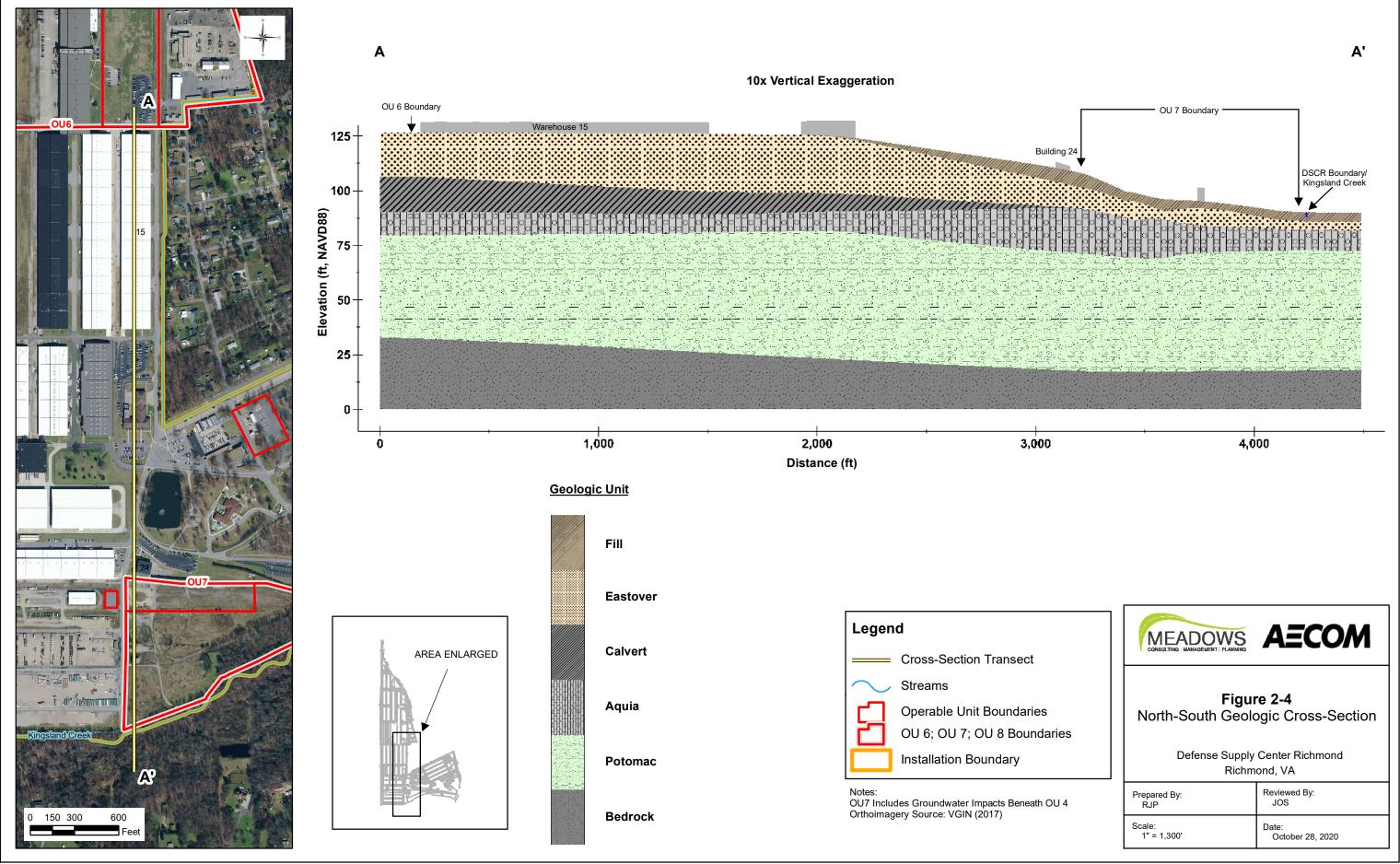
Notes: THK=thickness

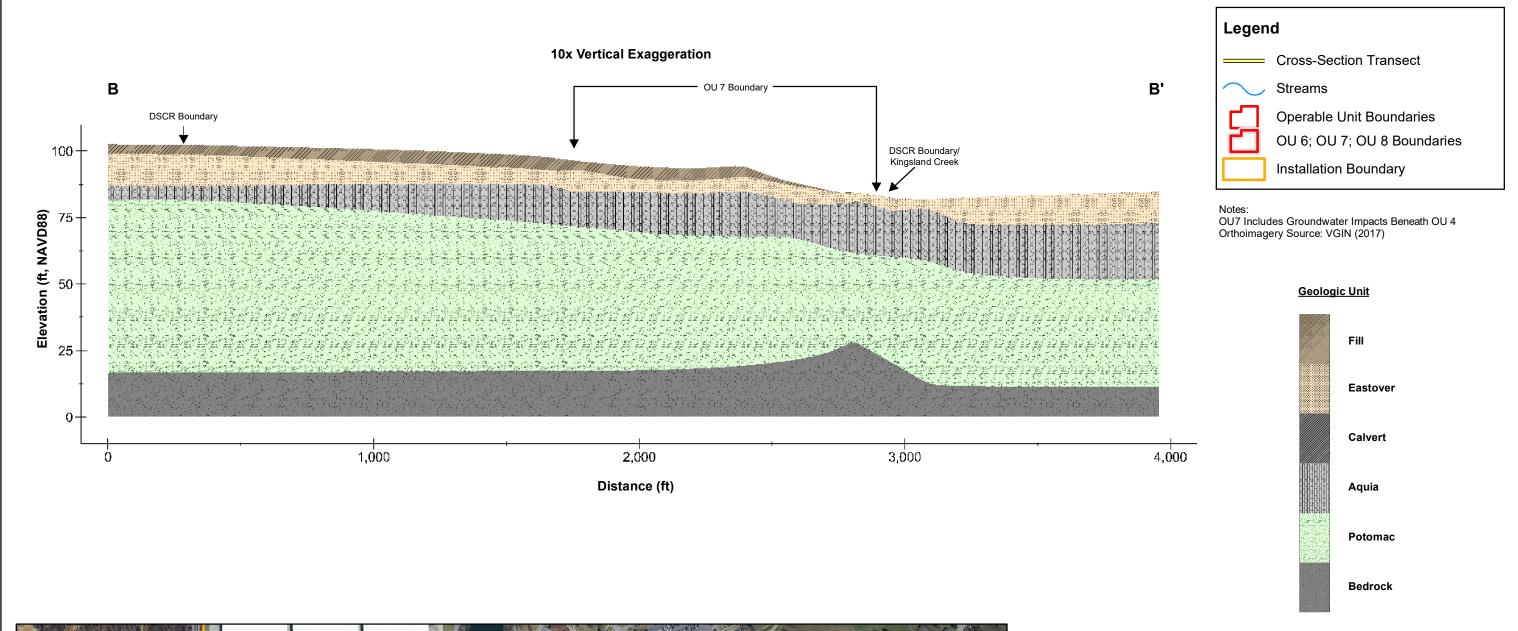
2.3.4.3 Aquia Formation

The Aquia Formation consists of a fining-upward, well graded, gray to dark green, glauconitic silty sand (SM) with a dense basal gravel stratum (GM). The Aquia Formation also contains finer grained clayey strata. Thickness of the Aquia Formation generally ranges from 15 to 22 ft.

2.3.4.4 Potomac Formation

Potomac Formation sediments consist of interbeds of light gray to greenish gray well graded sand (SW) with silt or clay, silty sand (SM), poorly graded sand (SP) with silt, clayey sand (SC), and clay (CL-CH). Standard penetration test sampling indicates dense to very dense consistency for coarse grained sediments and hard consistency for fine-grained sediments. The approximate thickness of the Potomac Formation ranges from 36 to 52 ft.







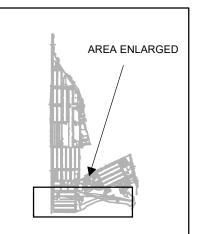




Figure 2-5 East-West Geologic Cross-Section

Defense Supply Center Richmond Richmond, VA

Prepared By:	Reviewed By:
RJP	JOS
Scale:	Date:
1" = 1,300'	October 28, 2020

2.3.4.5 Petersburg Granite

Bedrock in the study area is the Petersburg Granite described by the United States Geological Survey (USGS) as chlorite rich granodiorite. Dames & Moore performed rock coring at boring DMW-23A of the uppermost 15 ft. of bedrock (55 to 70 ft. below ground surface, BGS) and described bedrock as a granite with phaneritic, subhedral to euhedral texture primarily consisting of quartz and chlorite, some potassium feldspar, little muscovite and biotite, and trace garnet. Rock coring results indicated a vertical fractured zone at 58-59 ft., a highly fractured zone at 60-65 ft., and highly fractured zone at 68-70 ft. partially filled by green clay. Rock-quality designation values determined included 95 for core run 1, 40 for core run 2, and 90 for core run 3.

2.3.5 Hydrostratigraphy

The digital 3D CSM developed for the OU 7 area includes a refined hydrostratigraphy model developed by integrating the geologic model with existing hydrogeologic data and HRSC data obtained in 2019. A key aspect of the refined hydrostratigraphy is the semi-continuous measurement of hydraulic properties of the four unconsolidated geologic formations and integrated this data with other lines of data such as laboratory testing for physical properties. Table 2-3 summarizes the refined hydrostratigraphy for OU 6 and associated hydraulic properties defined in the 3D model.

Table 2-3 OU 7 Hydrostratigraphy

Hydrostratigraphic Unit	Туре	Description	Relative Permeability	Estimated K (cm/sec)
Eastover	Aquifer	Unconfined, discrete zones, highly variable, perched water	Low-High	<1.0E-04 to 3.5E-02 ² 1.52E03 to 9.12E-03 ⁴
Calvert	Confining Zone	Leaky unit	Very Low	4.8E-08 to 1.8E-06 ¹
Aquia	Aquifer	Semi-confined, bulk matrix of formation	Low-High	<1.0E-04 to 3.5E-02 ² 1.76E-06 to 1.55E-05 ¹
Potomac	Aquifer	Confined, bulk matrix of formation	Very Low- Moderate	4.9E-03 ⁵ 2.3E-07 to 3.5E-05 ³
Bedrock	Aquifer	Confined in fractures	Not determined	

Notes ¹ Laboratory core testing (vertical), ² Field testing with HPT, ³ Laboratory core testing (horizontal), K= hydraulic conductivity, cm/sec = centimeters per second, ⁴ Field slug tests at wells at OU 7., ⁵ USGS pumping test at OU 8 (USGS 1987)

2.3.5.1 Alluvium

Alluvium in the flood plain potentially contains a seasonally high-water table adjacent to Kingsland Creek. Primary flow and discharge into Kingsland Creek occur from the unconfined surficial aquifer within the uppermost portion of the Potomac.

2.3.5.2 Eastover-Aquia

This hydrostratigraphic unit corresponds to the unconfined surficial aquifer. Depth to groundwater is typically 10 ft. BGS) or less. The saturated thickness of this zone ranges from approximately 20 to 25 ft. The lithology of this zone varies between the upper portion (Eastover Formation) and lower portion (Aquia Formation) of this unit as described in Section 2.3.4.

2.3.5.3 Potomac

This hydrostratigraphic unit corresponds to the confined aquifer with dense strata. The lithology of this zone is heterogeneous as described in Section 2.3.4.

2.3.5.4 **Bedrock**

This hydrostratigraphic unit corresponds to the Petersburg Granite with an expected confined or semiconfined condition based on the overlying stratigraphy. Groundwater within unweathered bedrock will occur primarily in fractures.

2.3.6 Groundwater Flow

Lateral groundwater flow patterns and direction in the surficial aquifer reflect surface topography and surface water drainage in the Kingsland Creek area. The direction of groundwater flow is generally toward the southeast and Kingsland Creek (see Section 2.4.1). Shallow groundwater from the surficial aquifer discharges into Kingsland Creek with an upward component of flow apparent in the area. Lateral

groundwater flow in the confined aquifer is toward the east and at depth beneath Kingsland Creek (see Section 2.4.1).

2.4 Current Conditions for Groundwater in Surficial Aguifer

Section 2.4 describes the current conditions for groundwater in the surficial aquifer from the March 2024 annual sampling event as described in the Fiscal Year 2024 Annual Report for OU 7 (AECOM-Meadows 2024b). The focus of the current conditions summary is the surficial aquifer that is targeted for ISB. Figure 2-6 (page 2-16) shows the current groundwater and surface water monitoring locations at OU 7.

2.4.1 Surficial Aquifer Potentiometric Surface and Groundwater Flow

Figure 2-7 (page 2-17) has a potentiometric surface elevation contour map (March 2024) for the surficial aquifer developed by geostatistical analysis (kriging) in Earth Volumetric Studio using site-wide groundwater elevation data for wells screened in the surficial aquifer. Groundwater flow direction in the Pit 1 and Pit 2 area is toward the southeast and Kingsland Creek with a southwest flow direction in the Pit 3 area. The average hydraulic gradients across the OU 7 ranged from 1.90E-02 ft./ft. (Pit 3 area) to 3.00E-02 ft./ft. (Pit 1 area). Shallow groundwater discharge with an upward component of flow occurs at Kingsland Creek.

Equation 1 calculates the approximate horizontal velocity range of groundwater flow within the surficial aquifer using a form of Darcy's velocity equation:

$$(1) V = \frac{Ki}{n}$$

Where:

V = groundwater flow velocity (ft./yr.)

K = hydraulic conductivity (feet per day [ft/day])

i = hydraulic gradient (ft/ft)n_e = effective porosity (unitless)

Input values for Equation 1 for the surficial aquifer include:

- Average K of 3.60E+00 ft./day (Table 2-3).
- Estimated effective porosity of 0.27 based on average physical test data for total porosity, air filled porosity, and clay content.
- Average hydraulic gradient ranging from 1.90E-02 ft./ft. to 3.00E-02 ft./ft. for the March 2024 sampling event (Figure 2-7).

The estimated horizontal groundwater flow velocity calculated for the surficial aquifer using the above input values ranges from 2.53E-01 ft./day (91.3 ft./yr.) to 4.0E-01 ft./day (146.0 ft./yr.).

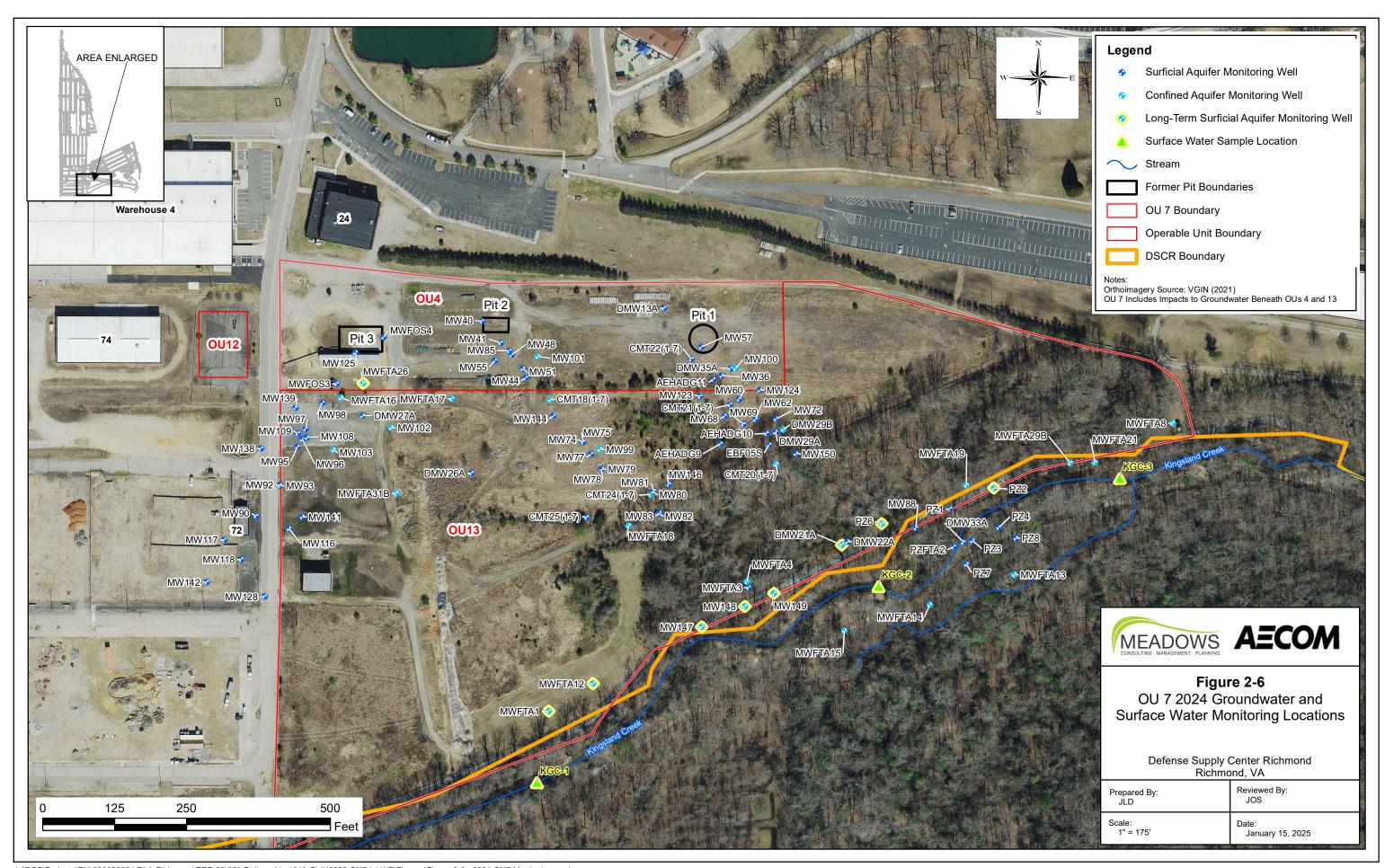
2.4.2 Water Quality Parameters for Surficial Aquifer (2024)

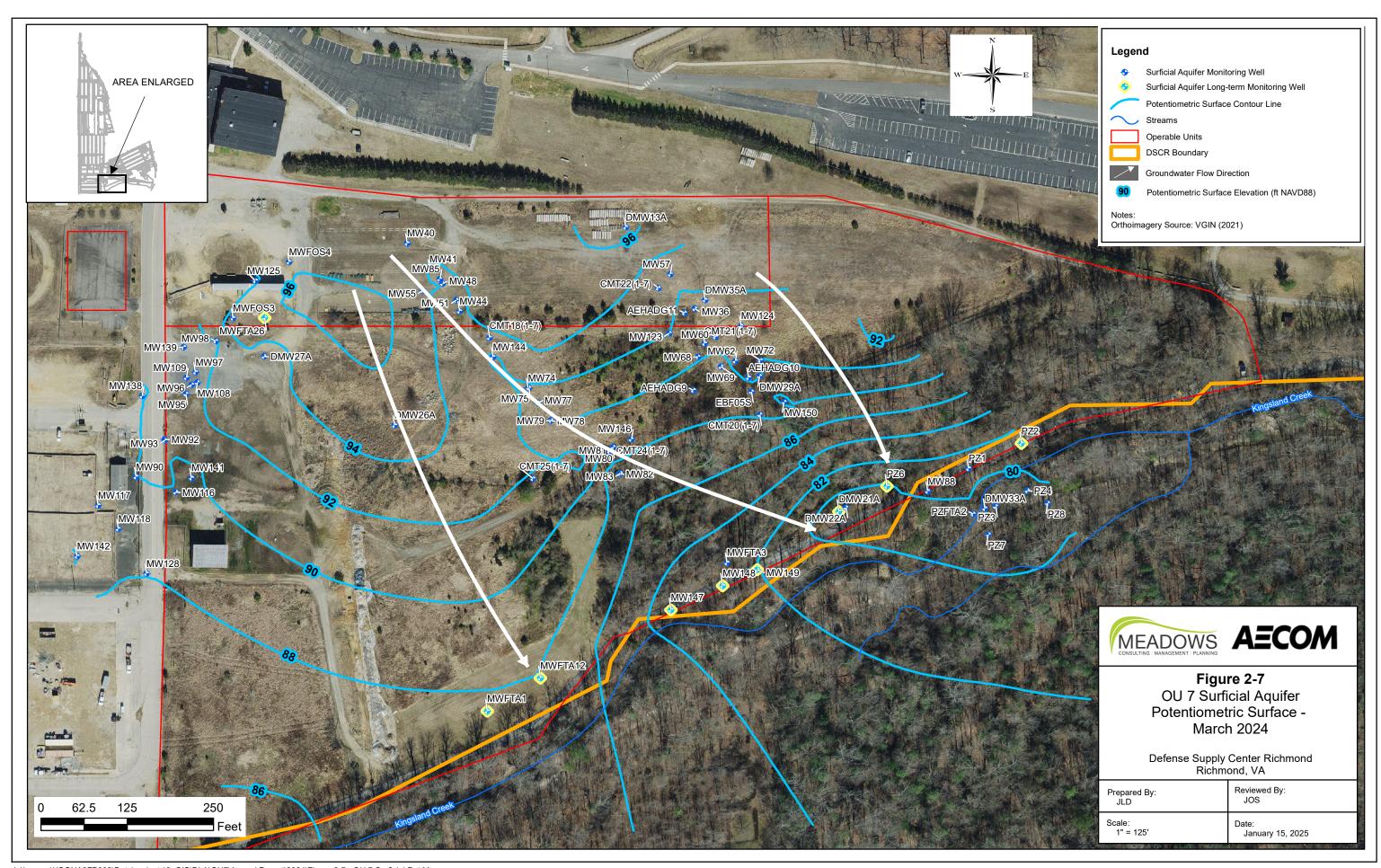
Section 2.4.2 and Exhibit 2-1 (page 2-18) have a summary of data distributions for water quality parameter results for the surficial aquifer for the Pit 1, Pit 2, and Pit 3 monitoring areas. The summaries use the OU 7 annual monitoring data from March 2024.

2.4.2.1 pH Results

The data distributions for pH vary by pit area with progressively higher pH levels moving across the site from Pit 1 to Pit 3 (see Exhibit 2-1). For Pit 1, the distribution range for pH is 3.58 with a mean of 5.81 and median of 5.69. The distribution 25th and 75th percentiles are 5.06 and 6.29, respectively. Seven of 36 data observations have a pH less than 5.

For Pit 2, the distribution range is 3.55 with a mean of 5.89 and a median of 5.84. The distribution 25th and 75th percentiles are 5.33 and 6.57, respectively. Four of 33 data observations have a pH less than 5.

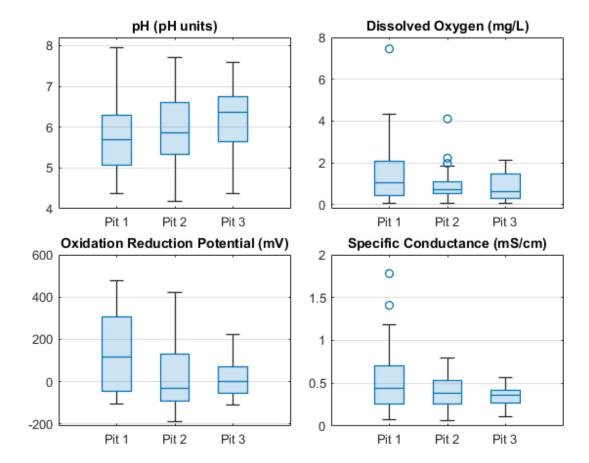




Defense Supply Center Richmond

OU 7 Remedial Action Work Plan Addendum

Exhibit 2-1 Water Quality Parameter Data Distribution for Surficial Aquifer: March 2024



pH (pH units)

Area	N	D	Min	Max	Range	Mean	Median	Std	Var	Skewness	Kurtosis	Outliers	PCTL25	PCTL75	pH< 5
Pit 1	36	36	4.38	7.96	3.58	5.81	5.69	0.93	0.87	0.6713	2.7056	0	5.06	6.29	7
Pit 2	31	31	4.17	7.72	3.55	5.92	5.85	0.83	0.69	0.0895	2.6678	0	5.33	6.61	4
Pit 3	20	20	4.36	7.60	3.24	6.18	6.37	0.90	0.81	-0.5816	2.6834	0	5.64	6.75	3

Dissolved Oxygen (milligrams per liter)

Area	N	D	Min	Max	Range	Mean	Median	Std	Var	Skewness	Kurtosis	Outliers	PCTL25	PCTL75	DO<=1
Pit 1	36	36	0.06	7.45	7.39	1.45	1.05	1.46	2.13	2.186	9.162	1	0.41	2.05	18
Pit 2	31	31	0.04	4.10	4.06	0.94	0.70	0.81	0.66	2.082	8.38	3	0.52	1.09	22
Pit 3	20	20	0.06	2.11	2.05	0.83	0.61	0.69	0.47	0.664	1.949	0	0.28	1.45	13

Oxidation Reduction Potential (millivolts)

Area	N	D	Min	Max	Range	Mean	Median	Std	Var	Skewness	Kurtosis	Outliers	PCTL25	PCTL75	ORP<-50
Pit 1	36	36	-106.2	477.9	584.1	146.5	115.4	194.8	37935	0.242	1.600	0	-45.0	307.3	9
Pit 2	31	31	-189	420.6	609.6	21.4	-33.1	149.3	22297	0.838	3.009	0	-94.0	129.5	12
Pit 3	20	20	-111.9	224.6	336.5	16.2	0.1	93.3	8696	0.664	2.589	0	-53.2	70.1	6

Specific Conductance (milliSiemens per centimeter)

Area	N	D	Min	Max	Range	Mean	Median	Std	Var	Skewness	Kurtosis	Outliers	PCTL25	PCTL75
Pit 1	36	36	0.075	1.78	1.705	0.526	0.446	0.390	0.152	1.338	4.614	2	0.261	0.702
Pit 2	31	31	0.067	0.790	0.731	0.397	0.378	0.198	0.039	0.363	2.442	0	0.252	0.526
Pit 3	20	20	0.108	0.564	0.456	0.349	0.364	0.118	0.014	-0.201	2.676	0	0.272	0.415

Notes: N = number of normal samples, D = number of detected results, Min = minimum detected result, max = maximum detected result, Std. Dev. = standard deviation, Var = variance, PCTL25 = 25th percentile, PCTL75. = 75th percentile, pH < 5 = number of results with pH less than 5, pH < 10 = number of results with dissolved oxygen level less than or equal to 1 milligram per liter, pH < 10 = number of results with oxidation reduction potential less than pH < 10 = number of results with oxidation reduction potential less than pH < 10 = number of pH < 10 = number

For Pit 3, the distribution range is 3.24 with a mean of 6.18 and median of 6.37. The distribution 25th and 75th percentiles are 5.64 and 6.75, respectively. Three of 20 data observations have a pH less than 5.

2.4.2.2 Dissolved Oxygen (DO) Results

For Pit 1, the distribution range for dissolved oxygen (DO) is 7.39 milligrams per liter (mg/L) with a mean of 1.45 mg/L and median of 1.05 mg/L. The distribution 25th and 75th percentiles are 0.41 mg/L and 2.05 mg/L, respectively. Eighteen of 36 data observations have a DO level less than or equal to 1 mg/L.

For Pit 2, the distribution range for DO is 6.96 mg/L with a mean of 1.12 mg/L and median of 0.70 mg/L. The distribution 25th and 75th percentiles are 0.53 mg/L and 1.20 mg/L, respectively. Twenty three of 35 data observations have a DO level less than or equal to 1 mg/L.

For Pit 3, the distribution range for DO is 2.05 mg/L with a mean of 0.83 mg/L and median of 0.61 mg/L. The distribution 25th and 75th percentiles are 0.28 mg/L and 1.45 mg/L, respectively. Thirteen of 20 data observations have a DO level less than or equal to 1 mg/L.

2.4.2.3 ORP Results

For Pit 1, the distribution range for oxidation reduction potential (ORP) is 584.1 millivolts (mV) with a mean of 146.5 mV and median of 115.4 mV. The distribution 25th and 75th percentiles are -45.0 mV and 307.3 mV, respectively. Nine of 36 data observations have an ORP level less than -50 mV.

For Pit 2, the distribution range for ORP is 609.6 mV with a mean of 26.5 mV and median of -33.1 mV. The distribution 25th and 75th percentiles are -84.7 mV and 141.2 mV, respectively. Thirteen of 33 data observations have an ORP level less than -50 mV.

For Pit 3, the distribution range for ORP is 336.5 mV with a mean of 16.2 mV and median of 0.1 mV. The distribution 25th and 75th percentiles are -53.2 mV and 70.1 mV, respectively. Six of 20 data observations have an ORP level less than -50 mV.

2.4.2.4 SC Results

For Pit 1, the distribution range for SC is 1.705 mS/cm with a mean of 0.526 mS/cm and median of 0.446 mS/cm. The distribution 25th and 75th percentiles are 0.261 mS/cm and 0.702 mS/cm, respectively.

For Pit 2, the distribution range for SC is 0.963 mS/cm with a mean of 0.424 mS/cm and a median of 0.320 mS/cm. The distribution 25th and 75th percentiles are 0.261 mS/cm and 0.552 mS/cm, respectively.

For Pit 3, the distribution range for SC is 0.456 mS/cm with a mean of 0.349 mS/cm and median of 0.364 mS/cm. The distribution 25th and 75th percentiles are 0.272 mS/cm and 0.415 mS/cm, respectively.

2.4.3 VOC Results for Surficial Aguifer (2024)

Section 2.4.3 and Exhibit 2-2 (page 2-20) have a summary of data distributions for VOC results (primary COCs) for the surficial aquifer. The summaries use the OU 7 annual monitoring data from March 2024.

2.4.3.1 Constituents of Concern

Table 2-4 on page 2-21 has a statistical summary of VOC constituents detected in one or more samples collected for annual groundwater monitoring. This table has descriptive statistics, identifies the location of the maximum detected concentrations, and presents screening comparisons to remedy cleanup levels (MCLs), EPA regional screening levels (RSLs), and ecological screening benchmarks as defined in the table notes.

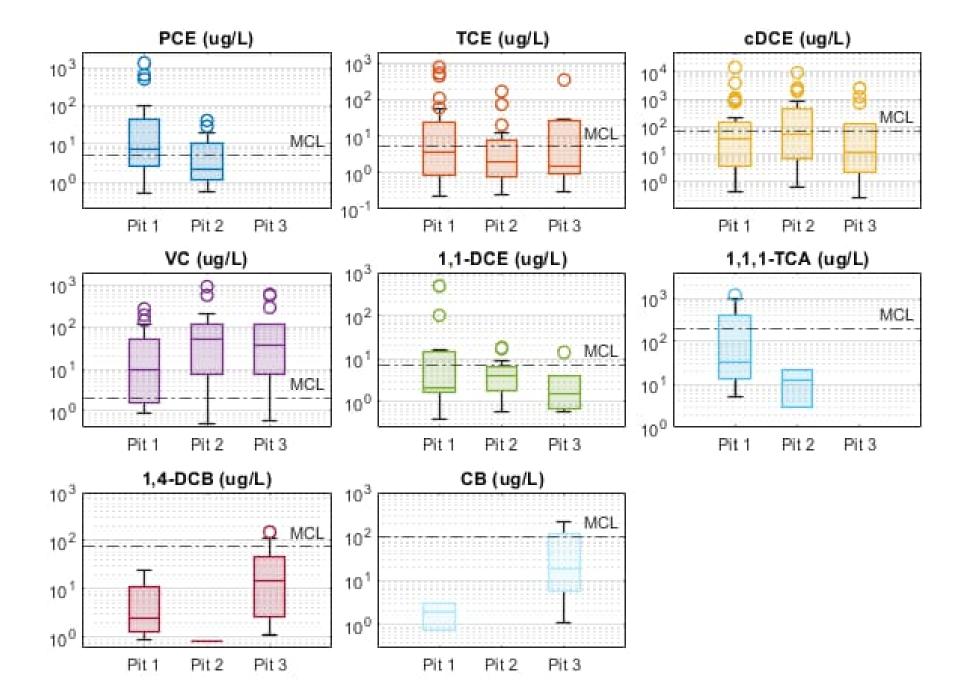
VOCs detected in groundwater for this sampling event at levels greater than MCLs include: tetrachloroethene (PCE), trichloroethene (TCE), cis-1,2-dichloroethene (cDCE), vinyl chloride (VC), carbon tetrachloride, 1,1-dichloroethene (1,1-DCE), 1,1,1-trichloroethane (1,1,1-TCA), 1,4-dichlorobenzene (1,4-DCB), chlorobenzene (CB), and benzene. For constituents without MCLs, VOCs detected at concentrations greater than RSLs include: 1,1-dichloroethane, 1,2,3-trichlorobenzene, 1,3,5-trichlorobenzene, 2-butanone, 2-hexanone, acetone, hexachlorobutadiene, naphthalene, and o-xylene. Ketone constituents such as 2-butanone, 2-hexanone, and acetone typically form as a temporary byproduct of engineered bioremediation actions.

Defense Supply Center Richmond

OU 7 Remedial Action Work Plan Addendum

Tetrachloroethene (micrograms per liter)

Exhibit 2-2 VOC Data Distribution Surficial Aquifer: Annual Monitoring March 2024



Skewness Kurtosis Outliers PCTL25 PCTL75 >MCL Area D Min Max Range | Mean | Median Std Var Pit 1 1300 1299.48 331.3 109739 2.6 0.52 143.1 2.640 Pit 2 0.55 12.83 13 8.84 2.2 164 1.619 4.228 10.2 41 40.45 1.2 Pit 3 Trichloroethene (micrograms per liter)

Area	N	D	Min	Max	Range	Mean	Median	Std	Var	Skewness	Kurtosis	Outliers	PCTL25	PCTL75	>MCL
Pit 1	36	27	0.22	780	779.78	79.2	3.6	191.6	36706	2.701	9.063	5	0.82	24.2	12
Pit 2	30	20	0.24	170	169.76	15.6	2.0	39.9	1590	3.267	12.71	3	0.80	7.4	7
Pit 3	21	13	0.31	350	349.69	34.3	1.5	95.6	9138	3.090	10.746	1	0.96	27.5	4

Cis-1	1,2-dic	<u>chloro</u>	ethene	(microg	rams p	er liter)										
Are	ea	N	D	Min	Max	Range	Mean	Median	Std	Var	Skewness	Kurtosis	Outliers	PCTL25	PCTL75	>MCL
Pit 1		36	32	0.43	14000	13999.57	706.6	37	2528	6391301	4.809	25.535	6	3.5	135	12
Pit 2	2	30	27	0.61	9200	9199.39	710.1	53	1832	3356600	3.936	18.496	4	6.9	445	13
Pit 3	3	21	19	0.27	2500	2499.73	379.5	12	791	625272	2.042	5.639	4	2.2	128	7

Vinyl Ch	nloride (microgr	ams pe	r liter)											
Area	Ν	D	Min	Max	Range	Mean	Median	Std	Var	Skewness	Kurtosis	Outliers	PCTL25	PCTL75	>MCL
Pit 1	36	24	0.88	270	269.12	46.3	9.5	70.4	4951	1.861	5.740	3	1.6	52	17
Pit 2	30	24	0.49	920	919.51	118.4	52	208.6	43496	2.907	10.936	2	7.4	120	21
Pit 3	21	15	0.59	600	599.41	126.5	36	198.9	39559	1.684	4.294	3	7.7	118	13

1,1-Dich	loroeth	ene (mi	crogran	ns per li	ter)										
Area	Ν	D	Min	Max	Range	Mean	Median	Std	Var	Skewness	Kurtosis	Outliers	PCTL25	PCTL75	>MCL
Pit 1	36	17	0.39	490	489.61	66.8	2.1	159.2	25332	2.2843	6.359	3	1.62	14.5	6
Pit 2	30	13	0.58	18	17.42	5.6	4.1	5.74	32.9	1.398	3.525	2	1.78	56.4	3
Pit 3	21	6	0.56	14	13.44	3.7	1.6	5.18	26.9	1.5795	3.781	1	0.7	4.0	1

<u>1,1,</u>	<u>,1-Tri</u>	chloroe	thane (microgr	ams pe	r liter)										
ıA	rea	N	D	Min	Max	Range	Mean	Median	Std	Var	Skewness	Kurtosis	Outliers	PCTL25	PCTL75	>MCL
Pit	1	36	9	5.4	1200	1194.6	283.3	33	470	221200	1.312	2.869	1	14	415	3
Pit	2	30	2	3	22	19	12.5	12.5	13.4	180.5	0	1	0	3	22	0
Pit	3	21	0													0

1,4-Dich	nlorober	zene (n	nicrogra	ams per	liter)										
Area	N	D	Min	Max	Range	Mean	Median	Std	Var	Skewness	Kurtosis	Outliers	PCTL25	PCTL75	>MCL
Pit 1	36	8	0.88	24	23.12	6.8	2.5	9.2	85	1.196	2.582	0	1.25	11.2	0
Pit 2	30	1	0.85	0.85	0	0.9	0.9	0.0	0			0	0.85	0.85	0
Pit 3	21	13	1.1	150	148.9	34.2	15	46.6	2169	1.553	4.198	1	2.55	47.2	2

Chlorob	enzene	(microg	grams p	er liter)											
Area	N	D	Min	Max	Range	Mean	Median	Std	Var	Skewness	Kurtosis	Outliers	PCTL25	PCTL75	>MCL
Pit 1	36	2	0.72	3.1	2.38	1.91	1.91	1.68	2.83	-2.6E-16	1.000	0	0.72	3.1	0
Pit 2	32	0										0			0
Pit 3	21	13	1.1	220	218.9	61.9	19.0	74.4	5529	1.004	2.657	0	5.8	112	4

Notes: N = number of normal samples, D = number of detected results, Min = minimum detected result, max = maximum detected result, Std. Dev. = standard deviation, Var = variance, PCTL25 = 25th percentile, PCTL75. = 75th percentile, >MCL = number of results with concentrations greater than Maximum Contaminant Level. Maximum contaminant level contained in Subpart G—National Primary Drinking Water Regulations: Maximum Contaminant Levels and Maximum Residual Disinfectant Levels, § 141.61 Maximum contaminant levels for organic contaminants

Table 2-4 Statistical Summary for VOCs: March 2024 GW Monitoring Surficial Aquifer (Detects)

									No.				Federal MCI	_1	EPA	RSL ² (N	lov 2024)	BTAG	FWSB	3 ³ (2006)
Chemical	Matrix	No. of Results	Unit	Min	Max	Mean	Median	95th Percentile	> LOD	% > LOD	Location of Max	No. > MCL	% > MCL	MCL ¹	No. > RSL	No. > RSL	EPA RSL ²	No. > BTAG	% > BTAG	BTAG FWSB ³
1,1,1-Trichloroethane	WG	91	ug/L	<0.21	1200	n.d.	n.d.	1100	11	12.1	AEHADG-11	3	3	2.00E+02	2	2	8.01E+02	8	9	1.10E+01
1,1-Dichloroethane	WG	91	ug/L	<0.33	450	14.5	n.d.	127	43	47.3	OU7-MW-68	-	-	-	25	27	2.75E+00	5	5	4.70E+01
1,1-Dichloroethene	WG	91	ug/L	<0.33	490	13.9	n.d.	195	38	41.8	CMT-21-2	11	12	7.00E+00	3	3	2.85E+01	3	3	2.50E+01
1,2,3-Trichlorobenzene	WG	91	ug/L	<0.81	1.5	n.d.	n.d.	n.d.	1	1.1	CMT-18-1	-	-	-	1	1	7.04E-01	0	0	8.00E+00
1,2,4-Trichlorobenzene	WG	91	ug/L	<0.53	3.8	n.d.	n.d.	3	7	7.7	OU7-MW-125	0	0	7.00E+01	7	8	3.99E-01	1	1	2.40E+01
1,2-Dichlorobenzene	WG	91	ug/L	<0.31	220	11.6	n.d.	170	37	40.7	OU7-MW-68	0	0	6.00E+02	6	7	3.04E+01	35	38	7.00E-01
1,2-Dichloroethene, Total	WG	91	ug/L	<0.37	14000	546.1	23	2510	79	86.8	CMT-21-2	-	1	-	-	-	-	15	16	5.90E+02
1,3,5-Trimethylbenzene	WG	91	ug/L	<0.28	7.1	n.d.	n.d.	n.d.	3	3.3	OU7-MW-68	-	1	-	1	1	6.03E+00	0	0	7.10E+01
1,3-Dichlorobenzene	WG	91	ug/L	<0.31	14	1.6	n.d.	11	15	16.5	OU7-MW-125	-	-	-	-	-	-	0	0	1.50E+02
1,4-Dichlorobenzene	WG	91	ug/L	<0.31	150	6.5	n.d.	107	22	24.2	OU7-MW-125	2	2	7.50E+01	22	24	4.82E-01	6	7	2.60E+01
2-Butanone (MEK)	WG	91	ug/L	<6.4	120	n.d.	n.d.	109	13	14.3	DMW-21A	-	-	-	1	1	5.57E+02	0	0	1.40E+04
2-Hexanone	WG	91	ug/L	<3.2	57	n.d.	n.d.	51	5	5.5	OU7-MW-109	-	-	-	1	1	3.80E+00	1	1	9.90E+01
4-Chlorotoluene	WG	91	ug/L	<0.41	0.53	n.d.	n.d.	n.d.	1	1.1	CMT-18-1	-	-	-	0	0	2.50E+01	-	_	-
4-Isopropyltoluene	WG	91	ug/L	<0.44	3.1	n.d.	n.d.	n.d.	3	3.3	OU7-MW-68	-	-	-	-	-	-	0	0	8.50E+01
4-Methyl-2-pentanone (MIBK)	WG	91	ug/L	<2.7	20	n.d.	n.d.	n.d.	2	2.2	OU7-MW-57	-	-	-	0	0	6.26E+02	0	0	1.70E+02
Acetone	WG	91	ug/L	<3.7	19000	403.8	n.d.	2975	36	39.6	CMT-21-2	-	-	-	5	5	1.80E+03	6	7	1.50E+03
Benzene	WG	91	ug/L	<0.27	5.2	n.d.	n.d.	4	11	12.1	OU7-MW-109	1	1	5.00E+00	9	10	4.55E-01	0	0	3.70E+02
Bromobenzene	WG	91	ug/L	<0.24	0.33	n.d.	n.d.	n.d.	1	1.1	CMT-18-1	-	-	-	0	0	6.22E+00	-	_	-
Carbon disulfide	WG	91	ug/L	<0.43	7.9	n.d.	n.d.	6	6	6.6	AEHADG-11	-	-	-	0	0	8.11E+01	3	3	9.20E-01
Carbon tetrachloride	WG	91	ug/L	<0.3	40	n.d.	n.d.	n.d.	1	1.1	AEHADG-11	1	1	5.00E+00	1	1	4.55E-01	1	1	1.33E+01
Chlorobenzene	WG	91	ug/L	<0.15	220	9.4	n.d.	192	15	16.5	OU7-MW-125	4	4	1.00E+02	9	10	7.77E+00	13	14	1.30E+00
Chloroethane	WG	91	ug/L	<4.6	460	n.d.	n.d.	256	13	14.3	CMT-21-1	-	,	-	0	0	8.34E+02	-	-	-
Chloroform	WG	91	ug/L	<0.27	42	n.d.	n.d.	37	7	7.7	CMT-21-2	0	0	8.00E+01	7	8	2.21E-01	6	7	1.80E+00
Chloromethane (Methyl chloride)	WG	91	ug/L	<0.54	1.1	n.d.	n.d.	n.d.	1	1.1	CMT-21-4	-	ı	-	0	0	1.88E+01	-	_	-
cis-1,2-Dichloroethene	WG	91	ug/L	<0.25	14000	544.3	23	2500	81	89.0	CMT-21-2	34	37	7.00E+01	65	71	2.52E+00	-	-	-
Ethylbenzene	WG	91	ug/L	<0.2	130	2.5	n.d.	37	19	20.9	CMT-21-2	0	0	7.00E+02	14	15	1.50E+00	1	1	9.00E+01
Hexachlorobutadiene	WG	97	ug/L	<0.22	1.5	n.d.	n.d.	1	5	5.2	CMT-18-1	-	-	-	4	4	1.39E-01	1	1	1.30E+00
Isopropylbenzene (Cumene)	WG	91	ug/L	<0.26	1.5	n.d.	n.d.	n.d.	3	3.3	OU7-MW-68	-	1	-	0	0	4.51E+01	0	0	2.60E+00
m+p-Xylenes	WG	91	ug/L	<0.49	460	9.5	n.d.	235	17	18.7	CMT-21-2	-	-	-	-	-	-	-	_	-
Naphthalene	WG	97	ug/L	<0.021	290	n.d.	n.d.	241	8	8.2	CMT-21-2	-	-	-	6	7	1.17E-01	6	7	1.10E+00
n-Butylbenzene	WG	91	ug/L	<0.52	1.2	n.d.	n.d.	n.d.	2	2.2	CMT-18-1	-	-	-	0	0	1.00E+02	-		-
n-Propylbenzene	WG	91	ug/L	<0.41	1.1	n.d.	n.d.	n.d.	2	2.2	OU7-MW-69	-	-	-	0	0	6.56E+01	0	0	1.28E+02
o-Xylene	WG	91	ug/L	<0.26	36	1.6	n.d.	23	17	18.7	CMT-21-2	-	-	-	1	1	1.93E+01	-	-	-
sec-Butylbenzene	WG	91	ug/L	<0.53	0.82	n.d.	n.d.	n.d.	2	2.2	OU7-MW-69	-	-	-	0	0	2.01E+02	-	_	-
Styrene	WG	91	ug/L	<0.27	0.28	n.d.	n.d.	n.d.	1	1.1	CMT-18-1	0	0	1.00E+02	0	0	1.21E+02	0	0	7.20E+01
tert-Butyl methyl ether (MTBE)	WG	91	ug/L	<0.81	1.9	n.d.	n.d.	n.d.	1	1.1	OU7-MW-116	-	_	-	0	5	1.43E+01	0	0	1.11E+04

									No.				Federal MCL	1	EPA	RSL ² (N	lov 2024)	ВТАС	FWSB	3 (2006)
Chemical		No. of Results	Unit	Min	Max	Mean	Median	95th Percentile	>	% > LOD	Location of Max	No. > MCL	% > MCL	MCL ¹	No. > RSL	No. > RSL	EPA RSL ²	No. > BTAG	% > BTAG	BTAG FWSB ³
tert-Butylbenzene	WG	91	ug/L	<0.43	0.55	n.d.	n.d.	n.d.	2	2.2	CMT-18-1	-	-	-	0	0	6.91E+01	-	-	-
Tetrachloroethene (PCE)	WG	91	ug/L	<0.35	1300	31.9	n.d.	556	33	36.3	AEHADG-11	20	21	5.00E+00	21	23	4.06E+00	3	3	1.11E+02
Toluene	WG	91	ug/L	<0.25	17	2.0	n.d.	10	34	37.4	OU7-MW-48	0	0	1.00E+03	0	0	1.10E+02	22	24	2.00E+00
trans-1,2-Dichloroethene	WG	91	ug/L	<0.34	34	2.0	n.d.	11	31	34.1	OU7-MW-85	0	0	1.00E+02	7	8	6.78E+00	0	0	9.70E+02
Trichloroethene (TCE)	WG	91	ug/L	<0.2	780	32.3	1.1	341	62	68.1	OU7-MW-36	24	26	5.00E+00	65	71	2.83E-01	15	16	2.10E+01
Trichlorofluoromethane	WG	91	ug/L	<0.33	8.6	n.d.	n.d.	n.d.	1	1.1	AEHADG-11	-	-	-	0	0	5.16E+02	-	-	-
Vinyl chloride	WG	91	ug/L	<0.4	920	67.0	6.9	492	66	72.5	CMT-18-5	54	59	2.00E+00	91	100	1.88E-02	0	0	9.30E+02
Xylene (Total)	WG	91	ug/L	<0.49	500	10.5	n.d.	220	18	19.8	CMT-21-2	0	0	1.00E+04	5	5	1.93E+01	5	5	1.30E+01

Notes: WG = groundwater, ug/L = micrograms per liter, Min = minimum, Max = maximum, No. > = number greater, LOD = limit of detection, % = percent, , ¹Subpart G—National Primary Drinking Water Regulations: Maximum Contaminant Levels and Maximum Residual Disinfectant Levels, § 141.61 Maximum contaminant levels for organic contaminants. > LOD = percent of detected results greater than laboratory limit of detection, ² EPA Regional Screening Level for Resident Tap Water with a target cancer risk (TR) of 1E-06 and target hazard quotient (THQ) of 0.1, November 2004 at https://www.epa.gov/risk/regional-screening-levels-rsls-generic-tables. ³EPA Region III BTAG Freshwater Screening Benchmarks, Number of results with concentrations greater than the MCL, Number of results with concentration greater than the RSL, Number of results with concentrations greater than the BTAG FWSB

For the ecological screening, groundwater constituents detected in surface water samples in 2024 included cDCE, total dichloroethene, and TCE in one sample (KGC-1) at concentrations less than the laboratory LOQ. Sample KGC-1 is located upstream of the monitoring areas for FTA Pits 1 and 2 (Figure 2-6).

2.4.3.2 Data Distribution for Annual Monitoring

Exhibit 2-2 (2-20) has box plots and tabular information displaying data distributions for primary VOC constituents for annual monitoring for the plume areas associated with FTA Pits 1, 2, and 3. Figure 3-1 shows the locations of monitoring wells referenced in this section.

PCE

PCE occurs in the plume areas associated with FTA Pits 1 and 2 with the data distribution for Pit 1 plotting at a higher concentration range than Pit 2 (see Exhibit 2-2). Samples collected from the Pit 3 area did not have detections of PCE.

Pit 1 has 19 of 36 samples with detections of PCE and a detected concentration range of 0.52 micrograms per liter (μ g/L) to 1,300 μ g/L. Mean and median concentrations for PCE at Pit 1 are 143.1 μ g/L and 7 μ g/L, respectively, with 11 samples having concentrations greater than the MCL of 5 μ g/L. Distribution 25th and 75th percentiles for PCE at Pit 1 are 2.6 μ g/L and 45 μ g/L, respectively.

Pit 2 has 13 of 30 samples with detections of PCE and a detected concentration range of 0.55 μ g/L to 41 μ g/L. Mean and median concentrations for PCE at Pit 2 are 8.84 μ g/L and 2.2 μ g/L, respectively, with five (5) samples having concentrations greater than the MCL of 5 μ g/L. Distribution 25th and 75th percentiles for PCE at Pit 2 are 1.2 μ g/L and 10 μ g/L, respectively.

TCE

TCE has data distributions in Pit 1, 2, and 3 areas that extend 1 to 2 orders of magnitude higher than the MCL level of 5 μ g/L (see Exhibit 2-2).

Pit 1 has 27 of 36 samples with detections of TCE and a detected concentration range of 0.22 μ g/L to 780 μ g/L. Mean and median concentrations for TCE at Pit 1 are 79.2 μ g/L and 3.6 μ g/L, respectively, with 12 samples having concentrations greater than the MCL of 5 μ g/L. Distribution 25th and 75th percentiles for TCE at Pit 1 are 0.82 μ g/L and 24.2 μ g/L, respectively.

Pit 2 has 20 of 30 samples with detections of TCE and a detected concentration range of 0.24 μ g/L to 170 μ g/L. Mean and median concentrations for TCE at Pit 2 are 15.6 μ g/L and 2.1 μ g/L, respectively, with seven (7) samples having concentrations greater than the MCL of 5 μ g/L. Distribution 25th and 75th percentiles for TCE at Pit 2 are 0.80 μ g/L and 7.4 μ g/L, respectively.

Pit 3 has 13 of 21 samples with detections of TCE and a detected concentration range of 0.31 μ g/L to 350 μ g/L. Mean and median concentrations for TCE at Pit 3 are 34.3 μ g/L and 1.5 μ g/L, respectively, with four (4) samples having concentrations greater than the MCL of 5 μ g/L. Distribution 25th and 75th percentiles for TCE at Pit 3 are 0.96 μ g/L and 27.5 μ g/L, respectively.

cDCE

cDCE has data distributions in the Pit 1, 2, and 3 areas that extend 1 to 2 magnitudes higher than the MCL level of 70 μ g/L (see Exhibit 2-2).

Pit 1 has 32 of 36 samples with detections of cDCE and a detected concentration range of 0.43 μ g/L to 14,000 μ g/L. Mean and median concentrations for cDCE at Pit 1 are 706.5 μ g/L and 73 μ g/L, respectively, with 12 samples having concentrations greater than the MCL of 70 μ g/L. Distribution 25th and 75th percentiles for cDCE at Pit 1 are 3.5 μ g/L and 135 μ g/L, respectively.

Pit 2 has 27 of 30 samples with detections of cDCE and a detected concentration range of 0.61 μ g/L to 9,200 μ g/L. Mean and median concentrations for cDCE at Pit 2 are 710.1 μ g/L and 53 μ g/L, respectively, with 13 samples having concentrations greater than the MCL of 70 μ g/L. Distribution 25th and 75th percentiles for cDCE at Pit 2 are 6.9 μ g/L and 445 μ g/L, respectively.

Pit 3 has 19 of 21 samples with detections of cDCE and a detected concentration range of 0.27 μ g/L to 2,500 μ g/L. Mean and median concentrations for cDCE at Pit 3 are 379.5 μ g/L and 12 μ g/L, respectively, with seven (7) samples having concentrations greater than the MCL of 70 μ g/L. Distribution 25th and 75th percentiles for cDCE at Pit 3 are 2.2 μ g/L and 128 μ g/L, respectively.

VC

VC has data distributions in the Pit 1, 2, and 3 areas that extend two (2) magnitudes higher than the MCL level of 2 µg/L (see Exhibit 2-2).

Pit 1 has 17 of 36 samples with detections of VC and a detected concentration range of 0.88 μ g/L to 270 μ g/L. Mean and median concentrations for VC at Pit 1 are 46.3 μ g/L and 9.5 μ g/L, respectively, with 17 samples having concentrations greater than the MCL of 2 μ g/L. Distribution 25th and 75th percentiles for VC at Pit 1 are 1.6 μ g/L and 52 μ g/L, respectively.

Pit 2 has 24 of 30 samples with detections of VC and a detected concentration range of 0.49 μ g/L to 920 μ g/L. Mean and median concentrations for VC at Pit 2 are 118.4 μ g/L and 52 μ g/L, respectively, with 21 samples having concentrations greater than the MCL of 2 μ g/L. Distribution 25th and 75th percentiles for VC at Pit 2 are 7.5 μ g/L and 120 μ g/L, respectively.

Pit 3 has 6 of 21 samples with detections of VC and a detected concentration range of 0.59 μ g/L to 600 μ g/L. Mean and median concentrations for VC at Pit 3 are 126.5 μ g/L and 35 μ g/L, respectively, with 13 samples having concentrations greater than the MCL of 2 μ g/L. Distribution 25th and 75th percentiles for VC at Pit 3 are 7.7 μ g/L and 118 μ g/L, respectively.

1,1-DCE

Pit 1 has 17 of 36 samples with detections of 1,1-DCE and a detected concentration range of 0.39 μ g/L to 49 μ g/L. Mean and median concentrations for 1,1-DCE at Pit 1 are 66.8 μ g/L and 2.1 μ g/L, respectively, with six (6) samples having concentrations greater than the MCL of 7 μ g/L. Distribution 25th and 75th percentiles for 1,1-DCE at Pit 1 are 1.62 μ g/L and 14.5 μ g/L, respectively.

Pit 2 has 13 of 30 samples with detections of 1,1-DCE and a detected concentration range of 0.58 μ g/L to 18 μ g/L. Mean and median concentrations for 1,1-DCE at Pit 2 are 5.6 μ g/L and 4.1 μ g/L, respectively, with three (3) samples having concentrations greater than the MCL of 7 μ g/L. Distribution 25th and 75th percentiles for 1,1-DCE at Pit 2 are 1.77 μ g/L and 4.0 μ g/L, respectively.

Pit 3 has 6 of 21 samples with detections of 1,1-DCE and a detected concentration range of 0.56 μ g/L to 14 μ g/L. Mean and median concentrations for 1,1-DCE at Pit 3 are 3.7 μ g/L and 1.6 μ g/L, respectively, with one (1) sample having a concentration greater than the MCL of 7 μ g/L. Distribution 25th and 75th percentiles for 1,1-DCE at Pit 3 are 0.7 μ g/L and 4.0 μ g/L, respectively.

1,1,1-TCA

Pit 1 has 9 of 36 samples with detections of 1,1,1-TCA and a detected concentration range of 5.4 μ g/L to 1,200 μ g/L. Mean and median concentrations for 1,1,1-TCA at Pit 1 are 283.3 μ g/L and 33 μ g/L, respectively, with three (3) samples having concentrations greater than the MCL of 200 μ g/L. Distribution 25th and 75th percentiles for 1,1,1-TCA at Pit 1 are 14 μ g/L and 415 μ g/L, respectively.

Pit 2 has 2 of 30 samples with detections of 1,1,1-TCA and a detected concentration range of 3 μ g/L to 22 μ g/L. Mean and median concentrations for 1,1,1-TCA at Pit 2 are 12.5 μ g/L and 12.5 μ g/L, respectively, with no samples having concentrations greater than the MCL of 200 μ g/L. Distribution 25th and 75th percentiles for 1,1,1-TCA at Pit 2 are 3 μ g/L and 22 μ g/L, respectively.

Samples collected from the Pit 3 area did not have detections of 1,1,1-TCA.

1,4-DCB

Pit 1 has 8 of 36 samples with detections of 1,4-DCB and a detected concentration range of 0.88 μ g/L to 24 μ g/L. Mean and median concentrations for 1,4-DCB at Pit 1 are 6.8 μ g/L and 2.5 μ g/L, respectively, with no samples having concentrations greater than the MCL of 75 μ g/L. Distribution 25th and 75th

percentiles for 1,4-DCB at Pit 1 are 1.25 μ g/L and 11.2 μ g/L, respectively. Pit 2 has 1 of 30 samples with a detection of 1,4-DCB at an estimated concentration of 0.85 μ g/L.

Pit 3 has 13 of 21 samples with detections of 1,4-DCB and a detected concentration range of 1.1 μ g/L to 150 μ g/L. Mean and median concentrations for 1,4-DCB at Pit 3 are 34.2 μ g/L and 15 μ g/L, respectively, with two (2) samples having a concentration greater than the MCL of 75 μ g/L. Distribution 25th and 75th percentiles for 1,4-DCB at Pit 3 are 2.55 μ g/L and 47.2 μ g/L, respectively.

CB

Pit 1 has 2 of 36 samples with detections of CB at concentrations of 0.72 μ g/L to 3.1 μ g/L. The detected concentrations of CB did not exceed the MCL of 100 μ g/L. Samples collected from the Pit 2 area did not have detections of CB.

Pit 3 has 13 of 21 samples with detections of CB and a detected concentration range of 1.1 μ g/L to 220 μ g/L. Mean and median concentrations for CB at Pit 3 are 61.9 μ g/L and 19.0 μ g/L, respectively, with four (4) samples having a concentration greater than the MCL of 100 μ g/L. Distribution 25th and 75th percentiles for CB at Pit 3 are 5.8 μ g/L and 112 μ g/L, respectively.

Carbon Tetrachloride

Carbon tetrachloride had a detection in 1 of 89 samples with reported concentration of 40 μ g/L at well AEHADG-11 in the Pit 2 area. The detected concentration is greater than the MCL of 5 μ g/L.

Benzene

Benzene had reported detections in 4 of 87 samples with one sample in Pit 3 (well MW-109, $5.2 \mu g/L$) with a reported concentration greater than the MCL of $5 \mu g/L$.

2.4.4 Geochemical Results for Surficial Aquifer (2024)

Section 2.4.4 and Exhibit 2-3 (page 2-26) have a summary of data distributions for geochemical results for the surficial aquifer targeted for ISB actions. The summaries use the OU 7 annual monitoring data from March 2024.

2.4.4.1 Chloride

Pit 1 has chloride concentrations ranging from 0.88 to 160 mg/L with a mean concentration of 54.3 mg/L and median concentration of 29 mg/L. Pit 2 has chloride concentrations ranging from 1.8 mg/L to 260 mg/L with a mean concentration of 96.1 mg/L and median concentration of 85.0 mg/L. Pit 3 has chloride concentrations ranging from 8 mg/L to 130 mg/L with a mean concentration of 58.7 mg/L and median concentration of 53.0 mg/L.

2.4.4.2 Nitrate

Minimal nitrate occurs in groundwater within the monitored areas at Pits 1, 2, and 3 with single detections in each monitoring area at levels less than 1 mg/L.

2.4.4.3 Sulfate

At Pit 1, 21 of 31 samples have detectable levels of sulfate ranging from 0.57 mg/L to 46 mg/L with mean and median concentrations of 12.5 mg/L and 6.5 mg/L, respectively. Pit 2 has 20 of 25 samples with detectable levels of sulfate ranging from 0.6 mg/L to 46 mg/L with mean and median concentrations of 13.3 mg/L and 6.0 mg/L, respectively. Pit 3 has 12 of 14 samples with detectable levels of sulfate ranging from 1 mg/L to 32 mg/L with mean and median concentrations of 6.7 mg/L and 2.8 mg/L, respectively.

2.4.4.4 Sulfide

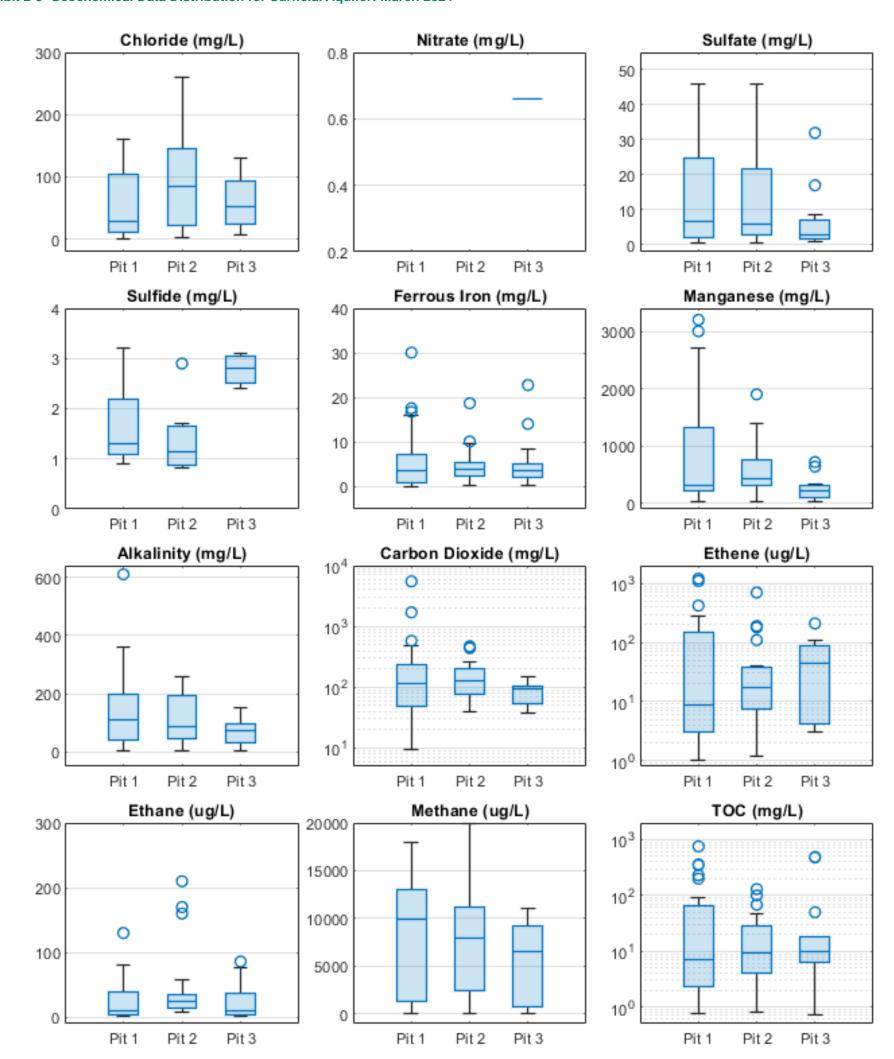
At Pit 1, 7 of 31 samples have detectable levels of sulfide ranging from 0.91 mg/L to 3.2 mg/L. Eight of 25 samples at Pit 2 have detectable levels of sulfide ranging from 0.83 mg/L to 2.9 mg/L. At Pit 3, four of 15 samples have detectable levels of sulfide ranging from 2.4 mg/L to 3.1 mg/L.

Defense Supply Center Richmond

OU 7 Remedial Action Work Plan Addendum

Chloride (milligrams per liter)

Exhibit 2-3 Geochemical Data Distribution for Surficial Aquifer: March 2024



illolla	,	iailis pe								1						
Area	N	D	Min	Max	Range		Median	Std	Var		_		Outliers I			
Pit 1	31	31	0.88		159.12	54.3	29	53.7	2886		_	2.0009	0	11.3	105	
Pit 2	25	25	1.8	260		96.2	85.0	74.0	5482			2.2672	0	21.8	145	
Pit 3	15	15	8	130	122.00	58.7	53.0	39.9	1595	0.272	5 1	1.7346	0	25.3	93.3	
_	milligra N			May	Dango	Moon	Modian	Std	Var	Skovupa	sec k	/urtocic	Outliers	DCTI 25	DCTL 75	<1
Area Pit 1	31	D 1	Min 0.95	Max 0.95	Range 0		Median 0.95	+	+	Skewne	255 K	Cur tosis	Outliers			<
Pit 2		1		0.95	0					0			-			1
	25		0.16										+			1
Pit 3	15	1	0.66	0.66	0	0.66	0.66	0		0			0	0.66	0.66	1
Area	(milligra N	ms per D	Min	Max	Range	Mean	Median	Std	Var	Skown	220	Kurtosi	s Outlier	s PCTI 2F	5 PCTI 75	=<50
Pit 1	31	23	0.57	46	45.4	12.5	6.5	13.6			072	2.80) 2.2		+
Pit 2	25	20	0.6	46	45.4	13.3	6.0	15.5			102	2.59) 2.8		
Pit 3	14	12	1	32	31.0	6.7	2.8	9.2)19			2 1.6		1
	(milligra			JZ	31.0	0.7	2.0	7.2	•	2.0	J 1 7	3.73	7 4	2 1.0	0.9	12
Area	N	D D	Min	Max	Range	Mean	Median	Std	Var	Skewnes	s Kı	urtosis	Outliers I	PCTL25 F	PCTL75	>1
Pit 1	31	7	0.91	3.2	2.29	1.66	1.30	0.84	0.71	0.98		2.518	0	1.10	2.20	6
Pit 2	25	8	0.83	2.9	2.07	1.38	1.14	0.70	0.49	1.33	3	3.744	1	0.88	1.65	4
Pit 3	15	4	2.4	3.1	0.7	2.78	2.80	0.33	0.11	-0.13		1.284	0	2.50	3.05	4
			s per lit		5.,			2 3]		'				
Area	N	D	Min	Max	Range	Mean	Median	Std	Var	Skewness	Ku	ırtosis	Outliers P	CTL25 P	CTL75	>1
Pit 1	31	31	0.01	48	47.99	6.9	3.6	10	101	2.724	1	0.720	4	1.01	7.20	23
Pit 2	27	27	0.26	18.72	18.46	4.5	3.8	3.7	13.5	2.289)	9.424	2	2.25	5.28	24
Pit 3	19	19	0.31	22.8	22.49	4.9	3.6	5.4	28.7	2.304		7.949	2	2.05	4.98	16
langan	ese (mi	crogran	ns per li	ter)			<u> </u>	l	I			I	<u> </u>			
Area	N	D	Min	Max	Range	Mean	Median	Std	Var	Skewnes	s Kı	urtosis	Outliers	PCTL25 I	PCTL75	
Pit 1	33	33	34	3200	3166	856	310	900	809687	1.29	2	3.591	2	228	1325	
Pit 2	26	26	35	1900	1865	566	435	426	181350	1.49	3	5.078	1	310	760	
Pit 3	15	15	22	720	698	243	210	203	41157	1.25	6	3.746	1	96	308	
lkalini	ty (millio	grams p	er liter)													
Area	N	D	Min	Max			Median	Std					Outliers P			
Pit 1	31	30	3.8	610		142	111	131.4	17253	1.546		6.372	1	41	200	
Pit 2	25	24	4	260	256.0	111	89	82.8	6864	0.496		1.834	0	44	195	
Pit 3	15	13	2.5	150	147.5	67	72	46.8	2188	0.015		2.077	0	29	98	
		,	ram per		Б	D 4	N 4 11	C1 1	\ \ \ \ \ \ \ \ \ \ \ \ \ \ \ \ \ \ \		1.		0 11	DOT! OF	DOTI 75	
Area	N	D	Min	Max			Median	Std	Var		-		Outliers			
Pit 1	31	31	9.7		5490.3	378	115	1018	103550		-	23.012		49	230	
Pit 2	25	24	40	470	430	151	125	112	1251			5.2638	1	76	205	
Pit 3	15	13	37	150	113	86	92	37	138	7 0.2	49	1.963	0	54	105	
Area	(microg N	rams pe D	Min	Max	Range	Mean	Median	Std	Var	Skewne	ss k	/urt∩sis	Outliers	PCTI 25	PCTI 75	>10
Pit 1	31	19	1	1200	1199	182	9	359	12907		-	6.328		-	148	9
Pit 2	25	19	1.2	700	698.8	74	17	162	2620		-	13.145		7.45		13
Pit 3	15	9	3	210	207	62	45	67	448			3.628		4.175		6
	(microg			210	201		70	07	770	1.2	-	5.020	<u>'</u>	1.173	00	U
Area	N	D D	Min	Max	Range	Mean	Median	Std	Var	Skewnes	s Kı	urtosis	Outliers I	PCTL25 I	PCTL75	>10
Pit 1	31	24	0.49	130		26.4	9.35	33.3	1111	1.60	_	4.979	1	3.5	39	12
Pit 2	25	20	6.7	210	203.3	45.9	23.5	59.6	3552	1.89		5.023	3	14	36	17
Pit 3	15	11	1.2	86		26.1	9.00	30.2	912			2.772	1	4.2	37	5
			per liter		5	_0.1		55.2	, , _	0	-	, ~	'		<u> </u>	
Area	N	D	Min	Max	Range	Mean	Median	Std	Var	Skew	ness	s Kurto	sis Outlie	ers PCTL2	25 PCTL7	5 >500
Pit 1	31	31	8.2	18000	17991.8	7557	7 9900	6035	363	670 -0.	0596	5 1.50	34	0 132	25 1300	0 24
Pit 2	25	25	51	20000	19949	7352	2 8000	5521	30488	000 0.	4503	3 2.28	64	0 26	75 1075	0 23
Pit 3	15	15	64	11000	10936	5398	3 6500	4192	17574	101 -0.	1898	3 1.35	67	0 802	2.5 922	5 12
otal Or	ganic C	arbon (milligra	m per lit	ter)											
Area	N	D	Min	Max		Mean	Median	Std	Var	Skewness	Ku	ırtosis	Outliers P	CTL25 P	CTL75	>20
Pit 1	31	31	0.77	760	759.23	74.1	7.2	161	25835	2.981	1	2.132	5	2.3	63.5	10
Pit 2	25	24	0.8	130	129.2	25.2	9.1	33.9	1154	1.836)	5.500	4	4.2	29	9
_											$\overline{}$					

Notes: N = number of normal samples, D = number of detected results, Min = minimum detected result, max = maximum detected result, Std. Dev = standard deviation, Var = variance, PCTL25 = 25th percentile, PCTL75 = 75th percentile, > greater than

70.8

9.8

162 26274

2.247

6.090

Pit 3

17

16

0.72

490 489.28

2.4.4.5 Ferrous Iron

At Pit 1, 31 of 31 samples have detectable levels of ferrous iron ranging from 0.01 mg/L to 48 mg/L with mean and median concentrations of 6.9 mg/L and 3.6 mg/L, respectively. Pit 2 has 27 of 27 samples with detectable levels of ferrous iron ranging from 0.26 mg/L to 18.72 mg/L with mean and median concentrations of 4.52 mg/L and 3.78 mg/L, respectively. Pit 3 has 19 of 19 samples with detectable levels of ferrous iron ranging from 0.31 mg/L to 22.80 mg/L with mean and median concentrations of 4.88 mg/L to 3.60 mg/L, respectively.

2.4.4.6 Manganese

Pit 1 has manganese concentrations ranging from 34 μ g/L to 3,200 μ g/L with mean and median concentrations of 856 μ g/L and 310 μ g/L. Pit 2 has manganese concentrations ranging from 35 μ g/L to 1,900 μ g/L with mean and median concentrations of 566 μ g/L and 435 μ g/L, respectively. Pit 3 has manganese concentrations ranging from 2 μ g/L to 720 μ g/L with mean and median concentrations of 242 μ g/L and 210 μ g/L, respectively.

2.4.4.7 Alkalinity

At Pit 1, 30 of 31 samples have detectable levels of alkalinity ranging from 3.8 mg/L to 610 mg/L with mean and median concentrations of 142 mg/L and 111 mg/L, respectively. Pit 2 has 24 of 25 samples with detectable levels of alkalinity ranging from 4 mg/L to 260 mg/L with mean and median concentrations of 111 mg/L and 88 mg/L, respectively. Pit 3 has 13 of 15 samples with detectable levels of alkalinity ranging from 2.5 mg/L to 150 mg/L with mean and median concentrations of 67 mg/L and 72 mg/L, respectively.

2.4.4.8 Carbon Dioxide

At Pit 1, 30 of 31 samples have detectable levels of carbon dioxide ranging from 9.7 mg/L to 5,500 mg/L with mean and median concentrations of 378 mg/L and 115 mg/L, respectively. Pit 2 has 24 of 25 samples with detectable levels of carbon dioxide ranging from 40 mg/L to 470 mg/L with mean and median concentrations of 151 mg/L and 125 mg/L, respectively. Pit 3 has 13 of 15 samples with detectable levels of carbon dioxide ranging from 37 mg/L to 150 mg/L with mean and median concentrations of 85 mg/L and 92 mg/L, respectively.

2.4.4.9 Ethene

At Pit 1, 19 of 31 samples have detectable levels of ethene ranging from 1 μ g/L to 1,200 μ g/L with mean and median concentrations of 182 μ g/L and 9 μ g/L, respectively. Pit 2 has 19 of 25 samples with detectable levels of ethene ranging from 1.2 μ g/L to 700 μ g/L with mean and median concentrations of 74 μ g/L and 17 μ g/L, respectively. Pit 3 has 9 of 15 samples with detectable levels of ethene ranging from 3 μ g/L to 210 μ g/L with mean and median concentrations of 62 μ g/L and 45 μ g/L, respectively.

2.4.4.10 Ethane

At Pit 1, 24 of 31 samples have detectable levels of ethane ranging from 0.49 μ g/L to 130 μ g/L with mean and median concentrations of 26.4 μ g/L and 9.35 μ g/L, respectively. Pit 2 has 20 of 25 samples with detectable levels of ethane ranging from 6.7 μ g/L to 210 μ g/L with mean and median concentrations of 45.9 μ g/L and 23.5 μ g/L, respectively. Pit 3 has 11 of 15 samples with detectable levels of ethane ranging from 1.2 μ g/L to 86 μ g/L with mean and median concentrations of 26.1 μ g/L and 9.00 μ g/L, respectively.

2.4.4.11 Methane

At Pit 1, 31 of 31 samples have detectable levels of methane ranging from 8.2 μ g/L to 18,000 μ g/L with mean and median concentrations of 7,557 μ g/L and 9,900 μ g/L, respectively. Pit 2 has 25 of 25 samples with detectable levels of methane ranging from 51 μ g/L to 20,000 μ g/ with mean and median concentrations of 7,352 μ g/L and 8,000 μ g/L, respectively. Pit 3 has 15 of 15 samples with detectable levels of methane ranging from 64 μ g/L to 11,000 μ g/L with mean and median concentrations of 5,398 μ g/L and 6,500 μ g/L, respectively.

2.4.4.12 Total Organic Carbon (TOC)

At Pit 1, 31 of 31 samples have detectable levels of total organic carbon (TOC) ranging from 0.77 mg/L to 760 mg/L with mean and median concentrations of 74.1 mg/L and 7.2 mg/L, respectively. Pit 2 has 24 of 25 samples with detectable levels of TOC ranging from 0.8 mg/L to 130 mg/L with mean and median concentrations of 25.2 mg/L and 9.0 mg/L, respectively. Pit 3 has 16 of 17 samples with detectable levels of TOC ranging from 0.72 mg/L to 490 mg/L with mean and median concentrations of 70.8 mg/L and 9.8 mg/L, respectively.

2.4.4.13 Plume Geochemical Environment

Table 2-5 has summary information for geochemical parameters at each monitored plume area used to characterize geochemical environments.

Table 2-5 Plume Geochemical Environments

		Median C	oncentra	tions (mg	/L)	Degradation	Geochemical Environment
Area	DO	NO3	Fe2	SO4	CH4	Products	in Plume Area > MCLs
Pit 1	1.05	0.95	3.6	6.5	9.90	Yes	Anaerobic
Pit 2	0.94	0.16	3.8	6.0	8.00	Yes	Anaerobic
Pit 3	0.61	0.66	3.6	2.8	6.50	Yes	Anaerobic

Notes: DO = dissolved oxygen, SO4 = sulfate, NO3 = nitrate, CH4 = methane, mg/L = milligrams per liter.

2.4.5 Microbial Results

Section 2.4.5 has a summary of microbial sample results for annual monitoring completed in March 2024. Appendix B.4 in the Fiscal Year 2024 OU 7 Annual Report has complete tables with microbial sample results.

2.4.5.1 Pit 1

Microbial sampling in the Pit 1 ISB treatment areas indicated the following results for sampled wells AEHADG-11, MW-36, and MW-124:

- Presence of Dehalococcoides (DHC) at a level of 3.60E+02 cells cells/mL at AEHADG-11, a level of 3.40E+03 cells/mL at MW-36, and a level of 3.20E+01 at MW-124.
- Presence of tceA Reductase (tceA) at an estimated level of 4.50E-01 cells/mL at MW-36, a level of 2.10E-00 at MW-124, with no detection reported for AEHADG-11.
- No detections of BAV1 Vinyl Chloride (BVC).
- Presence of Vinyl Chloride Reductase (VCR) at a level of 2.10E+02 cells/mL at AEHADG-11, a level
 of 4.80E+03 cells/mL at well MW-36, and a level of 1.40E+00 cells/mL for MW-124.

2.4.5.2 Pit 2

Microbial sampling in the Pit 2 ISB treatment areas indicated the following results for sampled wells CMT-18-3 and MW-48:

- Presence of DHC at a level of 9.50E+03 cell/mL at well CMT-18-4 and a level of 9.40E-04 cells/mL at MW-48.
- Presence of tceA at a level of 4.80E+01 cells/mL at well MW-48 with no detection reported for CMT-18-4.
- Presence of BVC at an estimated level of 7.40E+01 cells/mL at well CMT-18-4 and a level of 1.40E+02 cells/mL at well MW-48.
- Presence of VCR at a level of 2.81E+03 cells/mL at well CMT-18-4 and a level of 1.1E+04 cells/mL.

2453 Pit 3

Microbial sampling in the Pit 3 ISB treatment areas indicated the following results for sampled well MWFOS-3:

- Presence of DHC at a level of 2.90E+00 cells/mL.
- tceA, BVC, and VCR not detected.

2.4.6 Primary VOC Plume Areas for Surficial Aquifer (2024)

VOC plumes principally occur in the surficial aquifer at OU 7 with separate plume areas associated with the FTA Pit 1, 2, and 3 areas.

Figure 2-8 (page 2-30) depicts the lateral extent of PCE, TCE, cDCE, and VC plume areas within the surficial aquifer at concentrations equal to and greater than MCLs. VC generally defines the plume extent for Pits 1, 2, and 3. The Pit 1 VOC plume extends from the former pit area more than 600 ft. downgradient to the Kingsland Creek area beyond the installation fence line. At Pit 2, the VOC plume extends from the near the former pit area more than 600 ft. downgradient to the installation fence line at well MW-148. The VOC plume associated with Pit 3 extends approximately 500 ft. downgradient and southeast of the former pit area with the limits more than 300 ft. upgradient of installation fence line and Kingsland Creek.

2.4.6.1 PCE

Figure 2-9 (page 2-31) depicts the PCE plume areas associated with Pits 1 and 2. At Pit 1, a PCE plume area in the upper area has a limited source zone with concentrations greater than or equal to 500 μ g/L remaining in the area of AEHADG-11, MW-36, and CMT-21-2. A single detection of PCE at a concentration greater than the MCL (5 μ g/L) occurs downgradient of this plume area at well PZ-6 near the installation fence line.

Two isolated and degraded PCE plume areas with concentrations less than 50 μ g/L remain in the Pit 2 area. The isolated plumes are in the upper area and around well MW-148 in the fence line area.

2.4.6.2 TCE

Figure 2-10 (page 2-32) depicts the TCE plume areas associated with Pits 1, 2, and 3. At Pit 1, a TCE plume in the upper area has a limited source zone with concentrations greater than 500 μ g/L collocated with the TCE plume. A single detection of TCE at a concentration greater than MCL is collocated with PCE at PZ-6 with a second isolated plume area in the flood plain area of Kingsland Creek.

Two isolated TCE plume areas remain in the upper Pit 2 area with another isolated area encompassing wells MWFTA-3 and MW-148 in the lower area near the installation fence line. TCE concentrations are less than $50 \mu g/L$ in these plume areas except at well CMT-24-4 (170 $\mu g/L$) in the upper area.

At Pit 3, a TCE plume extends 120 ft. downgradient of the former pit with a limited area having concentrations greater than 50 μ g/L. A second plume area is located near Building 72 around well MW-117 (29 μ g/L).

2.4.6.3 cDCE

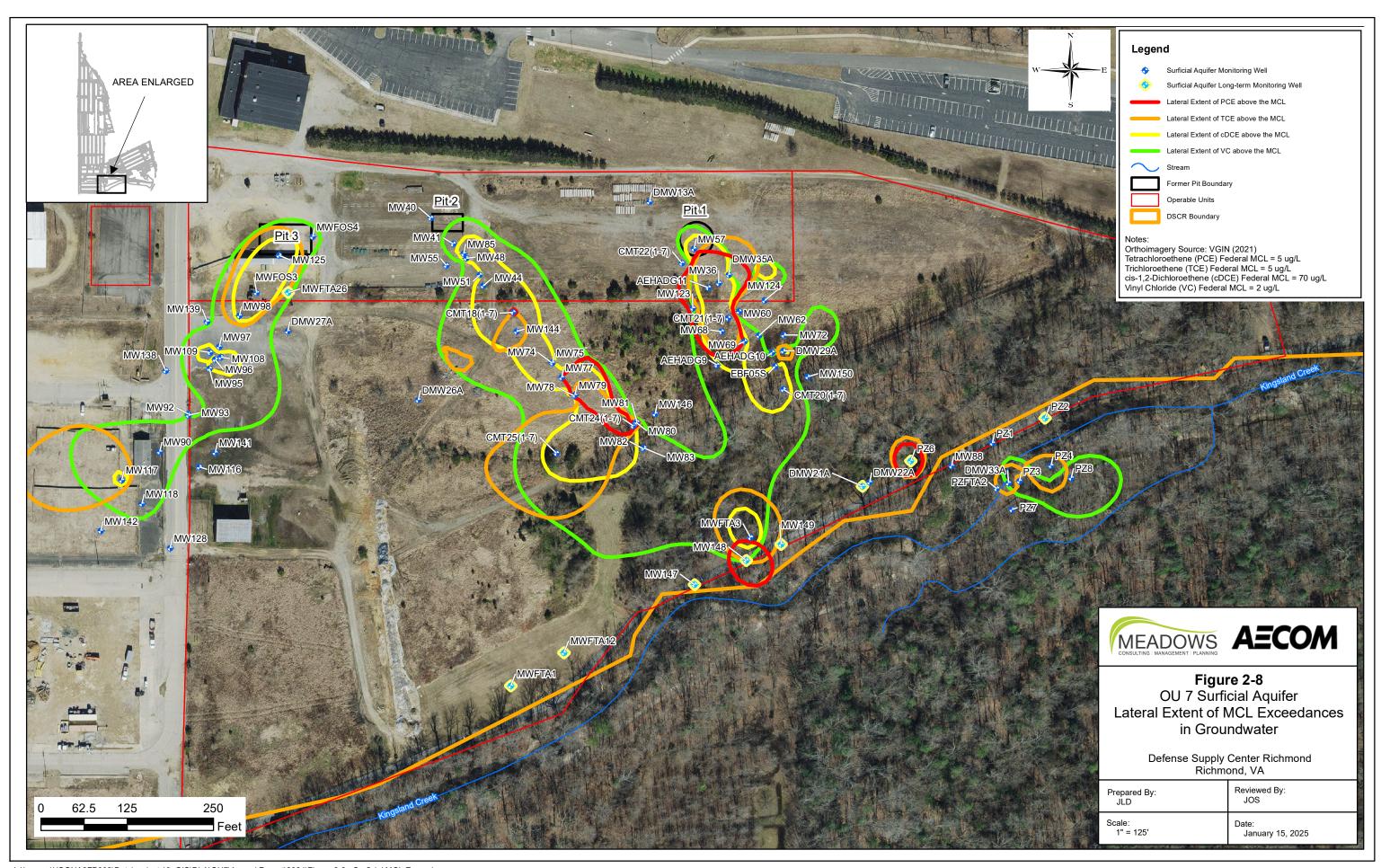
Figure 2-11 (page 2-33) depicts the cDCE plume areas associated with Pits 1, 2, and 3. At Pit 1, a cDCE plume extends 250 ft. downgradient of the former pit area with two limited areas within the plume having concentrations greater than 700 μ g/L. This plume area does not extend into the lower flood plain area at Kingsland Creek.

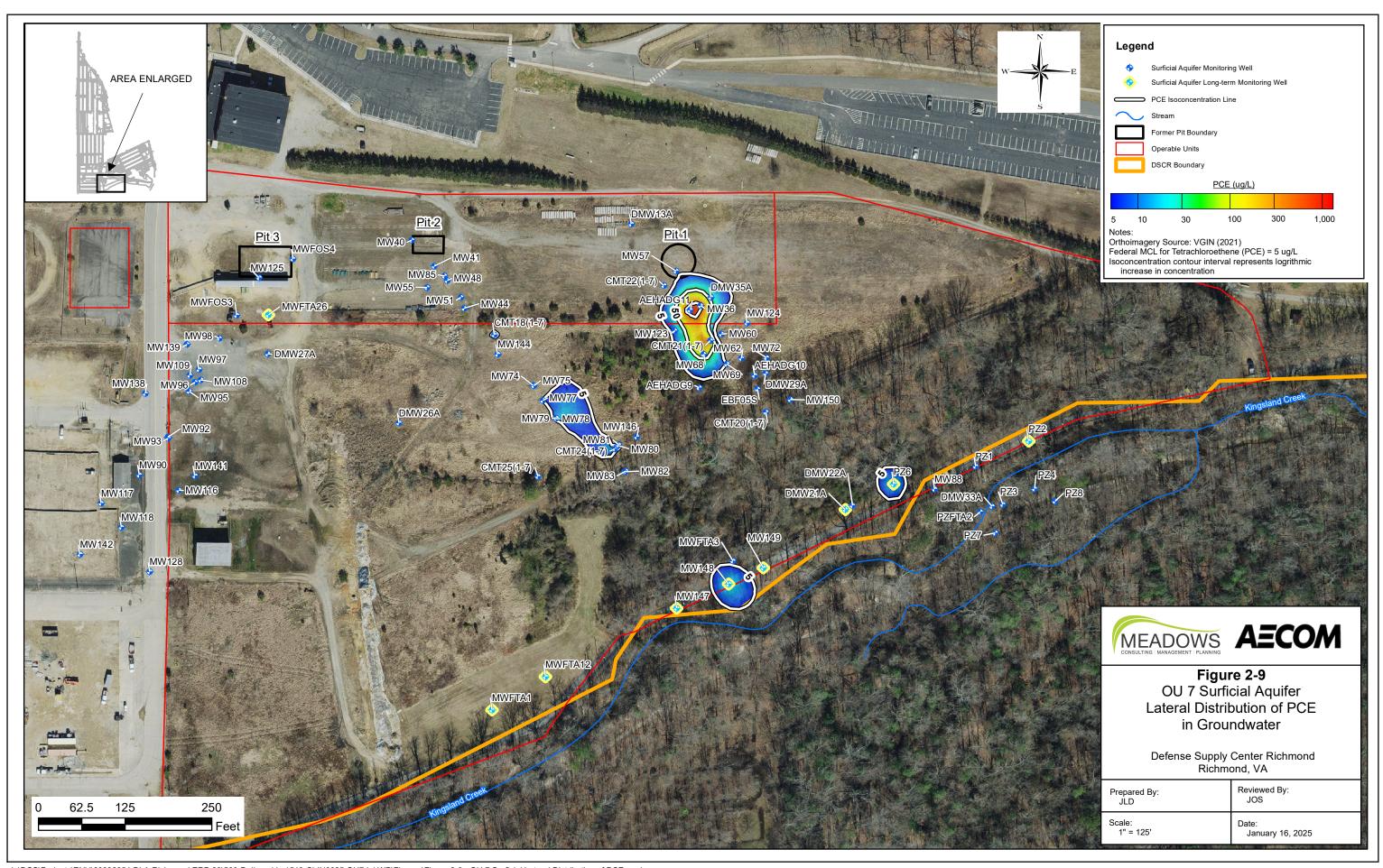
At Pit 2, a cDCE plume with an approximate length of 400 ft. is located in the upper area. This plume has limited areas with concentrations greater than 700 μ g/L. A isolated cDCE plume area is located around well MWFTA-3 in the lower area upgradient of the fence line.

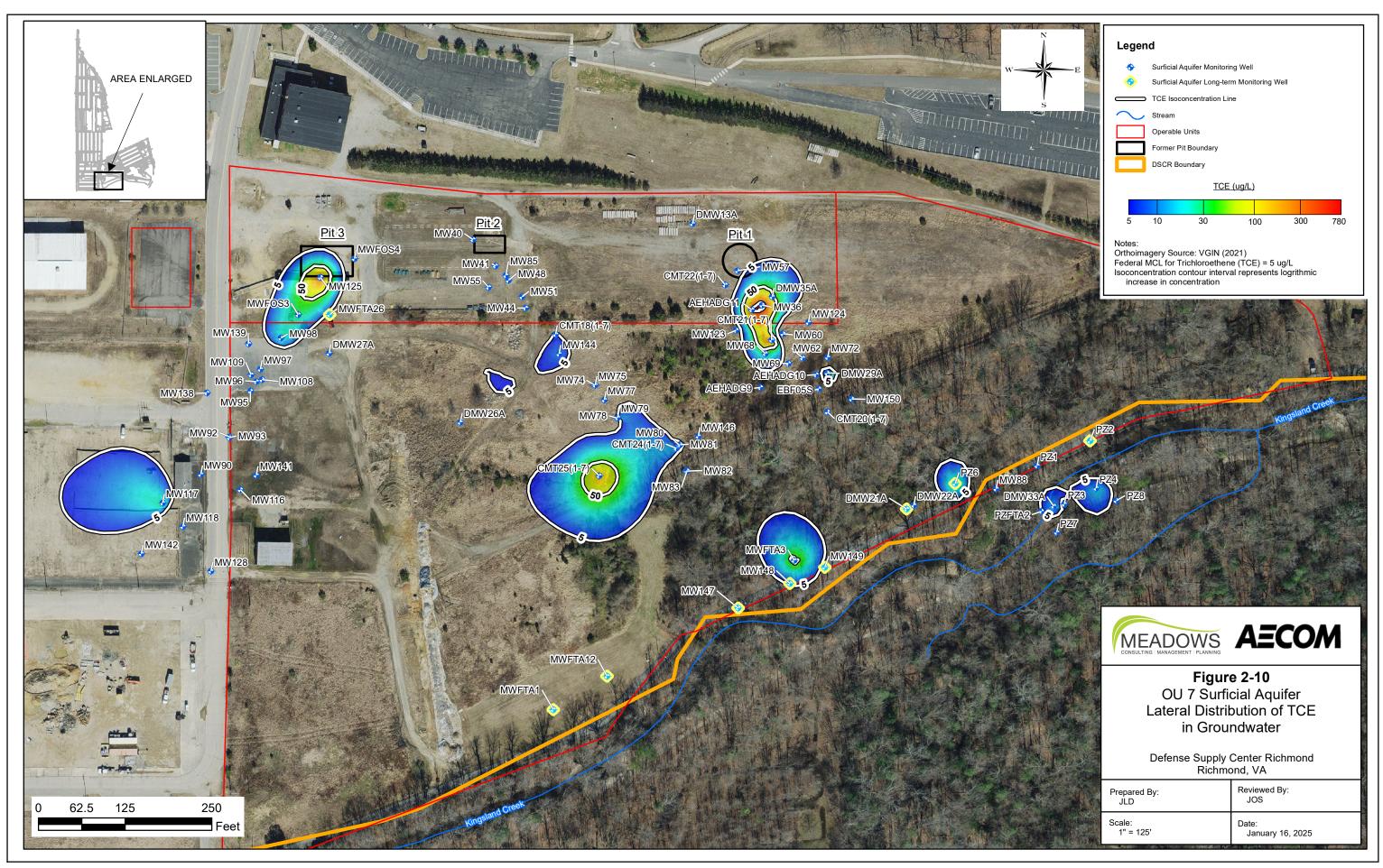
At Pit 3, a cDCE plume extends approximately 100 ft. downgradient of the former pit area with a limited area at MWFOS-3 having a concentration greater than 700 μ g/L. A second isolated cDCE plume area is located between this plume area and Building 72 with a single detection of cDCE at well MW-117 near Building 72.

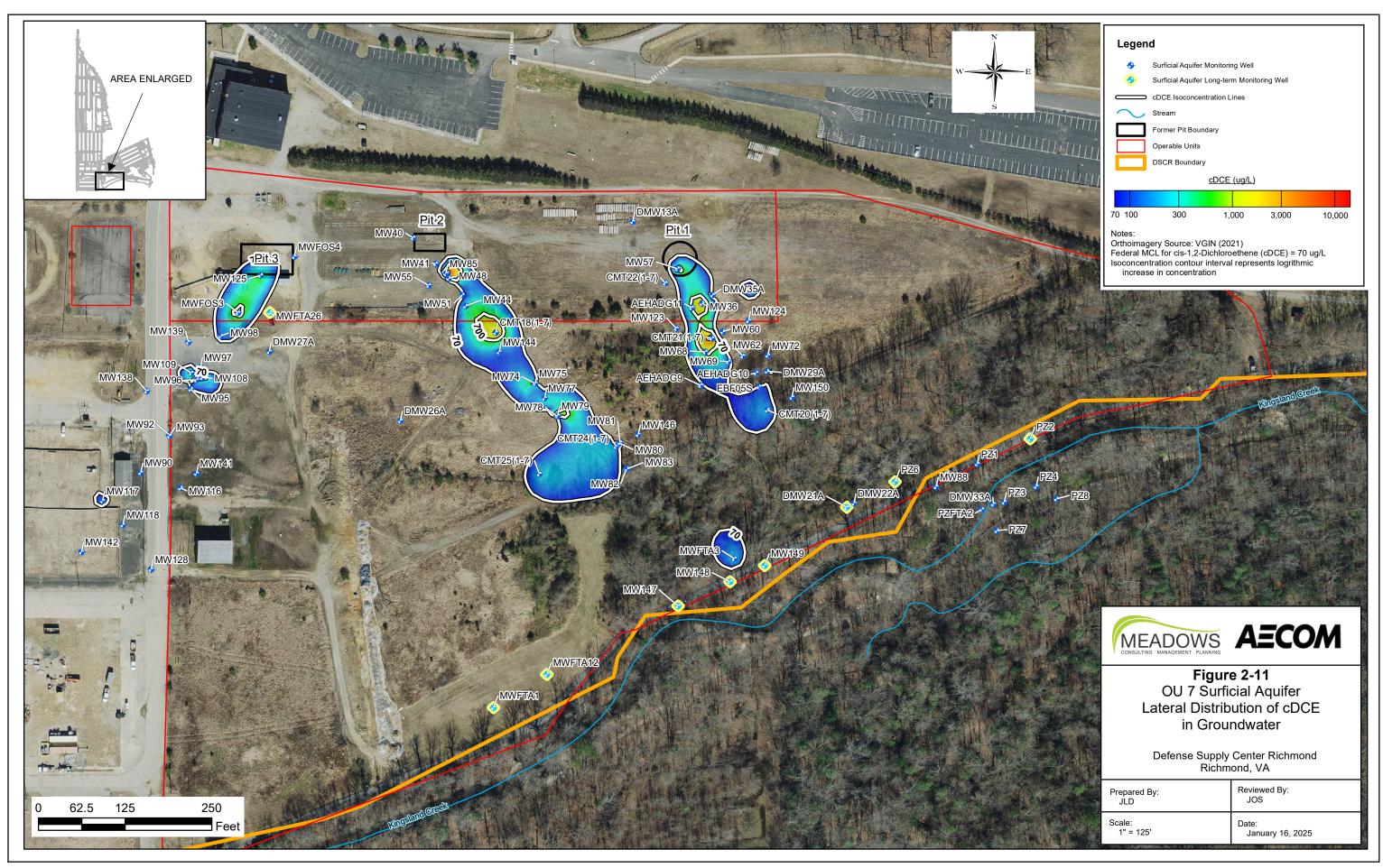
2.4.6.4 VC

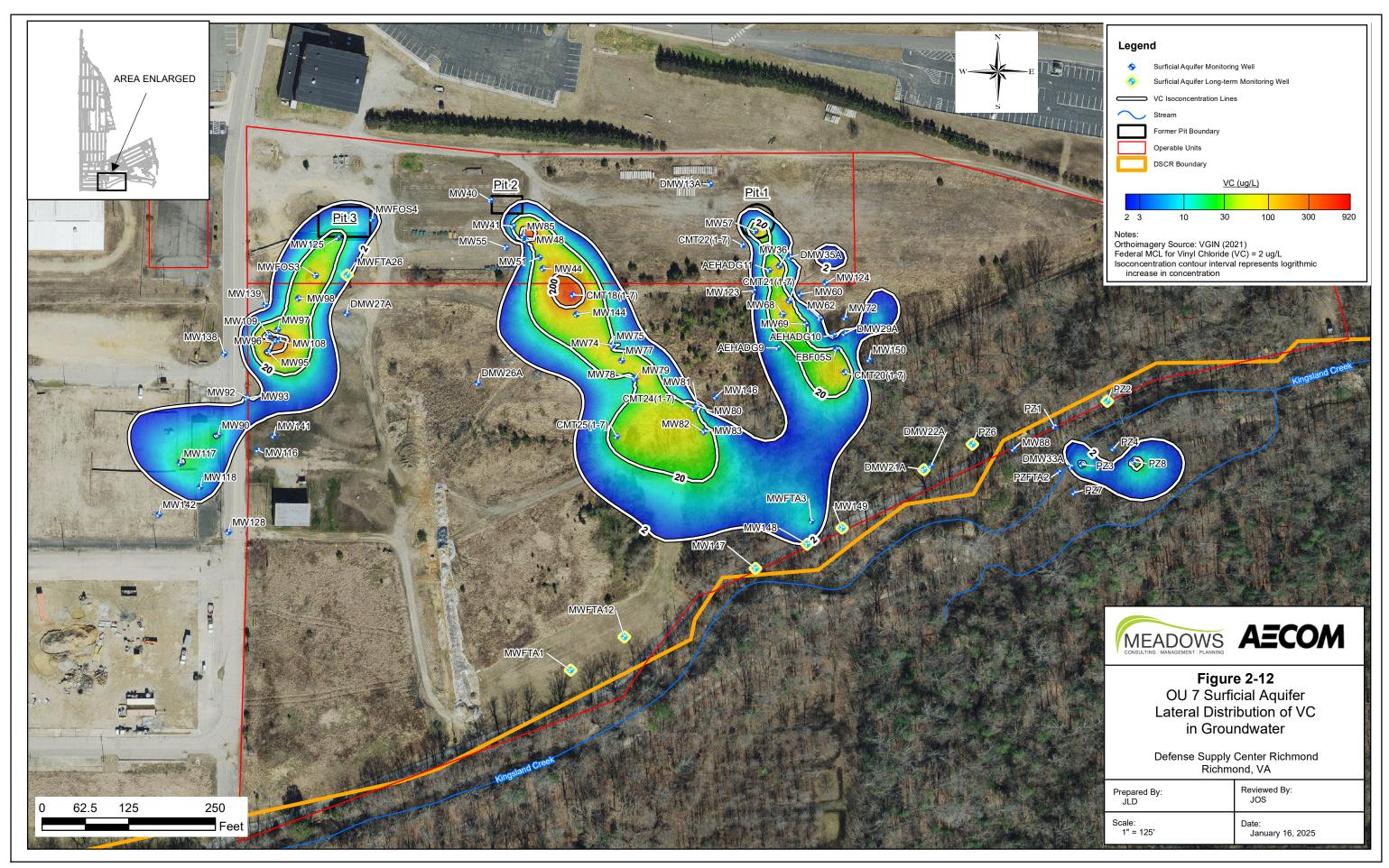
Figure 2-12 (page 2-34) depicts the VC plume areas associated with Pits 1, 2, and 3. Core plume areas generally have VC concentrations less than 20 μ g/L with isolated areas greater than 200 μ g/L. At Pit 1, a VC plume extends 300 ft. downgradient of the former pit area. A second isolated plume area with VC concentrations less than 20 μ g/L is located within the flood plain area of Kingsland Creek. At Pit 2, a VC plume extends more than 600 ft. from the former pit area to well MW-148 in the installation fence line area. In the upper area, the VC plume has two limited areas with concentrations greater than 200 μ g/L around wells MW-85 and CMT-18-5.











2.4.7 Distribution of Dissolved Gases in Surficial Aquifer (2024)

Section 2.4.7 describes the distribution of dissolved gases in the surficial aquifer for site-wide annual sampling performed in March 2024.

2.4.7.1 Carbon Dioxide

For the surficial aquifer at OU 7, background levels of carbon dioxide generally fall in the range of less than or equal to 15 mg/L outside of VOC plume areas and the influence of bioremediation processes. Carbon dioxide levels within the plume areas generally are 2 to 40 times higher than background with the highest concentrations in recent ISB injection areas.

2.4.7.2 **Methane**

For the 2024 annual sampling event, the number of samples having methane concentrations greater than 500 μ g/L is 24 of 31 samples for Pit 1, 25 of 27 samples for Pit 2, and 12 of 15 samples for Pit 3. Samples collected from the most recent bioremediation areas generally have methane concentrations greater than 1,000 μ g/L. Methane concentrations < 500 μ g/L generally occur at monitoring locations outside of VOC plumes and in the flood plain area of Kingsland Creek where oxidation is frequently a preferred reaction pathway.

2.4.7.3 Ethene

Figure 2-13 (page 2-36) shows contiguous plume areas of dissolved ethene at concentrations greater than or equal to 10 μ g/L, which generally occur in the core VOC plume areas in Pits 1, 2, and 3. Maximum ethene concentrations in these pit areas are 1,200 μ g/L for Pit 1 (MW-68), 700 μ g/L for Pit 2 (CMT-18-5), and 200 μ g/L for Pit 3 (MW-108).

2.4.7.4 Ethane

Figure 2-14 (page 2-37) shows contiguous plume areas of dissolved ethane at concentrations greater than or equal to 10 μ g/L, which generally occur in the core VOC plume areas in Pits 1, 2, and 3. Maximum ethane concentrations in these pit areas are 130 μ g/L for Pit 1 (MW-69), 210 μ g/L for Pit 2 (MW-48), and 86 μ g/L for Pit 3 (MW-108).

2.4.8 Distribution of Total Organic Carbon

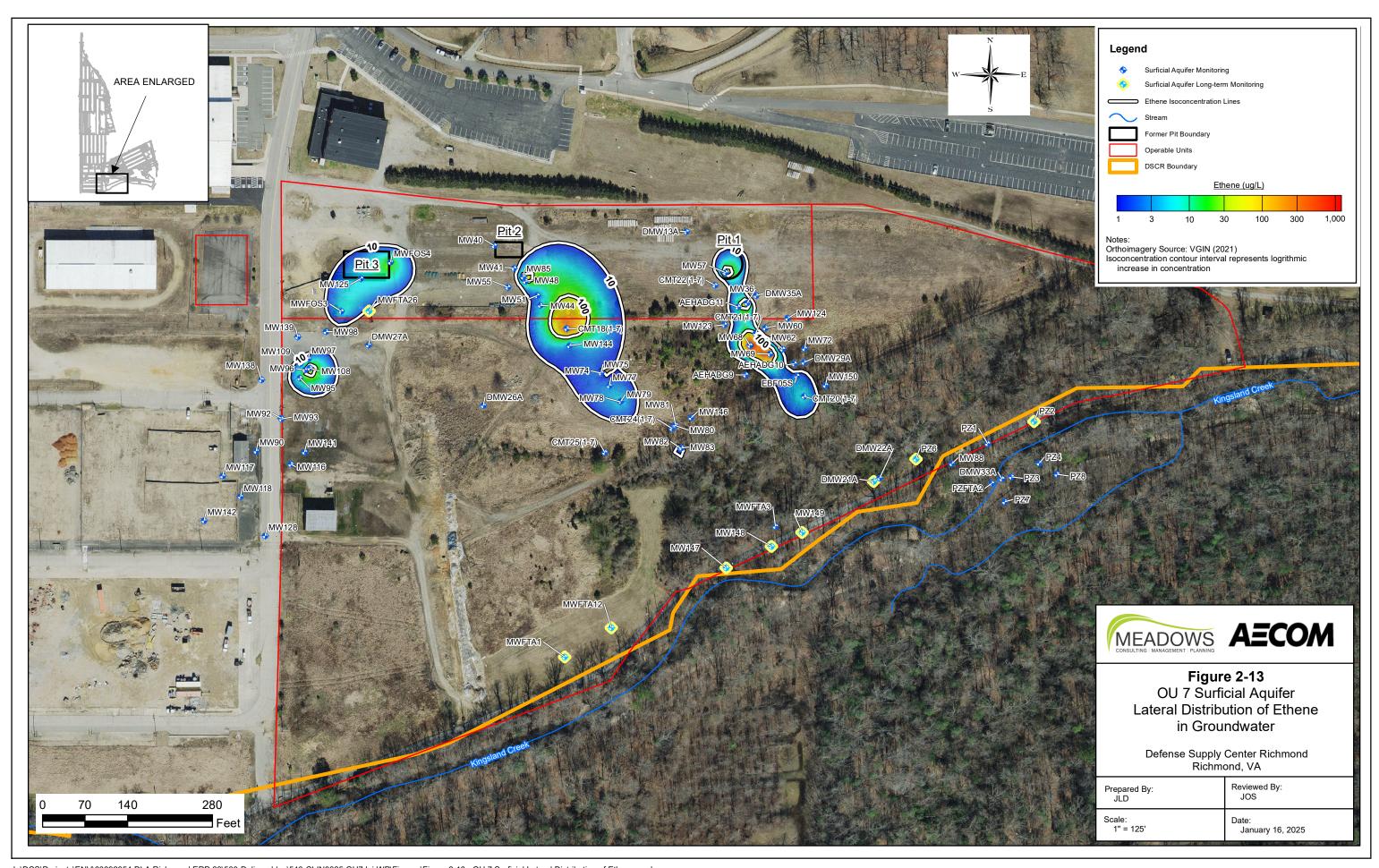
Figure 2-15 (page 2-38) shows areas of TOC in groundwater within the surficial aquifer at concentrations greater than or equal to 20 mg/L. At Pit 1, the area of TOC greater than or equal to 20 mg/L in groundwater extends from the former pit area approximately 300 ft. to the most downgradient ISB treatment area (TA-06). Ten of 31 wells monitored in the Pit 1 area have TOC concentrations greater than or equal to 20 mg/L with these wells having one or more COCs at concentrations greater than or equal to 10 times higher than cleanup levels (MCLs). At Pit 2, the primary area of TOC greater than or equal to 20 mg/L n groundwater extends from the most upgradient ISB treatment area (TA-07) approximately 200 ft. to the mid-plume ISB treatment area (TA-08). Single wells downgradient of this area have TOC concentrations greater than or equal to 20 mg/L including wells MW-75 (mid-plume), CMT-24-2 (downgradient), and well MW-148 (fence line area). Nine of 24 locations sampled including a well cluster with four samples had TOC concentrations greater than or equal to 20 mg/L. Pit 3 had two areas of groundwater with TOC concentrations greater than or equal to 20 mg/L corresponding to treatment areas at the former pit area (TA-10) and in the mid plume area (TA-11). Three of 17 samples had TOC concentrations greater than or equal to 20 mg/L.

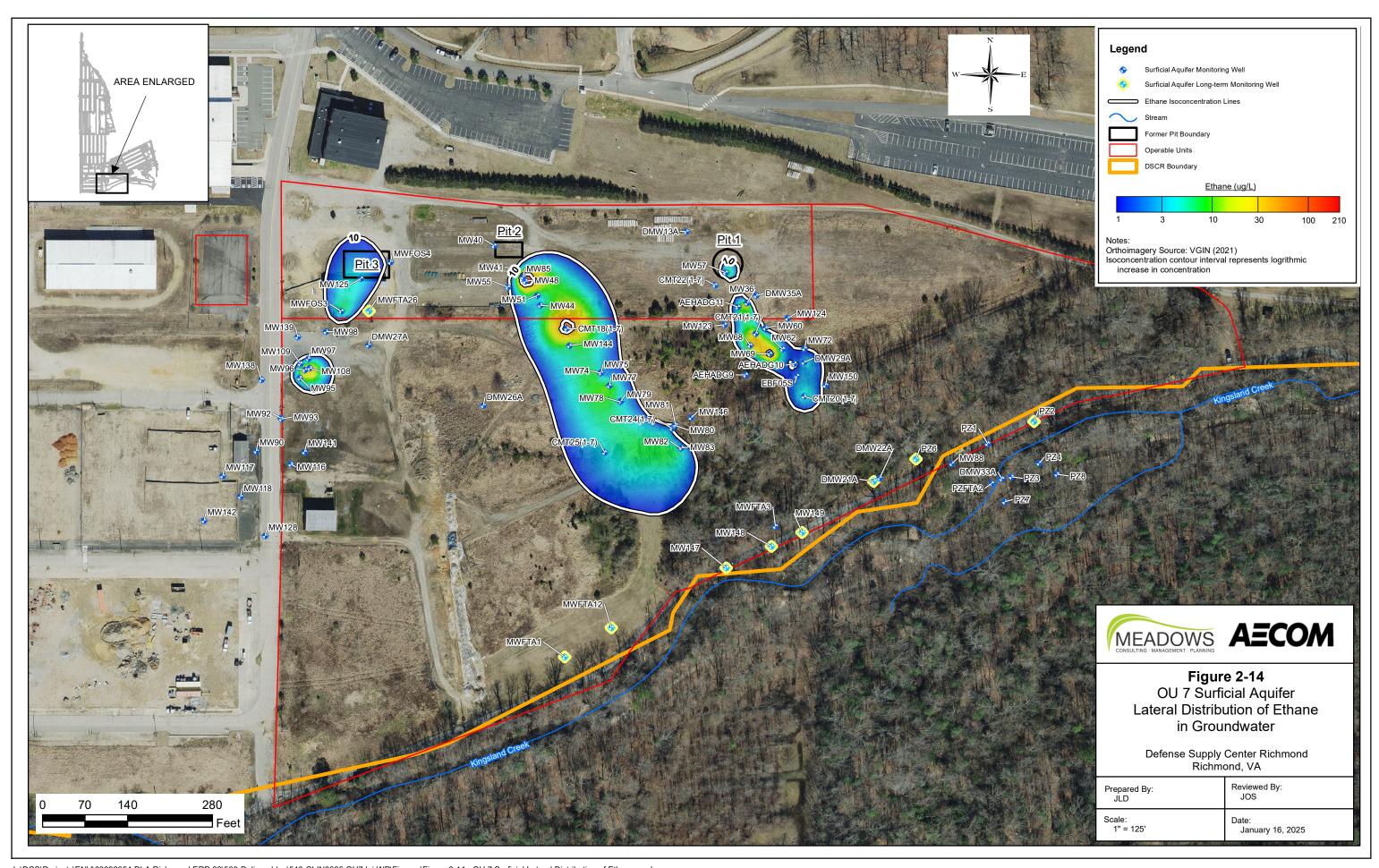
2.4.9 Bulk Plume Trends

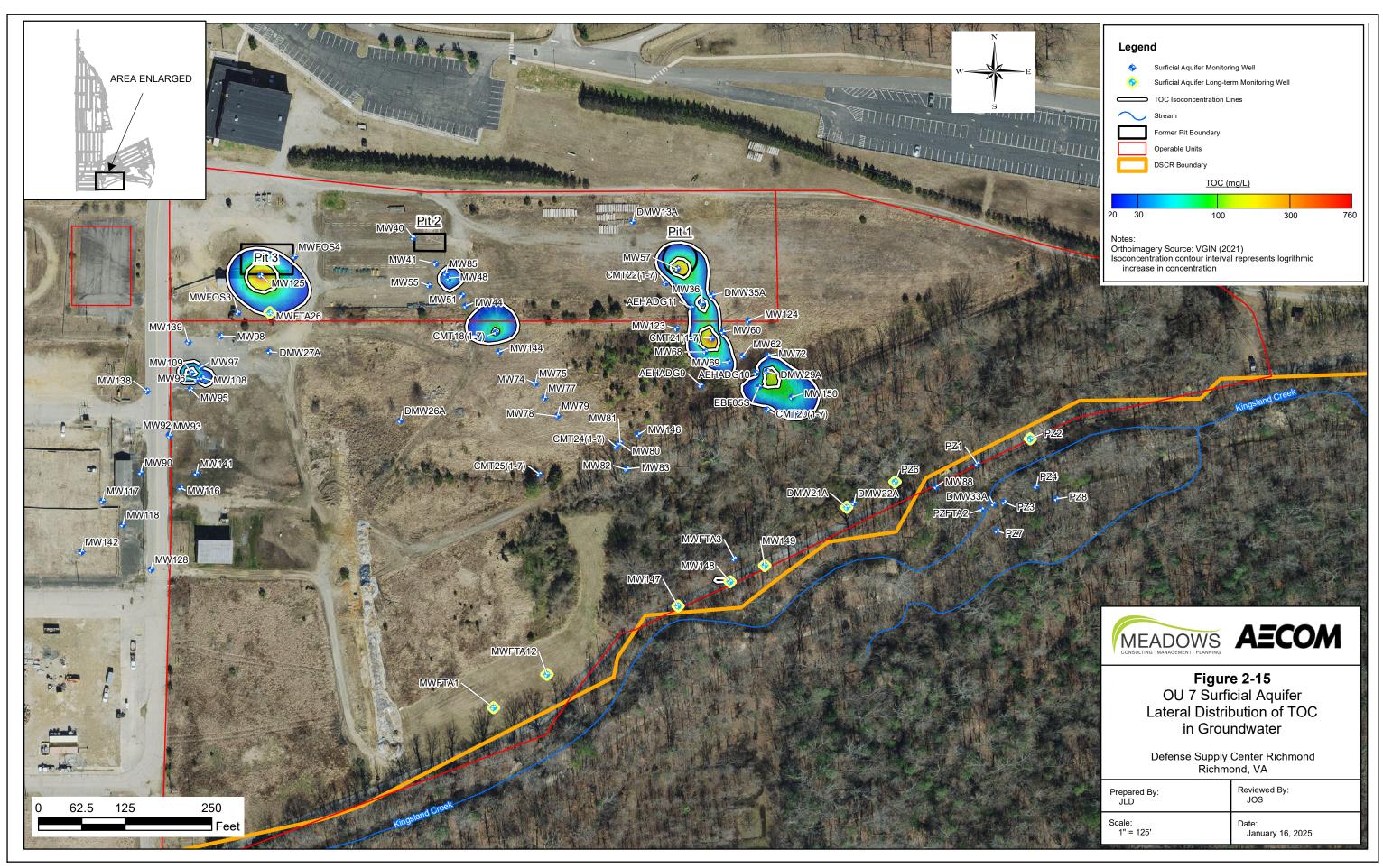
Section 2.4.9 summarizes bulk plume trends for the surficial aquifer at OU 7. Figures 2-16, 2-17, 2-18, and 2-19 (pages 2-39 through 2-42) have tiled layouts for PCE, TCE, cDCE, and VC plumes characterized for annual sampling events performed in 2021, 2022, 2023, and 2024, respectively. Exhibit 2-4 (page 2-43) has tiled stem plots for plume area data (acres) from 2009, 2012, 2015, and 2018 through 2024. Plots in Exhibit 2-4 use singular spectrum analysis (SSA¹) to find and plot long-term trends in the time series data. SAA calculates long-term trend as tabular and plotted data allowing for further

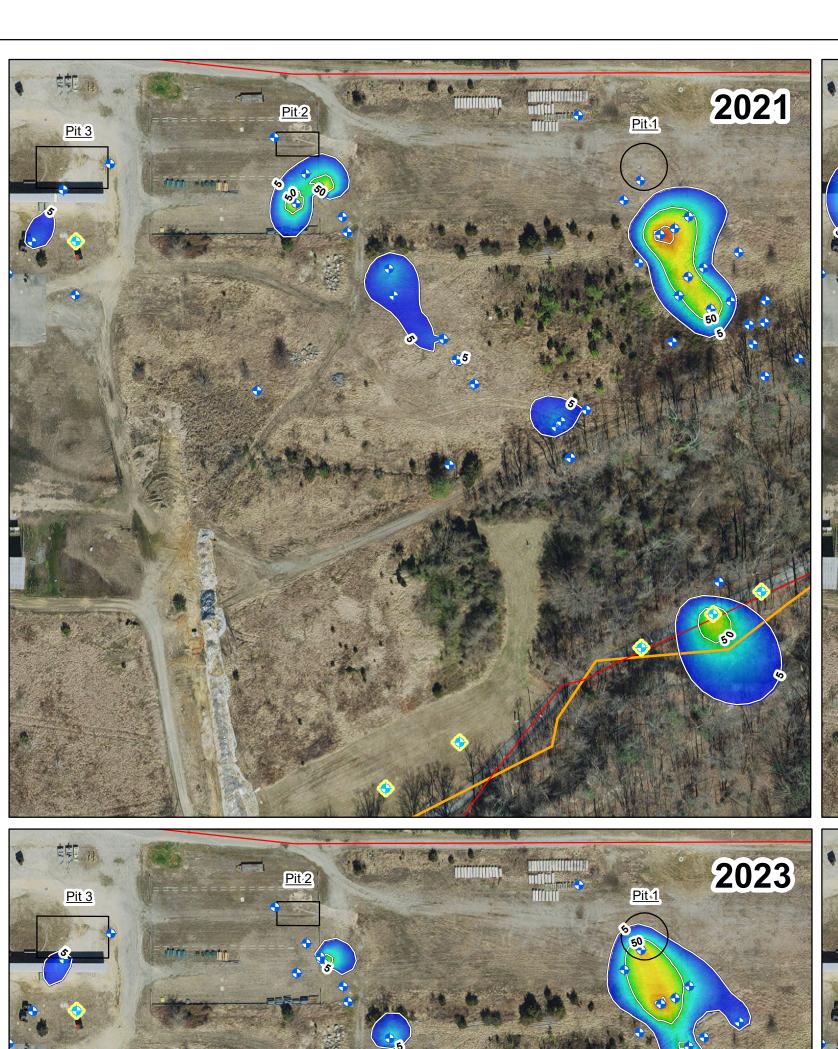
¹ Singular spectrum analysis performed using MATLAB 2024a

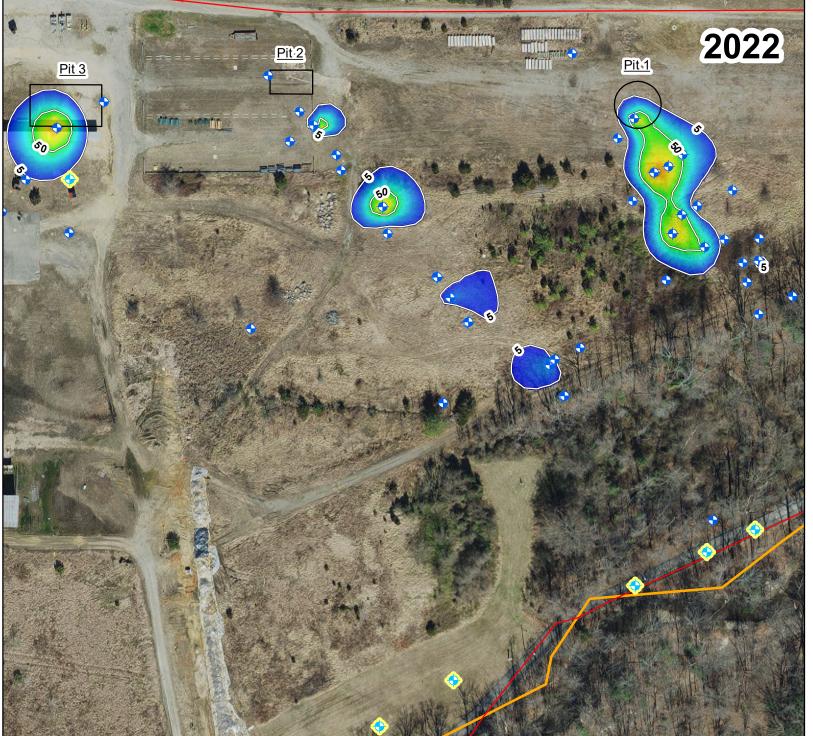
analysis of rates of change and degree of trend. Each plot has a percent change representing the change from the maximum plume area (pre-2024) to 2024.

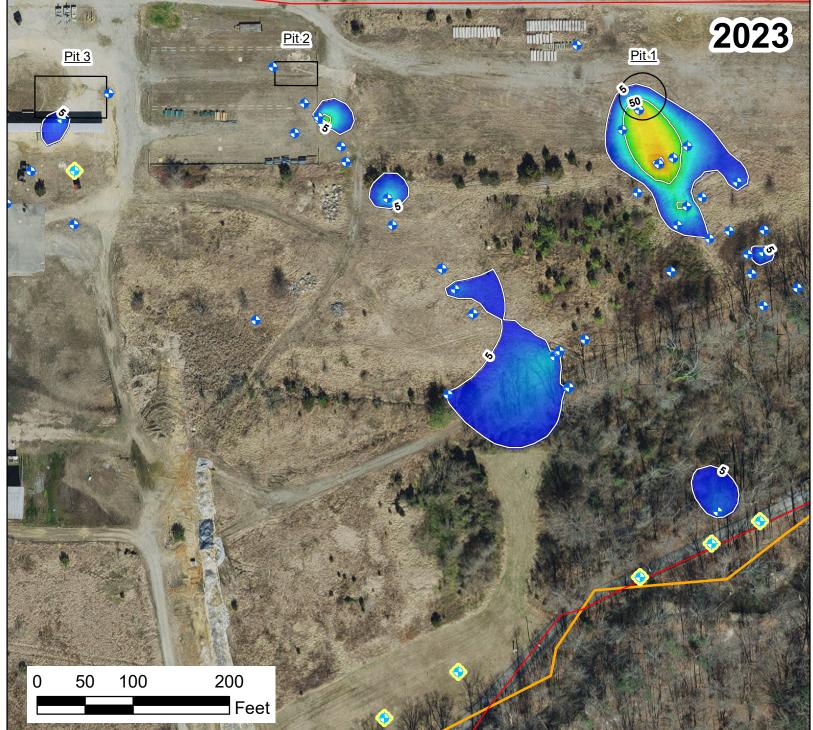






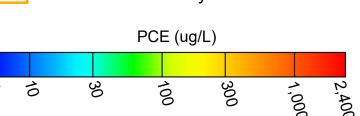








- Surficial Aquifer Monitoring Well
- Surficial Aquifer Long-term Monitoring Well
- PCE Isoconcentration Line
- OU7 Boundary
- Installation Boundary



Notes:

Orthoimagery Source: VGIN (2021)
Federal MCL for Tetrachloroethene (PCE) = 5 ug/L
Isoconcentration contour interval represents
logrithmic increase in concentration

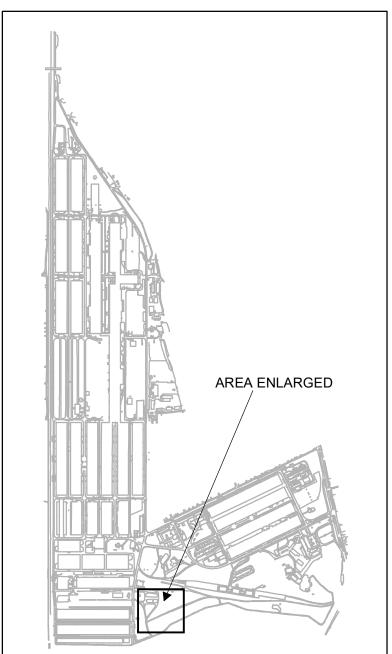
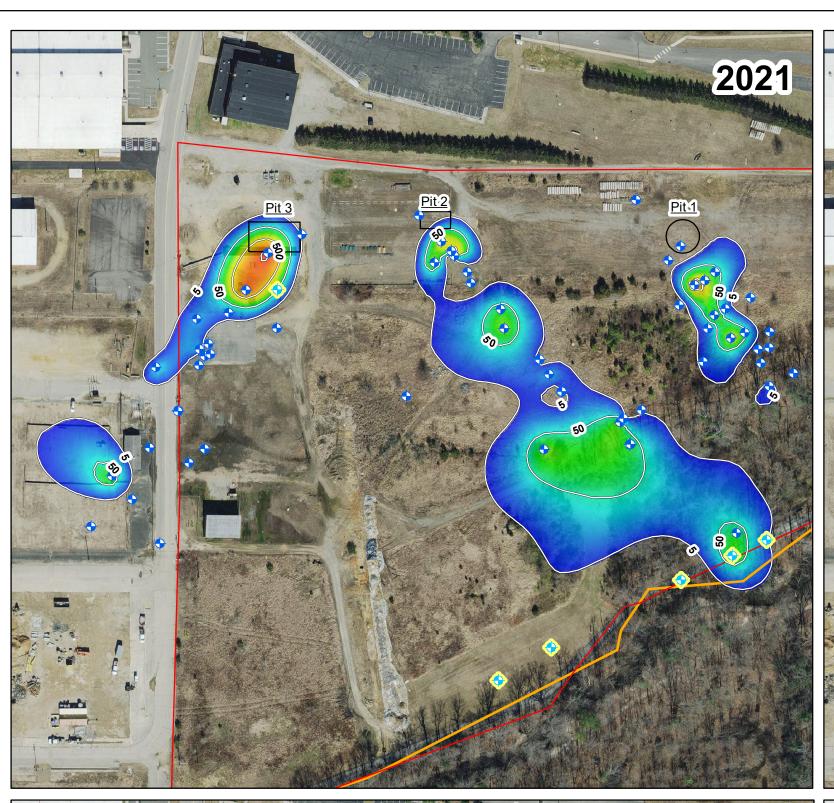


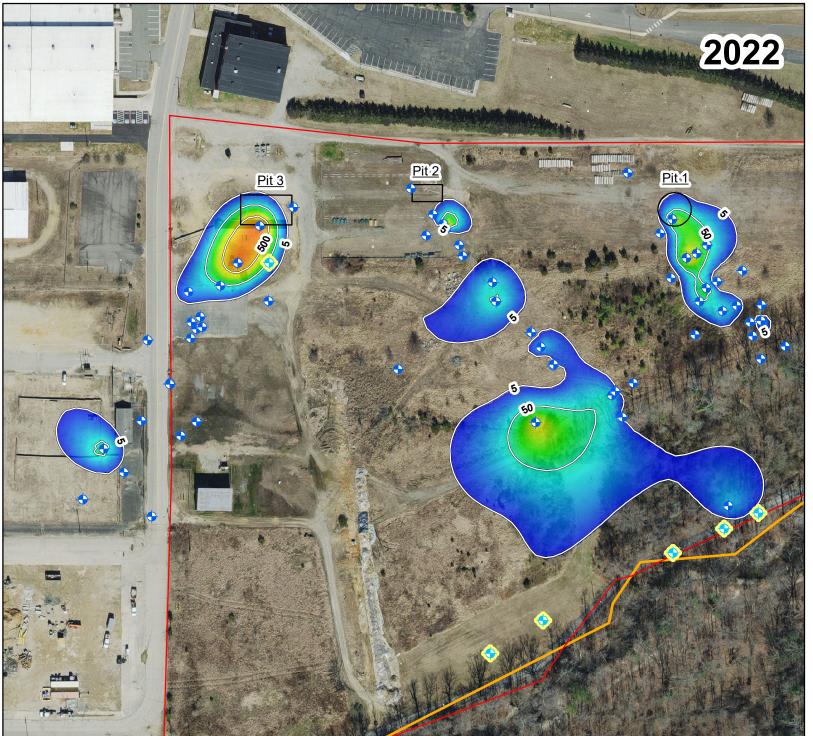


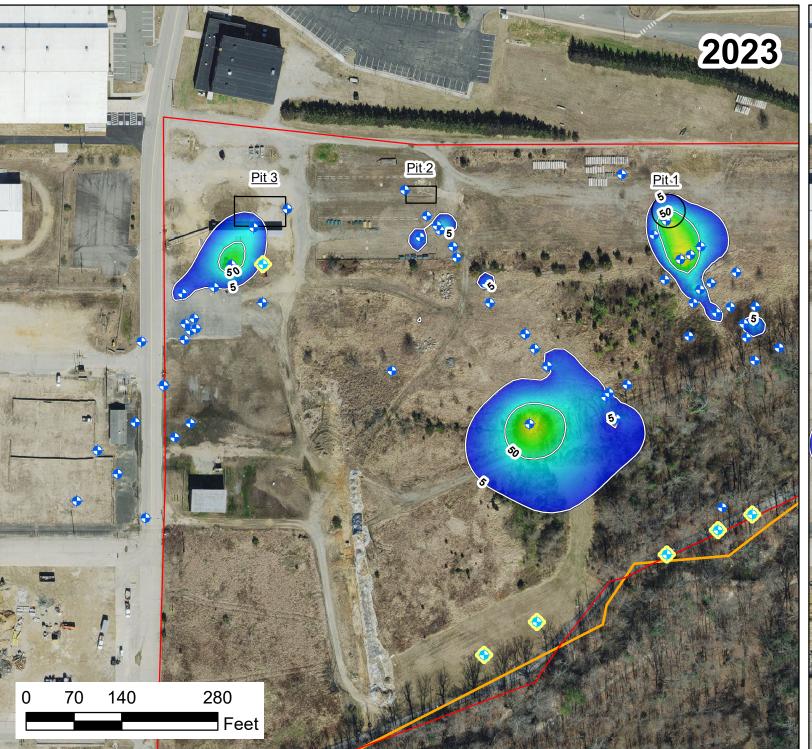
Figure 2-16

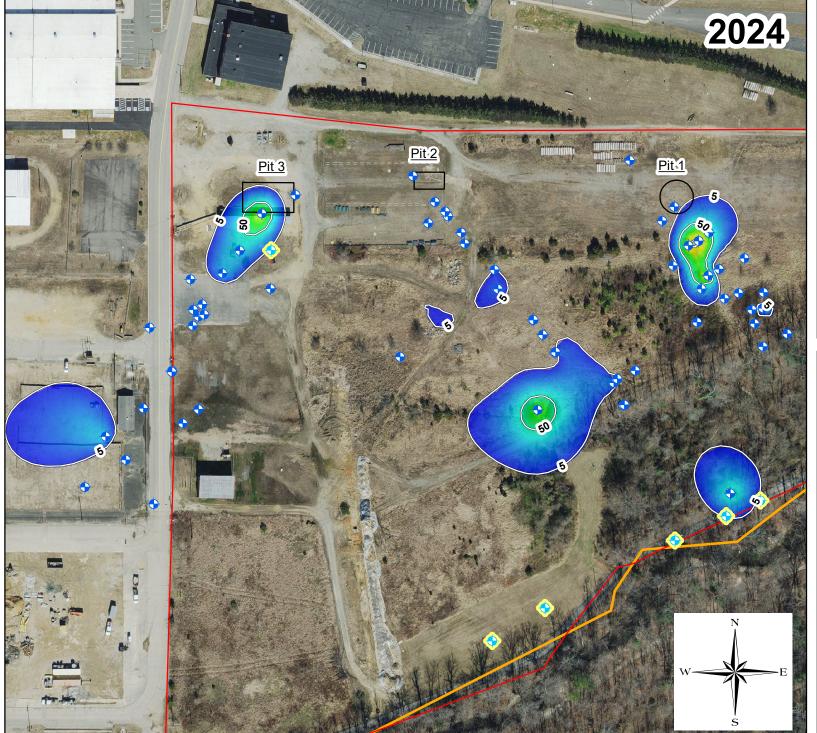
Changes in Lateral Distribution of PCE in Surficial Aquifer (2021-2024)

Prepared By: JLD	Reviewed By: JOS
Scale: 1 " = 100 '	Date: January 16, 2025

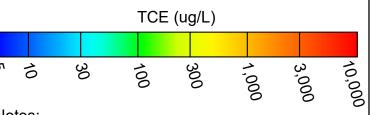




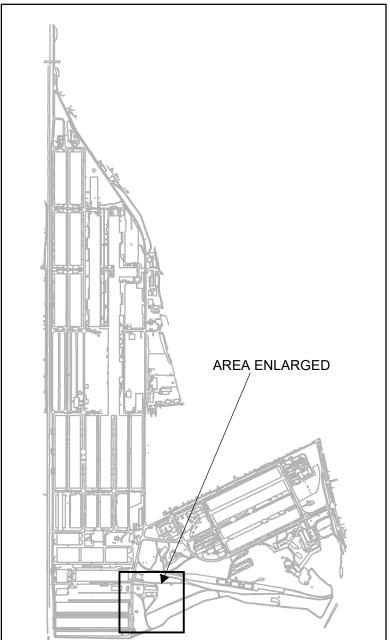




- Surficial Aquifer Monitoring Well
- Surficial Aquifer Long-term Monitoring Well
- ▼ TCE Isoconcentration Line
 - OU7 Boundary
- Installation Boundary



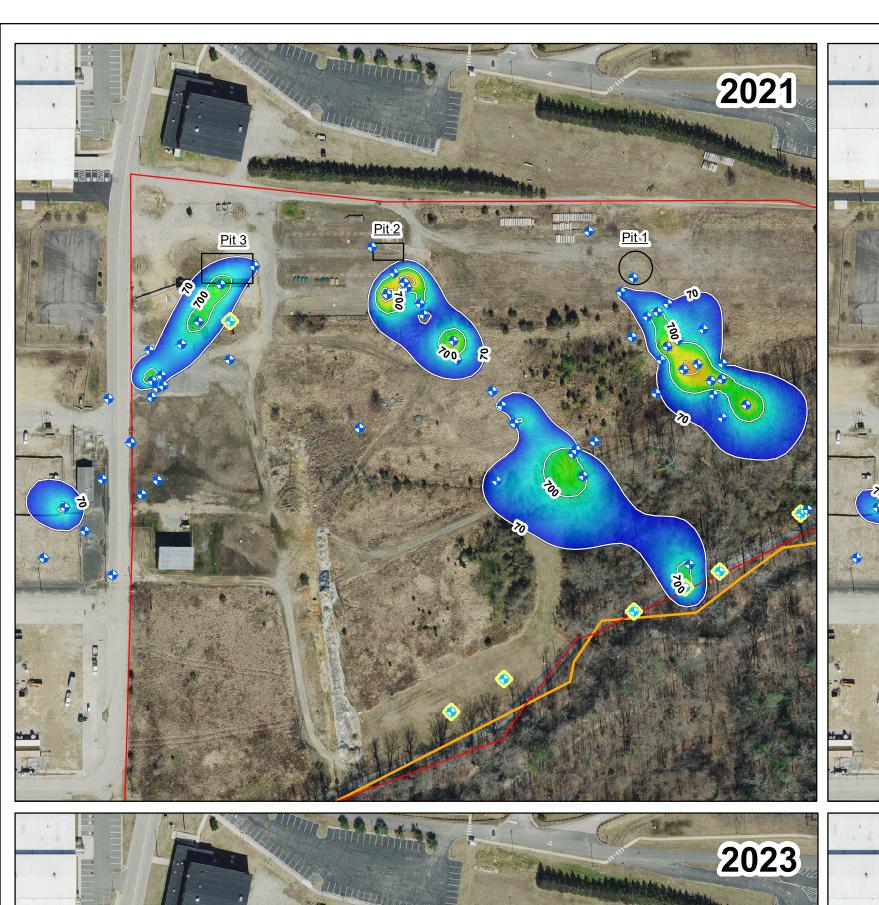
Notes:
Orthoimagery Source: VGIN (2021)
Federal MCL for Trichloroethene (TCE) = 5 ug/L
Isoconcentration contour interval represents
logrithmic increase in concentration

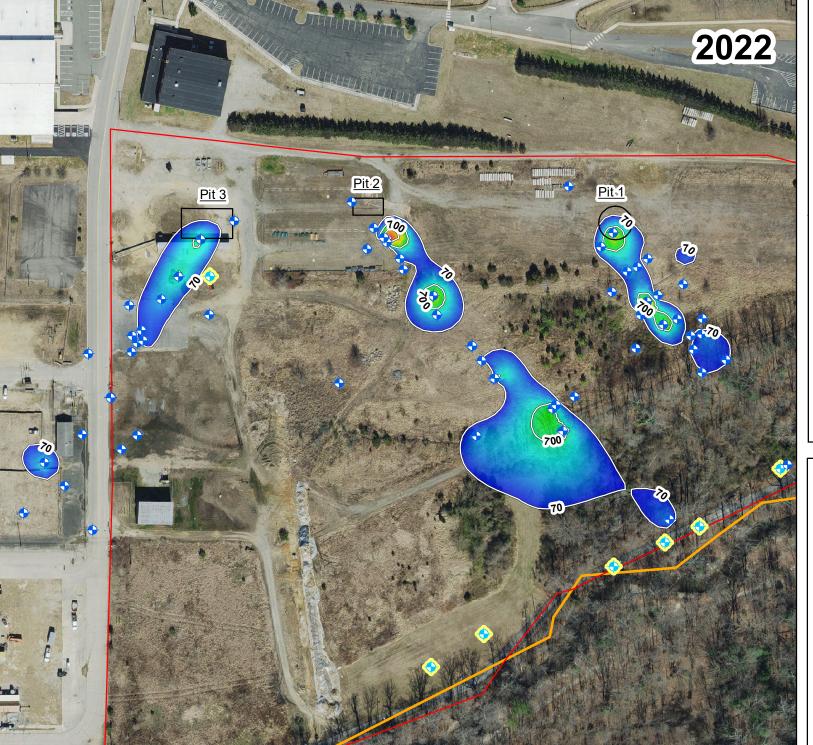


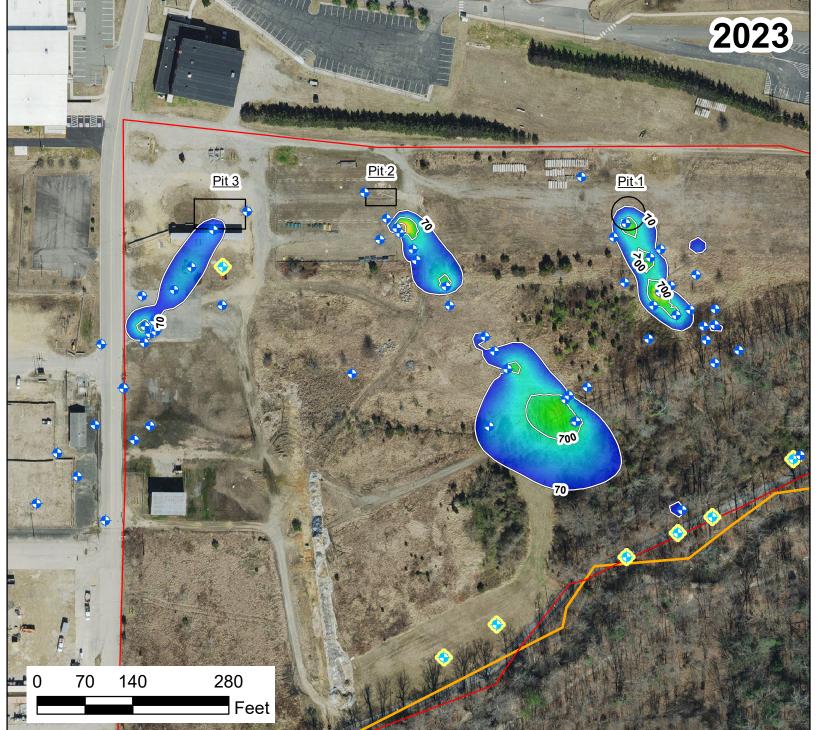
MEADOWS AECOM

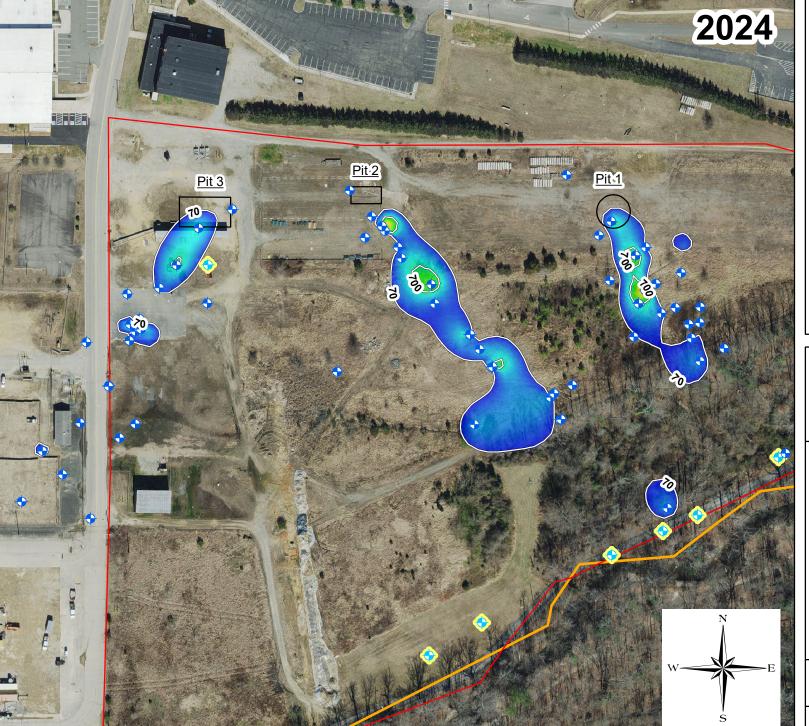
Figure 2-17
Changes in Lateral Distribution of TCE in Surficial Aquifer (2021-2024)

Prepared By JLD	/ :	Reviewed By: JOS
Scale: 1 " = 140		Date: January 16, 2025



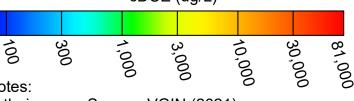






- Surficial Aquifer Monitoring Well
- Surficial Aquifer Long-term Monitoring Well
- cDCE Isoconcentration Line
- **OU7** Boundary
- Installation Boundary

cDCE (ug/L)



Notes:
Orthoimagery Source: VGIN (2021)
Federal MCL for
cis-1,2-Dichloroethene (cDCE) = 70 ug/L
Isoconcentration contour interval represents
logrithmic increase in concentration

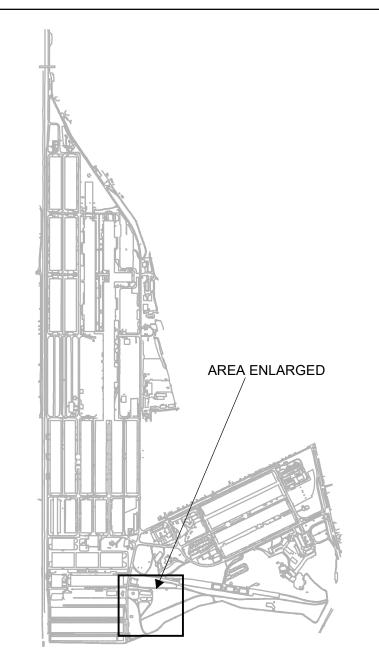
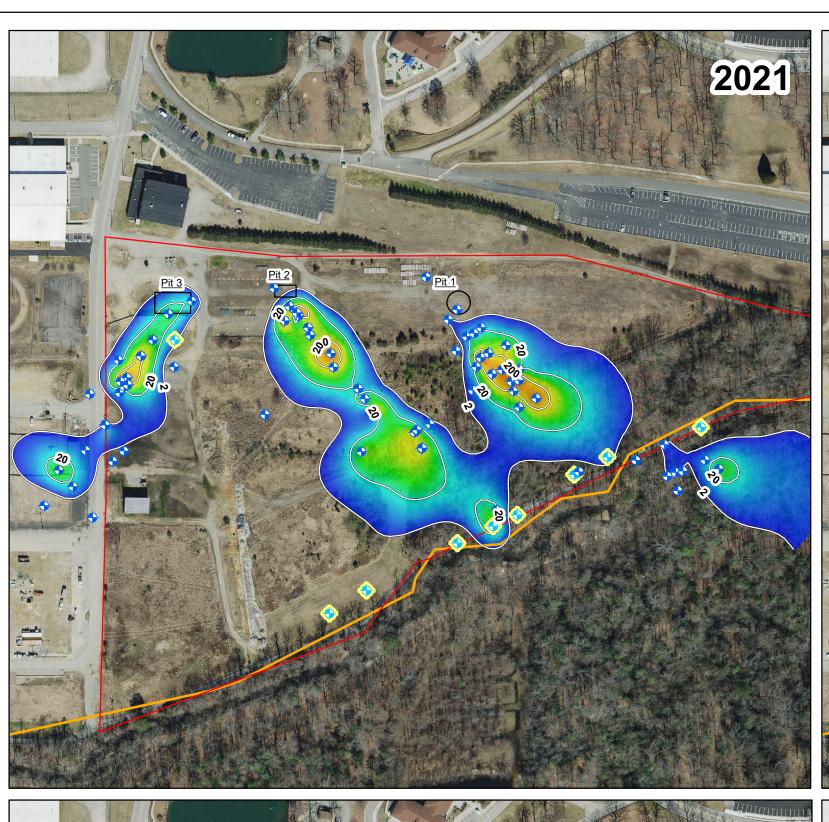


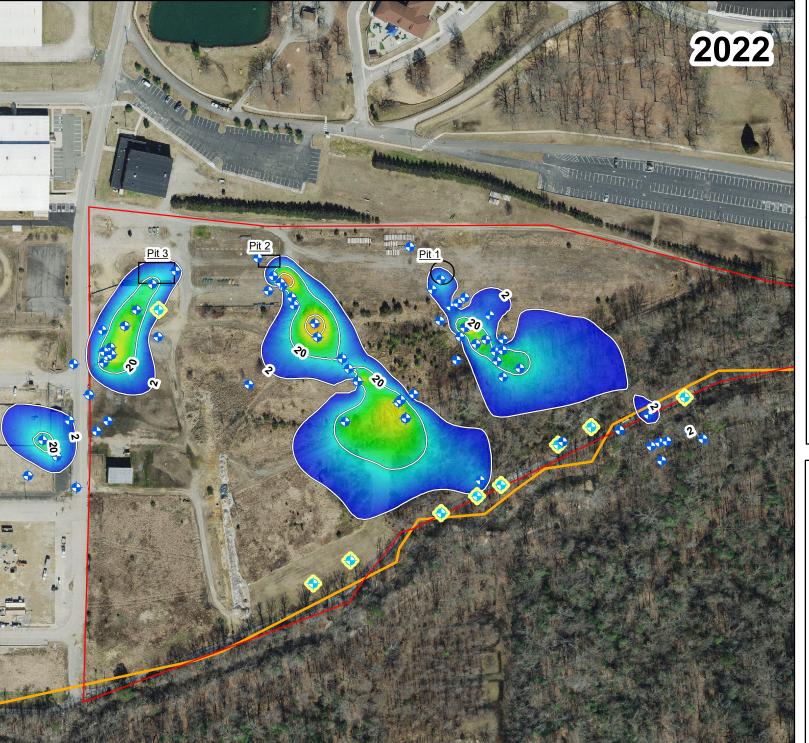


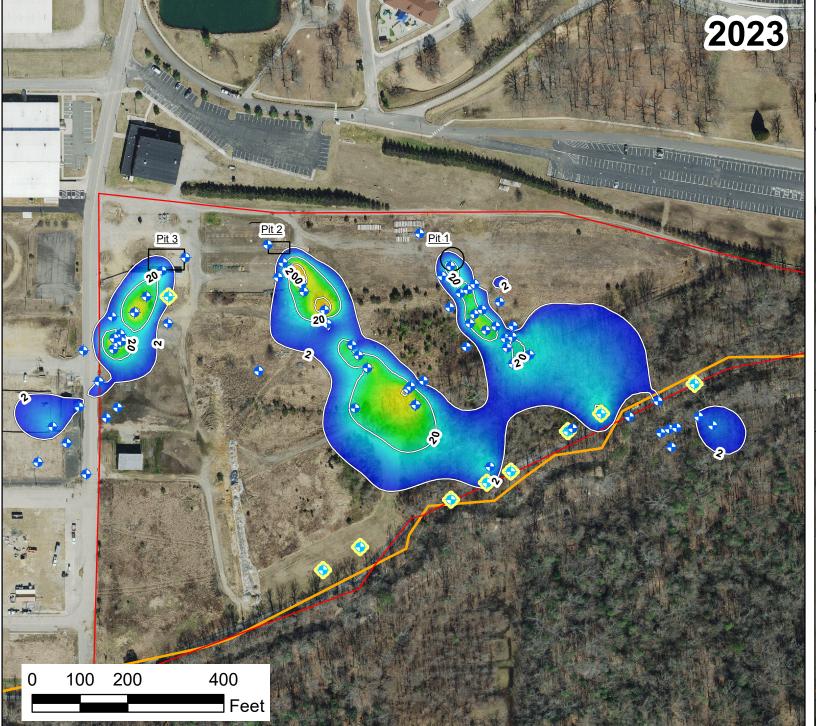
Figure 2-18

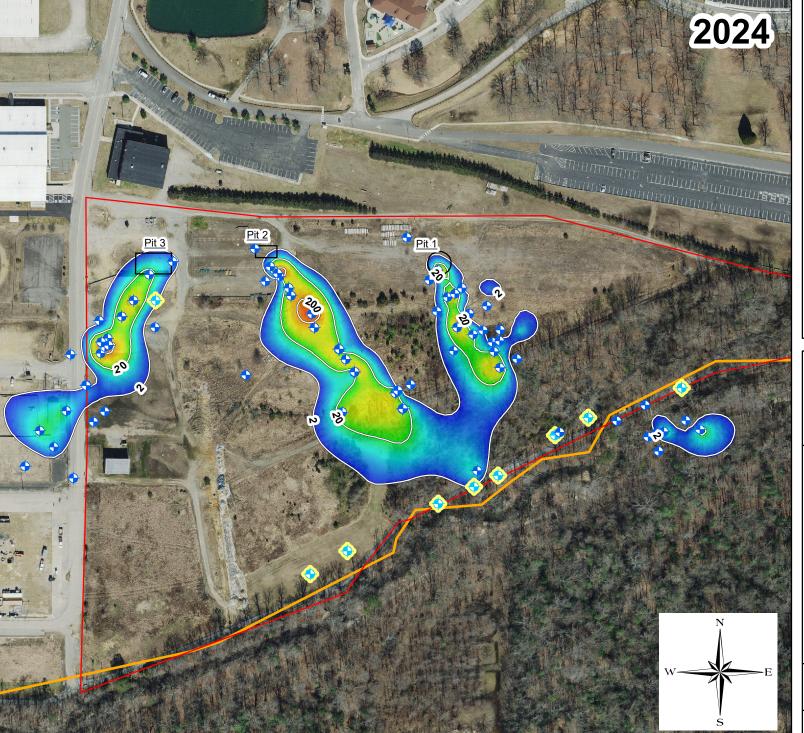
Changes in Lateral Distribution of cDCE in Surficial Aquifer (2021-2024)

Prepared By: JLD	Reviewed By: JOS
Scale: 1 " = 140 '	Date: January 16, 2025









- Surficial Aquifer Monitoring Well
- Surficial Aquifer Long-term Monitoring Well
- OU7 Boundary
- Installation Boundary

VC (ug/L)



Notes:
Orthoimagery Source: VGIN (2021)
Federal MCL for Vinyl Chloride (VC) = 2 ug/L
Isoconcentration contour interval represents
logrithmic increase in concentration

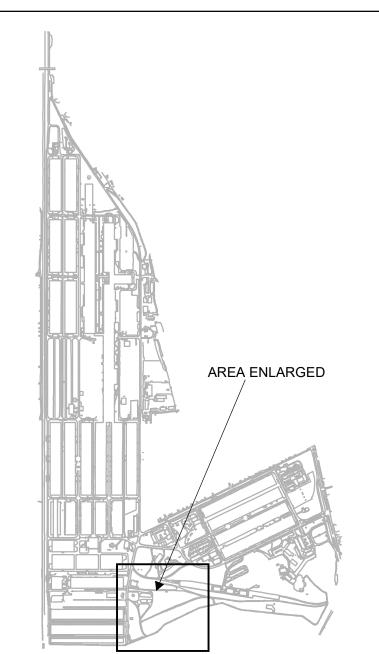


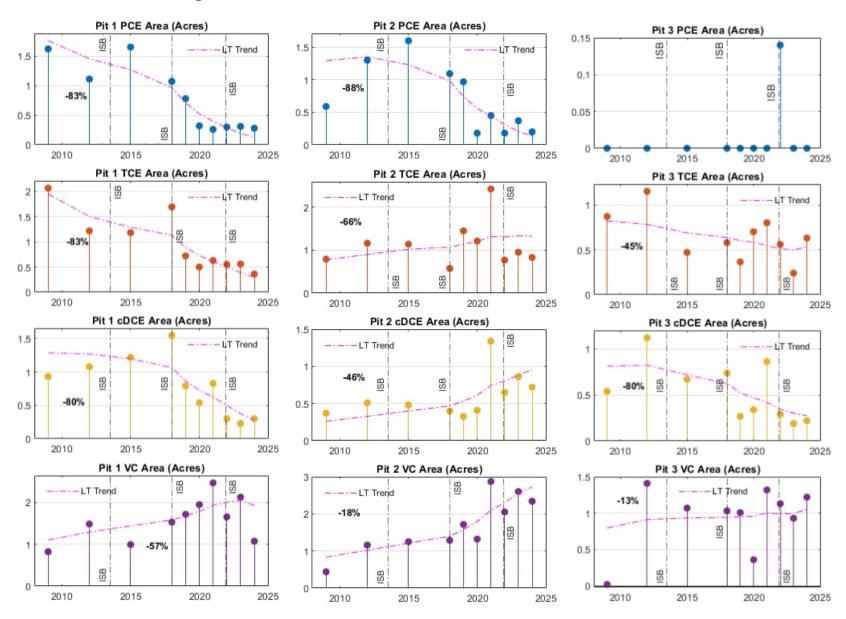


Figure 2-19
Changes in Lateral Distribution of VC

in Surficial Aquifer (2021-2024)

Prepared By: JLD	Reviewed By: JOS
Scale: 1 " = 200 '	Date: January 17, 2025

Exhibit 2-4 Plume Area Changes



Tiled Plume Visualizations (2021-2024)

Figure 2-16 shows a contiguous PCE plume in the Pit 1 area with plume areas ranging from 0.26 acres in 2021 to 0.31 acres in 2023. The PCE plume is disaggregated into separate areas at Pit 2 with a decrease in area from 2.42 acres in 2021 to 0.83 acres in 2024. Figure 2-17 shows reductions and disaggregation of TCE plume areas over time within the three pit areas.

At Pit 1, the TCE plume area decreased from 0.63 acres in 2021 to 0.36 acres in 2024. At Pit 2, the TCE plume area decreased from 2.42 acres in 2021 to 0.83 acres in 2024. Pit 3 had a TCE plume reduction from 0.80 acres in 2021 to 0.63 acres in 2024.

Figure 2-18 shows degradation of cDCE plume areas over time. At Pit 1, the cDCE plume area decreased from 0.83 acres in 2021 to 0.30 acres in 2024. At Pit 2, the cDCE plume area decreased from 1.34 acres in 2021 to 0.72 acres in 2024. At Pit 3, the cDCE plume area decreased from 0.86 acres in 2021 to 0.22 acres in 2024.

VC plumes in the Pit 1, 2, and 3 areas had overall reductions in area over time. At Pit 1, the VC plume area decreased from 2.46 acres in 2021 to 1.07 acres in 2024. At Pit 2, the VC plume area decreased from 2.87 acres in 2021 to 2.34 acres in 2024. At Pit 3, the VC plume area decreased from 1.32 acres in 2021 to 1.22 acres in 2024.

Tiled Stem Plots and Trends

Tiled plots for Pit 1 in Exhibit 2-4 show decreasing trends for PCE, TCE, and cDCE plume areas with percent changes from plume area local maxima of -83 percent (%) for PCE and cDCE, and -75 % for TCE. The slope of the trend line increased for PCE, TCE, and cDCE after completion of ISB actions in late 2017 and early 2018 indicating an increasing rate of plume degradation. VC shows an increasing trend with a local maxima area observed in 2022 with an apparent change to a decreasing trend corresponding to a 57% decrease in plume area from 2022 to 2024.

Tiled plots for Pit 2 show a decreasing trend for the PCE plume area. The slope of the trend line increased for PCE after completion of ISB actions in late 2017 and early 2018 indicating an increasing rate of plume degradation. TCE trend analysis for Pit 2 showed a leveling trend and change to a decreasing trend in 2023 after ISB injections. Plume areas for TCE, cDCE, and VC had reached local maxima in 2021 and by 2024 had decreased by 66% for TCE, 46% for cDCE, and 18 percent for VC. Plume areas for cDCE and VC showed an increasing trend for Pit 2.

Tiled plots for Pit 3 show decreasing trends for TCE and cDCE plume areas with percent changes from plume area local maxima of -45% for TCE and -80% for cDCE. The VC plume reached a local maxima for area in 2012 with a slight upward trend beginning in 2019.

Summary of Plume Areas Changes (2021 to 2024)

Table 2-6 has as summary of plume area changes from 2021 (September) to 2024 (March) illustrating overall decreases in plume areas from baseline sampling performed for the ISB injections in 2021 and the post-ISB sampling event performed in March 2024. The 2024 sampling occurred 27 months after completion of the 2021 ISB injections corresponding to 75% of design treatment period. The post 2021 ISB injection monitoring data indicates a trend change point where post injection monitoring for the first time indicates overall reductions of cDCE and VC plumes at a point more than 2 years after completion of the injections.

Table 2-6 Plume Area Changes (2021 to 2024)

Plume Area		PCE			TCE			cDCE		VC			
(Acres)	2021	2024	%∆	2021	2024	%∆	2021	2024	%∆	2021	2024	%∆	
Pit 1	0.26	0.28	7.7%	0.63	0.36	-42.9	0.83	0.30	-63.9	2.46	1.07	-56.5	
Pit 2	0.45	0.20	-56.5	2.42	0.83	-65.7	1.34	0.72	-46.3	2.87	2.34	-18.5	
Pit 3	0.00	0.00		0.80	0.63	-21.2	0.86	0.22	-74.4	1.32	1.22	-7.8	

Notes: % percent change from 2021 to 2024, increase in plume area, decrease in plume area.

2.4.10 Bulk Plume Distribution Trends

This assessment evaluates changes in data distributions for bulk plume areas associated with Pits 1, 2, and 3. Exhibit 2-5 (page 2-46) has time series, line data plots for PCE, TCE, cDCE, and VC at each pit area for the period 2019-2024 that include number of detections, maximum concentration, mean of detected results, and number of results greater than the MCLs.

2.4.10.1 Number of Detections

Table 2-7 has summary data from the tiled plots in Exhibit 2-5 for each plume area for number of detections of PCE, TCE, cDCE, and VC in 2019 and 2024 with the percent change from 2019 to 2024. Reductions in detection frequency along the reaction pathway in the Pit 1 and 2 areas provides evidence of complete reductive processes. At Pit 3, the detection frequency for TCE increased with the detections decreasing for cDCE and VC. cDCE had the most frequent detections in each monitored area for 2019 and 2024.

Table 2-7 Number of PCE, TCE, cDCE, and VC Detections and Percent Change

		PCE			TCE			cDCE			VC	
Plume Area	2019	2024	%∆	2019	2024	%∆	2019	2024	%∆	2019	2024	%∆
Pit 1	28	18	-35.7	45	27	-40.0	46	32	-30.4	36	24	-33.3
Pit 2	24	14	-42.7	36	22	-38.9	46	29	-37.0	36	24	-33.3
Pit 3	0	0		11	23	109	23	19	-17.4	21	15	-28.6

Notes: % percent change from 2021 to 2024, increase in frequency of detection, decrease in frequency of detection.

2.4.10.2 Maximum Concentrations

Local maxima concentrations for PCE, TCE, cDCE, and VC occurred before ISB injections completed in November-December 2021, with the exception of cDCE at Pit 2 where the local maxima occurred in 2022 after ISB injections. Table 2-8 has summary data from the tiled plots in Exhibit 2-5 for each plume area including local maxima concentrations for 2019-2024, the detected maximum concentrations for 2024, and the percent change from the local maxima to the 2024 maximum. Pit 1 had percent changes in concentrations greater than 91% except for cDCE (-47.1%). Pit 2 had percent changes in PCE and TCE greater than -96% with a percent change of -88.6% for cDCE and a percent change of -42.5% for VC. Pit 3 had percent changes in concentrations greater than -90% for TCE, cDCE, and VC.

Table 2-8 Maximum Concentrations and Percent Change

		PCE			TCE			cDCE			VC	
Plume Area	Local Max	2024	% ∆	Local Max	2024	%∆	Local Max	2024	% ∆	Local Max	2024	% ∆
Pit 1	15,000	1,300	-91.3	21,100	780	-96.3	24,000	14,000	-41.7	5,200	270	-94.8
Pit 2	1,300	41	-96.8	20,000	170	-99.1	81,000	9,200	-88.6	1,600	920	-42.5
Pit 3				15,000	350	-97.7	43,000	2,500	-94.1	5,300	600	-90.6

Notes: local max = local maxima concentration for 2019-2024, $\%\Delta$ percent change from 2021 to 2024, decrease in local maxima concentration.

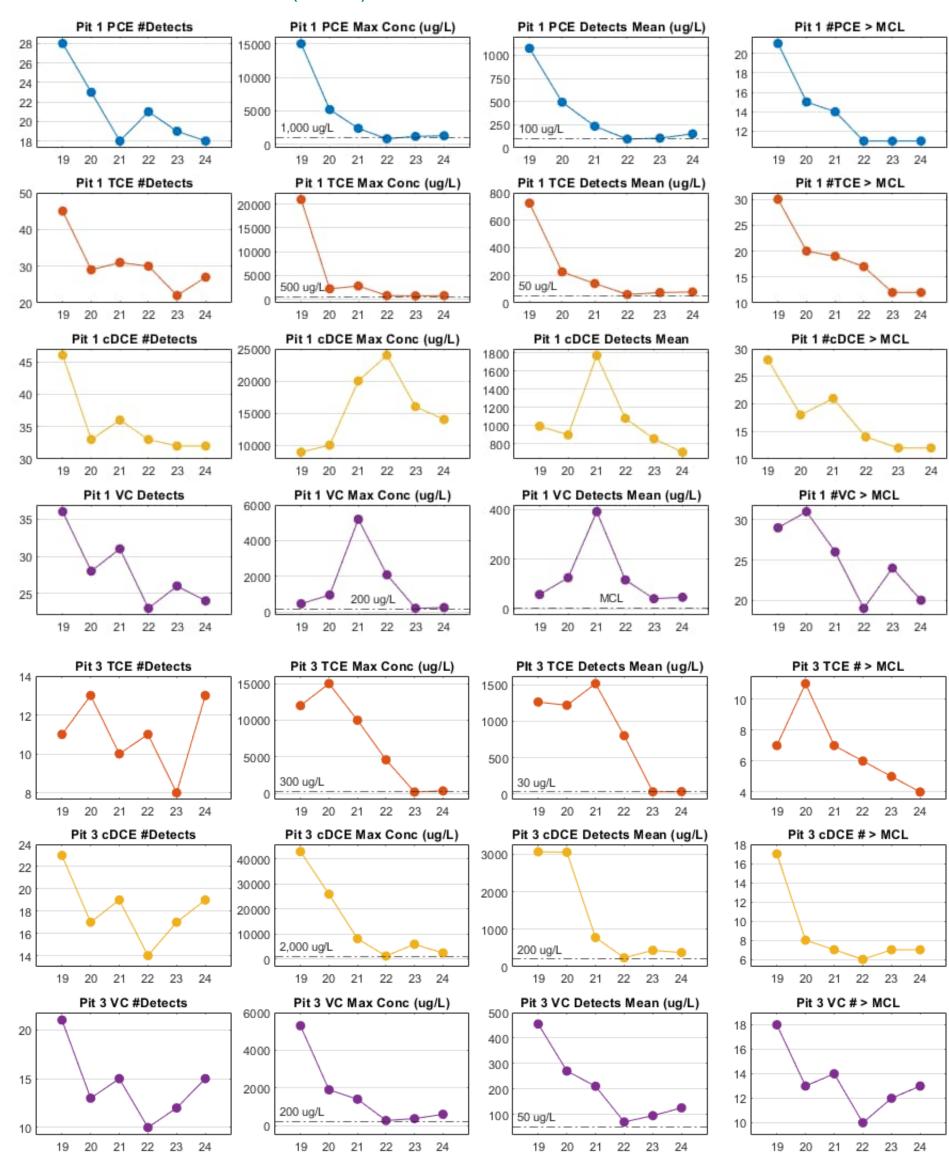
2.4.10.3 Mean of Detected Concentrations

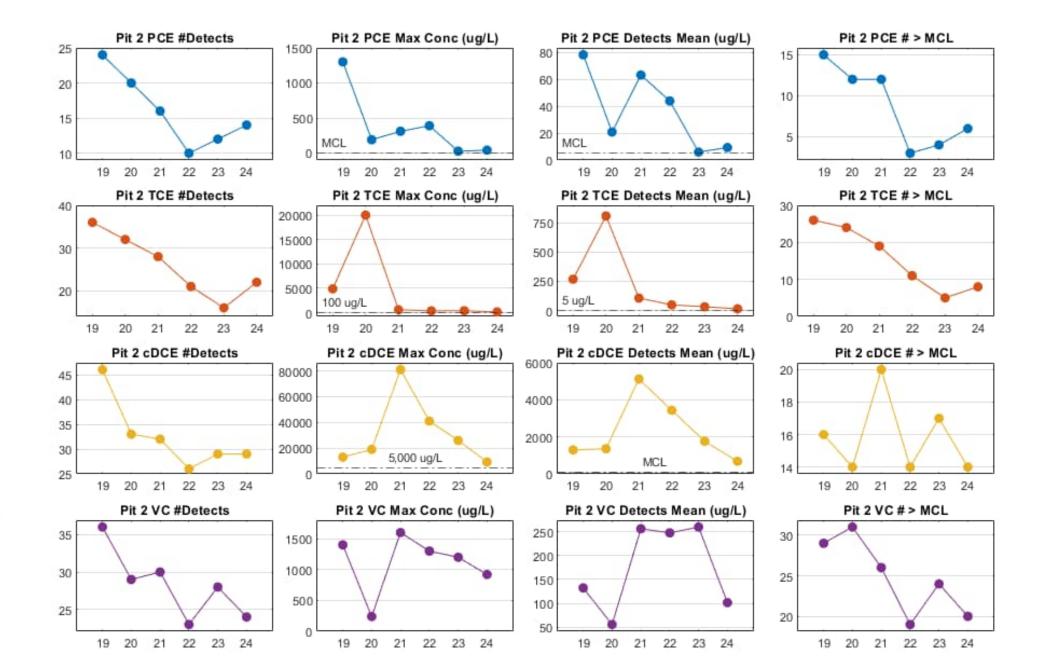
For time series data, mean concentrations of PCE, TCE, cDCE, and VC had similar changes in timeseries data as local maxima concentrations for 2019-2024.

2.4.10.4 Maximum Contaminant Levels (MCLs)

Local maxima for the number of detected results for PCE and TCE greater than MCLs occurred in 2019 except at Pit 3 (2020). For cDCE, the local maxima for the number of detected results greater than MCLs occurred in 2019 at Pits 1 and 2 and in 2022 at Pit 3 (after ISB injections). For VC, the local maxima for the number of detected results greater than MCLs occurred in 2019 in Pit 3 and in 2020 at Pits 1 and 2. Table 2-9 (page 2-47) has summary data from the tiled plots in Exhibit 2-5 for each plume area including the local maxima for number of results greater than MCLs, the number of results greater than MCLs for 2024, and the percent change from the local maxima for results greater than the MCL to the 2024 results.

Exhibit 2-5 Bulk Plume Distribution Trends (2019-2024)





PCE had percent changes in the number of results greater than MCLs from the local maxima to 2024 of -47.6% and -60.0% for Pits 1 and 2, respectively. TCE had the highest percent changes ranging from -63.3% at Pit 1 to -69.2% at Pit 2. cDCE had percent changes ranging from -30.0% at Pit 2 to -58.8% at Pit 3. VC had percent changes in the number of results greater than MCLs from the local maxima to 2024 at levels ranging from -27.8% at Pit 3 to -35.5% at Pit 2.

Table 2-9 Number of Samples with Concentrations > MCLs and Percent Change

		PCE			TCE			cDCE	VC			
Plume Area	Local Max	2024	% ∆									
Pit 1	21	11	-47.6	30	11	-63.3	28	12	-57.1	31	21	-32.2
Pit 2	15	6	-60.0	26	8	-69.2	20	14	-30.0	31	20	-35.5
Pit 3	0	0		11	4	-63.6	17	7	-58.8	18	13	-27.8

Notes: Max = local maxima for number of results detected at concentrations greater than maximum contaminant levels (MCLs), % Δ percent change from 2021 to 2024, decrease in number of results reported at concentrations > MCL.

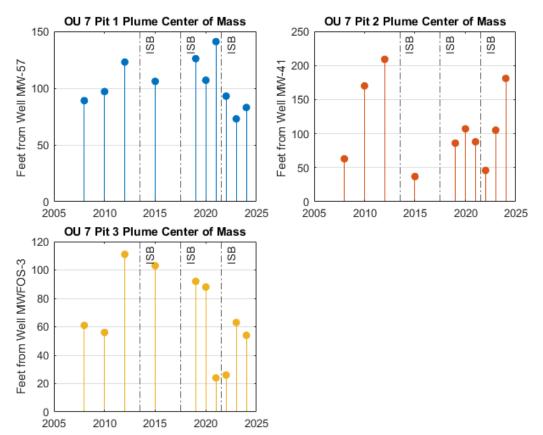
2.4.11 Plume Concentration vs. Distance Plots

This assessment uses monitoring well transects in each pit plume area to evaluate changes in plume molar mass, plume attenuation with distance from former pit areas, and changes in plume center of mass over time.

2.4.11.1 Plume Center of Molar Mass

Exhibit 2-6 plots the calculated plume center of mass for the Pit 1, 2, and 3 areas using well transect data for 2008, 2010, 2012, 2015, and 2019 through 2024. Center of molar mass for this calculation represents the distance along the well transect from the identified well located in or near the former pit areas for each plume. Calculations use micromolar concentrations for PCE, TCE, cDCE, 1,2-trans-dichloroethene (tDCE), and VC at each well along the transect. The stem plots have vertical reference lines for ISB injection events completed at OU 7.

Exhibit 2-6 Plume area Center of Molar Mass for Pits 1, 2, and 3



The Pit 1 well transect covers a distance of 605 ft. from well MW-57 in the former pit area to PZ-4 in the Kingsland Creek flood plain across Kingsland Creek in the off-installation area. Exhibit 2-7 (page 2-49) has the well locations for this flow path above the tiled plots. For Pit 1, the plotted plume center of mass reached its farthest downgradient point in 2021 (pre-ISB injection monitoring event) at a distance of 126 ft. from well MW-57. After injections, the calculated center of mass has moved upgradient to a distance of 83 ft. from well MW-57 as of the 2024 annual monitoring event.

The Pit 2 well transect covers a distance of 588 ft. from well MW-41 near the former pit area to MWFTA-3 in the lower area approximately 30 ft. upgradient of the installation fence line. Exhibit 2-9 (page 2-50) has the well locations for this flow path above the tiled plots. For Pit 2, the plotted plume center of mass reached its farthest downgradient point (209 ft. from well MW-41) in 2012 before the ISB remedial action in 2013. After this remedial action, the calculated plume center of mass has varied between 37 and 181 ft. from well MW-41.

The Pit 3 well transect covers a distance of 348 ft. from well MWFOS-3 near the former pit area to MW-118 south of the Building 72 area. Exhibit 2-11 (page 2-51) has the well locations for this flow path above the tiled pilots. For Pit 3, the plotted plume center of mass reached its farthest downgradient point (111 ft. from well MWFOS-3) in 2012 before the ISB remedial action in 2013. As of 2024, the calculated plume center of mass is 54 ft. downgradient of MWFOS-3.

2.4.11.2 Plume Molar Percentage Plots

Exhibits 2-7, 2-9, and 2-11 (pages 2-49, 50, 51) have relative molar percentage of constituents vs. distance plots that illustrate the relative degree of plume transformation along the groundwater flow path and changes over time. Molar percentages derive from constituent concentrations expressed as micromolar concentrations. Thus, the molar percentages for PCE, TCE, cDCE, tDCE, and VC sum to 100 percent for each stacked stem plot location.

Molar plots for Pit 1 show transformation from predominantly a TCE and cDCE plume in 2008 to predominantly a cDCE and VC plume in the main plume area. For 2024, cDCE predominates in the upper 200 ft. of the transect with VC becoming predominate downgradient to the fence line area. Molar plots for Pit 2 show transformation from predominately a PCE plume in 2008 to predominantly a cDCE and VC plume in 2024. Molar plots for Pit 3 show transformation from predominantly a TCE and cDCE plume in 2008 to predominately a cDCE and VC plume in 2024. For 2024, cDCE predominates over VC closer to the former pit area with the ISB injections in late 2021 reducing TCE in this area. VC is more predominant in downgradient areas.

2.4.11.3 Concentration vs. Distance Plots (2024)

Exhibits 2-8, 2-10, and 2-12 (pages 2-49, 50, 51) have concentration vs. distance plots (2024) that display changes in PCE and degradation by-products along plume centerlines for Pits 1, 2, and 3 (surficial aquifer). Each tile plot has stem plots for PCE, TCE, cDCE, tDCE, and VC with the sixth tile containing a stem plot displaying molar percentages for these constituents along the plume transect.

Pit 1

Concentration peaks for PCE, and TCE occur 60 ft. downgradient of MW-57 with the peak concentrations for cDCE and VC occurring at MW-57. Constituent concentrations in the main plume decrease to levels less than MCLs when reaching the installation fence line at 468 ft. (MW-88). Isolated plume areas with PCE and VC/or TCE concentrations greater than MCL occur in the Kingsland Creek at distances of 579 ft. (PZ-3) and 605 ft. (PZ-4). The molar plot for the Pit 1 plume shows that cDCE predominates on a percentage basis except where location concentrations fall below the LOQ or limit of detection (LOD) at the fence line and at PZ-3.

Exhibit 2-7 Pit 1 Plume Molar Percent vs. Distance Plots vs. Time

Pit 1 MW-57 \rightarrow AEHADG-11 \rightarrow CMT-21-1 \rightarrow MW-69 \rightarrow MW-EBF-05S \rightarrow MW-86 \rightarrow DMW-33A \rightarrow PZ-3 \rightarrow PZ-4

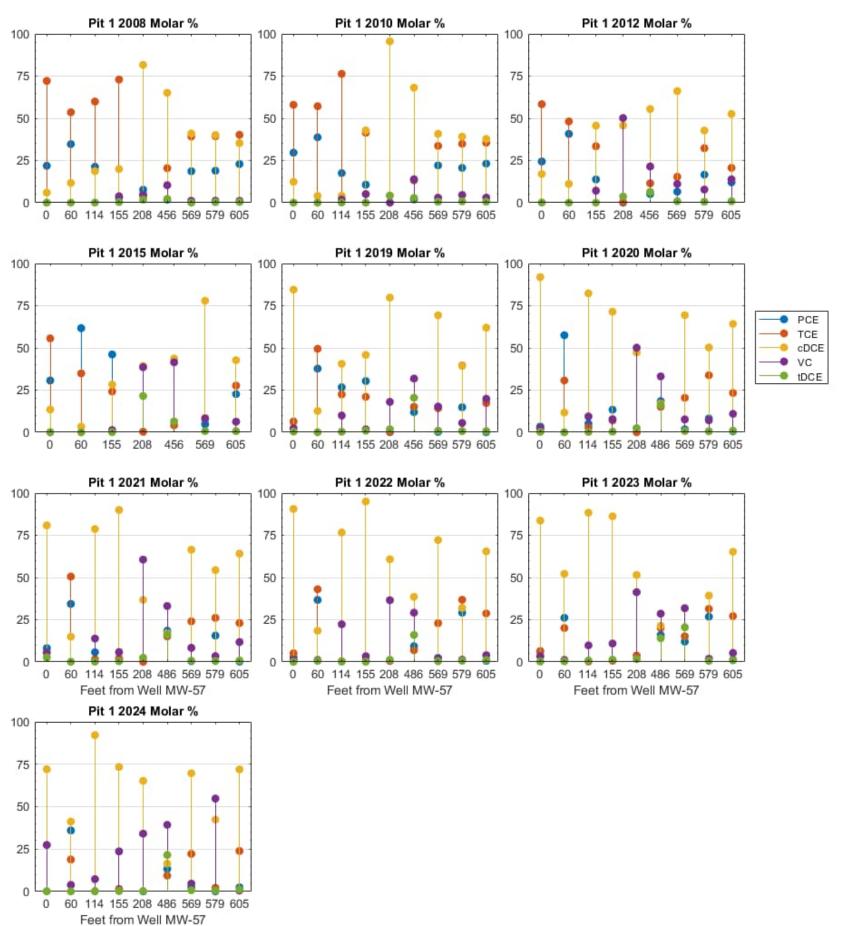


Exhibit 2-8 Pit 1 Concentration vs. Distance (2024)

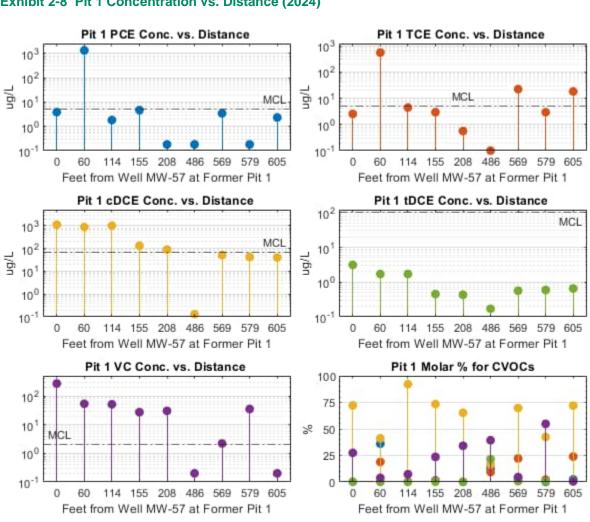


Exhibit 2-9 Pit 2 Plume Molar Percent vs. Distance Plots vs. Time

$Pit\ 2\ MW-41 \rightarrow MW-48 \rightarrow MW-51 \rightarrow MW-53 \rightarrow MW-54 \rightarrow CMT-18-6 \rightarrow MW-74 \rightarrow MW-79 \rightarrow MW-83 \rightarrow MWFTA-3 \rightarrow MW-74 \rightarrow MW-79 \rightarrow MW-83 \rightarrow MW-74 \rightarrow MW-79 \rightarrow$

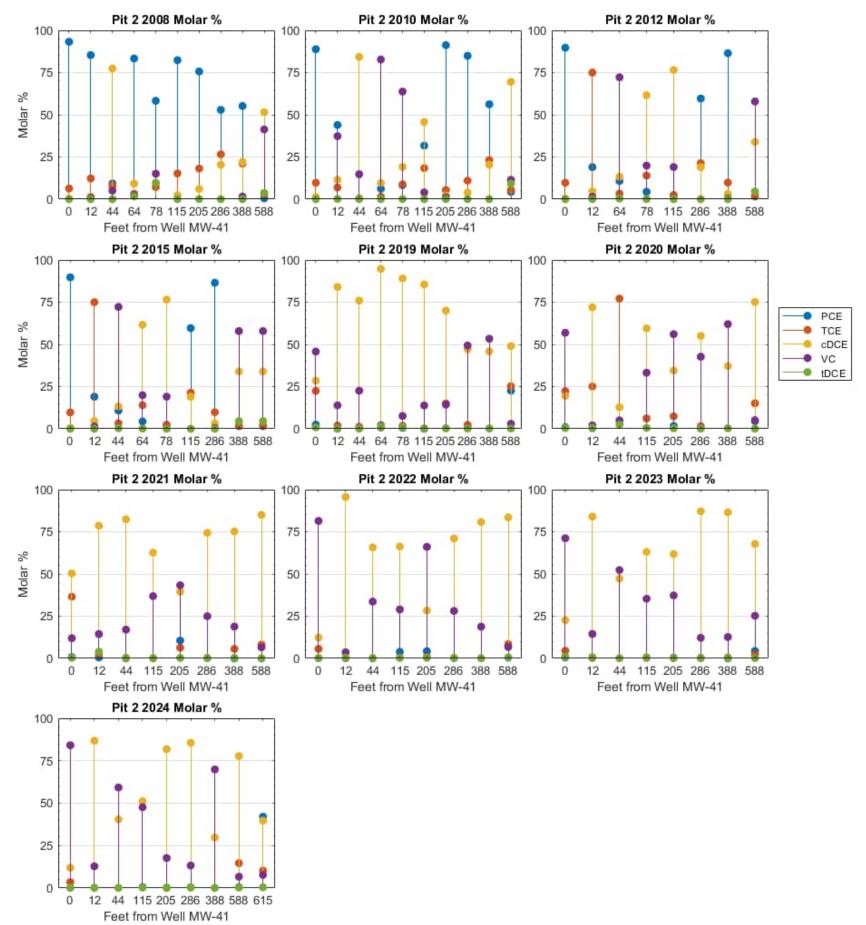


Exhibit 2-10 Pit 2 Concentrations vs. Distance (2024)

Pit 2 MW-41→MW-48→MW-51→MW-53→MW-54→CMT-18-6→MW-74→MW-79→MW-83→MWFTA-3

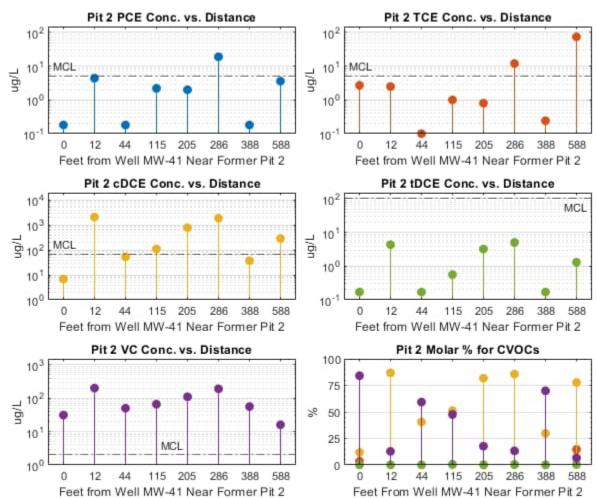


Exhibit 2-11 Pit 3 Plume Molar Percent vs. Distance Plots vs. Time

Pit 3 MWFOS-3→MW97→MW-96→MW95→MW-93→MW90 (2008-2019)→MW-118

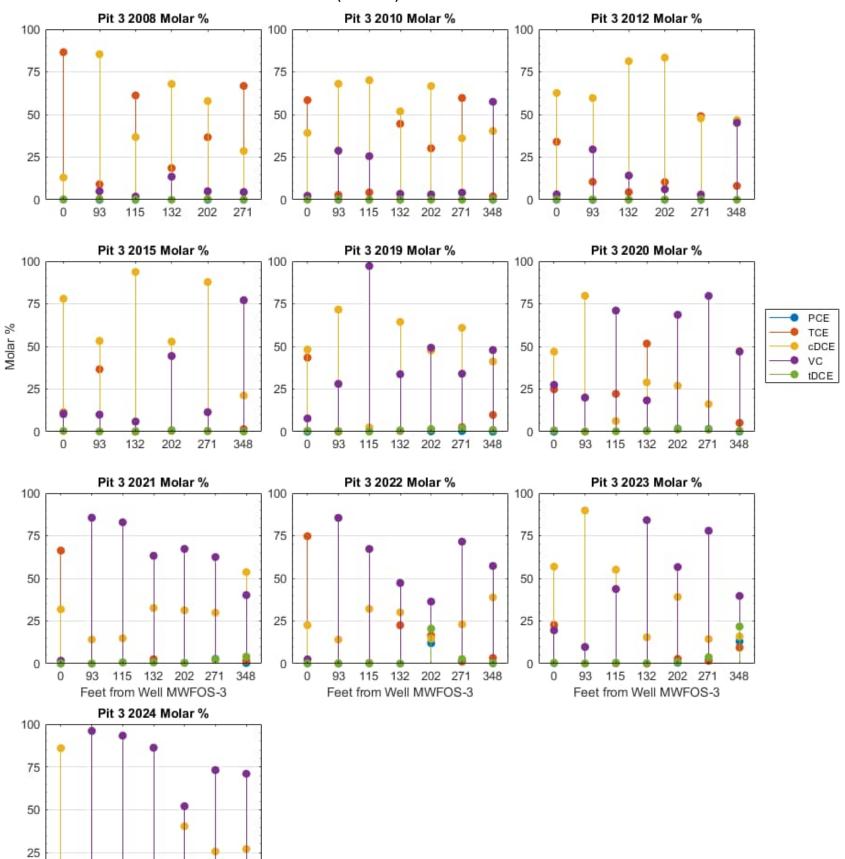


Exhibit 2-12 Pit 3 Concentration vs. Distance (2024)

202 271

115

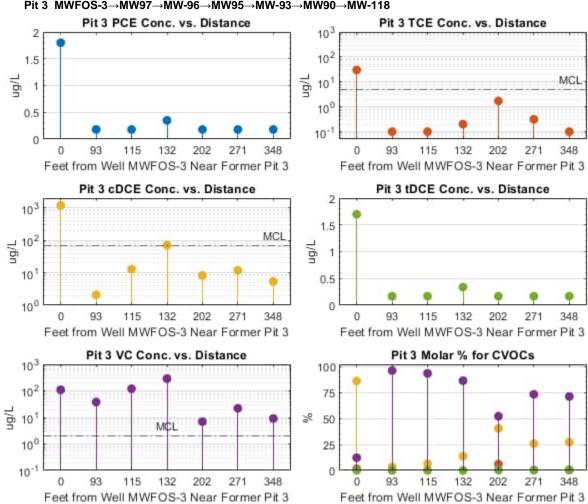
132

Feet from Well MWFOS-3

0

Pit 3 MWFOS-3 \rightarrow MW97 \rightarrow MW-96 \rightarrow MW95 \rightarrow MW-93 \rightarrow MW90 \rightarrow MW-118

348



Pit 2

Along the transect, PCE has concentrations less than the MCL except at 278 ft. downgradient of MW-41 at CMT-18-6 with the peak TCE concentration occurring at well MW-FTA-3 near the installation fence line. TCE concentrations are less than the MCL along the transect except at wells CMT-18-6 and MW-FTA-3. cDCE has concentrations greater than MCLs at 5 of 8 well locations along the transect with concentrations less than MCLs at well 41 (0 ft.), 44 ft., and 388 ft. within the upper area. VC concentrations at 8 of 8 locations along the profile exceed the MCL. The molar plot for the Pit 2 plume shows that cDCE or VC predominate on a percentage basis along the transect.

Pit 3

PCE data plotted reflect left-censored data (non-detects). TCE concentrations peak at MWFOS-3 (0 ft.) with concentrations at downgradient profile locations less than the MCL. cDCE concentrations peak at MWFOS-3 (0 ft.) and with only the downgradient profile location at 132 ft. (71 µg/L) greater than MCL. VC concentrations peaked at 132 ft. downgradient of MWFOS-3 with concentrations decreasing to a level below MCLs at 348 ft. The molar plot for the Pit 3 plume shows that VC predominates on a percentage basis along the transect except at 0 ft. (MWFOS-3) closest to the former pit area.

2.4.12 Well Trends

Section 2.4.12 has monitoring well trend analyses for wells in the Pit 1, 2, and 3 areas at select locations from the former pit areas to downgradient monitored areas. Several lines of evaluation used include coefficient of variation (CV), linear correlation (Pearson's linear correlation), Mann-Kendall (M-K) test for trend, sequential M-K test for trend change points, and SAA.

2.4.12.1 Pit 1 Well Trends

Table 2-10 (page 2-53) has a descriptive statistical summary and trend analysis for wells in the Pit 1 plume area at locations from the former pit area downgradient to Kingsland Creek. Exhibit 2-13 (page 2-54) has plots for these wells showing timeseries data and SSA trends as applicable. SSA trend analysis for the Pit 1 monitored area indicated decreasing trends for primary COCs for evaluated locations except for VC at well MW-57 (former pit area) and at well AEHADG-11 located 60 ft. downgradient of MW-57. M-K trend tests indicated statistical evidence of increasing trends for VC at wells MW-57 and AEHAG-11 with no other increasing trends detected. M-K trend tests indicated decreasing trends for one or more COCs at 6 of 8 well locations along the transect.

2.4.12.2 Pit 2 Well Trends

Table 2-11 (page 2-55) has a descriptive statistical summary and trend analysis for wells in the Pit 2 plume area at locations from the former pit area downgradient to the installation fence line at well MW-148. Exhibit 2-14 (page 2-56) has plots for these wells showing timeseries data and SSA trends as applicable. SSA trend tests for the Pit 2 monitored area indicated decreasing trends at 19 of 26 evaluated location/COC combinations with two (2) locations having no trend, and five (5) locations having an increasing trend. M-K trend analysis indicated decreasing concentrations at 8 of 28 evaluated location/COC combinations with 15 locations having no trend, and five (5) locations having an increasing trend. SSA showed decreasing trends at the most downgradient location closest the installation fence line.

2.4.12.3 Pit 3 Well Trends

Table 2-12 (page 2-57) has a descriptive statistical summary and trend analysis for wells in the Pit 3 plume area at locations from the former pit area to well MW-118 downgradient of Building 72. Exhibit 2-15 (page 2-58) has plots for these wells showing timeseries data and SSA trends as applicable. SSA trend analysis for the Pit 3 monitored area indicated decreasing trends at 18 of 20 evaluated location/COC combinations with cDCE and VC at well MW-96 having increasing trends. M-K trend tests indicated decreasing trends at 13 of 20 evaluated location/COC combination with no detected increasing trends. The most downgradient well in the transect (MW-118) had decreasing trends for COCs.

Table 2-10 Statistical Summary and Trend Analysis for Pit 1 Surficial Aquifer Wells (2015-2024)

		Date	Distance from	Primary		ribution riance	(α=0	orrelation ¹	(Or	(Trend ne Way) =0.05)	2024 Result	MCL			M-K Sequential for Trend	M-K Trend – One Way
Well ID	Area	Range	MW-57	COC	CV	Variance	r¹	p-value	Н	p-value	(µg/L)	(µg/L)	>MCL	SSA Trend	Change	(Increasing or Decreasing)
				PCE	1.0492	High	-0.9032	5.69E-05	1	-1.60E-03	3.8	5	No	SSA↓,	No trend change	M-K SS Decreasing at α = 0.05
MW-57	Pit 1	2015-2024	0 ft.	TCE	1.3115	High	-0.8875	1.17E-04	1	-3.70E-03	2.5	5	No	SSA↓	No trend change	M-K SS Decreasing at α = 0.05
			0	cDCE	0.9041	Moderate	0.5773	4.93E-02	0	1.08E-01	1100	70	Yes	SSA↓	No trend change	No M-K SS trend at α = 0.05
				VC	1.4024	High	0.6888	1.32E-02	1	7.84E-04	270	2	Yes	SSA↑	No trend change	M-K SS Increasing at $\alpha = 0.05$
				PCE	0.3464	Low	-0.4448	1.27E-01	0	-2.14E-01	1300	5	Yes	SSA↓	No trend change	No M-K SS trend at $\alpha = 0.05$
AEHADG-11	Source Zone	2015-2024	60 ft.	TCE	0.9120	Moderate	-0.3971	1.79E-01	0	-7.12E-02	540	5	Yes	SSA↓	No trend change	No M-K SS trend at $\alpha = 0.05$
ALTIADO-11	Source Zone	2013-2024	00 11.	cDCE	0.8063	Moderate	0.6325	2.03E-02	1	1.39E-01	870	70	Yes	SSA↓	No trend change	No M-K SS trend at $\alpha = 0.05$
				VC	1.0770	High	0.7726	2.00E-03	1	4.20E-03	53	2	Yes	SSA↑	No trend change	M-K SS Increasing at $\alpha = 0.05$
				PCE	NaN		NaN		1	-3.01E-02	ND	5	No	SSA↓	No trend change	M-K SS Decreasing at $\alpha = 0.05$
CMT21-1	Mid-Plume	2019-2024	114 ft.	TCE	NaN	-	NaN	-	1	-6.30E-03	4.4	5	No	SSA↓	No trend change	M-K SS Decreasing at $\alpha = 0.05$
CIVITZ 1-1	Mid-Flume	2019-2024	11411.	cDCE	0.1995	Low	0.5162	0.1548	1	5.589E-02	1000	70	Yes	SSA↑	No trend change	No M-K SS trend at α = 0.05
				VC	0.4601	Low	0.2168	0.5753	0	4.58E-01	51	2	Yes	SSA ↓ Mar. 2024	No trend change	No M-K SS trend at α = 0.05
				PCE	1.2095	High	-0.9112	1.47E-05	1	-6.11E-04	4.6	5	No	SSA↓	No trend change	M-K SS Decreasing at α = 0.05
MW-69	Mid-Plume	2015 2022	1 F F #	TCE	1.1940	High	-0.8658	1.30E-04	1	-1.40E-03	2.9	5	No	SSA↓	No trend change	M-K SS Decreasing at α = 0.05
10100-09	iviid-Piume	2015-2023	155 ft.	cDCE	0.7713	Moderate	0.1222	6.91E-01	0	4.76E-01	130	70	Yes	SSA ↓ Mar. 2024	No trend change	No M-K SS trend at α = 0.05
				VC	0.8131	Moderate	0.3933	1.84E-01	0	1.50E-01	27	2	Yes	SSA ↓ Mar. 2024	No trend change	No M-K SS trend at α = 0.05
				PCE	2.0641	High	-0.4070	1.68E-01	1	-4.98E-02	6	5	Yes	SSA↓ Nov 2022	Decreasing trend (Sept 2021)	M-K SS Decreasing at $\alpha = 0.05$
D14144 00 4	Downgradient	0045 0004	400 (TCE	1.4107	High	-0.8486	2.44E-04	0	-8.03E-02	110	5	Yes	SSA↓ Nov 2022	No trend change	No M-K SS trend at α = 0.05
DMW-29A	Upper Area	2015-2024	196 ft.	cDCE	1.2828	High	-0.2917	3.33E-01	1	-1.40E-03	110	70	Yes	SSA↓	No trend change	M-K SS trend at α = 0.05
				VC	2.5515	High	0.0928	7.63E-01	0	1.23E-01	6.9	2	Yes	SSA ↓ Mar. 2023	Decreasing trend (Oct 2023)	No M-K SS trend at α = 0.05
				PCE	NaN		NaN				ND	5	No	Insufficient data	Insufficient data	Insufficient detected observations
EDE 050	Downgradient	0045 0004	000 (1	TCE	1.8454	High	-0.4150	0.1798	0	-3.40E-01	0.56	5	No	SSA↓	No trend change	No M-K SS trend at α = 0.05
EBF-05S	Upper Area	2015-2024	206 ft.	cDCE	1.7754	High	0.3791	0.2243	0	-1.51E-01	89	70	Yes	SSA↓	No trend change	No M-K SS trend at α = 0.05
				VC	1.3168	High	-0.4248	0.1686	0	-1.51E-01	30	2	Yes	SSA↓	No trend change	No M-K SS trend at α = 0.05
				TCE	1.7469	High	NaN		1	-9.40E-03	ND	5	No	SSA↓	No trend change	M-K SS Decreasing at $\alpha = 0.05$
MW-150	Downgradient Upper Area	2015-2024	253 ft.	cDCE	1.1864	High	-0.7834	1.50E-03	1	-9.90E-05	2.9	5	No	SSA↓	No trend change	M-K SS Decreasing at $\alpha = 0.05$
	Орреі Ліва			VC	1.4053	High	-0.3944	1.82E-01	1	-3.96e-04	0.4	70	No	SSA↓	No trend change	M-K SS Decreasing at α = 0.05
				PCE	1.8057	High	-0.4081	1.66E-01	0	4.27E-01	3.4	5	No	SSA↓	No trend change	No M-K SS trend at α = 0.05
DMW-33A	Off-Installation Kingsland Creek	2015-2024	569 ft.	TCE	0.7530	Moderate	-0.0362	9.07E-01	0	-2.51E-01	22	5	Yes	SSA↓	No trend change	M-K SS Decreasing at $\alpha = 0.05$
	Kingsianu Creek			VC	1.0476	High	0.5622	4.59E-02	1	-2.20E-02	2.2	2	Yes	SSA↓	No trend change	No M-K SS trend at $\alpha = 0.05$

Notes: $NaN = not\ a\ number$, not calculated because of left-censored data, ¹Pearson's linear correlation coefficient, $r = correlation\ coefficient$, $r = correlation\ coefficient$, r

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Exhibit 2-13 Pit 1 Well Trend Plots

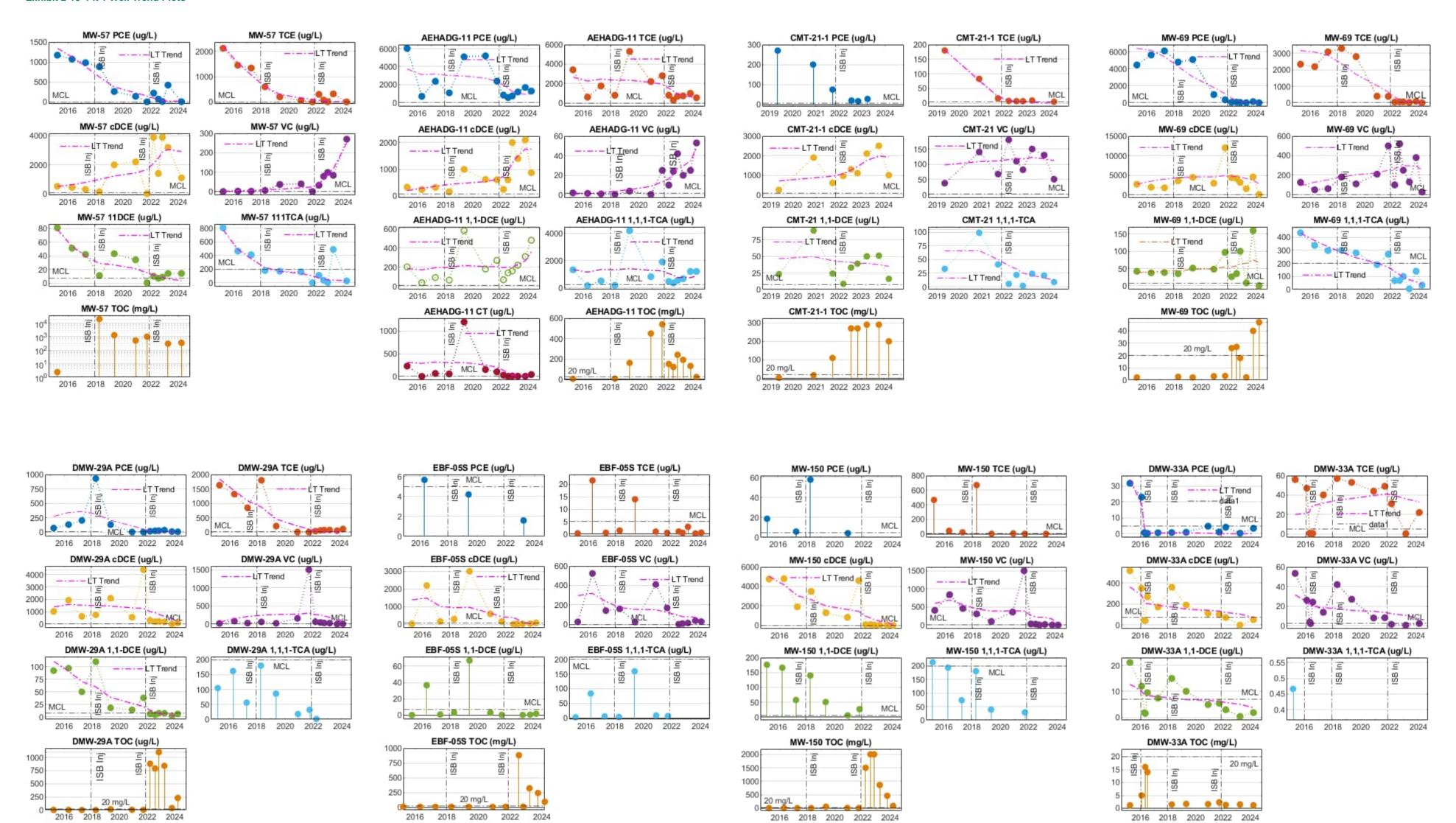
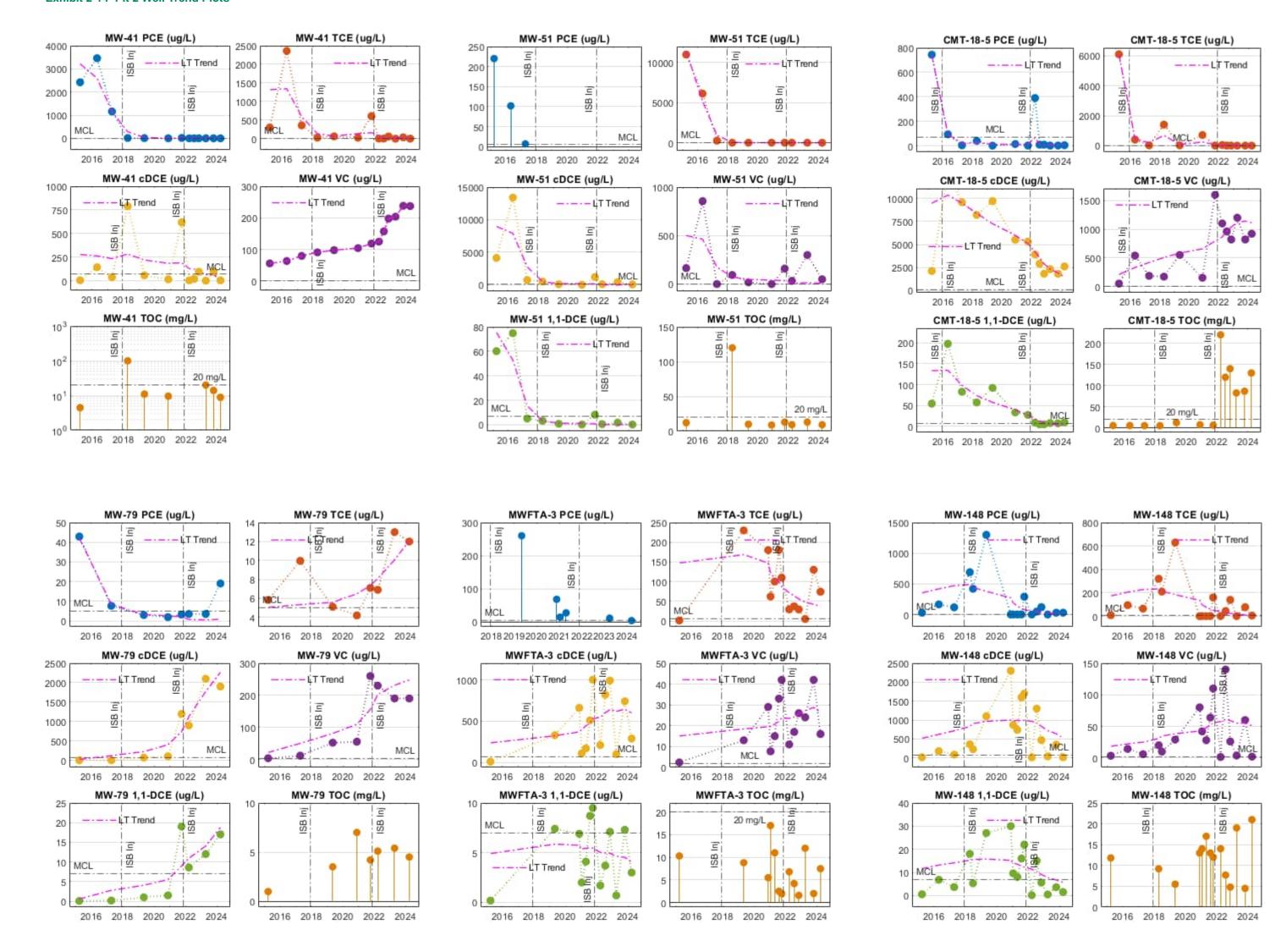


Table 2-11 Statistical Summary and Trend Analysis for Pit 2 Surficial Aquifer Wells

		Date	Distance from	Primary	Va	ribution riance		orrelation ¹ 0.05)	(O (d	K Trend one Way) a=0.05)	2024 Result	MCL				M-K Trend (One-Way for Increasing or Decreasing
Well ID	Area	Range	MW-57	COC	CV	Variance	r¹	p-value	Н	p-value	(µg/L)	(µg/L)	>MCL	SSA Trend ¹	Sequential M-K ²	Trend)
				PCE	2.0734	High	-0.7798	1.70E-03	1	-7.81E-04	0.18	5	No	SSA↓,	No trend change detected	M-K SS Decreasing at $\alpha = 0.05$
MW-41	Pit 2	2015-2024	O ft.	TCE	2.1775	High	-0.5202	6.84E-02	1	-8.70E-03	2.7	5	No	SSA↓	No trend change detected	M-K SS Decreasing at $\alpha = 0.05$
	Source Zone	2010 2021	0 11.	cDCE	1.7176	High	-0.1521	6.20E-01	0	-2.24E-01	6.8	70	Yes	SSA↓	No trend change detected	No M-K SS trend at α = 0.05
				VC	1.3056	High	0.2414	4.27E-01	0	2.14E-01	31	2	Yes	SSA↑	No trend change detected	No M-K SS trend at α = 0.05
				PCE	NaN		NaN		0	-1.49E-01	ND	5	No	Insufficient data	No trend change detected	No M-K SS trend at $\alpha = 0.05$
MW-51	Source Zone	2015-2024	44 ft.	TCE	2.1685	High	-0.7083	2.19E-02	0	-1.48E-01	0.1	5	No	SSA↓	No trend change detected	No M-K SS trend at $\alpha = 0.05$
IVIVV-51	Source Zone	2010-2024	7710.	cDCE	2.0210	High	-0.5709	8.48E-02	1	-2.45E-02	53	70	No	SSA↓	No trend change detected	M-K SS Decreasing at $\alpha = 0.05$
				VC	1.5490	High	-0.3315	3.49E-01	0	-4.29E-01	50	2	Yes	SSA↓	No trend change detected	No M-K SS trend at $\alpha = 0.05$
				PCE	2.1825	High	-0.4797	9.71E-02	1	-4.35E-02	4.4	5	No	SSA↓	No trend change detected	M-K SS Decreasing at $\alpha = 0.05$
CMT18-5	Downgradient	2015-2024	115 ft.	TCE	2.5077	High	-0.6200	2.38E-02	1	-3.10E-04	2.5	5	No	SSA↓	No trend change detected	M-K SS Decreasing at $\alpha = 0.05$
CIVIT 10-3	Source Zone	2013-2024	11311.	cDCE	0.6955	Moderate	-0.6434	1.77E-02	1	-3.60E-03	2600	70	Yes	SSA↓	No trend change detected	M-K SS Decreasing at $\alpha = 0.05$
				VC	0.6790	Moderate	0.7109	6.50E-03	1	2.52E-02	920	2	Yes	SSA→ March 2024	No trend change detected	M-K SS Increasing at $\alpha = 0.05$
				PCE	1.3422	High	-0.5417	1.65E-01	0	5.00E-01	19	5	Yes	SSA↓	No trend change detected	No M-K SS trend at α = 0.05
MW-79	Mid-Plume	2015-2024	206 #	TCE	0.4070	Low	0.4692	2.41E-01	0	1.33E-01	12	5	Yes	SSA↑	No trend change detected	No M-K SS trend at α = 0.05
10100-79	iviid-Plume	2015-2024	286 ft.	cDCE	1.1070	High	0.8386	9.30E-03	1	4.70E-03	1900	70	Yes	SSA↑	No trend change detected	M-K SS Increasing at $\alpha = 0.05$
				VC	0.8379	Moderate	0.8232	1.21E-02	1	2.30E-02	190	2	Yes	SSA↑	No trend change detected	M-K SS Increasing at $\alpha = 0.05$
				PCE	NaN		NaN		1	-4.47E-02	ND	5	No	Insufficient data	Insufficient data	M-K SS Decreasing at $\alpha = 0.05$
MM 00	Edge Upper	2045 2024	200 #	TCE	NaN		NaN	-	0	-3.94E-01	0.24	5	No	SSA↓ Mar 2024	Insufficient data	No M-K SS trend at α = 0.05
MW-83	Area	2015-2024	388 ft.	cDCE	1.3178	Low	0.5673	4.32E-02	1	8.70E-03	37	70	No	SSA↓ Mar 2024	No trend change detected	M-K SS Increasing at $\alpha = 0.05$
				VC	1.0290	Low	0.6813	1.04E-02	1	6.20E-03	56	2	Yes	SSA↓ Mar 2024	No trend change detected	M-K SS Increasing at $\alpha = 0.05$
				PCE	NaN		NaN		1	-1.21E-02	3.6	5	No	SSA↓	No trend change detected	M-K SS Decreasing at $\alpha = 0.05$
NAVA/ETA O	Flood Plain	0045 0004	500 ft	TCE	0.8164	Moderate	-0.0297	0.9234	0	-1.79E-01	74	5	Yes	SSA↓	No trend change detected	No M-K SS trend at α = 0.05
MWFTA-3	Near Perimeter Road	2015-2024	588 ft.	cDCE	0.7659	Moderate	0.3912	1.86E-01	0	1.50E-01	290	70	Yes	SSA→	No trend change detected	No M-K SS trend at α = 0.05
				VC	0.5862	Moderate	0.5110	7.43E-02	0	5.60E-02	16	2	Yes	SSA↑	No trend change detected	No M-K SS trend at α = 0.05
				PCE	1.7942	High	-0.3091	2.27E-01	0	-7.28E-02	29	5	Yes	SSA↓	Decreasing trend (Jan 2021)	No M-K SS trend at α = 0.05
N A) A 4 C	Installation	0045 0007	050 %	TCE	1.5792	High	-0.2590	3.15E-01	0	-2.42E-01	5.6	5	Yes	SSA↓	Decreasing trend (Jan 2021)	No M-K SS trend at α = 0.05
MW-148	Fence Line	2015-2024	253 ft.	cDCE	1.0504	High	0.1996	4.42E-01	0	4.51E-01	16	70	Yes	SSA↓ Mar 2022	Decreasing trend (Oct 2023)	No M-K SS trend at α = 0.05
				VC	1.0822	High	0.3341	1.90E+01	0	1.94E-01	2	2	No	SSA↓ Mar 2024	No trend change detected	No M-K SS trend at α = 0.05

Notes: $NaN = not\ a\ number$, not calculated because of left-censored data, ¹Pearson's linear correlation coefficient, $r = correlation\ coefficient$, $r = correlation\ coefficient$, r

Exhibit 2-14 Pit 2 Well Trend Plots



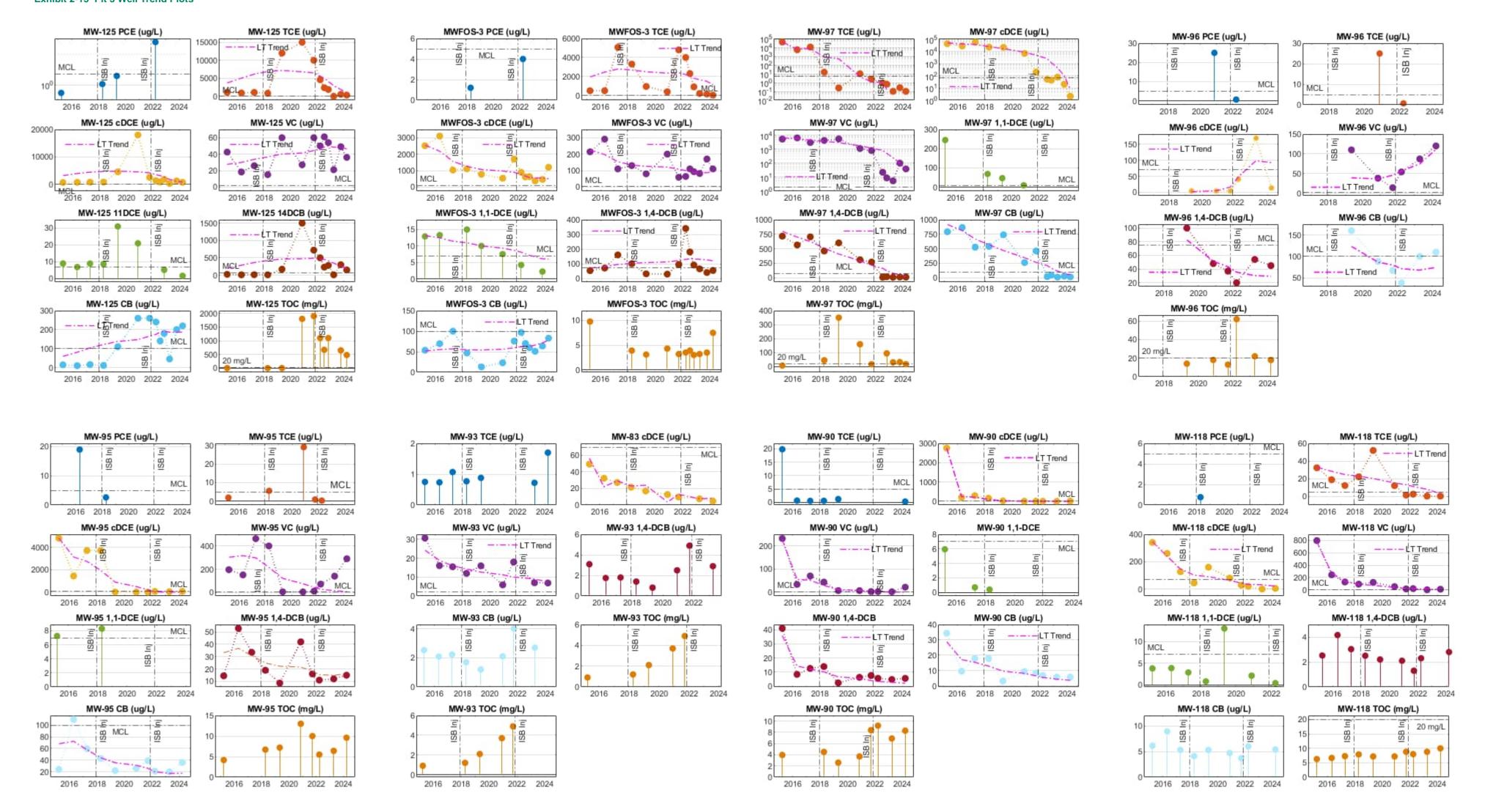
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Table 2-12 Statistical Summary and Trend Analysis for Pit 3 Surficial Aquifer Wells

	Area	Date Range	Distance from MW-41	Primary	Distribution Variance		Linear Correlation¹ (α=0.05)		M-K Trend (One Way) (α=0.05)		2024 Result	MCL				M-K Trend (One-Way for Increasing or Decreasing	
Well ID				COC	CV	Variance	r ¹	p-value	Н	p-value	(µg/L)	(µg/L)	>MCL	SSA Trend ¹	Sequential M-K ²	Trend)	
MW-125	Pit 3 Source Zone	2015-2024	0 ft.	TCE	1.2890	High	0.0220	9.43E-01	0	-1.11E-01	350	5	Yes	SSA↓ Mar 2019	Decreasing Trend (Mar 2023)	No M-K SS trend at $\alpha = 0.05$	
				cDCE	1.8191	High	0.0349	9.10E-01	0	3.80E-01	710	70	Yes	SSA↓ Mar 2019	Decreasing Trend (Oct 2023)	No M-K SS trend at $\alpha = 0.05$	
				VC	0.4254	Low	0.3782	2.03E-01	0	2.32E-01	36	2	Yes	SSA↓ Mar 2024	Decreasing Trend (Oct 2023)	No M-K SS trend at $\alpha = 0.05$	
MWFOS-3	Downgradient of Pit	2015-2024	44 ft.	TCE	1.0616	High	-0.1627	5.94E-01	1	-3.84E-02	30	5	Yes	SSA↓ Apr 2017	No trend change detected	M-K SS Decreasing at $\alpha = 0.05$	
				cDCE	0.7220	Moderate	-0.7021	7.50E-03	1	-1.64E-02	1200	70	Yes	SSA↓	No trend change detected	M-K SS Decreasing at $\alpha = 0.05$	
				VC	0.5194	Moderate	-0.5569	4.80E-02	0	-1.34E-01	110	2	Yes	SSA↓	Decreasing Trend (Oct 2023)	No M-K SS trend at $\alpha = 0.05$	
MW-97	Downgradient TA-11 area	2015-2024	114 ft.	TCE	2.4789	High	-0.6740	1.62E-02	1	-2.94E-04	0.1	5	No	SSA↓	Insufficient data	M-K SS Decreasing at $\alpha = 0.05$	
				cDCE	1.3071	High	-0.8948	8.50E-05	1	-4.69E-05	2.1	70	No	SSA↓	No trend change detected	M-K SS Decreasing at $\alpha = 0.05$	
				VC	1.1369	High	0.9064	4.82E-05	1	3.20E-03	38	2	Yes	SSA↓	No trend change detected	M-K SS Decreasing at $\alpha = 0.05$	
MW-96	Downgradient TA-11 area	2019-2024	155 ft.	TCE	NaN		NaN				ND	5	No	Insufficient data	Insufficient data	Insufficient detected observations	
				cDCE	1.6591	High	0.4415	3.81E-01	0	1.30E-01	13	70	Yes	SSA↑	No trend change detected	No M-K SS trend at α = 0.05	
				VC	0.5952	Moderate	0.2046	6.97E-01	0	2.25E-01	120	2	Yes	SSA↑	No trend change detected	No M-K SS trend at α = 0.05	
MW-95	Downgradient	2015-2024	196 ft.	TCE	NaN		NaN				ND	5	No	Insufficient data	Insufficient data	Insufficient detected observation	
				cDCE	0.2110	Low	-0.7697	1.1E-04	1	-4.40E-02	71	70	Yes	SSA↓	No trend change detected	M-K SS Decreasing at $\alpha = 0.05$	
				VC	1.2614	High	-0.7217	4.9E-04	0	-3.60E-01	290	2	Yes	SSA↓	No trend change detected	No M-K SS trend at α = 0.05	
MW-93	Downgradient	2015-2024	222 (cDCE	0.7717	Moderate	-0.8213	6.60E-03	1	-2.38E-02	8.3	70	No	SSA↓	No trend change detected	M-K SS Decreasing at $\alpha = 0.05$	
			206 ft.	VC	0.5408	Moderate	0.7364	2.36E-02	1	-3.82E-02	6.9	2	Yes	SSA↓	No trend change detected	M-K SS Decreasing at $\alpha = 0.05$	
MW-90	Downgradient near Building 72	2015-2024	253 ft.	cDCE	2.4730	High	-0.6164	5.77E-02	1	-1.50E-03	12	70	No	SSA↓	No trend change detected	M-K SS Decreasing at $\alpha = 0.05$	
				VC	1.6520	High	-0.6885	2.77E-02	1	-6.10E-03	22	2	Yes	SSA↓	No trend change detected	M-K SS Decreasing at $\alpha = 0.05$	
MW-118	Downgradient	2015-2024	569 ft.	TCE	1.1002	High	-0.6289	5.14E-02	1	-4.60E-03	0.1	5	No	SSA↓	No trend change detected	M-K SS Decreasing at $\alpha = 0.05$	
				cDCE	1.0864	High	0.3961	9.51E-04	1	-6.41E-04	5.4	70	No	SSA↓	No trend change detected	M-K SS Decreasing at $\alpha = 0.05$	
				VC	1.6326	High	-0.3487	1.30E-02	1	-3.38E-04	9.1	2	Yes	SSA↓	No trend change detected	M-K SS Decreasing at $\alpha = 0.05$	

Notes: $NaN = not\ a\ number$, not calculated because of left-censored data, ¹Pearson's linear correlation coefficient, $r = correlation\ coefficient$, $r = correlation\ coefficient$, r

Exhibit 2-15 Pit 3 Well Trend Plots



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2.5 Current Conditions for Surface Water

Table 2-13 on page 2-60 has summary results for detected VOCs in surface water samples collected from Kingsland Creek for the monitoring period 2021-2024. This table compares these results to MCLs, EPA RSLs, and ecological benchmarks. VOCs detected for the four (4) annual LTM events included 1,4-dichlorobenzene, cDCE, chlorobenzene, and TCE. Detected concentrations are less than MCLs and ecological benchmarks. At stations KCG-1, single detections of chlorobenzene and TCE at levels less than the laboratory LOD had reported results greater than their RSL. The RSLs for chlorobenzene and TCE are less than the laboratory LOQ. KGC-1 is located in the stream segment between the monitoring areas of Pit 3 and Pit 2 (see Figure 2-6).

2.6 Building 72 Subsurface Vapor

Annual sampling of sub-slab vapor at Building 72 at VMP-116 began in 2016 after converting over from monitoring external soil gas monitoring points that frequently accumulated perched water. None of the sub-slab sampling results for 2016-2024 at VMP-116 had concentrations greater than the EPA vapor intrusion screening levels (VISLs)². Field monitoring of VMP-116 has not indicated accumulation of methane beneath the building slab.

Sampling for the full list of TO-15 compounds began in 2017. The first full-list sampling had 37 VOC analytes detected, 20 analytes detected in 2018 and 2020-2022, 32 analytes detected in 2019, and six (6) analytes detected in December 2023. The reduced number of VOC detections reflects VOC plume conditions that have degraded over time in the Building 72 area from ISB remedial actions and intrinsic processes. The monitoring has indicated continued remedy protectiveness for workers that occasionally occupy Building 72.

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United States Environmental Protection Agency Vapor Intrusion Screening Level, November 2024 Updated Calculator, target cancer risk (TR) of 1E-06 and target hazard quotient (THQ) of 0.1, commercial exposure scenario, site-specific groundwater temperature of 15.2 Celsius at well OU7-MW-117, adjacent to Building 72, receptor, VISL calculator at Vapor Intrusion Screening Level Home (ornl.gov)

Table 2-13 OU 7 VOC Summary Statistics for Kingsland Creek Surface Water (Detects 2021-2024)

									Federal MCL ¹			EPA RSL² (November 2024)						BTAG FWSB ³ (2006)		
Chemical	Matrix	No. of Results	Unit	Min	Max	No. > LOD	% > LOD	Location of Max	No. > MCL	% > MCL	MCL ¹	No. > RSL	% > RSL	EPA RSL ²	Location >RSL	Yr. >RSL	No. > BTAG	% > BTAG	BTAG FWSB ³	
1,2-Dichloroethene, Total	WS	12	ug/L	<0.25	1.5	7	58.3	KGC-3									0	0	5.90E+02	
cis-1,2-Dichloroethene	WS	12	ug/L	<0.37	1.5	5	41.7	KGC-3	0	0	7.00E+01	0	0	2.52E+00			0	0		
Chlorobenzene	WS	12	ug/L	<0.15	1.1	3	25.0	KGC-1	0	0	1.00E+02	0	0	7.77E+00			0	0	1.30E+00	
1,4-Dichlorobenzene	WS	12	ug/L	<0.31	0.87 J	1	8.3	KGC-1	0	0	7.50E+01	1	8	4.82E-01	KGC-1	2021	0	0	2.60E+01	
Trichloroethene (TCE)	WS	12	ug/L	<0.2	0.84 J	1	8.3	KGC-1	0	0	5.00E+00	1	8	2.83E-01	KGC-1	2024	0	0	2.10E+01	

Notes: WS = surface water ug/L = micrograms per liter, Min = minimum, Max = maximum, No. > = number greater, LOD = limit of detection, % = percent, J = Estimated: the analyte is positively identified; the quantitation is an estimation, ¹Subpart G—National Primary Drinking Water Regulations: Maximum Contaminant Levels and Maximum Residual Disinfectant Levels, § 141.61 Maximum contaminant levels for organic contaminants. > LOD = percent of detected results greater than laboratory limit of detection, 2 EPA Regional Screening Level for Resident Tap Water with a target cancer risk (TR) of 1E-06 and target hazard quotient (THQ) of 0.1, November 2004 at https://www.epa.gov/risk/regional-screening-levels-rsls-generic-tables. ³EPA Region III BTAG Freshwater Screening Benchmarks, 7/2006. https://www.epa.gov/risk/freshwater-screening-benchmarks, Number of results with concentrations greater than the MCL, Number of results with concentrations greater than the BTAG FWSB

3. Remedial Design

Section 3 presents the remedial design for follow-up enhanced ISB actions at OU 7. Follow-up ISB actions at OU 7 will target accelerated treatment of remaining hot spot areas in the surficial aquifer. Addressing these hot spot areas will reduce contaminant mass and contaminant flux to downgradient areas. ISB actions proposed in this work plan will target additional treatment of diffuse VOC plumes farther downgradient of source zones with the objective of reducing overall plume areas.

3.1 Remedial Design Basis

The process option of the ISB design in this work plan follows the remedial design/remedial action work plan for OU 7 (AECOM 2013) using metabolic anaerobic reductive dechlorination as the targeted degradation process to treat the chlorinated solvents. In this reaction, microorganisms gain energy as one or more chlorine atoms on a chlorinated ethene or ethane compound molecule are replaced with hydrogen atoms in an anaerobic environment. The chlorinated compound serves as the electron acceptor and molecular hydrogen usually serves as the electron donor (source of energy). Hydrogen used in this reaction is supplied by fermentation of organic substrates or a direct electron donor. Biodegradation of an organic substrate depletes the aquifer of DO and sequentially reduces native electron acceptors nitrate, manganese, iron, sulfate, and carbon dioxide. In general, metabolic anaerobic reductive dechlorination occurs by sequential removal of chlorine atoms. Exhibit 3-1 illustrates the reductive dechlorination pathway for PCE the parent compound. Primary COCs for the targeted constituent plume for ISB treatment include TCE, cDCE, and VC.

Exhibit 3-1 Reductive Dechlorination Treatment Pathway

The specific ISB design in this work plan considers the results from previous treatability studies (AECOM, 2010) and remedial implementation for the confined aquifer at OU 7 (AECOM 2013, USACE 2017, AECOM 2021b) and the current conditions presented in Section 2.4.

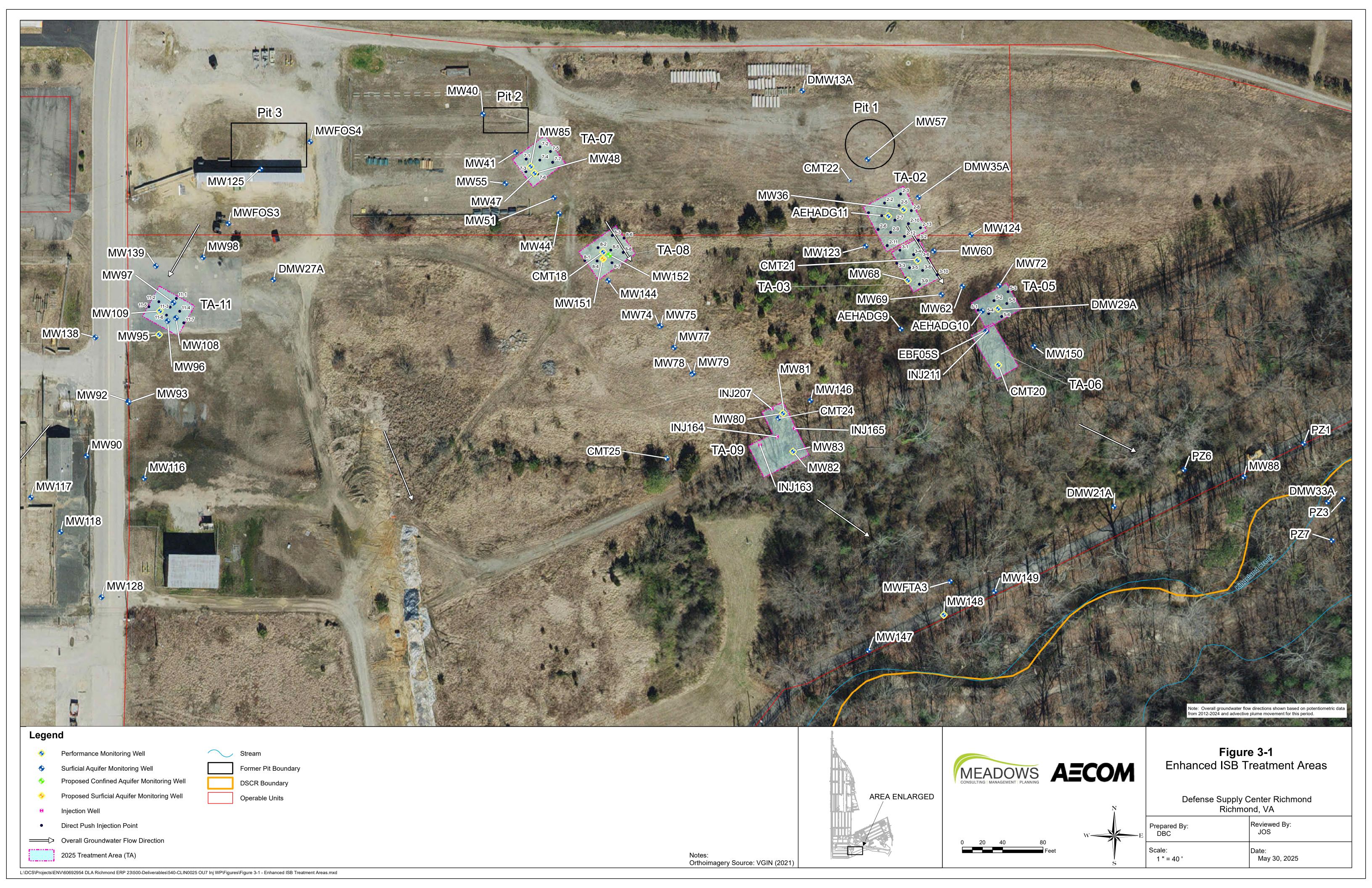
3.2 Enhanced ISB Treatment Areas

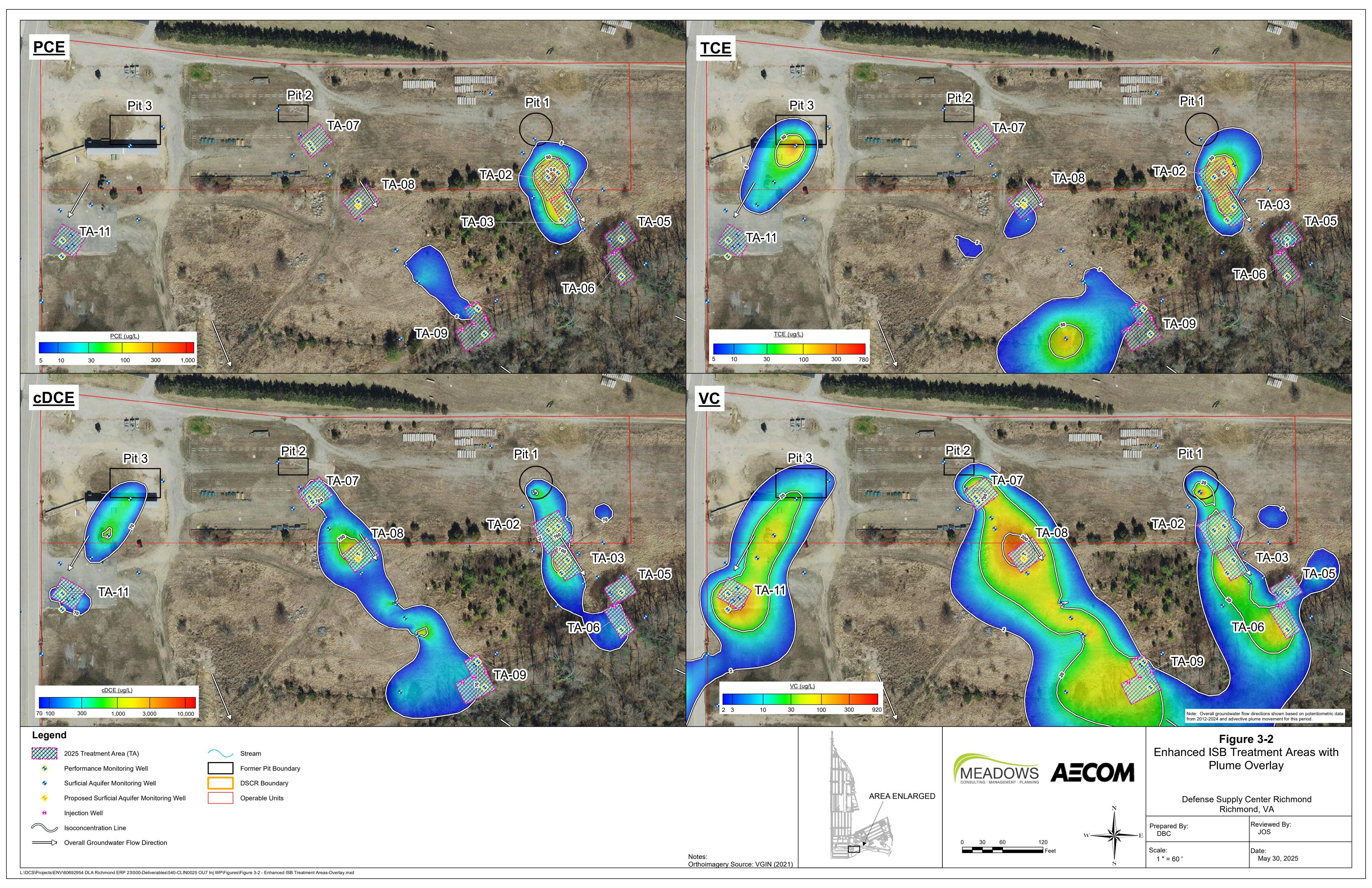
Eight enhanced ISB treatment areas are proposed for the surficial aquifer VOC plumes in the Pit 1, 2, and 3 areas as shown in Figure 3-1 (page 3-2). Figure 3-2 (page 3-3) has a tile layout showing the ISB treatment areas with plume overlays for PCE, TCE, cDCE, and VC. Treatment area nomenclature follows designations established for the 2021 RAWP Addendum (AECOM-Meadows 2021b).

3.2.1 Treatment Area TA-02 (Pit 1)

Treatment area TA-02 targets a hot spot area near the Pit 1 area that includes wells AEHADG-11 and OU7-MW-36, which have the highest PCE concentrations (1,300 μ g/L, March 2024) and TCE concentrations (780 μ g/L, March 2024) in the Pit 1 plume area. The design dimensions for TA-2 are 45 ft. x 50 ft. The 45 ft. width is oriented perpendicular to the direction of groundwater flow with a 50 ft. length parallel to the direction of groundwater flow. The target depth interval for treatment is 10 to 26 ft. (85 to 69 ft. NAVD88) following the treatment area approach implemented in 2021, which is based on well data and high resolution PDI completed in the area for contaminant profiling by MIP and hydraulic properties profiling by HPT³.

³ Design based on PDI location OU7-PDI-10-MHP and well data from AEHADG-11 and MW-36.





3.2.2 Treatment Area TA-03 (Pit 1)

Treatment area TA-03 targets well cluster CMT-21 a hot spot in the Pit 1 plume and nearby well OU7-MW-68. CMT-21 has the highest cDCE concentration (14,000 μ g/L, March 2024) in the Pit 1 plume area. The design dimensions for TA-3 are 45 ft. x 40 ft. The 45 ft. width is oriented perpendicular to the direction of groundwater flow with a 40 ft. length parallel to the direction of groundwater flow. The target depth interval for treatment is 9 to 24 ft. (85 to 70 ft. NAVD88) following the treatment area approach implemented in 2021, which is based on high resolution PDI completed in the area for contaminant profiling by MIP and hydraulic properties profiling by HPT⁴.

3.2.3 Treatment Area TA-05 (Pit 1)

Treatment area TA-05 targets well DMW-29A in the Pit 1 plume area, with reported TCE concentrations of 110 μ g/L for March 2024. This well is located 180 ft. downgradient of the former FTA Pit 1 area. The design dimensions for TA-3 are 40 ft. x 30 ft. The 50 ft. width is oriented perpendicular to the direction of groundwater flow with a 20 ft. length parallel to the direction of groundwater flow. The target depth interval for treatment is 10 to 28 ft. (85.5 to 66.5 ft. NAVD88) following the treatment area approach implemented in 2021, which is based on high resolution PDI completed in the area for contaminant profiling by MIP and hydraulic properties profiling by HPT⁵.

3.2.4 Treatment Area TA-06 (Pit 1)

Treatment area TA-06 targets well cluster CMT-20 in the Pit 1 plume area, with reported cDCE and VC concentrations of 220 μ g/L and 120 μ g/L, respectively, for March 2024. This well cluster is located 228 ft. downgradient of the former FTA Pit 1 area. The design dimensions for TA-6 are 25 ft. x 50 ft. The 25 ft. width is oriented perpendicular to the direction of groundwater flow with a 50 ft. length parallel to the direction of groundwater flow. The target depth interval for treatment is 10 to 30 ft. (84 to 64 ft. NAVD88) following the treatment area approach implemented in 2021 modified to extend the treatment from 26 ft. to 30 ft., which corresponds to the construction of injection well OU7-INJ-211 proposed for implementing the targeted treatment.

3.2.5 Treatment Area TA-07 (Pit 2)

Treatment area TA-7 targets a hot spot area in the Pit 2 area (near former FTA Pit 2) that includes wells MW-85 and MW-48, which has the highest cDCE concentration (9,200 µg/L, March 2024) in the Pit 2 plume area. The design dimensions for TA-07 are 40 ft. x 34 ft. The 40 ft. width is oriented perpendicular to the direction of groundwater flow with a 34 ft. length parallel to the direction of groundwater flow. The target depth interval for treatment is 20 to 32 ft. (80 to 68 ft. NAVD88) following the treatment area approach implemented in 2021, which is based on well data and high resolution PDI completed in the area for contaminant profiling by MIP and hydraulic properties profiling by HPT⁶.

3.2.6 Treatment Area TA-08 (Pit 2)

Treatment area TA-08 targets well cluster CMT-18 in the Pit 2 plume area, with the highest VC concentration (920 μ g/L) in the Pit 2 plume area and the second highest cDCE concentration (2,600 μ g/L) in the Pit 2 plume area. This well cluster is located 136 ft. downgradient of the former FTA Pit 2 area. The design dimensions for TA-8 are 45 ft. x 34 ft. The 45 ft. width is oriented perpendicular to the direction of groundwater flow with a 34 ft. length parallel to the direction of groundwater flow. The target depth interval for treatment is 17 to 28 ft. (84 to 64 ft. NAVD88) following the treatment area approach implemented in 2021, which is based on well data and high resolution PDI completed in the area for contaminant profiling by MIP and hydraulic properties profiling by HPT⁷.

Design based on PDI location OU7-PDI-11-MHP and well data from CMT-21 and MW-68.

⁵ Design based on PDI location OU7-PDI-15-MHP and well data from DMW-29A.

⁶ Design based on PDI location OU7-PDI-06-MHP and well data from MW-85 and MW-48.

⁷ Design based on PDI location OU7-PDI-07A-MHP and well data from CMT-18.

3.2.7 Treatment Area TA-09 (Pit 2)

Treatment area TA-09 targets the VC plume (greater than cleanup levels) that extends into lower Pit 2 monitored area near the installation fence line. The application area is 365 ft. to 400 ft. downgradient of the former Pit 2 area. The design dimensions for TA-9 are 40 ft. x 30 ft. The 40 ft. width is oriented perpendicular to the direction of groundwater flow with a 30 ft. length parallel to the direction of groundwater flow. The target depth interval for treatment is 16 to 26 ft. (78.5 to 68.5 ft. NAVD88) following the treatment area approach implemented in 2021 slightly modified for the construction of the injection wells proposed for implementation.

3.2.8 Treatment Area TA-11 (Pit 3)

Treatment area TA-11 targets wells OU7-MW-97, OU7-MW-108, and OU7-MW-109 in the Pit 3 plume area, with the highest cDCE concentration (2,500 μ g/L) and highest VC concentration (600 μ g/L) in the Pit 3 plume area. This group of wells is closely spaced and approximately 170 ft. downgradient of the former FTA Pit 3 area. The design dimensions for TA-11 are 40 ft. x 34 ft. The 40 ft. width is oriented perpendicular to the direction of groundwater flow with a 30 ft. length parallel to the direction of groundwater flow. The target depth interval for treatment is 11 to 25 ft. (88 to 74 ft. NAVD88) following the treatment area approach implemented in 2021, which is based on well data and high resolution PDI completed in the area for contaminant profiling by MIP and hydraulic properties profiling by HPT)8.

3.3 Substrate Selection

Section 3.3 describes the selection of substrates for enhanced ISB treatment at OU 7.

3.3.1 Source Zone and Hot Spot Treatment

For source zone and hot sport treatment, primary criteria to select a substrate for the ISB design is sustaining treatment for up to three (3) years, compatibility with a DPT treatment approach, and cost effectiveness. The substrate used for the ISB design must support reductive dechlorination; this includes implementable amendments for aquifer buffering and bioaugmentation to support complete reduction of COCs.

EVO is the selected ISB substrate comprised of food-grade soybean oil, emulsifiers, and amendments (e.g., mono and diglycerides, lactate, whey, etc.); it is widely available with demonstrated effectiveness to support enhanced reductive dechlorination (ERD). Evidence of complete ERD pathways to ethene and methane is apparent for previous EVO injections at OU 7 and treatability studies. The low solubility of EVO provides for a long-lasting carbon source due to its slow rate of chemical dissolution into groundwater. EVO can also help sequester chlorinated VOC compounds, which will further reduce their mobility in the aquifer (Borden 2006,5).

Terra Systems, Wilmington Delaware, will provide slow-release, emulsified vegetable oil substrate (small droplet identified as SRS®-SD EVO (60% soybean oil). Table 3-1 provides data on this EVO product. Bioaugmentation of the solution will use Terra Systems TSI DC (dehalococcoides mccartyi) to support consistent dechlorination across the treatment area and address existing cDCE and degradation products in the treatment area. This enriched culture contains greater than1E+11 Dehalococcoides cells per liter. The culture degrades PCE and TCE to ethene. The injection process will include sodium ascorbate (L-ascorbic acid, Vitamin C) as an additive to drive the injection water anaerobic for bioaugmentation injections.

Low alkalinity in groundwater and pH levels less than 6 in soil/groundwater will require buffering to maintain a near-neutral pH for enhanced ISB treatment. Previous buffering studies for treatability studies and injections recommended and have used of sodium bicarbonate for pH buffering. Buffering studies performed in 2021 recommended buffering dosage of 0.16 pounds of sodium bicarbonate per cubic foot of aquifer (Terra Systems 2021).

⁸ Design based on PDI location OU7-PDI-03-MHP and well data from MW-97, MW-108, MW-109.

Table 3-1 Terra Systems Inc. 60% Small Droplet Slow Release EVO Substrate (SRS® SD EVO)

Ingredient	Synonyms	CAS No.	Percent
Soybean oil	Soya oil	8001-22-7	60%
Emulsifiers and proprietary Nutri Plus nutrient package containing nitrogen, phosphorus, and vitamin B ₁₂		Mixture	5-15%
Sodium lactate	2-hydroxropionnic acid sodium salt	72017-3	<5%
Sodium bicarbonate ¹	Baking soda	144-5-8	0-1%
Calcium carbonate ¹	Lime	471-34-1	0-1%
Sodium carbonate ¹	Soda Ash	497-19-8	0-1%
Magnesium oxide ¹	Magnesia	1309-48-4	0-1%
Water		7732-18-5	20-26%

Notes: Source: Terra Systems, Inc. Safety Data Sheet for SRS®B) in Appendix C.1 ¹ Depending on the pH of the aquifer one or more of the above buffers (sodium bicarbonate, calcium carbonate, sodium carbonate or magnesium oxide) will be selected to adjust the pH of acidic aquifers to optimal levels for biodegradation.

Appendix A.1 has technical data sheets and safety data sheets for the selected EVO substrate, bioaugmentation, sodium ascorbate, and pH buffering components.

3.3.2 Diffuse Plume Treatment

For diffuse plume treatment, primary criteria to select a substrate for the ISB design is sustaining treatment for 1 or more years, substrate mobility with advective groundwater flow, compatibility with gravity feed injection in fixed-based injection wells, and cost effectiveness. The substrate used for the ISB design must support reductive dechlorination.

Sodium lactate is the selected ISB substrate comprised of sodium lactate, proprietary nutrients, and Vitamin B12; it is widely available with demonstrated effectiveness to support ERD). Evidence of complete ERD pathways to ethene and methane is apparent for previous sodium lactate injections at OU 7. The low viscosity and high solubility of sodium lactate in water allow for rapid transport with groundwater, which enhances distribution in the aquifer and minimizes the number of injection points.

Terra Systems, Wilmington Delaware, will provide sodium lactate (identified as 60% QRS™-Plus SL). Table 3-2 provides data on this sodium lactate product.

Table 3-2 Terra Systems Inc. 60% Sodium Lactate with Nutrients (60% QRS™-Plus SL)

Ingredient	Percent	Description.	Benefit
Sodium lactate	60%	Rapidly biodegradable soluble substrate; miscible in water	Fast release source of carbon and hydrogen. Rapidly generates reducing conditions. Provides 60% fermentable carbon.
Proprietary Nutrients	<5%		
pH	6.5-7	6.5-7	Optimum microbial activity

Notes: Source: Terra Systems, Inc. Technical Data Sheet for 60%-QRS™ Plus Sodium Lactate in Appendix C.1

Appendix A.2 has the technical data sheet and safety data sheet for the selected sodium lactate substrate.

3.4 Injection Process Option Selection

Section 3.4 describes the selection of injection process options for source zone/hot spot treatment and diffuse plume treatment.

3.4.1 Source Zone/Hot Spot Treatment

The injection process option selected for enhanced ISB for source zone/hot spot treatment at OU 7 is DPT injection using a pressure activated injection probe. The specified equipment is a Geoprobe® 7822DT Drill Rig with 1.5 inch probe rods. Pre-design investigations and testing performed at OU 7 indicate that the high density of the mid-to-lower intervals of surficial aquifer will require injection pressures greater than 100 pounds per square inch (psi) to distribute reagents in this zone. This 2021 ISB actions effectively used this process option to EVO with ZVI reagents in the surficial aquifer at OU 7.

3.4.2 Diffuse Plume Treatment

The injection process option selected for enhanced ISB for diffuse plume areas downgradient of the source zone/hot spot areas at OU 7 is existing injection wells used most recently for the 2017-2018 injections of sodium lactate.

3.5 Substrate Loading Rates and Injection Volume Estimates

Section 3.5 has design information for substrate loading rates and injection volume estimates.

3.5.1 Substrate Loading Rates

Enhanced ISB substrate mass and loading rates will need to satisfy native and contaminant electron acceptor demand in the reactive treatment zone to stimulate anaerobic reductive dechlorination processes. Too low of a substrate loading rate may result in reducing conditions that are insufficient to support anaerobic dechlorination of COCs. Too high of a substrate loading rate can lead to inefficiencies and uncontrolled reactions that lower pH and result in excessive methanogenesis, degradation of groundwater quality and/or accumulation of methane in the vadose zone. Determining appropriate substrate loading rates is therefore a primary objective of the enhanced ISB design.

Substrate demand for enhanced ISB of chlorinated VOCs is a function of: (1) contaminant electron acceptor supply, (2) native electron receptor supply, and (3) non-specific demands (microbial cell growth, etc.). Following previous pilot tests and remedial designs for OU 7, the theoretical demand for substrate is determined in this work plan through stoichiometric calculations using site data; these calculations quantify the amount of electron donor (hydrogen) required to completely reduce contaminant and native electron receptors based on the substrate used and levels of acceptors present.

The pore water of the aquifer and the solid aquifer matrix contain native electron receptors (such as DO and iron hydroxide materials) that the electron donor may use preferentially over chlorinated VOCs. Substrate loading rates in the enhanced ISB design account for the stoichiometric demand to completely reduce these native electron receptors before complete reductive dechlorination of COCs can occur.

Calculation of Substrate Demand and Loading Rates

The enhanced ISB design for this work plan addendum uses the Substrate Estimating Tool for Enhanced Anaerobic Bioremediation of Chlorinated Solvents Version 1.2 (ESTCP, 2010) to calculate substrate requirements, demand, and loading rates. Each source zone/hot spot treatment area has a design specific to the conditions found in the respective target treatment zones. The DPT injection treatment areas using EVO have a 3 year design period of performance assuming a single application event. The injection well treatment areas using sodium lactate to treat diffuse plume areas have a 1.5 year design period of performance assuming a single application event. Appendices B.1 through B.8 have the treatment design workbooks for TA-02, TA-03, TA-05, TA-06, TA-07, TA-08, TA-09, and TA-11. Table 3-2 (pages 3-7 through 3-10) has a summary of the enhanced ISB design parameter for each treatment area. Table 3-3 (page 3-11) has a summary of design workbook outputs including electron receptor demand and substrate requirements in hydrogen equivalents.

Table 3-3 Enhanced In Situ Bioremediation Design Parameter Summary

Treatment Zone	TA-02	TA-02 Notes	TA-03	TA-03 Notes
Area Description:		Hot spot area, wells AEHADG-11, MW-36, 35 ft. downgradient of Former Pit 1		CMT-21 Hot spot and MW-68, 100 ft. downgradient of Former Pit 1
Width (ft) ⊥ to GW flow x Length (ft) ∥ to GW flow x TZ Thickness (ft)	45' x 50' x 16'	Vertical treatment interval based on PDI MIP, HPT	45' x 40' x 15'	Vertical treatment interval based on PDI MIP, HPT
Design Period (yrs.)	3	Design period for enhanced in situ bioremediation	3	Design period for enhanced in situ bioremediation
Design Factor (times the electron acceptor hydrogen demand)	10	Electron acceptor (4X), microbial efficiency (4X), loss of substrate leaving reaction zone (2X)	10	Electron acceptor (4X), microbial efficiency (4X), loss of substrate leaving reaction zone (2X)
Aquifer Total Porosity (%) / Aquifer Effective Porosity (%)	0.36 / 0.27	PDI Eastover Formation OU 7/ average based on TP, AP, clay content	0.36 / 0.27	PDI Eastover Formation OU 7/ average based on TP, AP, clay content
Average Hydraulic Conductivity (ft/day)	5	OU7 PDI	5	OU7 PDI
Average Hydraulic Gradient (ft/ft)	1.61E-02	2024 Annual Report	1.61E-02	2024 Annual Report
Soil Bulk Density (gm/cm³)	1.63	Average OU- PDI for Eastover and Aquia Formations	1.663	Average OU- PDI for Eastover and Aquia
Soil Fraction of Organic Carbon (%)	0.28	PDI-38 Eastover Formation Saturated Zone OU-7 x 2	0.28	PDI-38 Eastover Formation Saturated Zone OU-7 x 2
Substrate	EVO	Terra Systems SRS®-SD EVO (small droplet, 60% soybean oil)	EVO	Terra Systems SRS®-SD EVO (small droplet, 60% soybean oil)
Native Electron Acceptors	TA-02		TA-03	Notes
Dissolved Oxygen (mg/L)	4.3	Data from 11/2020 (Average AEHADG-11, MW-36)	0.2	CMT-21-2, MW-68 (average) 03/2024
Nitrate (mg/L)	0.03	AEHADG-11 03/2024	0.02	CMT-21-2, MW-68 (average) 03/2024
Manganese (IV) (mg/L)	0.22	AEHADG-11 03/2024	1.3	CMT-21-2, MW-68 (average) 03/2024
Iron (III) (mg/L) (estimated as the amount of Fe II produced)	3	AEHADG-11 03/2024	3.43	Average MW-157 and MW-291 (May 2024)
Sulfate (mg/L)	16	AEHADG-11 03/2024	5	MW-68 03/2024
Carbon Dioxide (mg/L)	8	Estimate based on previous EISB injections	10	Estimated based on previous EISB injections
Contaminant Electron Acceptors	TA-02		TA-03	Notes
Tetrachloroethene (mg/L)	1.250	AEHADG-11 03/2024	0.500	Average MW-157 and MW-291 (May 2024)
Trichloroethene (mg/L)	0.520	AEHADG-11 03/2024	0.450	MW-157 (May 2024)
Dichloroethenes (mg/L)	1.300	AEHADG-11 03/2024	14.490	MW-291 (May 2024)
Vinyl Chloride (mg/L)	0.050	AEHADG-11 03/2024	0.140	MW-291 (May 2024)
Carbon Tetrachloride (mg/L)	0.040	AEHADG-11 03/2024	0.000	Not detected (2024)
Chloroform (mg/L)	1.250	AEHADG-11 03/2024	0.042	Not detected (2024)
Methylene chloride (mg/L)	0.000	AEHADG-11 03/2024	0.000	Not detected (2024)
Tetrachloroethanes (mg/L)	0.001	AEHADG-11 03/2024	0.000	Not detected (2024)
Trichloroethanes (mg/L)	1.100	AEHADG-11 03/2024	0.000	Not detected (2024)
Dichloroethanes (mg/L)	0.040	AEHADG-11 03/2024	0.003	MW-291 (May 2024)
Chloroethane (mg/L)	0.120	AEHADG-11 03/2024	0.000	Not detected (2024)
Aquifer Geochemistry	TA-02		TA-03	Notes
Oxidation Reduction Potential (mV)	234	AEHADG-11 03/2024	23	MW-283 (May 2024)
Temperature (°C)	12	AEHADG-11 03/2024	19	MW-283 (May 2024)
pH (standard units)	5.1	AEHADG-11 03/2024	5.4	Average MW-157 and MW-291 (May 2024)
Alkalinity (mg/L)	55	AEHADG-11 03/2024	64	Average MW-157 and MW-291 (May 2024)
Total Dissolved Solids (mg/L)	100	No data	100	No data
Specific Conductance (µs/cm)	158	AEHADG-11 03/2024	441	Average MW-157 and MW-291 (May 2024)
Chloride (mg/L)	6	AEHADG-11 03/2024	104	Average MW-157 and MW-291 (May 2024)
Sulfide – Pre Injection (mg/L)	0.1	Estimate Not Detected	0.1	Estimated
Sulfide – Post Injection (mg/L)	1.7	AEHADG-11 03/2024	2.6	MW-291 (May 2024)
Aquifer Matrix	TA-02		TA-03	Notes
Total Iron (mg/kg)	18000	CSM 2006 Soil Eastover Formation	18000	CSM 2006 Soil Eastover Formation
Cation Exchange Capacity meq/100 g	1	CSM 2006 Soil Eastover Formation	1	CSM 2006 Soil Eastover Formation

Notes: yrs. = years, mg/L = milligrams per liter, meq/100 g = milliequivalents per 100 grams, µs/cm = microsiemens per centimeter, CaCO3 = calcium carbonate, ft./ft. = feet per foot, ft./day = fee

Table 3-2 (Cont'd)

Treatment Zone	TA-05	TA-05 Notes	TA-06	TA-06 Notes
Area Description:		Targets well DMW-29A area, 180 ft. downgradient of Former Pit 1		Targets well cluster CMT-20, Uses existing injection well INJ-211, 230 ft. downgradient of Former Pit 1
Width (ft) ⊥ to GW flow x Length (ft) ∥ to GW flow x TZ Thickness (ft)	40' x 30' x 16'	Vertical treatment interval based on PDI MIP, HPT	25' x 50' x 20'	Vertical treatment interval based on PDI MIP, HPT
Design Period (yrs.)	3	Design period for enhanced in situ bioremediation	1.5	Design period for enhanced in situ bioremediation
Design Factor (times the electron acceptor hydrogen demand)	10	Electron acceptor (4X), microbial efficiency (4X), loss of substrate leaving reaction zone (2X)	10	Electron acceptor (4X), microbial efficiency (4X), loss of substrate leaving reaction zone (2X)
Aquifer Total Porosity (%) / Aquifer Effective Porosity (%)	0.36 / 0.27	PDI Eastover Formation OU 7/ average based on TP, AP, clay content	0.36 / 0.27	PDI Eastover Formation OU 7/ average based on TP, AP, clay content
Average Hydraulic Conductivity (ft/day)	5	OU7 PDI	5	OU7 PDI
Average Hydraulic Gradient (ft/ft)	1.61E-02	2024 Annual Report	1.61E-02	2024 Annual Report
Soil Bulk Density (gm/cm³)	1.63	Average OU- PDI for Eastover and Aquia Formations	1.663	Average OU- PDI for Eastover and Aquia Formations
Soil Fraction of Organic Carbon (%)	0.28	PDI-38 Eastover Formation Saturated Zone OU-7 x 2	0.28	PDI-38 Eastover Saturated Zone OU-7 x 2
Substrate	EVO	Terra Systems SRS®-SD EVO (small droplet, 60% soybean oil)	Sodium Lactate	Terra Systems 60% QRS-SL™ Plus Sodium Lactate
Native Electron Acceptors	TA-05		TA-06	Notes
Dissolved Oxygen (mg/L)	1.1	DMW-29A 03/2024	0.1	CMT-20-3 03/2024
Nitrate (mg/L)	0.05	DMW-29A 03/2024	0.03	CMT-20-3 03/2024
Manganese (IV) (mg/L)	0.46	DMW-29A 03/2024	1.3	CMT-20-3 03/2024
Iron (III) (mg/L) (estimated as the amount of Fe II produced)	9	DMW-29A 03/2024	10	Estimated
Sulfate (mg/L)	1	DMW-29A 03/2024	2	CMT-20-3 03/2024
Carbon Dioxide (mg/L)	10	Estimate based on previous EISB injections	10	Estimated based on previous EISB injections
Contaminant Electron Acceptors	TA-05		TA-06	Notes
Tetrachloroethene (mg/L)	0.006	DMW-29A 03/2024	0.000	CMT-20-3 03/2024
Trichloroethene (mg/L)	0.100	DMW-29A 03/2024	0.001	CMT-20-3 03/2024
Dichloroethenes (mg/L)	0.126	DMW-29A 03/2024	0.221	CMT-20-3 03/2024
Vinyl Chloride (mg/L)	0.007	DMW-29A 03/2024	0.120	CMT-20-3 03/2024
Carbon Tetrachloride (mg/L)	0.000	Not detected	0.000	Not detected (2024)
Chloroform (mg/L)	0.000	Not detected	0.000	Not detected (2024)
Methylene chloride (mg/L)	0.000	Not detected	0.000	Not detected (2024)
Tetrachloroethanes (mg/L)	0.000	Not detected	0.000	Not detected (2024)
Trichloroethanes (mg/L)	0.000	Not detected	0.000	Not detected (2024)
Dichloroethanes (mg/L)	0.000	Not detected	0.024	CMT-20-3 03/2024
Chloroethane (mg/L)	0.010	DMW-29A 03/2024	0.000	Not detected (2024)
Aquifer Geochemistry	TA-05		TA-06	Notes
Oxidation Reduction Potential (mV)	310	DMW-29A 03/2024	-64	CMT-20-3 03/2024
Temperature (°C)	14	DMW-29A 03/2024	16	CMT-20-3 03/2024
pH (standard units)	5.5	DMW-29A 03/2024	6.4	CMT-20 03/2024
Alkalinity (mg/L)	78	DMW-29A 03/2024	153	CMT-20 03/2024
Total Dissolved Solids (mg/L)	100	No data	100	No data
Specific Conductance (µs/cm)	750	DMW-29A 03/2024	479	CMT-20-3 03/2024
Chloride (mg/L)	19	DMW-29A 03/2024	125	CMT-20 03/2024
Sulfide – Pre Injection (mg/L)	0.1	Estimate Not Detected	0.1	Estimate
Sulfide – Post Injection (mg/L)	1.2	DMW-29A 03/2024	0.2	Estimate
Aquifer Matrix	TA-02		TA-06	Notes
Total Iron (mg/kg)	18000	CSM 2006 Soil Eastover Formation	11145	CSM 2006 Soil Eastover Formation
Cation Exchange Capacity meq/100 g	1	CSM 2006 Soil Eastover Formation	1	CSM 2006 Soil Eastover Formation
Neutralization Potential (percent as CaCO3)	1.0%	CSM 2006 Soil Eastover Formation	1.0%	CSM 2006 Soil Eastover Formation
/				

Table 3-2 (Cont'd)

Treatment Zone	TA-07	TA-07 Notes	TA-08	TA-08 Notes
Area Description:		Targets hot spot at well MW-85 and MW-48, 35 ft. downgradient of Former Pit 2		Targets well cluster CMT-20, Uses existing injection well INJ-211, 145 ft. downgradient of Former Pit 2
Width (ft) ⊥ to GW flow x Length (ft) ∥ to GW flow x TZ Thickness (ft)	40' x 34' x 12'	Vertical treatment interval based on PDI MIP, HPT	55' x 34' x 11'	Vertical treatment interval based on PDI MIP, HPT
Design Period (yrs.)	3	Design period for enhanced in situ bioremediation	1.5	Design period for enhanced in situ bioremediation
Design Factor (times the electron acceptor hydrogen demand)	10	Electron acceptor (4X), microbial efficiency (4X), loss of substrate leaving reaction zone (2X)	10	Electron acceptor (4X), microbial efficiency (4X), loss of substrate leaving reaction zone (2X)
Aquifer Total Porosity (%) / Aquifer Effective Porosity (%)	0.36 / 0.27	PDI Eastover Formation OU 7/ average based on TP, AP, clay content	0.36 / 0.27	PDI Eastover OU 7/ average based on TP, AP, clay content
Average Hydraulic Conductivity (ft/day)	5	OU7 PDI	5	OU7 PDI
Average Hydraulic Gradient (ft/ft)	1.20E-02	2024 Annual Report for Pit 2	1.61E-02	2024 Annual Report
Soil Bulk Density (gm/cm³)	1.63	Average OU- PDI for Eastover and Aquia Formations	1.663	Average OU- PDI for Eastover and Aquia Formations
Soil Fraction of Organic Carbon (%)	0.28	PDI-38 Eastover Saturated Zone OU-7 x 2	0.28	PDI-38 Eastover Formation Saturated Zone OU-7 x 2
Substrate	EVO	Terra Systems SRS®-SD EVO (small droplet, 60% soybean oil)	Sodium Lactate	Terra Systems 60% QRS-SL™ Plus Sodium Lactate
Native Electron Acceptors	TA-07		TA-08	Notes
Dissolved Oxygen (mg/L)	3.1	MW-85, MW-48 (average), 3/2024	0.1	CMT-18 03/2024
Nitrate (mg/L)	0.05	MW-85, MW-48 (average), 3/2024	0.03	CMT-18 03/2024
Manganese (IV) (mg/L)	0.44	MW-85, MW-48 (average), 3/2024	1.3	CMT-18 03/2024
Iron (III) (mg/L) (estimated as the amount of Fe II produced)	4	MW-85, MW-48 (average), 3/2024	10	CMT-18 03/2024
Sulfate (mg/L)	3.95	MW-85, MW-48 (average), 3/2024	0.5	CMT-18 03/2024
Carbon Dioxide (mg/L)	10	Estimate based on previous EISB injections	10	Estimate based on previous EISB injections
Contaminant Electron Acceptors	TA-07		TA-08	Notes
Tetrachloroethene (mg/L)	0.007	MW-85 3/2024	0.007	CMT-18 03/2024
Trichloroethene (mg/L)	0.007	MW-85 3/2024	0.006	CMT-18 03/2024
Dichloroethenes (mg/L)	9.318	MW-85 3/2024	2.609	CMT-18 03/2024
Vinyl Chloride (mg/L)	0.570	MW-85 3/2024	0.920	CMT-18 03/2024
Carbon Tetrachloride (mg/L)	0.000	Not detected	0.000	Not detected
Chloroform (mg/L)	0.000	Not detected	0.010	CMT-18 03/2024
Methylene chloride (mg/L)	0.000	Not detected	0.000	Not detected
Tetrachloroethanes (mg/L)	0.000	Not detected	0.000	Not detected
Trichloroethanes (mg/L)	0.023	MW-85 3/2024	0.000	Not detected
Dichloroethanes (mg/L)	0.018	MW-85 3/2024	0.027	CMT-18 03/2024
Chloroethane (mg/L)	0.000	Not detected	0.000	Not detected
Aquifer Geochemistry	TA-07		TA-08	Notes
Oxidation Reduction Potential (mV)	49	MW-85, MW-48 (average), 3/2024	249	CMT-18 03/2024
Temperature (°C)	16	MW-85, MW-48 (average), 3/2024	15	CMT-18 03/2024
pH (standard units)	5.8	MW-85, MW-48 (average), 3/2024	5.2	CMT-18 03/2024
Alkalinity (mg/L)	160	MW-85, MW-48 (average), 3/2024	93	CMT-18 03/2024
Total Dissolved Solids (mg/L)	100	No data	100	No data
Specific Conductance (µs/cm)	547	MW-85, MW-48 (average), 3/2024	339	CMT-18 03/2024
Chloride (mg/L)	110	MW-85, MW-48 (average), 3/2024	121	CMT-18 03/2024
Sulfide – Pre Injection (mg/L)	0.1	Estimate Not Detected	0.1	Estimate Not Detected
Sulfide – Post Injection (mg/L)	2.9	Data from MW-85 3/2024	1.6	CMT-18 03/2024
Aquifer Matrix	TA-07		TA-08	Notes
Total Iron (mg/kg)	18000	CSM 2006 Soil Eastover Formation	11145	CSM 2006 Soil Eastover Formation
Cation Exchange Capacity meq/100 g	1	CSM 2006 Soil Eastover Formation	1	CSM 2006 Soil Eastover Formation
Neutralization Potential (percent as CaCO3)	1.0%	CSM 2006 Soil Eastover Formation	1.0%	CSM 2006 Soil Eastover Formation

Table 3-2 (Cont'd)

Treatment Zone	TA-09	TA-09 Notes	TA-11	TA-11 Notes
Area Description:		Targets lower diffuse plume area at Pit 2, INJ-163, INJ-164, INJ-165, INJ-207		Targets hot spot in well MW-109, MW-108, and MW-97 area, 155 ft. downgradient of Former Pit 3
Width (ft) ⊥ to GW flow x Length (ft) ∥ to GW flow x TZ Thickness (ft)	40' x 30' x 10'	Vertical treatment interval based on injection well construction	40' x 34' x 14'	Vertical treatment interval based on PDI MIP, HPT
Design Period (yrs.)	1.5	Design period for enhanced in situ bioremediation (soluble substrate)	3	Design period for enhanced in situ bioremediation
Design Factor (times the electron acceptor hydrogen demand)	10	Electron acceptor (4X), microbial efficiency (4X), loss of substrate leaving reaction zone (2X)	10	Electron acceptor (4X), microbial efficiency (4X), loss of substrate leaving reaction zone (2X)
Aquifer Total Porosity (%) / Aquifer Effective Porosity (%)	0.36 / 0.27	PDI Eastover Formation OU 7/ average based on TP, AP, clay content	0.36 / 0.27	PDI Eastover Formation OU 7/ average based on TP, AP, clay content
Average Hydraulic Conductivity (ft/day)	5	OU7 PDI	5	OU7 PDI
Average Hydraulic Gradient (ft/ft)	2.23E-02	Between injection transect and MW-149 (03/2024)	1.80E-02	Flow path in injection area 03/2024
Soil Bulk Density (gm/cm³)	1.63	Average OU- PDI for Eastover and Aquia Formations	1.663	Average OU- PDI for Eastover and Aquia Formations
Soil Fraction of Organic Carbon (%)	0.28	PDI-38 Eastover Saturated Zone OU-7 x 2	0.28	PDI-38 Eastover Saturated Zone OU-7 x 2
Substrate	Sodium Lactate	Terra Systems 60% QRS-SL™ Plus Sodium Lactate	EVO	Terra Systems SRS®-SD EVO (small droplet, 60% soybean oil)
Native Electron Acceptors	TA-09		TA-11	Notes
Dissolved Oxygen (mg/L)	0.5	CMT-18 03/2024	0.4	MW-97 03/2024
Nitrate (mg/L)	0.05	CMT-18 03/2024	0.05	MW-97 03/2024
Manganese (IV) (mg/L)	0.726	CMT-18 03/2024	0.72	MW-95 03/2024
Iron (III) (mg/L) (estimated as the amount of Fe II produced)	3	CMT-18 03/2024	10	Estimated
Sulfate (mg/L)	0.5	CMT-18 03/2024	2	MW-95 03/2024
Carbon Dioxide (mg/L)	10	Estimate based on previous EISB injections	10	Estimated based on previous EISB injections
Contaminant Electron Acceptors	TA-09		TA-11	Notes
Tetrachloroethene (mg/L)	0.041	MW-80 03/2024	0.000	Not detected
Trichloroethene (mg/L)	0.020	MW-80 03/2024	0.001	MW-109 03/2024
Dichloroethenes (mg/L)	0.224	MW-80 03/2024	2.404	MW-109 03/2024
Vinyl Chloride (mg/L)	0.054	MW-80 03/2024	0.560	MW-109 03/2024
Carbon Tetrachloride (mg/L)	0.000	Not detected	0.000	Not detected
Chloroform (mg/L)	0.000	Not detected	0.000	Not detected
Methylene chloride (mg/L)	0.000	Not detected	0.000	Not detected
Tetrachloroethanes (mg/L)	0.000	Not detected	0.000	Not detected
Trichloroethanes (mg/L)	0.003	MW-148 03/2024	0.000	Not detected
Dichloroethanes (mg/L)	0.003	Average for plume area	0.000	Not detected
Chloroethane (mg/L)	0.000	Not detected	0.000	Not detected
Aquifer Geochemistry	TA-09		TA-11	Notes
Oxidation Reduction Potential (mV)	71	MW-80, MW-83, MW-148 (average) 03/2024	-56	MW-98 03/2024
Temperature (°C)	13	MW-80, MW-83, MW-148 (average) 03/2024	19	MW-98 03/2024
pH (standard units)	5.9	MW-80, MW-83, MW-148 (average) 03/2024	6.4	MW-98 03/2024
Alkalinity (mg/L)	54	MW-80, MW-83, MW-148 (average) 03/2024	72	MW-98 03/2024
Total Dissolved Solids (mg/L)	100	Estimate	100	Estimated
Specific Conductance (µs/cm)	533	MW-80, MW-83, MW-148 (average) 03/2024	269	MW-98 03/2024
Chloride (mg/L)	178	MW-80, MW-83, MW-148 (average) 03/2024	24	MW-98 03/2024
Sulfide – Pre Injection (mg/L)	0.1	Estimate	0.1	Estimate
Sulfide – Post Injection (mg/L)	0.2	Estimate	0.2	Estimate
Aquifer Matrix	TA-09		TA-11	Notes
Total Iron (mg/kg)	18000	CSM 2006 Soil Eastover Formation	11145	CSM 2006 Soil Eastover Formation
Cation Exchange Capacity meq/100 g	1	CSM 2006 Soil Eastover Formation	1	CSM 2006 Soil Eastover Formation
Neutralization Potential (percent as CaCO3)	1.0%	CSM 2006 Soil Eastover Formation	1.0%	CSM 2006 Soil Eastover Formation

Table 3-4 Summary of Electron Receptor Requirements in Hydrogen Equivalents

	TA	-02	TA	-03	TA	-05	TA	\-06	TA	-07	TA	-08	TA	-09	TA	-11
Parameter	% of Total	H Demand (lbs.)														
Aerobic Respiration	7.5%	2.475	0.5%	0.121	2.6%	0.599	0.2%	0.018	6.2%	0.876	1.9%	0.285	0.9%	0.074	0.9%	0.180
Nitrate Reduction	0.0%	0.011	0.0%	0.009	0.1%	0.021	0.0%	0.004	0.1%	0.011	0.3%	0.043	0.1%	0.006	0.1%	0.018
Sulfate Reduction	18.8%	6.234	6.2%	1.646	0.8%	0.182	3.4%	0.332	5.3%	0.744	0.8%	0.119	18.8%	1.504	2.6%	0.538
Manganese Reduction	0.1%	0.037	0.8%	0.199	0.3%	0.073	0.9%	0.087	0.3%	0.036	0.4%	0.064	0.4%	0.032	0.5%	0.100
Iron Reduction	0.9%	0.285	1.0%	0.258	3.0%	0.703	3.3%	0.324	1.1%	0.150	1.1%	0.164	2.8%	0.220	3.3%	0.680
Methanogenesis	69.2%	22.961	79.3%	20.954	93.0%	21.741	91.9%	9.032	80.0%	11.276	92.8%	14.213	76.7%	6.124	90.5%	18.947
Dechlorination	3.6%	1.184	12.3%	3.253	0.2%	0.051	0.3%	0.029	7.1%	0.999	2.8%	0.427	0.3%	0.021	2.2%	0.470
Total	100%	33.19	100%	26.4	100%	23.37	100%	9.83	100%	14.09	100%	15.32	100%	7.98	100%	20.93
Hydrogen Demand (lbs./gal)		6.06E-05		5.29E-05		4.51E-05		4.56E-05		5.24E-05		4.52E-05		5.46E-05		4.63E-05
Hydrogen Demand (g/L)		7.26E-03		6.34E-03		5.40E-03		5.47E-03		6.28E-03		5.41E-03		6.55E-03		5.55E-03
Substrate		EVO		EVO		EVO	Sodiu	ım Lactate		EVO		EVO	Sodi	um Lactate		EVO
Substrate equivalents (10X)		4,810 lbs.		3,832 lbs.		3,387 lbs.		4,554 lbs.		2,043 lbs.		2,220 lbs.		3,699 lbs.		3,034 lbs.
Substrate equivalents (10X)		617 gal		491 gal		434 gal		414 gal		262 gal		285 gal.		336 gal.		389 gal
Effective Concentration ¹		632 mg/L		551 mg/L		470 mg/L	1	,221 mg/L		546 mg/L		471 mg/L	,	1,463 mg/L		483 mg/L

Notes: % = percent, lbs. = pounds, lbs./gal = pounds per gallon, g/L = grams per liter, mg/L = milligrams per liter., EVO = emulsified vegetable oil, 10X = 10 times design factor, ¹effective concentration is for total volume of groundwater treated.

For this work plan addendum, the enhanced ISB designs use data from each treatment area when available. The design for this work plan also applies a design factor of 10 to the calculated total hydrogen demand to account for microbial efficiency (4-times design factor), uncertainties in electron acceptor demand (4-times design factor), and loss of substrate leaving the reaction zone (2-times). The three individual design factors sum to a total design factor of 10. The design or safety factor used for enhanced ISB designs typically ranges from 2 to 10 (AFCEC 2004).

3.5.2 Design Radius of Influence and Injection Volume

Section 3.5.2 describes enhanced ISB design parameters for substrate distribution. In conjunction with sufficient substrate loading, substrate distribution is another critical design parameter. Distribution design parameters ROI, mobile porosity of the targeted formation zone, and injection volume corresponding to the design ROI and mobile porosity.

Design Radius of Influence (ROI)

A design ROI of 12 ft is established for the proposed treatment areas in this work plan addendum. This ROI is based on using the higher-pressure DPT tooling targeting mid-to-deeper intervals of the surficial aquifer that have higher density than shallower intervals and results of previous DPT injections at OU 7. Injection well spacing in proposed treatment areas TA-6 and TA-9 is 18 ft. This well spacing corresponds to a 25 percent overlap for the design radius of influence.

Injected fluids travel principally through the mobile fraction of the surficial aquifer, which is a fraction (or percentage) of the total porosity of the bulk matrix. The mobile fraction or porosity serves as a correction factor to determine the distance injection fluids travel based on the injection volume introduced into the aquifer bulk matrix (Suthersan et al, 2017, 177). Injection tracer testing performed at various locations for porous aquifer media have been used to estimate mobile porosity and established empirical relationships between aquifer mobile fraction, target radial distribution for injection, and injection volume (Suthersan et al. 2017, 177). Equation 3.1 illustrates the mathematical relationship between these parameters (Suthersan et al. 2017, 177).

Equation 3.1:
$$r_{inj} = \sqrt{\frac{V_{inj}}{\pi x h x \theta_m}}$$

Where:

h = injection zone thickness $\theta_m = mobile fraction (porosity)$ $V_{inj} = injected volume$

 π = pi mathematical constant approximately equal to 3.14159

A mobility porosity of 0.02 (2%) is estimated based on the 2007 treatability study performed at OU 7. This falls within the expected range of 0.02 to 0.10 (Payne et al. 2007, 67)⁹.

Injection Volume

The estimated mobile porosity (0.02) determined from the 2007 pilot test data is used as a design parameter input along with ROI and target injection interval to determine target injection volumes for the treatment area. Equation 3.2 is a form of Equation 3.1 to solve for injection volume.

Equation 3.2:
$$V_{inj} = \pi \times h \times r_{inj}^2 \times \theta_m$$

Where:

h = injection zone thickness

 θ_m = mobile fraction (porosity)

V_{ini} = injected volume

 π = pi mathematical constant approximately equal to 3.14159

 r_{inj}^2 = radius occupied by the injected fluid immediately after injection is completed to the second power.

⁹ Payne F.C., J. A. Quinnan, and S. T. Potter 2007. Remediation Hydraulics. CRC Press. Page 67, 432 p.

Appendix B.9 uses Equation 3.2 to calculate injection volumes for each treatment area. Each vertical interval for DPT injection points will receive approximately 68 gallons per linear ft. Table 3-4 has a summary of the injection design and volumes for each injection point and treatment intervals with dosage levels for EVO and sodium lactate.

Table 3-5 Injection Design, Volumes, and Substrate Loading Rates

Treatment Area	Process Option	Dim.	No. of IPs	IP Spacing (ft)	Row Spacing (ft)	Injection Interval (ft BGS)	Vol. per LF (gal)	Vol. per IP (gal)	Total Vol. (gal)	EVO Vol. (gal)	Sodium Lactate Vol. (gal)	Reagent Dosage (%)
TA-02	DPT	45' x 50'	13	20	10	10-26	68	1,088	14,144	617		4.6%
TA-03	DPT	45' x 40'	10	20	10	9-24	68	1,020	10,200	491		5.1%
TA-05	DPT	40' x 30'	6	20	10	10-28	68	1,224	7,344	434		6.3%
TA-06	IW/GF	20' x 50'	1			10-30			475		425	
TA-07	DPT	40' x 34'	7	20	10	20-32	68	816	5,712	262		4.0%
TA-08	DPT	45' x 34'	8	20	10	17-28	68	748	5,236	233		4.1%
TA-09	IW/GF	40' x 30'	4	18		16-26	68		450		400	
TA-11	DPT	40' x 34'	7	20	10	11-25	68	952	6,664	389		6.2%
Total			56						56,372	2,426	825	

Reagent Amendments

Amendments to the prepared dilute EVO solution will include sodium bicarbonate for pH buffering and sodium ascorbate to drive the injection water anaerobic for bioaugmentation culture injections. Amendments amounts calculated for sodium bicarbonate and sodium ascorbate are as follows:

•

- 1,555 pounds of sodium bicarbonate for TA-02 corresponding to 0.16 pounds of sodium bicarbonate per cubic foot of aquifer for a treatment zone effective pore volume of 9,720 cubic feet (72,725 gallons¹⁰). The corresponding load rate is 0.12 pounds per gallon of dilution/chase water.
- 1,166 pounds of sodium bicarbonate for TA-03 corresponding to 0.16 pounds of sodium bicarbonate
 per cubic foot of aquifer for a treatment zone effective pore volume of 7,290 cubic feet (55,544
 gallons¹¹). The corresponding load rate is 0.12 pounds per gallon of dilution/chase water.
- 933 pounds of sodium bicarbonate for TA-05 corresponding to 0.16 pounds of sodium bicarbonate per cubic foot of aquifer for a treatment zone effective pore volume of 5,831 cubic feet (43,635 gallons¹²). The corresponding load rate is 0.14 pounds per gallon of dilution/chase water.
- 705 pounds of sodium bicarbonate for TA-07 corresponding to 0.16 pounds of sodium bicarbonate per cubic foot of aquifer for a treatment zone effective pore volume of 4,406 cubic feet (32,969 gallons¹³). The corresponding load rate is 0.13 pounds per gallon of dilution/chase water.
- 722 pounds of sodium bicarbonate for TA-08 corresponding to 0.16 pounds of sodium bicarbonate
 per cubic foot of aquifer for a treatment zone effective pore volume of 4,544 cubic feet (33,999
 gallons¹⁴). The corresponding load rate is 0.14 pounds per gallon of dilution/chase water.

¹⁰ Treatment zone effective pore volume for Appendix B.1 Part 1

¹¹ Treatment zone effective pore volume for Appendix B.2 Part 1

¹² Treatment zone effective pore volume for Appendix B.3 Part 1

¹³ Treatment zone effective pore volume for Appendix B.5 Part 1

¹⁴ Treatment zone effective pore volume for Appendix B.6 Part 1

- 822 pounds of sodium bicarbonate for TA-011 corresponding to 0.16 pounds of sodium bicarbonate
 per cubic foot of aquifer for a treatment zone effective pore volume of 5,140 cubic feet (38,463
 gallons¹⁵). The corresponding load rate is 0.13 pounds per gallon of dilution/chase water.
- 135 pounds of sodium ascorbate for TA-02 for making 13,527 gallons of anaerobic water at the rate of 10 pounds per 1,000 gallons of injection water.
- 97 pounds of sodium ascorbate for TA-03 for making 9,709 gallons of anaerobic water at the rate of 10 pounds per 1,000 gallons of injection water.
- 69 pounds of sodium ascorbate for TA-05 for making 6,910 gallons of anaerobic water at the rate of 10 pounds per 1,000 gallons of injection water.
- 54 pounds of sodium ascorbate for TA-07 for making 5,450 gallons of anaerobic water at the rate of 10 pounds per 1,000 gallons of injection water.
- 50 pounds of sodium ascorbate for TA-08 for making 5,003 gallons of anaerobic water at the rate of 10 pounds per 1,000 gallons of injection water.
- 63 pounds of sodium ascorbate for TA-11 for making 6,275 gallons of anaerobic water at the rate of 10 pounds per 1,000 gallons of injection water.

3.6 Treatment Area Configuration and Injection Points

Figure 3-1 shows the layout and configuration of the enhanced ISB treatment areas TA-02, TA-03, TA-05, TA-06, TA-07, TA-08, TA-09, and TA-11. DPT treatment areas have offset rows of injection points with each row having a spacing of 20 ft. between injection points with a row spacing of 10 ft. This provides for a 20% overlap along each row for the design ROI of 12 ft. (perpendicular to groundwater flow) with the row spacing providing a 50% overlap parallel to the direction of groundwater flow. TA-06 uses a single injection well to target well cluster CMT-20 and TA-09 uses four injection wells to target diffuse plume areas in the downgradient area of the VOC plume associated with Pit 2.

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¹⁵ Treatment zone effective pore volume for Appendix B.8 Part 1

4. Remedial Action Field Activities

Section 4 describes field activities associated the proposed remedial actions described in the RAWP Addendum. The scope of work for proposed remedial actions will include installation of two monitoring wells for site groundwater monitoring, ISB injections, and remedy performance monitoring.

4.1 Utility Clearance

Utility avoidance will include marking of proposed monitoring well and DPT injection point locations for utility clearance following the DSCR dig permit process including:

- Meadows or their designated contractor will contact the Virginia One Call Center (811) for mark out of utility locations. A minimum of three-day notice is required for 811 notification.
- Meadows will coordinate and provide notification to DSCR Installation Management for utility designation and location in the proposed disturbance areas.
- Meadows will contract with a private utility locating company to survey and mark the proposed disturbance areas (20 ft. scan radius) using ground penetrating radar and magnetic locating equipment.
- The project team will review of available utility maps and other information when proposing subsurface intrusion and disturbance locations (i.e., boring and wells).

The planned locations for utility clearance are in the area of existing well cluster CMT-18 in the Pit 2 monitored area and in the proposed enhanced ISB treatment areas (DPT areas) shown in Figure 3-1.

Per previous regulatory correspondence, an underground injection control permit is not required for the proposed ISB injections (Appendix C).

4.2 Field Survey of Locations and Monitoring Well Survey

The project geodatabase in the geographic information system (GIS) will contain the spatial location information for the proposed monitoring wells and design locations for DPT injection points. For each location, this will include: 1) horizontal coordinates (northing and easting) using the North American Datum of 1983, State Plan – Virginia South, and 2) vertical elevation (North American Vertical Datum of 1988) estimated using the horizontal coordinates in the digital elevation model¹⁶ for OU 7.

The field team will locate the proposed monitoring wells and established DPT injection points in the field using a Trimble handheld global positioning system (GPS) unit with submeter accuracy. The GPS unit has a general design accuracy of 10 millimeters (mm) + 1 part per million for horizontal and 15 mm + 1 ppm for vertical. If boring offsets are required, the project team will use the GPS to determine the revised horizontal coordinates to update the project GIS geodatabase.

Surveying of new monitoring wells by a Commonwealth of Virginia licensed surveyor will occur and include: 1) survey of the horizontal coordinates (northing and easting) of each monitoring well using the North American Datum of 1983, State Plane – Virginia South, and 2) survey of the vertical elevation of each monitoring well using the North American Vertical Datum of 1988 (NAVD88) including the elevation of the top of the inner well casing used for measuring water levels and the ground surface adjacent to the well location.

4.3 Monitoring Well Installation

Proposed actions for this RAWP addendum will include installation of two monitoring wells for remedy monitoring at OU 7 in the area of existing well cluster CMT-18 (see Figure 3-1). Monitoring wells designated OU7-MW-151 and OU7-MW-152 are proposed at adjacent locations to monitor groundwater in the surficial aquifer and confined aquifer, respectively. The proposed spacing of these wells is 7.5 ft. on center east to west perpendicular to groundwater flow. Table 4-1 (page 4-2) has summary information for

¹⁶ Digital Elevation Model, Virginia Geographic Information Network, https://vgin.vdem.virginia.gov/search?tags=dem

the proposed monitoring well construction including boring depth, well completion depth, well type, well screen specifications, and screen interval.

Table 4-1 Monitoring Well Construction

	Dept	th (ft. BGS)		Well Construction	
Well ID	Boring	Well Completion	Well Type	Well Screen	Screen Interval (ft. BGS)
OU7-MW-151	28.5	28	FM 2-inch Sch 40 PVC	2.0-inch ID, 5 ft. pre-pack 20/40 mesh sand, 0.010-inch slots	23 to 28
OU7-MW-152	50.5	50	FM 2-inch Sch 40 PVC	2.0-inch ID, 5 ft. pre-pack 20/40 mesh sand, 0.010-inch slots	45 to 50

Notes: FM = flush mount, Sch 40 PVC = schedule 40 polyvinyl chloride, ID = inside diameter, ft. BGS = feet below ground surface.

The drilling method for monitoring well installation is hollow-stem auger (HSA) drilling following the procedures outlined in ASTM designation D5784/D5784M-18 Standard Guide for Use of Hollow-Stem Augers for Geoenvironmental Exploration and the Installation of Subsurface Water Quality Monitoring Devices. The drilling procedure for borings will use 4.25-inch inner diameter HSA (9-inch borehole diameter) to provide for sufficient borehole diameter for installation of single cased monitoring wells (Type II) constructed with 2-inch inner diameter well casing and screen.

HSA operations will involve soil sampling with a 2-ft split barrel sampler following ASTM designation D1586/D1586M-18 Standard Test Method for Standard Penetration Test (SPT) and Split-Barrel Sampling of Soils to characterize lithologic conditions and verify well element installation zones. Drilling for OU7-MW-151 will use continuous SPT sampling to 28 ft BGS. Drilling for adjacent OU7-MW-152 will use continuous SPT soil sampling from 20 ft. BGS to 50 ft. bgs.

Monitoring well installation procedures will follow ASTM designation D5092/D5092M-16 Standard Practice for Design and Installation of Groundwater Monitoring Wells. A field geologist will oversee drilling operations, sampling, and monitoring well installation and prepare boring logs and construction documentation for each monitoring well location. Boring logs will include the location, geotechnical data, and sample description information for each material identified in the borehole using symbol and word descriptions. Description and identification of soils will follow ASTM designation D2488-17e1 Standard Practice for Description and Identification of Soils (Visual-Manual Procedures). Schematic completion diagrams will document well construction.

Well construction will consist of 2-inch inner diameter, Schedule 40 polyvinyl chloride (PVC) with 10-slot (0.010 inch slotted) PVC screen. Placement of an approximate 3 ft. thick fine-sand filter above the prepack screen will create an annular seal, followed by 2-ft. thick bentonite seal, and a bentonite-cement grout within the remaining annulus to the surface. Surface completion with include a flush mouth manhole to match existing grade with affixed well identification labels.

4.4 Well Development

Development of newly installed monitoring wells will commence 24 to 48 hours after emplacement of the final bentonite grout annual seal to allow sufficient time for curing. Well development will remove suspended solids from disturbance of geologic materials during installation and improve the hydraulic characteristics of the filter pack and the hydrologic unit adjacent to the intake. The well development process will include two phases: preliminary development and development. Development will follow procedures identified in ASTM designation D5521/D5521M-18 Standard Guide for Development of Groundwater Monitoring Wells in Granular Aquifers.

Preliminary development will involve mechanical surging, bailing, and potentially other techniques such as air lift pumping to apply sufficient energy in the well to address potential formation damage from the drilling process and remove fine-grained sediment from the screen, filter pack, and geologic formation adjacent to the filter pack. Gradual application of the well development will occur with an increase in intensity, if the well responds to the processes with an increase yield of water and fine-grained sediment.

The final phase of development will involve well pumping with the degree of over pumping, surging, and backwashing determined based on the results of the predevelopment. Development will continue until the discharge water from the well is visibly clear, or until the turbidity of water is reduced to the extent practical. Additional criteria for completion of well development is the stabilization of water quality indicator parameters pH, temperature, specific conductance, oxidation-reduction potential, and dissolved oxygen.

4.5 Enhanced ISB Injection Field Implementation

This section describes field implementation activities and methods for DPT ISB injections. Section 3 describes the proposed injection point locations and specifications for implementation.

4.5.1 Mobilization and Setup

Site mobilization will include delivery of reagents and rental equipment to the site at OU 7. The injection subcontractor will mobilize to the site with personnel, drill rig equipment, injection system equipment, and support materials. Initial activities at the site will include setup of the work area and construction of a secondary containment area to house the injection system. Water tests of all equipment and pumps will verify system integrity. The contractor will use a forklift to stage equipment and reagents within the injection area. EVO is containerized in intermediate bulk container (IBC) totes with a 275-gallon capacity.

Components of the injection system include a high-pressure injection pump equipped with two mix tanks with pneumatic paddle mixers, and a single point manifold equipment with a flow meter and pressure gauge. A single air compressor will power this system. Site control measures will include traffic cones and cone bars to delineate the work area exclusion zone. The injection subcontractor will have spill kits and portable vacuums in the work area for immediate deployment, if needed.

4.5.2 Injection/Reagent Application

Section 4.3.2 describes injection procedures and reagent application for DPT injections and injection wells.

4.5.2.1 **DPT Injections**

Reagent amendment preparation will include dilution of the vendor provided EVO substrate to the design loading rate (proportions) in Table 3-4. The amount of water needed for the injection will require the use of the installation water system supplied by a hydrant in the OU 7 work area. Following preparation of the injection substrate, the injection subcontractor will thoroughly batch mix in the appropriate mass of sodium bicarbonate for pH buffering and sodium ascorbate to drive the water anaerobic for bioaugmentation culture injections.

DPT drilling (Geoprobe 7822® track unit) will advance temporary injection points and use 1.5-inch diameter, pressure-activated injection probe (nozzles) tooling for reagent application pumped through the drilling rods. Each injection point location will have a 1.5-inch-high pressure, stainless steel threaded injection caps and 1-inch diameter high pressure injection hose connected to the cap. Each cap will have a pressure gauge and pressure relief valve.

Use of the pressure-activated injection probe (activated at a pressure of 100 to 120 psi) will allow for targeted placement of the reagent laterally into the dense strata of the confined aquifer (Potomac Formation). Distribution of the reagent into the confined aquifer will require injection pressures greater than 100 psi because of the higher density of the strata. Once flow is established and the tool is open, pressure may increase or decrease depending on the subsurface conditions. Anticipated injection pressures are in the range of 150 psi to 250 psi. The probe assembly prevents backflow of injection material through the tool string and keep soil out of the tool string during advancement and retraction.

The pressure-activated injection probe can perform top-down or bottom-up injections, with a bottom-up approach planned for the site. Bottom-up injections will start by advancing the tool string to the bottom of the injection interval. Injections will occur at this interval by pumping reagents through the tool string under pressure that in turn activates the injection probe for 360-degree reagent distribution through the probe nozzles. The injection tooling will then work incrementally upward through entire injection interval in

each injection point using the same injection process to provide overlapping coverage. A 2-ft. injection interval is anticipated for the site.

If the injectate delivery is not successful to a selected depth interval, injection of the remaining volume will occur at an adjacent depth interval within the same injection point or to the same depth interval at an adjacent injection point. Adjustment of the injection depths and/or volumes will occur in real-time throughout the injections to optimize reagent delivery into the subsurface while limiting the potential for surfacing of the injectate. If daylighting occurs, the injection contractor will discontinue injections at that interval causing daylighting. Site conditions may require adjustment of the conceptual injection layouts and corresponding injection activities if conditions vary significantly from design and implementation assumptions.

The scope of work will include performance of water injection test at the first DPT injection location with approximately 15 gallons of potable water to establish flow rates/pressures and confirm integrity of the injection system and hoses.

DPT injection point abandonment will occur after completion of the injection activities and include removal of the downhole rod string and probe assembly and completely plugging/sealing the boring with bentonite-cement grout. Locations performed in pavement areas will include asphalt or concrete patch to match existing grade. Offsite locations will place native soil on top of the grout to match existing grade.

4.5.2.2 Injection Wells

Preparation for injections at TA-06 and TA-09 will include well inspections to verify integrity. Redevelopment of these wells will occur after inspection to restore any reduced hydraulic connection to the surficial aquifer from previous use in 2018¹⁷. The field team will containerize the redevelopment water and transfer this water to the Building 40 waste area.

The process option selected for injection of soluble substrate (sodium lactate) is gravity feed following the successful use of this method for the 2017-2018 injections in these injection wells. Gravity feed will include flush water at each well location with the planned quantities of reagent and water identified in Table 3-4.

Setup for gravity feed injections will occur at individual well locations (INJ-211 in TA-06 and INJ-207 in TA-09) and at the well transect in TA-09. The gravity flow setup will include the use of an IBC spill containment pallet when performing gravity feed injections. Appendix A.6 has a manufacturer's technical data sheet for proposed the spill containment pallet. This pallet has a secondary containment capacity of 365 gallons to contain 132% of the full contents of the 275-gallon tote with accommodation for precipitation. The spill pallet has dimensions of 62-inches by 62 inches, a 52-inch by 52-inch deck and has a 28-inch height that when combined with the height of tote drain valve will facilitate gravity flow and adjustment of flow rate. A bucket shelve provides additional containment for connection and dispensing. The spill pallet will also have a drain valve and pullover cover to prevent precipitation from entering the sump when not in use. Forklift slots on the spill pallet facilitate movement and placement of the pallet at the desired locations.

IBC setup for gravity feed operations will include installation of a liquid level gauge in the tote to monitor flow rates, reagent usage, and control application to the design volume (see Appendix A.6). Gravity feed operations setup at each well will include connection of a feed assembly from the IBC tote to the injection well. This assembly will consist of 0.75-inch braided PVC hose that has a 2-inch female National Pipe Straight (NPS) fitting to connect directly to the IBC tote that has a 2-inch drain valve with male-quick disconnect coupler. This PVC hose will extend from the tote and connect to a well head assembly with a coupled fitting configured with a drop tube for gravity feed. The well head assembly will have a fitting to attach a pressure indicator for injection process monitoring.

Setup for gravity injections will include traffic cones and cone bars to delineate the work area exclusion zone. Other actions for setup will include staging and deployment of spill prevention measures in addition to the IBC spill pallet.

¹⁷ Injection well locations for inspection include INJ-211 in TA-06 and INJ-163, INJ-164, INJ-165, and INJ-207 in TA-09. USACE used these wells for gravity feed injections of sodium lactate followed by flush water to clear well screens.

4.6 Injection Process Monitoring

Table 4-2 describes process monitoring that will occur during the injections performed at OU 7. Injection data will track injection progress relative to the design and identify variations in physical and hydraulic properties of the confined aquifer. Water level measurements (electronic water level indicator) at monitoring well locations in the vicinity of the injection areas will monitor hydraulic influence from injections. Visual checks and water quality measurements at monitoring wells in the vicinity of the injection areas will evaluate distribution of reagents in the targeted areas. Leading indicators at monitoring wells include visual evidence of reagents (cloudy, watercolor change, and odor) and changes in water quality parameters including increased specific conductivity and turbidity levels.

Table 4-2 Remedy Installation Monitoring

Monitoring Element	Parameters	Measures	Locations	Frequency
Injection data	Daily field conditions, injection intervals, measured pressures, injection volumes, and flows at each injection location	Injection performance vs. design, variations in hydrostratigraphy	Equipment system and injection points	Daily and cumulative subcontractor injection logs
Hydraulic data	Water level measurement	Injection effects on aquifer	TA-02: AEHDG-11, MW-36, DMW-35A, MW-123	
Visual parameters	Bailer checks of monitoring wells in vicinity of injection area	Distribution of injectate in treatment area	TA-03: CMT-21, MW-60, MW-68 TA-05: DMW-29A, AEHDG-10, EBF-05S, INJ-154, INJ-155	Baseline before
Water quality parameters	pH, SC, DO, ORP, temperature, and turbidity	Injectate lateral and vertical distribution and radius of influence	TA-06: CMT-20, EBF-05S TA-07: MW-85, MW-48, MW-47, MW-55, MW-41, INJ-205 TA-08: CMT-18, INJ-206, MW-144, MW-151 TA-09: CMT-24, MW-80, MW-82, MW-83 TA-11: MW-97, MW-108, MW-109, MW-96, MW-95, INJ-202, MW-139	injections Minimum daily during injections
Aboveground	Inspection of surface around injection areas	Reagent surfacing	Injection areas and vicinity	During active DPT injection

Notes: gpm = gallons per minute, psi = pounds per square inch, SC = specific conductance, DO = dissolved oxygen, ORP = oxidation reduction potential, temp = temperature.

4.7 Investigative Derived Material Management

Investigative derived material (IDM) generated during implementation of remediation injection related activities will include empty reagent intermediate bulk container (IBC) totes, containerized rinse water from totes, personal protective equipment, packaging materials, etc. Monitoring and sampling activities will include purge and decontamination water, personal protective equipment, and disposable materials used during sampling activities.

Table 4-3 identifies planned IDM containerization and disposal based on previous work conducted at OU 7 and DSCR.

Table 4-3 Investigative Derived Material Containerization and Disposal

IDM Type	Container	Expected Waste Characterization	Anticipated Transportation and Disposal
Soil cuttings from monitoring well drilling	Place in UN certified (Solid) 55-gallon, open head drum	Non-Hazardous Waste (Aqueous), Waste	Shamrock Richmond VA

IDM Type	Container	Expected Waste Characterization	Anticipated Transportation and Disposal
		characterization testing in Table 4-3.	
Personal protection equipment	Place in trash bag and dispose as general solid waste in dumpster at Bldg.40	General solid waste (no testing)	Solid waste for DSCR
Excess packaging materials and disposable items	Place in trash bag and dispose as general solid waste in dumpster at Bldg.40	General solid waste (no testing)	Solid waste for DSCR
IBC rinse water, decontamination water, purge water, new well development water, and injection well redevelopment water	Consolidate into holding containers at Building 40 for Vacuum truck pump out	Non-Hazardous Waste (Aqueous), Waste characterization testing in Table 4-4.	Shamrock Richmond VA
Empty reagent 275 gallon totes	Empty reagent 275 gallon totes, pickup at OU 7	Offsite recycling	Shamrock Richmond VA

Notes: IDM = investigative derived material, OU 7 = operable unit 7, PPE = personal protection equipment, Bldg. = building

Waste characterization will include composite sampling and field subsampling following ASTM Designation *D6051-15 Standard Guide for Composite Sampling and Field Subsampling for Environmental Waste Management Activities*. This sampling will determine if IDM is non-hazardous or hazardous according to the Code of Federal Regulations (CFR) 40 CFR Part 261 – Identification and Listing of Hazardous Waste and also include parameter testing required by the local non-hazardous treatment, storage, and disposal facility (Shamrock Environmental Richmond Virginia).

Table 4-4 Waste Characterization Parameter Analysis

Characteristic	Regulatory	Method	Parameters	Matrix
Ignitability	40 CFR §261.21	SW846 Method 1030 SW846 Method 1010A	Ignitability Ignitability	Solid Aqueous
Corrosivity	40 CFR §261.22	SW846 Method 9045D SW846 Method 9040C	pH pH	Solid Aqueous
Reactivity	40 CFR §261.23	No test	No test	No reactive media identified at site
Toxicity	40 CFR §261.24	SW846 Method 1311 SW846 8260 SW846 8270 SW846 8081 SW846 8051 SW846 6010 SW846 7470/7471	Table 1 - 40 CFR §261.24 Volatile organics Semi-volatile organics Pesticides Chlorinated herbicides Metals/metalloids Mercury	Solid and Aqueous
Other		SM 2320B SM 2540C SW846 8082A	Alkalinity Total dissolved solids Polychlorinated biphenyls (PCBs)	Aqueous Aqueous Solid and Aqueous

Notes: 40 CFR = Title 40 Code of Federal Regulations

4.8 Spill Response Procedures

The injection related work will include implementation of appropriate product handling procedures and spill response procedures, as applicable. Planned measures will include setup of a secondary containment area to house the injection system including mixing equipment and transfer hoses. Planned measures for gravity feed injections will include the use of IBC spill containment pallets.

The injection contractor will have additional containment/berming materials in the case where injected reagents reach the ground surface. If daylighting occurs, the contractor will place containment/berm materials around the affected area until all reagents are properly removed. The injection contractor will have spill kits and portable vacuums in the work area for immediate deployment if a spill or injectate surfacing occurs during site operations. Kingsland Creek is located approximately 360 ft. south (downgradient) of TA-09 and 310 ft. south (downgradient) of TA-06. Injections are not expected to daylight based on the gravity feed methods used and depth. During injection operations, the field team will implement measures to monitor for potential daylighting and have spill containment, sorbent, other materials, and recovery equipment available for deployment. If daylighting occurs, the injection contractor will immediately discontinue injection operations at the location causing daylighting. The field team will maintain spill containment until all reagents are properly removed.

Remedy Verification and Performance Monitoring

Remedy performance monitoring will evaluate the enhanced ISB actions for the confined aquifer at OU 7. The technical approach will include baseline monitoring before injections and post-injection performance monitoring at 14 monitoring wells. The proposed network of monitoring wells for performance monitoring is shown in Figure 3-1 (page 3-2) includes:

- Hot spot monitoring wells AEHADG-11 and MW-36 in treatment area TA-02 (Pit 1 monitored area).
- Hot spot monitoring well CMT-21-2 and well MW-68 in treatment area TA-03 (Pit 1 monitored area).
- Monitoring well DMW-29A in treatment area TA-05 (Pit 1 monitored area).
- Monitoring well CMT-20-3 in treatment area TA-06 (Pit 1 monitored area).
- Monitoring wells MW-85 and MW-48 in treatment area TA-07 (Pit 2 monitored area).
- Monitoring well CMT-18-5 in treatment area TA-08 (Pit 2 monitored area).
- Monitoring wells MW-80 and MW-82 in treatment area TA-09 (Pit 2 monitored area).
- Monitoring well MW-148 located near the installation fence line downgradient of treatment area TA-09 (Pit 2 monitored area).
- Monitoring wells MW-109 and MW-95 in treatment area TA-11 (Pit 3 monitored area).

5.1 Baseline Monitoring

Baseline groundwater monitoring for the enhanced ISB injections will occur as part of the 2025 Annual Monitoring Event for OU 7 scheduled for March 2025. This monitoring event includes 87 surficial aquifer monitoring wells including the 14 performance monitoring wells identified in Table 5-1 (page 5-4) identifies the 14 monitoring well locations and scope of baseline monitoring. Proposed new monitoring well OU7-MW-151 will have baseline sampling completed after installation/development and before ISB injections. The baseline data for 2025 will provide comparative data for ISB performance evaluations. The scope of analysis for baseline monitoring will include field water quality parameters, VOCs, ferrous iron, and geochemical parameters including TOC, anions, sulfide, alkalinity, manganese, ethene, ethane, methane, and carbon dioxide.

5.2 Performance Monitoring

ISB performance monitoring will occur after completion of the injections and include sample collection in the fourth quarter 2025 (calendar year), March 2026 as part of annual monitoring, and the fourth quarter 2026 (calendar year). An adaptive approach will determine the frequency and scope of sampling beyond 2026. Table 5-1 (page 5-4) identifies the 15 monitoring well locations and the scope and schedule for performance monitoring.

5.3 Monitoring Procedures

Prior to sampling, the annual sampling events performed in May 2025 and May 2026 will include a synoptic round of water level measurements at all monitoring well locations screened in the surficial aquifer at OU 7. The water level data will input into development of potentiometric surface contour maps to characterize groundwater flow patterns, hydraulic gradient, and to calculate the velocity of groundwater flow. The field team will collect water level measurements at each performance monitoring well prior to sampling.

Exhibit 5-1 (page 5-7) has summary information on groundwater sampling procedures for baseline and performance monitoring that references detailed information contained in the project QAPP (AECOM-Meadows 2024a).

Table 5-1 ISB Baseline and Performance Monitoring Program: OU 7 Surficial Aquifer (2025-2026)

	Screen	ed Interval		Baseline Q1	Inj. Event		Q2		Field Water	VOCs Method	TOC Method	Ferrous Iron	Anions Method	Sulfide Method	Mn Method	Alkalinity Method	Dissolved Gases	CO2 Method	
Well ID	Depth (ft. BGS.)	Elev ft. (NAVD88)	Location	Annual 2025	Q3 2025	Q4 2025	Annual 2026	Q4 2026	Quality Parameters	SW846 8260D	SW846 9060A	Method 8146	SW846 9065A	SM4500- S2-F-2011	SW846 6020	SM2320B- 2011	Method RSK-175	SM4500- CO2-D	qPCR
AEHADG-11	6.1-16.1	90.2-80.2	TA-02	X	Process Monitoring	X	X	X	X	X	X	Х	X	X	X	X	X	X	X
OU7-MW-36	9.0-19.0	87.3-77.3	TA-02	х	Process Monitoring	Х	Х	Х	Х	Х	Х	Х	Х	х	Х	Х	Х	Х	
CMT-21-2	18.5-20.0	76.0-74.5	TA-03	Х	Process Monitoring	Х	Х	Х	Х	Х	Х	Х	Х	Х	Х	х	Х	Х	Х
OU7-MW-68	5.0-15.0	89.9-79.9	TA-03	Х	Process Monitoring	Х	х	Х	Х	Х	Х	Х	Х	X	Х	x	х	Х	
DMW-29A	23.0-28.0	71.7-66.7	TA-05	х	Process Monitoring	Х	Х	Х	Х	Х	Х	Х	Х	Х	Х	Х	Х	Х	
CMT-20-3	18.5-20.0	77.7-75.70	TA-06	Х	Process Monitoring	Х	Х	Х	Х	Х	Х	Х	Х	Х	Х	Х	Х	Х	
OU7-MW-85	22.5-25.0	76.5-74.0	TA-07	Х	Process Monitoring	Х	Х	Х	Х	Х	Х	Х	Х	Х	Х	х	х	Х	
OU7-MW-48	15.0-25.0	83.9-73.9	TA-07	Х	Process Monitoring	Х	Х	Х	Х	Х	Х	Х	Х	Х	Х	х	х	Х	
CMT-18-5	26.5-28.0	70.1-68.8	TA-08	Х	Process Monitoring	Х	Х	Х	Х	Х	Х	Х	Х	Х	Х	х	х	Х	
OU7-MW-151	23.0-28.0	73.8-68.8	TA-08	X ¹	Process Monitoring	Х	Х	Х	Х	Х	Х	Х	Х	Х	Х	х	х	Х	
OU7-MW-80	16.0-21.0	78.9-73.9	TA-09	Х	Process Monitoring	Х	Х	Х	Х	Х	Х	Х	Х	Х	Х	х	х	Х	
OU7-MW-82	8.0-13.0	86.6-81.6	TA-09	Х	Process Monitoring	Х	Х	Х	Х	Х	Х	Х	Х	Х	Х	х	х	Х	
OU7-MW-148	8.0-18.0	76.8-66.8	TA-09	х	None	Х		Х	Х	Х	Х	Х	Х	х	Х	Х	х	Х	
OU7-MW-109	9.0-14.0	89.6-84.6	TA-11	х	Process Monitoring	Х	Х	Х	Х	Х	Х	Х	Х	х	Х	Х	Х	Х	
OU7-MW-95	9.0-14.0	88.1-83.1	TA-11	Х	Process Monitoring	Х	х	Х	х	Х	Х	Х	х	Х	Х	х	Х	Х	

Notes: ¹ new monitoring well location with planned installation before injections, baseline sampling will occur after well development and before injections, ft. = feet, ft. BGS. = feet below ground surface, Elev NAVD88 = elevation North American Vertical Datum of 1988, UG = upgradient, DG = downgradient, TAC-1 = treatment area confined aquifer in OU 3 upgradient of installation fence line, TAC-2 = treatment area confined aquifer in offsite OU 7 area. Q1 = quarter 1 of calendar year, Q2 = quarter 2 of calendar year, quality parameters = water quality parameters = water quality parameters including pH, dissolved oxygen, specific conductivity, oxidation reduction potential, and turbidity, VOC = volatile organic compound, TOC = total organic compound, Anions incudes chloride, nitrite as nitrogen, nitrite as nitrogen, and sulfate, Mn = manganese, alkalinity includes total alkalinity as CaCO3, CO2 = carbon dioxide, qPCR = quantitative polymerase chain reaction with analysis parameters including Dehalococcoides, tceA Reductase, BAV1 Vinyl Chloride Reductase, and Vinyl Chloride Reductase.

Exhibit 5-1 Summary of Monitoring Procedures

Procedure Element	Description	Reference			
Logs and record keeping	Sampling documentation in logbooks, recordkeeping, sample labeling, and chain of custody	QAPP Worksheets #26 and #27 QAPP Worksheet #21 SOPs P-01, P-02			
Sample handling, storage, and shipping	Methods for sample handling, storage, and shipping	QAPP Worksheets #26 and #27, #29 QAPP Worksheet #21 SOP P-03			
Planning, preparing, and documenting groundwater sampling events	Methods for planning, preparing, and documenting groundwat3er sampling events	QAPP Worksheets #21 SOPs P- 04, P-05, QAPP Worksheet #29 QAPP Appendix B.1 Groundwater sampling form			
Groundwater purging and sampling method	Low-flow purging and sampling using a peristaltic pump, and new disposable tubing for each location	QAPP Worksheet #21, SOPs P- 07, P-08, P-12, P-13, P-14, P-15, P-16			
Field preservation of samples	Methods for preserving samples	QAPP Worksheets #19 and #20 QAPP Worksheet #21, SOP P-09			
Field analysis for Ferrous Iron	Field analysis of ferrous iron by Method 8146	QAPP Worksheet #21, SOPs P-11, P-12			
Field measurement of water quality parameters and use of flow-through cell	Field measurement of pH, temp, DO, SC, and ORP	QAPP Worksheet #21, SOPs P- 19, P-20			
Field measurement of turbidity	Field measurement of turbidity	QAPP Worksheet #21, SOPs P- 21, P-22			
Equipment decontamination	Decontamination of field equipment	QAPP Worksheet #21, SOP P-25			
Field measurements with photoionization detector	Use of photoionization detector for field screening of VOCs	QAPP Worksheet #21, SOP P-30			

Notes: temp = temperature, DO = dissolved oxygen, SC = specific conductivity, ORP = oxidation reduction potential, SOP = standard operating procedure, QAPP = quality assurance project plan.

Exhibit 5-2 identifies the quality assurance and quality control samples established in the project QAPP (Worksheet #20) for annual monitoring at OU 7. The baseline and annual sampling will follow the established locations in the QAPP. Quarterly performance monitoring events will use the same matrix spike/matrix spike duplicate and field equipment blank locations for annual sampling (Work Sheet #18) and reduce the number of field duplicates to one at location AEHADG-11.

Exhibit 5-2 Quality Assurance/Quality Control Samples: Baseline/Performance Monitoring

Sample Event	Field Duplicate ¹	MS/MSD ²	Field EB ³	Trip Blank⁴
Annual Events	AEHADG-11	OU7-MW-148	NA	1 per cooler of VOCs
Performance Monitoring	AEHADG-11	OU7-MW-148	NA	1 per cooler of VOCs

Notes: MS/MSD = matrix spike/matrix spike duplicate., EB = equipment blank, ¹duplicate samples analyzed for the same parameters as associated normal sample, ²matrix spike/matrix spike duplicate analyzed for the same parameters as normal samples except for microbial analysis, ³equipment blank samples analyzed for the same parameters as normal samples except for alkalinity and microbial analysis, ⁴trip blank sample analysis for volatile organic compounds, NA = equipment not applicable based on sampling method used (peristaltic pump with new disposable tubing at each location).

5.4 ISB Performance Evaluation

Performance evaluations for the proposed ISB actions in this work plan will use multiple lines of evaluation as generally described in the 2013 RD/RAWP for OU 7 (AECOM 2013). Table 5-2 has a summary of planned ISB performance evaluations relative to ISB objectives.

Table 5-2 Enhanced ISB Performance Evaluation

Evaluation Element	Description
Reagent distribution	Evaluate reagent distribution and persistence relative to treatment design Perform injection process monitoring to evaluate reagent distribution Perform post-injection WQP measurements and sampling (TOC, geochemical) ISB objectives: distribute reagents across design treatment areas, enhance downgradient treatment zones for diffuse VC plumes in downgradient areas of Pit 1 and Pit 2 monitored plume areas.
Post-injection concentration trends	Evaluate parameter trends along groundwater flow path across barrier areas and at each performance well (WQP, VOCs, TOC, geochemical) Compare parameter concentrations to baseline + historical results Time series analysis: visualizations, exploratory data analysis, statistics, trend analysis ISB objectives: accelerate reduction of PCE, TCE, cDCE, and VC concentrations, reduce accumulation of cDCE/VC in plume areas.
Contaminant mass	Evaluate reduction of contaminant mass using chemical and geochemical data Time series analysis: evaluate changes in molar concentrations and ratios along flow paths across treatment areas, at individual wells, plume area analysis Evaluate depletion of electron acceptors and donors Evaluate increases in metabolic by-product concentrations Favorable succession of redox conditions ISB objective: reduce contaminant mass (molar) in target treatment/plume areas.
Contaminant flux	Evaluate changes in contaminant flux across treatment areas using well transects by integrating concentration and flow data Time series evaluation: individual monitoring events, changes over time ISB objective: reduce overall contaminant flux from remaining source areas and to downgradient areas near the fence line.
Plume stability and extent	Evaluate changes in plume extent (area) by comparing pre-and post-ISB modeled plumes Perform time series statistical evaluations for plume stability ISB objective: mitigate plume instability and reduce plume extent in the offsite area.
Biodegradation rates	Use data modeling to calculate rate of change of contaminant mass over time Compare estimates of pre-ISB biodegradation rates with update estimates after ISB actions Microbiological laboratory or field data that support the occurrence of biodegradation and provide estimated rates of biodegradation ISB objective: increase biodegradation rates for PCE, TCE, cDCE, and VC.

Notes: WQP = water quality parameters including pH, dissolved oxygen, specific conductivity, oxidation reduction potential, and turbidity, VOCs = volatile organic compounds, TCE = trichloroethene, cDCE = cis-1,2-dichloroethene, VC = vinyl chloride, ISB = in situ bioremediation.

6. Permitting

Section 6 discusses permitting requirements and activities for implementing ISB actions at OU 7.

6.1 Drilling and Subsurface Installations

Monitoring well drilling, DPT drilling, and subsurface disturbance are subject to the DSCR permitting system requirements for underground facilities protection. Meadows will clear drilling and subsurface injection activities through the DSCR excavation permitting system and obtain an excavation permit prior to commencing work. Subsurface utility mark outs and clearing will occur prior to commencing any intrusive activities as described in Section 4.1.

Monitoring well installation will not require permitting with local or Commonwealth of Virginia regulatory agencies.

6.2 Site Security and Communications

Meadows will coordinate all remedial activities with DSCR operations personnel to ensure compliance with DSCR physical and operational security requirements. This will include developing transit corridors for vehicles and transport of equipment and materials, participating in training, and participating in security briefings, as appropriate.

Oversight personnel and the project management (PM) Team will coordinate with DSCR personnel to establish specific lines of communication during remedial activities. These will include providing specific contacts for each phase of work.

6.3 Health and Safety

Remediation work at OU 7 will occur under the project health and safety plan and accident prevention plan, which complies with the applicable requirements of the Occupational Safety and Health Agency General Industry Standards (29 Code of Federal Regulations 1910), Construction Safety Standards (29 Code of Federal Regulations 1926), and Hazardous Waste Operations and Emergency Response Standards (29 Code of Federal Regulations 1910.120) and applicable requirements of USACE Engineer Manual 385-1-1. In addition, the safety program for all work activities will coordinate with applicable DSCR operational and emergency response policies and programs.

The PM Team will designate a task Site Safety Officer (SSO). The SSO will oversee health and safety requirements for task related field activities. The SSO will confer and coordinate with DSCR and/or USACE Safety Officer to identify hazards associated with the planned remedial activities and will ensure any concurrent activities and field work do not interfere with installation activities (in cooperation with the PM Team).

7. Reporting

A project technical memorandum will summarize completed remedial action installation activities. Annual reports for OU 7 will report the results of remedy implementation, performance monitoring, MNA and LTM and include data evaluations described in Table 5-2 and integrated analysis of remedy performance. Periodic updates of remedy performance and progress will occur during regulatory planning team meetings and for semi-annual restoration advisory board meetings.

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Appendix A Remedy Design Support Information

A.1 EVO Technical Data Sheet and Safety Data Sheet









Patented Injection Ready 60% SRS®-SD Small Droplet Emulsified Vegetable Oil (EVO) Substrate for Maximum Radius of Influence United States Patent #RE40,448

Terra Systems patented "*injection ready*" <u>60% SRS®-SD</u> Small Droplet Emulsified Vegetable Oil Substrate is added to the groundwater to rapidly generate reducing conditions and provide the necessary carbon and hydrogen to support native or introduced microorganisms (*Dehalococcoides*) for the biodegradation of chlorinated solvents such as tetrachloroethene (PCE) and trichloroethene (TCE) to innocuous end products including ethene and ethane.

Key Communication Points

- The 0.6 um droplet size results in better substrate distribution for the client, easier substrate injectability for the driller and fewer injection points for the consultant thereby lowering costs
- Provides 73% fermentable carbon
- Has >98% biobased content
- Includes sodium or potassium lactate to kick-start the anaerobic degradation process, nutrients and Vitamin B₁₂ a micronutrient, which *He et al.* 2007 demonstrated is an important micronutrient to enhance dechlorination activity.
- The nonionic emulsifier (does not have a charge) results in better distribution and bacteria contact for the client because the substrate does not readily stick to the positively charged soil particles.
- It arrives as a homogenous *injection ready substrate*, which results in lower field labor costs from inefficient field mixing.
- Proven effective with PCE, TCE, TECA, DNAPL (Sabre Project), Perchlorate, TCA, Cr⁶⁺, TNT, Uranium and Nitrate.
- Proven effective at military installations (Andrews AFB, Dover AFB, Beale AFB, Ft. Gillem, Fort Dix, Camp Bullis, Aberdeen Proving Ground, etc.), dry cleaners, semiconductor manufacturers, fabricators and manufacturing firms that use and clean metal parts (air conditioners, dishwashers, etc.).

Table I: SRS®-SD Small Droplet Emulsified Vegetable Oil Substrate Specifications

Ingredient	Percent	Description	Benefit
Food Grade U.S. Grown	60%	Locally sourced soybean	Long lasting slow release source of carbon and
Soybean Oil	0070	oil.	hydrogen.
Food Grade Soluble	5.5%	Rapidly biodegradable	Fast release source of carbon and hydrogen to
Substrate	3.370	soluble substrate	rapidly generate anaerobic conditions
Proprietary Food Grade Nutrients	<1%	Proprietary organic and inorganic nutrients such as yeast extract, nitrogen and phosphorus.	Nutrients have been demonstrated to support the growth of the anaerobic microbial population.









			INCORPORATED
Proprietary Food Grade Emulsifiers, Preservatives and other Organics	7.5%	Proprietary nonionic emulsifier and other organics	Maximum radius of influence due to small droplet size and nonionic emulsifier in moderate to fine sand, silt and clay aquifers
Vitamin B ₁₂	<1%	250 μg/L of Vitamin B ₁₂	He et al. 2007 demonstrated Vitamin B ₁₂ to be an important micronutrient to enhance dechlorination activity with 25 µg/L providing maximum stimulation
Median Oil Droplet Size (microns)	NA	0.6 µm	Maximum radius of influence due to small droplet size and nonionic emulsifier in moderate to fine sand, silt and clay aquifers
рН	6.5 - 7	6.5 - 7	Optimum microbial activity
Organic Carbon (wt%)	73%		60% soy bean oil and 13% from lactate, nutrients, emulsifiers and VB ₁₂
Zero Carbon Footprint	0%		Certified by The CarbonNeutral Co., SRS® has a carbon neutral footprint when it arrives at the job site.
Biobased Content	98%		Certified under USDA Biopreferred Program

Injection Ready Manufactured Emulsion

Terra Systems *Family* of patented SRS[®] emulsified vegetable oil substrates

- Arrives injection ready
- Nutrients are premixed into the SRS during the manufacturing process ensuring a homogenous substrate and avoiding the additional labor cost of mixing in the field
- Vitamin B_{12} is premixed into the SRS[®] during the manufacturing process ensuring a homogenous substrate and avoiding the additional labor cost of mixing in the field
- Sodium lactate, which kick starts the anaerobic process is premixed into the SRS[®] during the
 manufacturing process ensuring a homogenous substrate and avoiding the additional labor cost of
 mixing in the field
- Arrives at the site with a zero carbon footprint
- Certified under the USDA Biopreferred Program with >98% biobased content

<u>Result</u>: A consistent emulsified vegetable oil substrate, which arrives *ready to inject* for maximum distribution in the aquifer.

It Avoids Field Mixing and Their Hidden Costs Such As:

- The cost of inadequate distribution due to variable droplet size and emulsion inconsistency
- The inability to accurately determine if you have 100% emulsification.







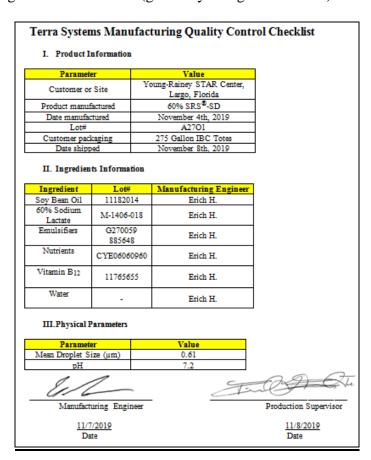


The lack of QA/QC in the field

Terra Systems QA/QC

Terra Systems owns and operates a state-of-the-art US based "*just-in-time*" manufacturing plant with an inhouse quality control laboratory for strict quality assurance of the emulsion, droplet size and pH. A Microscope with "*Droplet Size Calculation Software*" calculates the "*mean*" droplet size for each batch of SRS[®] before we transfer to a bucket, drum, tote or tanker for shipment to the customer. With every shipment, we include a QA/QC sheet for the actual batch that the customer receives. Included are:

- **Date Manufactured**: Freshly manufactured products have a longer shelf life in the field. Avoid buying substrates that have been stored for >1 month as fermentation can start and the pH will be negatively impacted.
- **pH**: We provide the pH of the product the day it is shipped
- **Droplet Size**: is a key measure of how effective the client can distribute the substrate in the sub-surface. The smaller the droplet, the more effective the distribution and ease of injection.
- Lot#'s for all the ingredients: This is especially useful if the driller accidentally hits a discharge pipe and the consultant needs to provide documentary evidence of what exactly was injected to the regulatory agency. All of our ingredients are GRAF (generally recognized as safe).











<u>Packaging</u>: Terra Systems patented SRS[®] Family of EVO substrates can be shipped in 5-gallon buckets, 55-gallon drums, 275-gallon IBC totes, 275-gallon cardboard totes or bulk tankers.











Patented *Injection Ready* 60% SRS[®]-SD Small Droplet Emulsified Vegetable Oil (EVO) Substrate for Maximum Radius of Influence

United States Patent #RE40,448 SAFETY DATA SHEET

1. Product Identification

Synonyms: 60% Small Droplet Slow Release Substrate (SRS[®]-SD)

Emulsified Vegetable Oil Substrate (EVO)

Recommended Use: Treatment of groundwater contaminated with chlorinated

solvents and other anaerobically degradable compounds.

Supplier: Terra Systems, Inc.

130 Hickman Road, Suite 1 Claymont, Delaware 19703 Telephone (302) 798-9553 Fax (302) 798-9554

www.terrasystems.net

2. Hazards Identification

Emergency Overview

Caution: May cause eye irritation.

Health Rating:1 - SlightFlammability Rating:1 - SlightReactivity Rating:1 - SlightContact Rating:1 - Slight

Protective Equipment: Goggles; Proper Gloves **Storage Color Code:** Green (General Storage)

Potential Health Effects

Inhalation: Not expected to be a health hazard. If heated, may produce

vapors or mists that irritate the mucous membranes and cause irritation, dizziness, and nausea. Remove to fresh air.

Ingestion: Not expected to be a health hazard via ingestion. Large

doses may produce abdominal spasms or diarrhea.

Skin Contact: No adverse effects expected. May cause irritation or

sensitization in sensitive individuals.

Eye Contact: May cause mild irritation, possible reddening.

Chronic Exposure: No information found.

Aggravation of Pre-existing

Conditions: No information found.









3. Composition/Information on Ingredients

Ingredient	Synonyms	CAS#	Percent	Hazardous
Soy bean oil	Soya oil	8001-22-7	60%	No
Emulsifiers and proprietary nutrient package containing nitrogen, phosphorus and vitamin B ₁₂		Mixture	7.5 - 10%	No
Sodium lactate	2- hydroxpropionic acid sodium salt	72-17-3	5.5%	Yes
Water		7732-18-5	Difference	No

The emulsifiers and nutrient package mixture is a trade secret and consists of ingredients of unknown acute toxicity.

4. First Aid Measures

Inhalation: Not expected to require first aid measures. Remove to fresh air.

Get medical attention for any breathing difficulty.

Ingestion: If large amounts were swallowed, give water to drink and get

medical advice.

Skin Contact: Not expected to require first aid measures. Wash exposed area

with soap and water. Get medical advice if irritation develops.

Eye Contact: Immediately flush eyes with plenty of water for at least 15

minutes, lifting upper and lower eyelids occasionally. Get

medical attention if irritation persists.

5. Fire Fighting Measures

Fire: Flash point: >200 C (>392 F). Not considered to be a fire

hazard. Isolate from heat and open flame.

Explosion: Not considered to be an explosion hazard. Closed containers

may explode if exposed to extreme heat.

Fire Extinguishing Media: Dry chemical, foam, or carbon dioxide. Water spray may be

ineffective on fire but can protect fire-fighters and cool closed

containers. Use fog nozzles if water is used.

Special Information: In the event of a fire, wear full protective clothing and NIOSH-

approved self-contained breathing apparatus with full face piece operated in the pressure demand or other positive

pressure mode.

6. Accidental Release Measures

Clean-up personnel may require protective clothing. Absorb in sand, paper towels, "Oil Dry", or other inert material. Scoop up and containerize for disposal. Flush trace residues to sewer with soap and water. Containerized waste may be sent to an approved waste disposal facility.









7. Handling and Storage

Store in a cool, dry, ventilated area. Do not store in sunlight or above 32 C (90 F). Keep container tightly closed and upright when not in use to prevent leakage. Observe all warnings and precautions listed for the product. Protect against physical damage.

If container begins to bulge, open cap slowly to release carbon dioxide from biological activity on the SRS-SD and call TSI.

Containers of this material are not hazardous when empty since they do not contain vapors or harmful substances; if drum or tote is observed to bulge, keep cap off as pressurization can occur on empty container with caps in place unless container is thoroughly rinsed.

8. Exposure Controls/Personal Protection

Airborne Exposure Limits: None established.

Ventilation System: Not expected to require any special ventilation.

Personal Respirators (NIOSH

Approved): Not expected to require personal respirator usage.

Skin Protection: Wear protective gloves and clean body-covering clothing.

Eye Protection: Use chemical safety goggles and/or a full-face shield where

splashing is possible. Provide readily accessible eye wash

stations and safety showers.

Slips, Trips, and Falls: Material is slippery when spilled. Clean up with sand, paper

towels, "Oil Dry", or other inert material.

9. Physical and Chemical Properties

Appearance:White liquid.Odor:Vegetable oil.Solubility:Miscible in water.

Specific Gravity (water=1): 0.95-0.98. 8.09 pounds per gallon.

pH: 6-7 (40% aqueous solution)

% Volatiles by volume

@ 21C (70F): Negligible. $> 100C (\ge 212F)$ **Boiling Point: Melting Point:** No information found. **Flash Point (F):** No information found. **Autoignition Temperature:** No information found. **Decomposition Temperature:** No information found. Vapor Density (Air=1): No information found. **Vapor Pressure (mm Hg):** < 1.0 @ 20C (68F). **Evaporation Rate (BuAc=1):** No information found.

Viscosity @23 C (73 F): 213 centipoises (1.2 centipoises diluted 1:10)

Partition Coefficient

(octanol/water): No information found.









10. Stability and Reactivity

Stability: Stable under ordinary conditions of use and storage.

Reactivity: Not reactive under ordinary conditions.

Hazardous Decomposition

Products: Carbon dioxide and carbon monoxide may form when

heated to decomposition.

Hazardous Polymerization: Will not occur.

Incompatibilities: Strong oxidizers, acids.

Conditions to Avoid: Incompatibles. Isolate from heat and open flame.

11. Toxicological Information

Soybean Oil: No information found on toxicology. It is not a carcinogen

listed by IARC, NTP, NIOSH, OSHA, or ACGIH.

Emulsifier/Nutrient Mixture: No information found on toxicology. It is not a carcinogen

listed by IARC, NTP, NIOSH, OSHA, or ACGIH.

Sodium Lactate: Oral rat LD50: 2,000 mg/kg. 100 mg caused mild irritation to

rabbit eye in Draize test. This compound is not listed as a carcinogen by IARC, NRP, NIOSH, OSHA, or ACGIM.

SRS-SD: The toxicity of the mixture has not been measured.

12. Ecological Information

Environmental Fate: No information found. **Environmental Toxicity:** No information found.

Degradability: This product is completely biodegradable under both aerobic

and anaerobic conditions.

Soil Mobility: This compound will move with groundwater until the adsorbed

onto the soil. Degradation products may be mobile.

Bioaccumulation Potential: No information found.

13. Disposal Considerations

Whatever cannot be saved for recovery or recycling should be managed in an appropriate and approved waste disposal facility. Processing, use or contamination of this product may change the waste management options. State and local disposal regulations may differ from federal disposal regulations. Dispose of container and unused contents in accordance with federal, state and local requirements.

14. Transport Information

Not regulated.

15. Regulatory Information









OSHA STATUS: This product is not hazardous under the criteria of the Federal OSHA hazard Communication Standard 29 CFR 1910.1200. However, thermal processing and decomposition fumes from this product may be hazardous as noted in Section 10.

TSCA STATUS: No component of this product is listed on the TSCA inventory.

CERCLA (Comprehensive Response Compensation, and Liability Act): Not reportable.

SARA TITLE III (Superfund Amendments and Reauthorization Act)

Section 312 Extremely Hazardous Substances: None

Section 311/312 Hazard Categories: Non-hazardous Under Section 311/312

Section 313 Toxic Chemicals: None

RCRA STATUS: If discarded in its purchased form, this product would not be a hazardous waste either by listing or by characteristic. However, under RCRA, it is the responsibility of the product user to determine at the time of disposal, whether a material containing the product or derived from the product should be classified as a hazardous waste. (40 CFR 261.20-24)

CALIFORNIA PROPOSITION 65: The following statement is made in order to comply with the California safe Drinking Water and Toxic Enforcement Act of 1986. The product contains no chemicals known to the State of California to cause cancer.

16. Other Information

NFPA Ratings: Health: **1** Flammability: **1** Reactivity: **1**

Date Prepared: September 11, 2019

Revision Information: SDS Section(s) changed since last revision of document

include: Updated Section 3 Composition/Information on

Ingredients.

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Prepared by: Terra Systems, Inc. **Phone Number:** (302) 798-9553 (U.S.A.)

A.2 Bioaugmentation Technical Data Sheet and Safety Data Sheet









TSI DC *Dehalococcoides mccartyi* Bioaugmentation Culture[®]

Containing >1 x 10¹¹ Dehalococcoides cells/L

Terra Systems TSI DC *Dehalococcoides mccartyi* Bioaugmentation Culture[®] is added to the groundwater at sites where the native microorganisms of Dehalococcoides are not present, are not in sufficient quantity, where the native population does not express all of the required functional genes for TCE and vinyl chloride reduction, or when the client wants to decrease the remediation time frame for the biodegradation of chlorinated solvents such as tetrachloroethene (PCE) and trichloroethene (TCE) to innocuous end products including ethene and ethane.

Key Communication Points

- TSI DC *Dehalococcoides mccartyi* Bioaugmentation Culture[®] is an enriched natural bacteria culture that contains *Dehalococcoides* species for bioaugmentation.
- TSI DC® contains >1 x 10¹¹ Dehalococcoides cells/L
- This culture dechlorinates tetrachloroethene (PCE) and trichloroethene (TCE) to the non-toxic product ethene.
- The culture also biodegrades 1,1,1-trichloroethane to 1,1-dichloroethene, 1,1-dichloroethane, and chloroethane.
- It also can biodegrade carbon tetrachloride and chloroform to methylene chloride and innocuous products.
- It can be used at sites where bacteria capable of complete reductive dechlorination are not present or there is a need to decrease the remediation time frame. It is estimated that *Dehalococcoides* are not present in 10 to 40 percent of chlorinated solvent contaminated sites.
- Key Benefits of TSI DC Dehalococcoides mccartyi Bioaugmentation Culture®
- The TSI-DC® Bioaugmentation Culture has been proven to be effective with a growing body of laboratory and field data demonstrating that the *Dehalococcoides* group of microorganisms is primarily responsible for the complete dechlorination of PCE and TCE to ethene. Some *Dehalogenimonas* species can also biodegrade PCE and TCE.
- At sites where *Dehalococcoides* microorganisms are not present or are found at low numbers, the process will often "**stall**" at cis-1,2-dichloroethene (cDCE). Low pH or insufficient substrate can also contribute to the cDCE stall.

On the Web: www.terrasystems.net











- The TSI-DC[®] Bioaugmentation Culture will promote the complete dechlorination of PCE or TCE.
- The TSI-DC[®] Bioaugmentation Culture contains greater than 1 x 10¹¹ Dehalococcoides/L.

Terra Systems QA/QC

Terra Systems owns and operates a state-of-the-art US based "*just-in-time*" manufacturing plant with an in-house quality control laboratory for strict quality assurance of our products. With every shipment, we include a QA/QC sheet for the actual batch that the customer receives. Included are the date manufactured, batch#, DHC concentration (cells/L), PCE dechlorination activity and cDCE dechlorination activity.

Manufacturing Quality Control Checklist for TSIDC Dehalococcoides mccartyi Bioaugmentation Culture®

I. Product Information

Parameter	Value
Product manufactured	TSI DC Dehalococcoides mccartyii Biosugmentation Culture [®]
Date manufactured	07/16/2019
Batch#	805-19
Customer packaging	Two (2) 19 L Kegs
Customer	Navarro Research and Engineering, Inc.
Volume of Culture	76 L concentrated 2X to fit into (2) 19 L Kegs
Date shipped	8/19/2019
Date delivered	8/21/2019
Site location	Largo, FL.

II. Ingredients Information

Test	Results	Acceptable Range	Date	Method
DHC content of Pre-concentrated culture (copies/L)	>1E11	1E11	07/25/2019	qPCR.
PCE dechlorination activity, mg/h per gram of dry weight	98	50	08/13/2019	Bottle Assay
cDCE declorination activity, mg/h per gram of dry weight	61	50	08/13/2019	Bottle Assay

michel I lee, PRI.

August 19th, 2019

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The TSI-DC® Bioaugmentation Culture is cost effective and is typically a minor component of the total remediation project cost. At sites where the *Dehalococcoides* is present, but at low numbers or poorly distributed, bioaugmentation can be used to reduce the treatment time. Bioaugmentation can also reduce the time required to grow the *Dehalococcoides* population to effective cell densities. Therefore, future costs can be reduced.

- The TSI-DC® Bioaugmentation Culture works with all commonly used electron donors.
- The TSI-DC® Bioaugmentation Culture is not genetically modified or engineered.
- The TSI-DC® Bioaugmentation Culture is certified to be free of known human pathogens.
- Each purchase comes with free technical phone support from an experienced Terra Systems microbiologist.
- The TSI-DC[®] Bioaugmentation Culture has rigorous quality control procedures in place to ensure that each shipment is of the highest quality, stable, safe, effective and free of chlorinated volatile organic compounds.
- The TSI-DC[®] Bioaugmentation Culture is shipped overnight in specially designed stainless-steel containers that prevent exposure to air and are safe & easy to handle.
- A senior level microbiologist is also available to be on-site to support the successful application at \$1,200 per day plus travel expenses





TERRA SYSTEMS, INC DECHLORINATING BIOAUGMENTATION CULTURE (TSI-DC) SAFETY DATA SHEET

1. Product Identification

Synonyms: Dehalococcoides or DHC Microbial Consortium (TSI-DC)

Recommended Use: Bioremediation of groundwater contaminated with

chlorinated solvents such as tetrachloroethene and

trichloroethene.

Supplier: Terra Systems, Inc.

130 Hickman Road, Suite 1 Claymont, Delaware 19703 Telephone (302) 798-9553

Fax (302) 798-9554 www.terrasystems.net

2. Hazards Identification

The available data indicates no known hazards associated with exposure to this product. Nevertheless, individuals who are allergic to enzymes or other related proteins should avoid exposure and handling. Health effects associated with exposure to similar organisms are listed below.

Emergency Overview

Caution: May cause eye irritation or discomfort if ingested or

inhaled or allergic reaction to sensitive individuals.

Health Rating: 1 - Slight
Flammability Rating: 0 - None
Reactivity Rating: 0 - None
Contact Rating: 1 - Slight

Protective Equipment: Goggles; Proper Gloves **Storage Color Code:** Green (General Storage)

Potential Health Effects

Inhalation: Not expected to be a health hazard. Hypersensitive

individuals may experience breathing difficulties after

inhalation of aerosols.

Ingestion: Not expected to be a health hazard via ingestion. Ingestion

of large quantities may result in abdominal discomfort including nausea, vomiting, cramps, diarrhea, and fever.

Skin Contact: No adverse effects expected. May cause irritation or

sensitization in sensitive individuals upon prolonged

contact.

Eye Contact: May cause mild irritation, possible reddening unless

immediately rinsed.



Chronic Exposure:
Aggravation of Pre-existing

No information found.

Aggravation of 1 re-existing Conditions:

No information found.

3. Composition/Information on Ingredients

Ingredient	Synonyms	CAS#	Percent	Hazardous
Non-hazardous ingredients	DHC	Not	100%	No
_		applicable		

4. First Aid Measures

Inhalation: Not expected to require first aid measures. Remove to fresh air.

Get medical attention for any breathing difficulty or if allergic

symptoms develop.

Ingestion: Thoroughly rinse mouth with water. Do not induce vomiting

unless directed to do so by medical personnel. Get immediate

medical attention. Never give anything by mouth to an

unconscious or convulsing person.

Skin Contact: Not expected to require first aid measures. Wash exposed area

with soap and water. Get medical advice if irritation develops.

Eye Contact: Immediately flush eyes with plenty of water for at least 15

minutes, lifting upper and lower eyelids occasionally. Get

medical attention if irritation persists.

Note to Physicians: All treatments should be based on observed signs and

symptoms of distress in the patient. Consideration should be given to the possibility that overexposure to materials other

than this material may have occurred.

5. Fire Fighting Measures

Fire: Non-flammable. Flash point and flammable limits are not

available.

Explosion: Not considered to be an explosion hazard. **Fire Extinguishing Media:** Dry chemical, foam, carbon dioxide, or water.

Special Information: In the event of a fire, wear full protective clothing and NIOSH-

approved self-contained breathing apparatus with full face piece operated in the pressure demand or other positive

pressure mode.

6. Accidental Release Measures

Clean-up personnel may require protective clothing and avoid skin contact. Absorb in sand, paper towels, or other inert material. Scoop up and containerize for disposal. Flush trace residues to sewer with soap and water. Containerized waste may be sent to an approved waste disposal facility. After clean-up, disinfect all cleaning materials and storage containers that come in contact with the spilled liquid.



7. Handling and Storage

Avoid breathing breathe aerosol. Avoid contact with skin. Use personal protective equipment recommended in Section 8. Keep containers tightly closed in a cool, well-ventilated area. The DHC microbial consortium (TSI-DC) can be supplied in stainless steel kegs designed for maximum working pressure of 130 psi and equipped with pressure relief valves. The kegs are pressurized with nitrogen gas up to the pressure of 15 psi. Do not exceed pressure of 15 psi during transfer of DHC microbial consortium (TSI-DC) from kegs. Don't open keg if content of the keg is under pressure. DHC microbial consortium (TSI-DC) may be stored for up to 3 weeks at temperature 2-4°C without aeration. Avoid freezing.

8. Exposure Controls/Personal Protection

Airborne Exposure Limits: None established.

Ventilation System: Not expected to require any special ventilation. Provide

adequate ventilation to remove odors.

Personal Respirators (NIOSH

Approved): Not expected to require personal respirator usage. If aerosols

might be generated, use N95 respirator.

Skin Protection: Wear protective rubber, nitrile, or vinyl gloves and clean body-

covering clothing.

Eye Protection: Use chemical safety goggles and/or a full face shield where

splashing is possible. Provide readily accessible eye wash

stations and safety showers.

9. Physical and Chemical Properties

Appearance: Light greenish, murky liquid.

Odor: Musty.

Soluble in water.

Specific Gravity (water=1): 1.0. 8.34 pounds per gallon.

OH: 6-8

% Volatiles by volume

 @ 21C (70F):
 Negligible.

 Boiling Point:
 100C (212F)

 Melting Point:
 0C (32F)

Flash Point (F):
Autoignition Temperature:
Decomposition Temperature:
Vapor Density (Air=1):
Vapor Pressure (mm Hg):
Evaporation Rate (BuAc=1):
No information found.

Viscosity @23 C (73 F): 1 centipoises



Partition Coefficient

(octanol/water): No information found.

10. Stability and Reactivity

Stability: Stable under ordinary conditions of use and storage.

Reactivity: Not reactive under ordinary conditions.

Hazardous Decomposition

Products: None.

Hazardous Polymerization: Will not occur.

Incompatibilities: Strong oxidizers, acids, water reactive materials.

Conditions to Avoid: Incompatibles. Isolate from heat and open flame.

11. Toxicological Information

TSI-DC No information found on toxicology. It is not a carcinogen

listed by IARC, NTP, NIOSH, OSHA, or ACGIH. It has tested negative for pathogenic microorganisms such as *Bacillus* cereus, Listeria monocytogens, Salmonella sp., Pseudomonas

sp., fecal coliform, total coliform, yeast, and mold.

12. Ecological Information

Environmental Fate: No information found. **Environmental Toxicity:** No information found.

Degradability: This product is completely biodegradable under both aerobic

and anaerobic conditions.

Soil Mobility: This compound will move with groundwater until the adsorbed

onto the soil.

Bioaccumulation Potential: No information found.

13. Disposal Considerations

Waste Disposal Method: No special disposal methods are required. The material is compatible with all known biological treatment methods. To reduce odors and permanently inactivate microorganisms, mix 100 parts (by volume) of TSI-DC consortium with 1 part (by volume) of bleach. Dispose of in accordance with local, state and federal regulations.

14. Transport Information

DOT Classification: N/A Labeling: NA

Shipping Name: Not regulated

15. Regulatory Information

OSHA STATUS: This product is not hazardous under the criteria of the Federal OSHA hazard Communication Standard 29 CFR 1910.1200.



TSCA STATUS: No component of this product is listed on the TSCA inventory. CERCLA (Comprehensive Response Compensation, and Liability Act): Not reportable.

SARA TITLE III (Superfund Amendments and Reauthorization Act)

Section 312 Extremely Hazardous Substances: None

Section 311/312 Hazard Categories: Non-hazardous Under Section 311/312

Section 313 Toxic Chemicals: None

RCRA STATUS: If discarded in its purchased form, this product would not be a hazardous waste either by listing or by characteristic. However, under RCRA, it is the responsibility of the product user to determine at the time of disposal, whether a material containing the product or derived from the product should be classified as a hazardous waste. (40 CFR 261.20-24)

CALIFORNIA PROPOSITION 65: The following statement is made in order to comply with the California safe Drinking Water and Toxic Enforcement Act of 1986. The product contains no chemicals known to the State of California to cause cancer.

16. Other Information

NFPA Ratings: Health: **1** Flammability: **0** Reactivity: **0**

Date Prepared: March 26, 2014

Revision Information: SDS Section(s) changed since last revision of document

include: None.

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UPON THIS INFORMATION.

Prepared by: Terra Systems, Inc. **Phone Number:** (302) 798-9553 (U.S.A.)

A.3 pH Buffer Technical Data Sheet and Safety Data Sheet









Terra Systems pH Buffers and Buffer Capacity Test

Emulsified Vegetable Oil Substrates, lactate and other carbon substrates are added to the groundwater to rapidly generate reducing conditions and provide the necessary carbon and hydrogen to support native or introduced microorganisms (*Dehalococcoides*) for the biodegradation of chlorinated solvents such as tetrachloroethene (PCE) and trichloroethene (TCE) to innocuous end products including ethene and ethane. Often pH at a site is below optimal levels of 6.5 to 8.5 and a buffer needs to be added to the aquifer for complete dechlorination to occur.

Key Communication Points

- 1. A combination of laboratory and field studies has indicated that the optimal pH range for anaerobic bioremediation of chlorinated solvents is between 6.5 and 8.5.
- 2. Based upon laboratory studies at Terra Systems, between 76.4 to 99.1% of the buffer demands (average 93.3%) are associated with the soil phase rather than the groundwater phase.
- 3. Since the pH of just the groundwater is an unreliable determinant of the buffer demand, if possible, we strongly recommend that a saturated soil sample be collected and sent to Terra Systems Treatability Lab for a pH Buffer Capacity Test.
- 4. Terra Systems will recommend a buffer to counter the natural drop in pH due to the acids produced during the reductive dechlorination process and to optimize pH conditions at the site

<u>Table I</u>: pH Buffer Options

Buffer	Effective in pH Range	Benefit
pH Buff-Up	3.0-5.5	Liquid slurry, easy to mix, long-lasting
Sodium Bicarbonate Powder	5.0-6.0	Can't take the pH to high, maximum pH is 8.3. Inexpensive.
Calcium Carbonate Powder	4.0-6.0	Low solubility contributes to enhanced longevity. Inexpensive.
Sodium Carbonate Powder	4.0-6.0	Higher solubility but can take pH to high if overdosed. Inexpensive.
Magnesium Oxide or Magnesium Hydroxide Powder	3.0-5.0	Higher solubility but can take pH to high if overdosed. Moderately expensive.

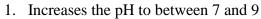




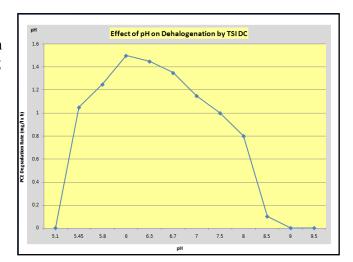




Terra Systems, Inc. (TSI) will conduct a test to determine the quantity of several potential amendments to neutralize the acidity of the groundwater at a potential bioremediation site. The objective of the evaluation is to select a buffering agent that can be added to increase the groundwater and soil pH and maintain neutral conditions needed for biological reductive dechlorination. The criteria for selecting the pH buffering agent are the following:



- 2. Does not exceed pH 10
- 3. The lowest price (either the lowest cost per unit or lower price for a larger quantity)
- 4. Is relatively soluble or has fine particles that can be suspended in the chase water



The quantities of the following buffering agents necessary to increase and maintain a neutral pH at the site will be determined:

- 1. Sodium bicarbonate or baking soda
- 2. Calcium carbonate or crushed limestone
- 3. Sodium carbonate or soda ash
- 4. Magnesium oxide

Technical References for the benefits of optimizing pH for in-situ bioremediation.

Alexander, M. L., R. Cronce, and T. Battenhouse. 2011. Differential Adjustment of pH for Optimal Reductive Dechlorination Conditions. *A-65*, in: H.V. Rectanus and R. Sirabian (Chairs), *Bioremediation and Sustainable Environmental Technologies—2011*. International Symposium on Bioremediation and Sustainable Environmental Technologies (Reno, NV; June 27–30, 2011). ISBN 978-0-9819730-4-3, Battelle Memorial Institute, Columbus, OH.

Lee, M. D., E. Hauptmann, R. L. Raymond, D. Ochs, R. Lake, and M. Selover. 2010. Buffering Acidic Aquifers with Soluble Buffer to Promote Reductive Dechlorination. *F-031*, in K.A. Fields and G.B. Wickramanayake (Chairs), *Remediation of Chlorinated and Recalcitrant Compounds—2010*. Seventh International Conference on Remediation of Chlorinated and Recalcitrant Compounds (Monterey, CA; May 2010). ISBN 978-0-9819730-2-9, Battelle Memorial Institute, Columbus, OH, www.battelle.org/chlorcon.









SODIUM BICARBONATE Safety Data Sheet

1. Product Identification

Synonyms: Sodium Hydrogen Carbonate, Baking Soda

CAS No: 144-55-8 Chemical Formula: NaHCO₃

Recommended Use: Food ingredient, pharmaceutical, water treatment

Supplier: Terra Systems, Inc.

130 Hickman Road, Suite 1 Claymont, Delaware 19703 Telephone (302) 798-9553 Fax (302) 798-9554

www.terrasystems.net

2. Hazards Identification

Emergency Overview

Caution:NoneHealth Rating:0 - NoneFlammability Rating:0 - NoneReactivity Rating:0 - NoneContact Rating:0 - None

Protective Equipment: Goggles; Proper Gloves **Storage Color Code:** Green (General Storage)

Potential Health Effects

Inhalation: Not expected to be a health hazard. If heated, may produce

vapors or mists that irritate the mucous membranes and cause irritation, dizziness, and nausea. Remove to fresh air. Possible

irritant.

Ingestion: Not expected to be a health hazard via ingestion. Material

is practically non-toxic. Small amounts (1-2

tablespoonfuls) swallowed during normal handling operations are not likely to cause injury as long as the stomach is not overly full; swallowing larger amounts may

cause injury.

Skin Contact: Not a skin irritant. **Eye Contact:** Not an eye irritant.

130 Hickman Road – Suite 1 – Claymont – Delaware – 19703

For More Information Call Michael Free at 302-798-9553 or Email: mfree@terrasystems.net











Chronic Exposure: Based on published studies on its effects in animals and

humans, sodium bicarbonate is not teratogenic or genotoxic. Only known subchronic effect is that of a marked systemic alkalosis. Not classified as carcinogenic

by NTP, IARC, OSHA, ACGIH or NIOSH.

Aggravation of Pre-existing

Conditions: No information found.

3. Composition/Information on Ingredients

Ingredient	Synonyms	CAS#	Percent	Hazardous
Sodium Bicarbonate	Baking soda	144-5-8	100	No

White crystalline powder; no odor.

4. First Aid Measures

Inhalation: Not expected to require first aid measures. Remove to fresh air.

Get medical attention for any breathing difficulty.

Ingestion: If large amounts were swallowed, do not induce vomiting.

Give water to drink if person is conscious and get medical

advice.

Skin Contact: Not expected to require first aid measures. Wash exposed area

with soap and water. Get medical advice if irritation develops.

Eye Contact: Check for and remove contacts. Immediately flush eyes with

plenty of water for at least 15 minutes, lifting upper and lower eyelids occasionally. Get medical attention if irritation persists.

Note to Physician: Large doses may produce systemic alkalosis and expansion in

extracellular fluid volume with edema.

5. Fire Fighting Measures

Fire: Not combustible. Not considered to be a fire hazard. Isolate

from heat and open flame.

Explosion: Not considered to be an explosion hazard.

Fire Extinguishing Media: Use extinguishing media suitable against surrounding fire or

the cause of the fire.

Special Information: Carbon Dioxide may be generated making necessary the use of

a self-contained breathing apparatus (SCBA) and full protective equipment (Bunker Gear). Carbon dioxide is an asphyxiant at levels over 5% w/w. Sodium oxide, another thermal decomposition product existing at temperatures above











1564°F is a respiratory, eye, and skin irritant. Avoid inhalation, eye and skin contact with sodium oxide dusts

6. Accidental Release Measures

Scoop up into dry, clean containers. Wash away small uncontaminated amounts of residue with water.

7. Handling and Storage

Keep in a tightly closed container, stored in a cool, dry, ventilated area. Protect against physical damage. Containers of this material are not hazardous when empty since they do vapors or harmful substances; observe all warnings and precautions listed for the product. Do not store above 49 C (120 F). Keep container tightly closed and upright when not in use to prevent leakage.

8. Exposure Controls/Personal Protection

Airborne Exposure Limits: None established.

Ventilation System: Not expected to require any special ventilation.

Personal Respirators (NIOSH

Approved): Dust mask required if total dust level exceeds 10 mg/m3.

Skin Protection: Wear protective gloves and clean body-covering clothing.

Eye Protection: Use chemical safety glasses when handling bulk material or

when dusts can be generated. Provide readily accessible eye

wash stations and safety showers.

9. Physical and Chemical Properties

Appearance: White crystalline.

Molecular Weight: 84.02 Odor: None.

Solubility: 86 g/L at 20 C.

Bulk Density: 9.94 g/cm³ or 62 pounds/ft³ **pH:** 8.2 (1% aqueous solution)

% Volatiles by volume

@ 21C (70F): Negligible.
Boiling Point: Not applicable.
Melting Point: Not applicable.
Flash Point (F): Not applicable.

Autoignition Temperature: Not flammable, will not support combustion.

Decomposition Temperature: 50 C.

Vapor Density (Air=1): No information found.











Vapor Pressure (mm Hg): Not applicable.

Evaporation Rate (BuAc=1): No information found.

Partition Coefficient

(octanol/water): No information found.

10. Stability and Reactivity

Stability: Stable under ordinary conditions of use and storage.

Reactivity: Not reactive under ordinary conditions. Reacts with acids

to yield carbon dioxide.

Hazardous Decomposition

Products: Carbon dioxide may form when heated to decomposition at

>100 C. If heated to >850 C, yields sodium oxide which should inhalation, eye and skin contact should be avoided.

Hazardous Polymerization: Will not occur. **Incompatibilities:** Strong acids.

Conditions to Avoid: Incompatibles. Isolate from heat and open flame.

11. Toxicological Information

Toxic Dose: 4,220 mg/kg (oral rat).

Inhalation: High concentrations of dust may cause transient irritation to

upper respiratory tract.

Ingestion: Ingestion of small amounts is unlikely to cause any adverse

effects. Ingestion of (excessive amounts) may cause

vomiting, nausea, convulsions

Skin: Repeated or prolonged contact may cause mild irritation

and/or drying (defatting) of skin.

Eyes: The material was minimally irritating to unwashed eyes and

practically non-irritating to washed eyes (rabbits).

12. Ecological Information

Environmental Fate: No information found.

Environmental Toxicity: 4,100 mg/L EC50 Daphnids. 7.100 mg/L LC50 Bluegills.

7,700 mg/L: LCt0 Rainbow trout.

Persistence: This product is expected to persist in the environment. It is

inorganic and not subject to biodegradation.

Soil Mobility: This compound will move with groundwater until it reacts with

acid.

Bioaccumulation Potential: This product is not expected to bioaccumulate











13. Disposal Considerations

Bury in a secured landfill in accordance with all local, state and federal environmental regulations. Empty containers may be incinerated or discarded as general trash.

14. Transport Information

Not regulated.

15. Regulatory Information

CLEAN AIR ACT SECTION 611: Material neither contains nor is it manufactured with ozone depleting substances (ODS).

FEDERAL WATER POLLUTION CONTROL ACT (40 CFR 401.15): Material contains no intentionally added or detectable (contaminant) levels of EPA priority toxic pollutants.

FOOD AND DRUG ADMINISTRATION: Generally Recognized As Safe (GRAS) direct food additive (21 CFR 184.1736).

US DEPARTMENT OF AGRICULTURE: List of Proprietary Substances - Permitted Use Codes 3A, J1, A1, G1, and L1.

CERCLA REPORTABLE QUANTITY: None

OSHA: Not hazardous under 29 CFR 1910.1200

RCRA: Not a hazardous material or a hazardous waste by listing or characteristic.

SARA TITLE III:

Section 302, Extremely Hazardous Substances: None

Section 311/312, Hazardous Categories: Non-hazardous

Section 313, Toxic Chemicals: None

Sodium Bicarbonate is reported in the EPA TSCA Inventory List.

Contains no VOCs.

NSF STANDARD 60: Corrosion and Scale Control in Potable Water. Max use 200 mg/l.

16. Other Information

NFPA Ratings: Health: **0** Flammability: **0** Reactivity: **0**

Date Prepared: July 18, 2014

Revision Information: SDS Section(s) changed since last revision of document

include: None.

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Prepared by: Terra Systems, Inc. **Phone Number:** (302) 798-9553 (U.S.A.)



A.4 Sodium Absorbate Technical Data Sheet and Safety Data Sheet

Chemical Safety Data Sheet MSDS / SDS

Sodium ascorbate

Revision Date: 2024-12-21 Revision Number: 1

SECTION 1: Identification of the substance/mixture and of the company/undertaking

Product identifier

Product name : Sodium ascorbate

CBnumber : CB8155737

CAS : 134-03-2

EINECS Number : 205-126-1

Synonyms: sodium ascorbate, sodium L-ascorbate

Relevant identified uses of the substance or mixture and uses advised against

Relevant identified uses : For R&D use only. Not for medicinal, household or other use.

Uses advised against : none

Company Identification

Company : Chemicalbook

Address : Building 1, Huihuang International, Shangdi 10th Street, Haidian District, Beijing

Telephone : 400-158-6606

SECTION 2: Hazards identification

GHS Label elements, including precautionary statements

Signal word No signal word

Hazard statement(s)

none

Prevention

none

Response

none

Storage

none

Disposal

none

SECTION 3: Composition/information on ingredients

Substance

Product name : Sodium ascorbate

Synonyms : sodium ascorbate, sodium L-ascorbate

CAS : 134-03-2
EC number : 205-126-1
MF : C6H7NaO6
MW : 198.11

SECTION 4: First aid measures

Description of first aid measures

If inhaled

After inhalation: fresh air.

In case of skin contact

In case of skin contact: Take off immediately all contaminated clothing. Rinse skin with water/ shower.

In case of eye contact

After eye contact: rinse out with plenty of water. Remove contact lenses.

If swallowed

After swallowing: make victim drink water (two glasses at most). Consult doctor if feeling unwell.

Most important symptoms and effects, both acute and delayed

The most important known symptoms and effects are described in the labelling (see section 2.2) and/or in section 11

Indication of any immediate medical attention and special treatment needed

No data available

SECTION 5: Firefighting measures

Extinguishing media

Suitable extinguishing media

Water Foam Carbon dioxide (CO2) Dry powder

Unsuitable extinguishing media

For this substance/mixture no limitations of extinguishing agents are given.

Special hazards arising from the substance or mixture

Carbon oxides Sodium oxides Combustible.

Development of hazardous combustion gases or vapours possible in the event of fire. Risk of dust explosion.

Advice for firefighters

In the event of fire, wear self-contained breathing apparatus.

Further information

Suppress (knock down) gases/vapors/mists with a water spray jet. Prevent fire extinguishing water from contaminating surface water or the ground water system.

NFPA 704



■ HEALTH 1 Exposure would cause irritation with only minor residual injury (e.g. acetone, sodium bromate, potassium chloride)

Materials that require considerable preheating, under all ambient temperature conditions, before ignition and combustion

- FIRE 1 can occur. Includes some finely divided suspended solids that do not require heating before ignition can occur. Flash point at or above 93.3 °C (200 °F). (e.g. mineral oil, ammonia)
- REACT 0 Normally stable, even under fire exposure conditions, and is not reactive with water (e.g. helium, N2)

SPEC.

HAZ.

SECTION 6: Accidental release measures

Personal precautions, protective equipment and emergency procedures

Advice for non-emergency personnel: Avoid inhalation of dusts. Evacuate the danger area, observe emergency procedures, consult an expert.

For personal protection see section 8.

Environmental precautions

Do not let product enter drains.

Methods and materials for containment and cleaning up

Cover drains. Collect, bind, and pump off spills. Observe possible material restrictions (see sections 7 and 10). Take up dry. Dispose of properly. Clean up affected area. Avoid generation of dusts.

Reference to other sections

For disposal see section 13.

SECTION 7: Handling and storage

Precautions for safe handling

For precautions see section 2.2.

Conditions for safe storage, including any incompatibilities

Storage conditions

Tightly closed. Dry.

Light sensitive.

Specific end use(s)

Apart from the uses mentioned in section 1.2 no other specific uses are stipulated

SECTION 8: Exposure controls/personal protection

control parameter

Hazard composition and occupational exposure limits

Does not contain substances with occupational exposure limits.

Exposure controls

Personal protective equipment

Eye/face protection

Use equipment for eye protection tested and approved under appropriate government standards such as NIOSH (US) or EN 166(EU). Safety glasses

Respiratory protection

required when dusts are generated.

Our recommendations on filtering respiratory protection are based on the following standards: DIN EN 143, DIN 14387 and other accompanying standards relating to the used respiratory protection system.

Recommended Filter type: Filter type P1

The entrepeneur has to ensure that maintenance, cleaning and testing of respiratory protective devices are carried out according to the instructions of the producer.

These measures have to be properly documented.

Control of environmental exposure

Do not let product enter drains.

SECTION 9: Physical and chemical properties

Information on basic physicochemical properties

Appearance	light yellow crystalline	
Odour	odorless	
Odour Threshold	Not applicable	
pH	7 - 8 at 100 g/l at 20 °C	
Melting point/freezing point	Melting point/range: 220 °C	
Initial boiling point and boiling range	235 °C	
Flash point	No data available	
Evaporation rate	No data available	
Flammability (solid, gas)	May form combustible dust concentrations in air.	

Upper/lower flammability or explosive	No data available	
limits		
Vapour pressure	No data available	
Vapour density	No data available	
Relative density	1,88 at 19,7 °C - OECD Test Guideline 109	
Water solubility	642,6 g/l at 20 °C - OECD Test Guideline 105- completely soluble	
Partition coefficient: n-octanol/water	log Pow:< -4,2 at 22 °C - Bioaccumulation is not expected.	
Autoignition temperature	No data available	
Decomposition temperature	232 °C -	
Viscosity	Viscosity, kinematic: No data available Viscosity, dynamic: No data available	
Explosive properties	No data available	
Oxidizing properties	No data available	
•		

Other safety information

Surface tension 74 mN/m at 20,3 °C

- OECD Test Guideline 115

SECTION 10: Stability and reactivity

Reactivity

The following applies in general to flammable organic substances and mixtures: in correspondingly fine distribution, when whirled up a dust explosion potential may generally be assumed.

Chemical stability

The product is chemically stable under standard ambient conditions (room temperature) .

Possibility of hazardous reactions

Violent reactions possible with:

Oxidizing agents

Conditions to avoid

Light.

no information available

Incompatible materials

No data available

Hazardous decomposition products

In the event of fire: see section 5

SECTION 11: Toxicological information

Information on toxicological effects

Acute toxicity

LD50 Oral - Rat - 11.900 mg/kg

Remarks: (calculated on the free acid)(RTECS)

Skin corrosion/irritation

Skin - Rabbit

Result: No skin irritation - 4 h (OECD Test Guideline 404)

Remarks: (in analogy to similar products)

The value is given in analogy to the following substances: ascorbic acid

Serious eye damage/eye irritation

Eyes - Rabbit

Result: No eye irritation (OECD Test Guideline 405)

Remarks: (in analogy to similar products)

The value is given in analogy to the following substances: ascorbic acid

Respiratory or skin sensitization

Local lymph node assay (LLNA) - Mouse

Result: negative

(OECD Test Guideline 429)

Germ cell mutagenicity

No data available

Carcinogenicity

No data available

Reproductive toxicity

No data available

Specific target organ toxicity - single exposure

No data available

Specific target organ toxicity - repeated exposure

No data available

Aspiration hazard

Toxicity

sce-hmn:lym 100 mmol/L MUREAV 60,321,79

SECTION 12: Ecological information

Toxicity

Toxicity to fish

static test LC50 - Oncorhynchus mykiss (rainbow trout) - > 1.020 mg/l - 96 h

(OECD Test Guideline 203)

Toxicity to daphnia and other aquatic invertebrates

semi-static test EC50 - Daphnia magna (Water flea) - 74 mg/l - 48 h (OECD Test Guideline 202)

Toxicity to algae

static test ErC50 - Pseudokirchneriella subcapitata - > 74 mg/l - 72 h

(OECD Test Guideline 201)

Persistence and degradability

Biodegradability aerobic - Exposure time 28 d

Result: > 99 % - Readily biodegradable. (OECD Test Guideline 301A)

Bioaccumulative potential

No data available

Mobility in soil

No data available

Results of PBT and vPvB assessment

This substance/mixture contains no components considered to be either persistent, bioaccumulative and toxic (PBT), or very persistent and very bioaccumulative (vPvB) at levels of 0.1% or higher.

Other adverse effects

No data available

SECTION 13: Disposal considerations

Waste treatment methods

Product

See www.retrologistik.com for processes regarding the return of chemicals and containers, or contact us there if you have further questions.

Incompatibilities

Incompatible with oxidizing agents, heavy metal ions, especially copper and iron, methenamine, sodium nitrite, sodium salicylate, and theobromine salicylate. The aqueous solution is reported to be incompatible with stainless steel filters.

SECTION 14: Transport information

UN number

ADR/RID: - IMDG: - IATA: -

UN proper shipping name

ADR/RID: Not dangerous goods IMDG: Not dangerous goods IATA: Not dangerous goods

Transport hazard class(es)

ADR/RID: - IMDG: - IATA: -

Packaging group

ADR/RID: - IMDG: - IATA: -

Environmental hazards

ADR/RID: no IMDG Marine pollutant: no IATA: no

Special precautions for user

Further information

Not classified as dangerous in the meaning of transport regulations.

SECTION 15: Regulatory information

Safety, health and environmental regulations/legislation specific for the substance or mixture

Regulations on the Safety Management of Hazardous Chemicals

China Catalog of Hazardous chemicals 2015:Not Listed. website: https://www.mem.gov.cn/

Measures for Environmental Management of New Chemical Substances

Philippines Inventory of Chemicals and Chemical Substances (PICCS):Listed. website: https://emb.gov.ph/

United States Toxic Substances Control Act (TSCA) Inventory:Listed. website: https://www.epa.gov/

Korea Existing Chemicals List (KECL):Listed. website: http://ncis.nier.go.kr

New Zealand Inventory of Chemicals (NZIoC):Listed. website: https://www.epa.govt.nz/

European Inventory of Existing Commercial Chemical Substances (EINECS):Listed. website: https://echa.europa.eu/

EC Inventory:Listed.

Vietnam National Chemical Inventory:Listed. website: https://chemicaldata.gov.vn/

Chinese Chemical Inventory of Existing Chemical Substances (China IECSC):Listed. website: https://www.mee.gov.cn/

SECTION 16: Other information

Abbreviations and acronyms

CAS: Chemical Abstracts Service

ADR: European Agreement concerning the International Carriage of Dangerous Goods by Road

RID: Regulation concerning the International Carriage of Dangerous Goods by Rail

IMDG: International Maritime Dangerous Goods

IATA: International Air Transportation Association

TWA: Time Weighted Average

STEL: Short term exposure limit

LC50: Lethal Concentration 50%

LD50: Lethal Dose 50%

EC50: Effective Concentration 50%

References

[1] CAMEO Chemicals, website: http://cameochemicals.noaa.gov/search/simple

[2] ChemlDplus, website: http://chem.sis.nlm.nih.gov/chemidplus/chemidlite.jsp

[3] ECHA - European Chemicals Agency, website: https://echa.europa.eu/

[4] eChemPortal - The Global Portal to Information on Chemical Substances by OECD, website:

http://www.echemportal.org/echemportal/index?pageID=0&request_locale=en

- [5] ERG Emergency Response Guidebook by U.S. Department of Transportation, website: http://www.phmsa.dot.gov/hazmat/library/erg
- [6] Germany GESTIS-database on hazard substance, website: http://www.dguv.de/ifa/gestis/gestis-stoffdatenbank/index-2.jsp
- [7] HSDB Hazardous Substances Data Bank, website: https://toxnet.nlm.nih.gov/newtoxnet/hsdb.htm
- [8] IARC International Agency for Research on Cancer, website: http://www.iarc.fr/
- [9] IPCS The International Chemical Safety Cards (ICSC), website: http://www.ilo.org/dyn/icsc/showcard.home
- 【10】 Sigma-Aldrich, website: https://www.sigmaaldrich.com/

Disclaimer:

The information in this MSDS is only applicable to the specified product, unless otherwise specified, it is not applicable to the mixture of this product and other substances. This MSDS only provides information on the safety of the product for those who have received the appropriate professional training for the user of the product. Users of this MSDS must make independent judgments on the applicability of this SDS. The authors of this MSDS will not be held responsible for any harm caused by the use of this MSDS.

A.5 Sodium Lactate Technical Data Sheet and Safety Data Sheet









Injection Ready 60% QRS-SLTM Sodium Lactate Quick Release Substrate For Aquifer Remediation and Conditioning

Terra Systems "*injection ready*" <u>60% QRSTM-SL</u> Sodium Lactate Quick Release Substrate is added to the groundwater to rapidly generate reducing conditions and provide the necessary carbon and hydrogen to support native or introduced microorganisms (*Dehalococcoides*) for the biodegradation of chlorinated solvents such as tetrachloroethene (PCE) and trichloroethene (TCE) to innocuous end products including ethene and ethane.

Key Communication Points

- 60% QRS[™]-SL sodium lactate is an inexpensive, soluble, food grade substrate
- It rapidly establishes reducing conditions to support the biodegradation of PCE, TCE, TECA, DNAPL (Sabre Project), Perchlorate, TCA, Cr⁶⁺, TNT, Uranium and Nitrate
- It is one of the most efficient electron donors available
- Provides 60% fermentable carbon
- 100% biobased content
- Its miscibility in water and low viscosity allow for effective transport with groundwater, enhancing subsurface radius of influence and minimizing the number of injection points.
- It arrives as a homogenous *injection ready substrate*, which results in lower field labor costs
- Proven effective at dry cleaners, semiconductor manufacturers, fabricators, manufacturing firms and military installations, that use and clean metal parts (air conditioners, dishwashers, etc.).

<u>**Table I:**</u> 60% QRS[™]-SL Sodium Lactate Specifications

Ingredient	Percent	Description	Benefit
Sodium lactate	60%	Rapidly biodegradable soluble substrate; miscible in water	Fast release source of carbon and hydrogen Rapidly generates reducing conditions High radius of influence Provides 60% fermentable carbon
pН	6.5 - 7	6.5 - 7	Optimum microbial activity
Organic Carbon (wt%)	60%		Provides 60% fermentable carbon
Biobased Content	100%		

Packaging: 5-gallon buckets, 55-gallon drums, 275-gallon IBC totes or bulk tankers.











"Injection Ready" 60% QRS-SL™ Plus Sodium Lactate

Quick Release Substrate with NutriPlus[™] a Proprietary Nutrient Package For Aquifer Remediation and Conditioning SAFETY DATA SHEET

Effective Date: 1/1/2020

1. Product Identification

Synonyms: Quick Release Substrate Plus (QRSTM-SL Plus); Sodium

Lactate; Propanoic acid, 2-Hydroxy Monosodium salt; L-

Lactic Acid, Sodium Salt

Recommended Use: Treatment of groundwater contaminated with chlorinated

solvents and other anaerobically degradable compounds.

Supplier: Terra Systems, Inc.

130 Hickman Road, Suite 1 Claymont, Delaware 19703 Telephone (302) 798-9553

Fax (302) 798-9554 www.terrasystems.net

2. Hazards Identification

Emergency Overview

Caution: May cause eye irritation.

Health Rating:1 - SlightFlammability Rating:0 - NoneReactivity Rating:0 - NoneContact Rating:1 - Slight

Protective Equipment: Goggles; Proper Gloves **Storage Color Code:** Orange (General Storage)

Potential Health Effects

Inhalation: Not expected to be a health hazard

Ingestion: Not expected to be a health hazard via ingestion

Skin Contact: No adverse effects expected

Eye Contact: May cause irritation, possible reddening

Chronic Exposure: No information found

Terra Systems, Inc.

130 Hickman Road, Suite 1, Claymont, Delaware 19703 Contact: Michael Free, VP, Sales and Marketing









Aggravation of Pre-existing Conditions:

No information found

3. Composition/Information on Ingredients

Ingredient	CAS#	Percent	Hazardous
Sodium Lactate	72-17-3	60	Yes
Organic Nutrients, Inorganic Nutrients and Surfactants	Mixture	1-5	No
Water	7732-18-5	35-39	No

Molecular Weight: No available Chemical Formula: Not available

The organic, inorganic nutrients and surfactant mixture is a trade secret and consists of ingredients of unknown acute toxicity.

4. First Aid Measures

Inhalation: Not expected to require first aid measures. Remove to fresh air.

Get medical attention for any breathing difficulty.

Ingestion: If large amounts were swallowed, give water to drink and get

medical advice.

Skin Contact: Not expected to require first aid measures. Wash exposed area

with soap and water. Get medical advice if irritation develops.

Eye Contact: Immediately flush eyes with plenty of water for at least 15

minutes, lifting upper and lower eyelids occasionally. Get

medical attention if irritation persists.

5. Fire Fighting Measures

Fire: Flash point: 110 C (230 F). Not considered to be a fire hazard.

Explosion: Not considered to be an explosion hazard.

Fire Extinguishing Media: Use any means suitable for extinguishing surrounding

fire.

Special Information: In the event of a fire, wear full protective clothing and NIOSH-

approved self-contained breathing apparatus with full face piece operated in the pressure demand or other positive

pressure mode.









6. Accidental Release Measures

Clean-up personnel may require protective clothing. Absorb in sand, paper towels, "Oil Dry", or other inert material. Scoop up and containerize for disposal. Flush trace residues to sewer with soap and water. Containerized waste may be sent to an approved waste disposal facility.

7. Handling and Storage

Keep in a tightly closed container, stored in a cool, dry, ventilated area. Protect against physical damage. Avoid long storage times. Containers of this material may be hazardous when empty since they do retain product residues (vapors, liquid). Observe all warnings and precautions listed for the product.

8. Exposure Controls/Personal Protection

Airborne Exposure Limits: None established.

Ventilation System: Not expected to require any special ventilation.

Personal Respirators (NIOSH

Approved): Not expected to require personal respirator usage.

Skin Protection: Wear protective gloves and clean body-covering clothing. **Eye Protection:** Use chemical safety goggles and/or a full face shield where

splashing is possible. Provide readily accessible eye wash

stations and safety showers.

Slips, Trips, and Falls: Material is slippery when spilled. Clean up with sand, paper

towels, or other inert material.

9. Physical and Chemical Properties

Appearance: Colorless to yellow liquid.

Odor: Odorless

Solubility: 100% soluble in water.

Specific Gravity (water=1): 1.32. (11.01 pounds per gallon)

pH: 6.5-8.5

% Volatiles by volume

@ **21C** (**70F**): No information found.

Boiling Point: 110 C (230 F) **Melting Point:** 17 C (63 F)

Flash Point (F):

Autoignition Temperature:

Decomposition Temperature:

No information found
No information found.

Vapor Density (Air=1): 0.7

Vapor Pressure (mm Hg): 14 @ 20 C (68 F) **Evaporation Rate (BuAc=1):** No information found

Viscosity @23 C (73 F): 100 centipoises

Terra Systems, Inc.

130 Hickman Road, Suite 1, Claymont, Delaware 19703 Contact: Michael Free, VP, Sales and Marketing









Partition Coefficient (octanol/water): No information found

10. Stability and Reactivity

Stability: Stable under ordinary conditions of use and storage.

Reactivity: Not reactive under ordinary conditions.

Hazardous Decomposition

Products: Carbon dioxide and carbon monoxide may form when

heated to decomposition.

Hazardous Polymerization: Will not occur.

Incompatibilities: Strong oxidizers, acids.

Conditions to Avoid: Incompatibles. Isolate from heat and open flame.

11. Toxicological Information

Oral rat LD50: 2000 mg/Kg. Irritation Data for Sodium Lactate: (Std Draize, rabbit, eye): 100 mg - mild.

12. Ecological Information

Environmental Fate: Mobile with water and readily biodegradable

Environmental Toxicity: Ecological injuries are not known or expected under

normal use; (No effect on Daphnia @ 10g/L)

Degradability: This product is completely biodegradable under both aerobic

and anaerobic conditions.

Soil Mobility: This compound will move with groundwater until the adsorbed

onto the soil. Degradation products may be mobile.

Bioaccumulation Potential: No information found.

13. Disposal Considerations

Whatever cannot be saved for recovery or recycling should be managed in an appropriate and approved waste disposal facility. Processing, use or contamination of this product may change the waste management options. State and local disposal regulations may differ from federal disposal regulations. Dispose of container and unused contents in accordance with federal, state and local requirements.









14. Transport Information

Not regulated.

15. Regulatory Information

\Chemical Inventory Sta Ingredient	ttus - Part 1\TSCA EC Japan Australia
· · · · · · · · · · · · · · · · · · ·	Yes Yes Yes Yes
Water (7732-18-5)	Yes Yes Yes Yes
\Chemical Inventory St	tatus - Part 2\
Ingradiant	Canada Korea DSL NDSL Phil.
Ingredient	
Sodium Lactate (72-17-3)	Yes Yes No Yes
Water (7732-18-5)	Yes Yes No Yes
•	national Regulations - Part 1\ARA 302SARA 313
Ingredient	RQ TPQ List Chemical Catg.
Sodium Lactate (72-17-3)	No No No
· · · · · · · · · · · · · · · · · · ·	No No No No
\Federal, State & Intern	national Regulations - Part 2\
Ingredient	CERCLA 261.33 8(d)
Sodium Lactate (72-17-3)	No No No
Water (7732-18-5)	No No No

Chemical Weapons Convention: No TSCA 12(b): No CDTA: No SARA 311/312: Acute: Yes Chronic: No Fire: No Pressure: No Pressur

Reactivity: No (Mixture / Liquid)









16. Other Information

NFPA Ratings: Health: **1** Flammability: **0** Reactivity: **0**

Date Prepared: March 28, 2014

Revision Information: SDS Section(s) changed since last revision of document

include: None.

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Prepared by: Terra Systems, Inc. **Phone Number:** (302) 798-9553 (U.S.A.)

A.6 Technical Data Sheets for Gravity Feed Injection



INTERSTATE PRODUCTS, INC.

"Your Road to Quality Environmental Products Since 1996"

https://store.interstateproducts.com

1-800-474-7294

Ultratech IBC Spill Pallet Plus With Drain 1158





Ultratech IBC Spill Pallet 1158 Specifications:

Dimensions: 62" x 62" x 28"

Uniformly Distributed Load: 8,500 lbs Containment Capacity: 365 Gallons

Weight: 324 lbs

Options: Ultra Bucket Shelf (Part#1160)

Contact us anytime for IPI customer service on all UltraTech IBC Spill Pallet & Spill Containment Products

Key Features:

- Low profile, 28" overall height allows safe and convenient IBC tank handling and dispensing.
- All polyethylene construction offers excellent chemical resistance and will not rust or corrode.
- Forkliftable allows convenient positioning to desired locations.
- Low-cost design with value-added features and benefits.
- Large 52" x 52" deck allows safe and convenient placement of IBC tanks.
- Small footprint 62" x 62" unit requires minimal floor space.
- 365-gallon sump capacity meets SPCC and EPA Container Storage and Spill Containment Regulations.
- Optional Pull Over Cover keeps rainwater out of the sump and helps comply with Stormwater Management Regulations.
- Five inner polyethylene columns support uniformly distributed loads of up to 8,500 lbs. All components are easily removed for cleaning.
- Optional Bucket Shelf captures spills from dispensing. Spills that exceed 3 gallons are channeled into the 365-gallon sump through a bulkhead fitting.



The Therma Gauge—Type H



Part Number— H-(size opening)-(tank depth)+(ext)-(list options) Sample—H-2-48+12

> The Cary Company 1195 W. Fullerton Ave. Addison, IL 60101

What it is:

Top mounted liquid level gauge that can measure from 6 inches to 144 inches in depth. Bushing size can be 2" or 1.5". Gauges are custom made in house to fit your tank. Can also accommodate for double walls and pipe risers. The Type H Gauge has a 2 piece bushing construction that allows you to aim your swing arm away from walls, corners, and other obstructions.

Additional Options:

Audible Alarm Accessory: This add on feature can turn your mechanical gauge into an audible Hi or Lo level alarm.

LED At-A-Glance Accessory: Another add on feature. This can give your mechanical gauge remote reading capability.

Gauge Guard: A cover that protects the exposed plastic components on top of the gauge.

Material Packages / Gauge Packages:

- Standard Therma Gauge-Type H-Rods are galvanized steel, floats are HDPE plastic, bushing is aluminum, calibration top is plastic.
- Stainless Level One-Type H-S1- Rods become stainless.
- Stainless Level Two-Type -H-S2- Rods and floats become stainless.
- Stainless Level Three-Type H-S3- This model comes standard with Glass Calibration, Aluminum Locking Nut, Stainless Steel Bushing, Stainless Steel Rods, and Stainless Steel Floats.
- All Plastic Model—Type PH—This model comes standard with Glass Calibration, PVC Gauge Guard, PVC Bushing, PEEK Plastic Rods, HDPE Plastic Float.

Contact us for more info: Ph: 630-629-6600

thecarycompany.com



Therma Gauge Accessories and Replacement Parts

Accessories



Glass Calibration

LED At A Glance Remote Display-Turns the direct reading Therma gauge into a wired remote reading gauge. (part # add -Ledaag)

Direct Mount Alarms - Audible alarm that mounts directly to the gauge and provides your choice of overfill or low level warnings. (part # add -Dalarm)

Remote Mount Alarms- Audible alarm that can be remotely wired and provides your choice of overfill or low level warnings. (part # add -Ralarm)

Aluminum Lock Nut- Replace the red lock nut for added durability. (part # add -ALN)

Gauge Guard- Protective Cover that replaces the red locking nut providing protection from physical damage and weathering

Glass Calibration - The internal piece of the calibration becomes glass. Provides protection from heat, fumes, weathering, and also



Direct Mount Alarm









Led At A Glance



Aluminum Lock Nut





- **HG-Kit** Replaces all the top components with standard materials and a glass calibration.
- **HALN-Kit-** Replaces all the top components with standard materials and an aluminum lock nut.
- **HGALN-Kit-** Replaces the top components with a glass calibration and aluminum lock nut.

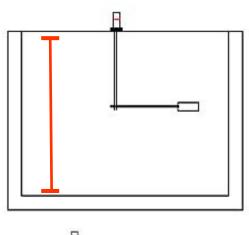


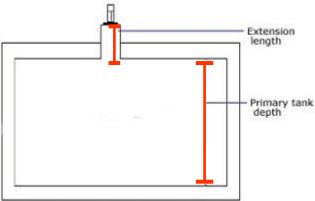
The Cary Company 1195 W. Fullerton Ave. Addison, IL 60101

Contact us for more info: Ph: 630-629-6600



Ordering, Installation, Maintenance and Operation





Installation Instructions:

When installing, unthread the red nut and remove the calibration and bushing. Thread in the bushing first. Drop the gauge through the center hole, make sure your swing arm is facing in the direction you want. Replace the calibration and tighten down your red locking nut.

Instructions for Operation:

The top of the red indicator is an indication of your approximate fuel level in your tank. Once installed, you simply view the calibration to monitor your tank level. Indicator is calibrated in eighths of a tank on one side, and quarters of a tank on the other as well as Gallons and Liters.

The Cary Company 1195 W. Fullerton Ave. Addison, IL 60101

Ordering Instructions:

Part Number Layout-

H-(Opening Size)-(tank depth)-(ext. length)-(material package)-(options)

- 1. State the gauge type (Type H for metal or Type PH for all plastic)
- 2. State the opening size your are using. (1.5"---2")
- 3. State the tank depth (pictured left). Tank depth is the distance from empty to full.
- 4. When necessary, provide the extension length (pictured left). Extension length is the distance from full to gauge threads. (sometimes zero)
- 5. List the material package you prefer for best compatibility with your liquid. Material packages are explained on page 1 of this document. Skip this if you want standard materials
- 6. List other material options and accessories that you prefer. These options and accessories can be located on page 2 of this document. Skip this if you want standard materials.

Example Part numbers

- H-2-48-12-S3
- H-2-24-GLC

Maintenance:

The Therma Gauge can fail in one of three ways.

- The calibration assembly on the outside of the tank can become weathered, unreadable, or broken.
- The connecting rods that extend into the tank can come apart or become damaged.
- The float that sits at the bottom of the gauge can come off or become saturated with product (leak).

Recommended Maintenance Procedures:

- Once per month: Inspect the top of the unit once per month to ensure that the calibration is visible, readable, and unbroken. (these top pieces can be easily replaced without having to replace the whole gauge)
- Once every 6 months: Unthread the unit and carefully remove it from the tank. Be sure all rods are connected as one piece from the red indicator all the way down to the float. Also inspect the float for damage or leakage. To ensure proper working order, manually raise the float arm from empty to full to be sure that the red indicator freely moves up and down with the motion of the float rod.
- When performing maintenance, be sure to follow the installation instructions above to prevent damaging the gauge

Contact us for more info: Ph: 630-629-6600



The Therma Gauge—FAQ, Troubleshooting





The Cary Company 1195 W. Fullerton Ave. Addison, IL 60101

Frequently Asked Questions:

Q: What is the difference between the Therma Gauge and the standard Direct Reading Gauge.

A: The two gauges are nearly identical, but the Therma gauge has a two piece bushing. This allows you to remove the bushing, thread it in, then drop the gauge in and point the swing arm in the direction of your choice. This makes installation easier when you fitting is near a wall/corner or if you have baffles/obstructions in your tank that you need to avoid.

Q: The top plastic portion of my gauge has become damaged or weathered, what can I do?

A: The top of the gauge is easily replaced, just order a H-Repair Kit which replaces all of the top components of the gauge.

Q: How can I prevent the top of my gauge from becoming weathered or warped in the future?

A: Several upgrades exist that improve the life and performance of the gauge. The glass calibration, which makes the internal part of the calibration into glass, increases the gauges resistance to heat, fumes, and sun.

Q: My gauge top keeps getting broken by hoses/weather/people, do you have a more durable option?

A: We offer something called a Gauge Guard. This is an aluminum cage that replaces the red lock nut and protects the top of your gauge from physical damage and weathering.

Q: The fire inspector says my gauge is not up to code? How can I satisfy their requirements.

A: If you get the glass calibration and the gauge guard, you are creating a glass and metal barrier between the inside of the tank and the external environment. Explain this to your inspector and see if it satisfies their requirements.

Q: I have aggressive chemicals that I need to gauge, will this work?

A: Our gauge has many material options, including all stainless components or all plastic components. We can help with determining what material option you need, but compatibility is ultimately the end users decision.

Contact us for more info: Ph: 630-629-6600

PRECISION IBC CALIBRATION CHART

for Supertainer IBC Tanks

FILL HEIGHT (inches)	42" x 48" (GALLONS)	42" x 42" (GALLONS)	FILL HEIGHT (inches)	42" x 48" (GALLONS)	42" x 42" (GALLONS)
3"	22.3	18	35"	306.6	258.8
4"	31	25.5	36"	316.9	266.3
5"	39.7	33.1	37"	325.6	273.9
6"	48.4	40.6	38"	334.3	281.4
7"	57.2	48.1	39"	343.1	288.9
8"	65.9	55.6	40"	351.8	296.4
9"	74.6	63.2	41"	360.5	304
10"	83.4	70.7	42"	370.2	311.5
11"	92.1	78.2	43"	379	319
12"	100.8	85.7	44"	387.7	326.5
13"	109.6	93.3	45"	396.4	334.1
14"	118.3	100.8	46"	405.2	341.6
15"	127	108.3	47"	413.9	349.1
16"	135.7	115.8	48"	422.6	356.7
17"	144.5	123.4	49"	431.3	364.2
18"	152.2	130.9	50"	440.1	371.7
19"	160.9	138.4	51"	448.8	379.2
20"	169.6	145.9	52"	457.5	386.8
21"	177.6	153.5	53"	466.3	394.3
22"	186.4	161	54"	477.6	401.8
23"	195.1	168.5	55"	486.3	409.3
24"	203.8	176	56"	495.1	416.9
25"	212.5	183.6	57"	503.8	424.4
26"	221.3	191.1	58"	512.5	431.9
27"	230	198.6	59"	521.2	439.4
28"	238.7	206.1	60"	530	447
29"	247.5	213.7	61"	538.7	454.5
30"	263	221.2	62"	547.4	462
31"	271.7	228.7	63"	556.2	469.5
32"	280.4	236.2	64"	564.9	477.1
33"	289.2	243.8	65"	573.6	484.6
34"	297.9	251.3	66"	582.4	492.1



*NOTE: All capacity values are approximate and are not to be used for dispensing. Values may vary from IBC to IBC.

Believe in better service.

PRECISIONIBC.COM

BROUSSARD, L. 888.805.1247

FAIRHOPE, AL 800.544.7069

MAY 2018

Appendix B Remedial Design

B.1 TA-02 Treatment Area Design

Site Name: OU7 F	Pit 1 TA-02 AEHAD	OG-11 Hotspot		RETURN TO COVER PAGE
		ded boxes are use	•	
Treatment Zone Physical Dimensions	Values	Range	Units	User Notes
Vidth (Perpendicular to predominant groundwater flow direction)	45	1-10,000	feet	Hotspot in AEHADG-1 and MW-36 Area, 14 DPT Points
ength (Parallel to predominant groundwater flow)	50	1-1,000	feet	40.00
Saturated Thickness Treatment Zone Cross Sectional Area	16 720	1-100	feet ft ²	10-26
Teatment Zone Volume	36,000		ft ³	
reatment Zone Total Pore Volume (total volume x total porosity)	96,967		gallons	
reatment Zone Effective Pore Volume (total volume x effective pore			gallons	
Design Period of Performance	3.0	.5 to 5	year	
Design Factor (times the electron acceptor hydrogen demand)	10.0	2 to 20	unitless	
Treatment Zone Hydrogeologic Properties				
Total Porosity	36%	.05-50	percent	PDI Eastover OU 7
Effective Porosity Average Aquifer Hydraulic Conductivity	27% 5	.05-50 .01-1000	percent ft/day	PDI Eastover OU 7, average based on TP, AP, clay conter OU7 PDI
Average Hydraulic Gradient	1.61E-02	0.0001-0.1	ft/ft	2024 Annual Report
Average Groundwater Seepage Velocity through the Treatment Zon			ft/day	2024 / William Report
Average Groundwater Seepage Velocity through the Treatment Zon			ft/yr	
Average Groundwater Discharge through the Treatment Zone	158,285		gallons/year	<u> </u>
Soil Bulk Density	1.63	1.4-2.0	gm/cm ³	Average OU- PDI for Eastover and Aquia
Soil Fraction Organic Carbon (foc)	0.28%	0.01-10	percent	PDI-38 Eastover Saturated Zone OU-7 x 2
Native Electron Acceptors				
A. Aqueous-Phase Native Electron Acceptors	- 10	0.011.15		Data from 14/0000 (August A 51/4 DO 14 A 18/4 C 5
Oxygen Vitrate	4.3	0.01 to 10 0.1 to- 20	mg/L	Data from 11/2020 (Average AEHADG-11, MW-36) AEHADG-11 03/2024
virrate Sulfate	0.03	10 to 5,000	mg/L mg/L	AEHADG-11 03/2024 AEHADG-11 03/2024
Carbon Dioxide (estimated as the amount of Methane produced)	10.0	0.1 to 20	mg/L	Estimate based on previous EISB injections
parbon bloxido (ostimated as the amount of methane produced)	10.0	0.1 to 20	mg/L	Estimate based on provides E105 injections
3. Solid-Phase Native Electron Acceptors				
Manganese (IV) (estimated as the amount of Mn (II) produced)	0	0.1 to 20	mg/L	AEHADG-11 03/2024
ron (III) (estimated as the amount of Fe (II) produced)	3	0.1 to 20	mg/L	AEHADG-11 03/2024
Contaminant Electron Acceptors				
etrachloroethene (PCE)	1.250		mg/L	AEHADG-11 03/2024
Trichloroethene (TCE) Dichloroethene (cis-DCE, trans-DCE, and 1,1-DCE)	0.520 1.300		mg/L	AEHADG-11 03/2024 AEHADG-11 03/2024
/inyl Chloride (VC)	0.050		mg/L mg/L	AEHADG-11 03/2024 AEHADG-11 03/2024
Carbon Tetrachloride (CT)	0.040		mg/L	AEHADG-11 03/2024
richloromethane (or chloroform) (CF)	0.024		mg/L	AEHADG-11 03/2024
Dichloromethane (or methylene chloride) (MC)	0.000		mg/L	AEHADG-11 03/2024
Chloromethane	0.000		mg/L	AEHADG-11 03/2024
etrachloroethane (1,1,1,2-PCA and 1,1,2,2-PCA)	0.001		mg/L	AEHADG-11 03/2024
richloroethane (1,1,1-TCA and 1,1,2-TCA)	1.100		mg/L	AEHADG-11 03/2024
Dichloroethane (1,1-DCA and 1,2-DCA)	0.040		mg/L	AEHADG-11 03/2024
Chloroethane	0.120		mg/L	AEHADG-11 03/2024
Perchlorate	0.000		mg/L	No data
Aquifer Geochemistry (Optional Screening Paramet	ers)			
A. Aqueous Geochemistry	224	400 to 1500	m\/	AEHADC 11 03/2024
Oxidation-Reduction Potential (ORP)	234	-400 to +500	mV °C	AEHADG-11 03/2024 AEHADG-11 03/2024
Temperature	12 5.1	5.0 to 30 4.0 to 10.0	su	AEHADG-11 03/2024 AEHADG-11 03/2024
Nkalinity	55	10 to 1,000	mg/L	AEHADG-11 03/2024 AEHADG-11 03/2024
otal Dissolved Solids (TDS, or salinity)	100	10 to 1,000	mg/L	No data
Specific Conductivity	158	100 to 10,000		AEHADG-11 03/2024
Chloride	6	10 to 10,000		AEHADG-11 03/2024
Sulfide - Pre injection	0.1	0.1 to 100	mg/L	Estimate Not Detected
Sulfide - Post injection	1.7	0.1 to 100	mg/L	AEHADG-11 03/2024
3. Aquifer Matrix				2014 2022 2 115
Total Iron	18000	200 to 20,000		CSM 2006 Soil Eastover
Cation Exchange Capacity Veutralization Potential	1.0%	1.0 to 10 1.0 to 100	meq/100 g Percent as CaCO ₃	CSM 2006 Soil Eastover CSM 2006 Soil Eastover
TOUT AND A TOUR THAT	1.070	1.0 to 100		CON 2000 CON ENSIONEN

Table S.2 S	Substrate Ca	llculations ir	n Hydrogen I	Equivalents		
Site Name:	OU7 Pit 1 T	A-02 AEHADG	-11 Hotspot		RETURN TO	COVER PAGE
<u>.</u>				NOTE: Open cells	•	
1. Treatment Zone Physical Dimensions				Values	Range	Units
Width (Perpendicular to predominant groundwater flow	w direction)			45	1-10,000	feet
Length (Parallel to predominant groundwater flow)	50	1-1,000	feet			
Saturated Thickness				16	1-100	feet
Treatment Zone Cross Sectional Area				720		ft ²
Treatment Zone Volume				36,000		ft ³
Treatment Zone Effective Pore Volume (total volume :	x effective porosit	y)		72,725		gallons
Design Period of Performance				3.0	.5 to 5	year
2. Treatment Zone Hydrogeologic Propertie	s					
Total Porosity				0.36	.05-50	
Effective Porosity				0.27	.05-50	
Average Aquifer Hydraulic Conductivity				5	.01-1000	ft/day
Average Hydraulic Gradient	-			0.0161	0.1-0.0001	ft/ft
Average Groundwater Seepage Velocity through the T				0.30		ft/day
Average Groundwater Seepage Velocity through the T		,		108.8 158,285		ft/yr
Average Groundwater Flux through the Treatment Zor	·	,				gallons/year
Soil Bulk Density Soil Fraction Organic Carbon (foc)				1.63 0.0028	1.4-2.0 0.0001-0.1	gm/cm ³
				0.0020	0.0001-0.1	
3. Initial Treatment Cell Electron-Acceptor I	Demand (one	total pore volu	me)			
				Stoichiometric	Hydrogen	Electron
A. Aqueous-Phase Native Electron Acceptors		Concentration	Mass	demand	Demand	Equivalents pe
•		(mg/L)	(lb)	(wt/wt h ₂)	(lb)	Mole
Oxygen		4.3	2.61	7.94	0.33	4
Nitrate (denitrification)		0.0	0.02	12.30	0.00	5
Sulfate		16.25	9.86	11.91	0.83	8
Carbon Dioxide (estimated as the amount of methane	produced)	10.0	6.07	1.99	3.05	8
		Soluble Compet	ing Electron Acc	eptor Demand (lb.)	4.21	
				Stoichiometric	Hydrogen	Electron
B. Solid-Phase Native Electron Acceptors		Concentration	Mass	demand	Demand	Equivalents pe
(Based on manganese and iron produced)		(mg/L)	(lb)	(wt/wt h ₂)	(lb)	Mole
Manganese (IV) (estimated as the amount of Mn (II) p	roduced)	0.2	1.01	27.25	0.04	2
Iron (III) (estimated as the amount of Fe (II) produced)	3.5	15.81	55.41	0.29	1	
	Soli	id-Phase Compet	ing Electron Acc	eptor Demand (lb.)	0.32	
				Stoichiometric	Hydrogen	Electron
C. Soluble Contaminant Electron Acceptors		Concentration	Mass	demand	Demand	Equivalents pe
		(mg/L)	(lb)	(wt/wt h ₂)	(lb)	Mole
Tetrachloroethene (PCE)		1.250	0.76	20.57	0.04	8
Trichloroethene (TCE)		0.520	0.32	21.73	0.01	6
Dichloroethene (cis-DCE, trans-DCE, and 1,1-DCE)		1.300	0.79	24.05	0.03	4
Vinyl Chloride (VC)		0.050	0.03	31.00	0.00	2
Carbon Tetrachloride (CT)		0.040	0.02	19.08	0.00	8
Trichloromethane (or chloroform) (CF)		0.024	0.01	19.74	0.00	6
Dichloromethane (or methylene chloride) (MC)		0.000	0.00	21.06	0.00	2
Chloromethane Tetrachloroethane (1,1,1,2-PCA and 1,1,2,2-PCA)		0.000	0.00	25.04 20.82	0.00	8
Trichloroethane (1,1,1,2-PCA and 1,1,2,2-PCA)		1.100	0.67	22.06	0.00	6
Dichloroethane (1,1,1-1CA and 1,1,2-1CA)		0.040	0.02	24.55	0.00	4
Chloroethane		0.120	0.07	32.00	0.00	2
Perchlorate		0.000	0.00	12.33	0.00	6
	Total S	oluble Contamin	ant Electron Acc	eptor Demand (lb.)	0.12	
				Stoichiometric	Hydrogen	Electron
D. Sorbed Contaminant Electron Acceptors	Koc	Soil Conc.	Mass	demand	Demand	Equivalents pe
(Soil Concentration = Koc x foc x Cgw)	(mL/g)	(mg/kg)	(lb)	(wt/wt h ₂)	(lb)	Mole
Tetrachloroethene (PCE)	263	0.92	3.37	20.57	0.16	8
Trichloroethene (TCE)	107	0.16	0.57	21.73	0.03	6
Dichloroethene (cis-DCE, trans-DCE, and 1,1-DCE)	45	0.16	0.60	24.05	0.02	4
Vinyl Chloride (VC)	3.0	0.00	0.00	31.00	0.00	2
Carbon Tetrachloride (CT)	224	0.03	0.09	19.08	0.00	8
Trichloromethane (or chloroform) (CF)	63	0.00	0.02	19.74	0.00	6
Dichloromethane (or methylene chloride) (MC)	28	0.00	0.00	21.06	0.00	4
Chloromethane	25	0.00	0.00	25.04	0.00	2
Tetrachloroethane (1,1,1,2-PCA and 1,1,2,2-PCA)	117	0.00	0.00	20.82	0.00	8
Trichloroethane (1,1,1-TCA and 1,1,2-TCA)	105	0.32	1.18	22.06	0.05	6
Dichloroethane (1,1-DCA and 1,2-DCA)	30	0.00	0.01	24.55	0.00	4
Chloroethane Perchlorate	0.0	0.00	0.00	32.00 12.33	0.00	2
reidiididle						6
	Total C	Sorbed Contamin	ant Electron Aca	antor Domand (IL \)	0.28	

		lculations in	Hydrogen E	Equivalents		
4. Treatment Cell Electron-Acceptor Flux (p	er year)					
				Stoichiometric	Hydrogen	Electron
A. Soluble Native Electron Acceptors		Concentration	Mass	demand	Demand	Equivalents per
		(mg/L)	(lb)	(wt/wt h ₂)	(lb)	Mole
Oxygen		4.3	5.68	7.94	0.72	4
Nitrate (denitrification)		0.0	0.03	10.25	0.00	5
Sulfate		16.25	21.46	11.91	1.80	8
Carbon Dioxide (estimated as the amount of Methane		10	13.21	1.99	6.64	8
	Tota	al Competing Elec	ctron Acceptor D	emand Flux (lb/yr)	9.2	
				Stoichiometric	Hydrogen	Electron
B. Soluble Contaminant Electron Acceptors		Concentration	Mass	demand	Demand	Equivalents per
		(mg/L)	(lb)	(wt/wt h ₂)	(lb)	Mole
Tetrachloroethene (PCE)		1.250	1.65	20.57	0.08	8
Trichloroethene (TCE)		0.520	0.69	21.73	0.03	6
Dichloroethene (cis-DCE, trans-DCE, and 1,1-DCE)		1.300	1.72	24.05	0.07	4
Vinyl Chloride (VC)		0.050	0.07	31.00	0.00	2
Carbon Tetrachloride (CT)		0.040	0.05	19.08	0.00	8
Trichloromethane (or chloroform) (CF)		0.024	0.03	19.74	0.00	6
Dichloromethane (or methylene chloride) (MC)		0.000	0.00	21.06	0.00	4
Chloromethane		0.000	0.00	25.04	0.00	2
Tetrachloroethane (1,1,1,2-PCA and 1,1,2,2-PCA)		0.001	0.00	20.82	0.00	8
Trichloroethane (1,1,1-TCA and 1,1,2-TCA)		1.100	1.45	22.06	0.07	6
Dichloroethane (1,1-DCA and 1,2-DCA)		0.040	0.05	24.55	0.00	4
Chloroethane		0.120	0.16	32.00	0.00	2
Perchlorate	T-1-10-1-11-	0.000	0.00	12.33	0.00	6
			-	emand Flux (lb/yr)	0.26	<u>]</u>
				t First Year (lb)		_
		i otai Lite-Cyci	e Hyarogen Ke	equirement (lb)	33.2	_
5. Design Factors						
Microbial Efficiency Uncertainty Factor					2X - 4X	
Methane and Solid-Phase Electron Acceptor Uncertaint					2X - 4X	
Remedial Design Factor (e.g., Substrate Leaving React	tion Zone)				1X - 3X	7
				Design Factor	10.0	
Total	Life-Cycle Hyd	Irogen Require	ment with Des	sign Factor (lb)	331.9	
6. Acronyns and Abbreviations						_
°C =degrees celsius	mea/100 a - mill	iequivalents per 10)() arams			
μs/cm = microsiemens per centimeter	mg/kg = milligran		o grains			
cm/day = centimeters per day	mg/L = milligram					
cm/sec = centimeters per second	m/m = meters pe					
ft ² = square feet	mV = millivolts	i illotois				
ft/day = feet per day	m/yr = meters pe	er vear				
ft/ft = foot per foot	su = standard pH					
ft/yr = feet per year		etration molecular h	nydrogen, weight p	er weight		
gm/cm ³ = grams per cubic centimeter			, 5-, -9	. 3		
kg of CaCO3 per mg = kilograms of calcium carbonat	e per milligram					
lb = pounds						

Table S.3

Hydrogen Produced by Fermentation Reactions of Common Substrates

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Substrate	Molecular Formula	Substrate Molecular Weight (gm/mole)	Moles of Hydrogen Produced per Mole of Substrate	Ratio of Hydrogen Produced to Substrate (gm/gm)	Range of Moles H ₂ /Mole Substrate
Lactic Acid	C ₃ H ₆ O ₃	90.1	2	0.0448	2 to 3
Molasses (assuming 100% sucrose)	C ₁₂ H ₂₂ O ₁₁	342	8	0.0471	8 to 11
High Fructose Corn Syrup (assuming 50% fructose and 50% glucose)	C ₆ H ₁₂ O ₆	180	4	0.0448	4 to 6
Ethanol	C ₂ H ₆ O	46.1	2	0.0875	2 to 6
Whey (assuming 100% lactose)	C ₁₂ H ₂₂ O ₁₁	342	11	0.0648	11
HRC [®] (assumes 40% lactic acid and 40% glycerol by weight)	C ₃₉ H ₅₆ O ₃₉	956	28	0.0590	28
Linoleic Acid (Soybean Oil, Corn Oil, Cotton Oil)	C ₁₈ H ₃₂ O ₂	281	16	0.1150	16

Table S.4 Estimated Substrate Requirements for Hydrogen Demand in Table S.3

Design Life (years): 3

Substrate	Design Factor	Pure Substrate Mass Required to Fulfill Hydrogen Demand (pounds)	Substrate Product Required to Fulfill Hydrogen Demand (pounds)	Substrate Mass Required to Fulfill Hydrogen Demand (milligrams)	Effective Substrate Concentration (mg/L)
Lactic Acid	10.0	7,415	7,415	3.36E+09	1,623
Sodium Lactate Product (60 percent solution)	10.0	7,415	15,383	3.36E+09	1,623
Molasses (assuming 6 0	10.0	7,044	11,740	3.20E+09	1,541
HFCS (assuming 40% fructose and 40% glucose by weight)	10.0	7,416	9,270	3.36E+09	1,623
Ethanol Product (assuming 80% ethanol by weight)	10.0	3,792	4,740	1.72E+09	830
Whey (assuming 100% lactose)	10.0	5,118	7,312	2.32E+09	1,120
HRC® (assumes 40% lactic acid and 40% glycerol by weight)	10.0	5,621	5,621	2.55E+09	984
Linoleic Acid (Soybean Oil, Corn Oil, Cotton Oil)	10.0	2,886	2,886	1.31E+09	632
Commercial Vegetable Oil Emulsion Product (60% oil by weight)	10.0	2,886	4,810	1.31E+09	632

NOTES: Sodium Lactate Product

- 1. Assumes sodium lactate product is 60 percent sodium lactate by weight.
- 2. Molecular weight of sodium lactate (CH_3 -CHOH-COONa) = 112.06.
- 3. Molecular weight of lactic Acid $(C_6H_6O_3) = 90.08$.
- 4. Therefore, sodium lactate product yields 48.4 (0.60 x (90.08/112.06)) percent by weight lactic acid.
- 5. Weight of sodium lactate product = 11.0 pounds per gallon.
- 6. Pounds per gallon of lactic acid in product = 1.323×8.33 lb/gal H2O x $0.60 \times (90.08/112.06) = 5.31$ lb/gal.

NOTES: Standard HRC Product

- 1. Assumes HRC product is 40 percent lactic acid and 40 percent glycerol by weight.
- 2. HRC® weighs approximately 9.18 pounds per gallon.

NOTES: Vegetable Oil Emulsion Product

- 1. Assumes emulsion product is 60 percent soybean oil by weight.
- 2. Soybean oil is 7.8 pounds per gallon.
- 3. Assumes specific gravity of emulsion product is 0.96.

Table S.5 Output for Substrate Requirements in Hydrogen Equivalents

Site Name: OU7 Pit 1 TA-02 AEHADG-11 Hotspot

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1. Treatment Zone Physical Dimensions

Width (perpendicular to groundwater flow) Length (parallel to groundwater flow) Saturated Thickness Design Period of Performance

Values	Units
45	feet
50	feet
16	feet
3	years

Values	
14	
15.2	
4.9	
3	

Units meters meters meters years

2. Treatment Zone Hydrogeologic Properties

Total Porosity
Effective Porosity
Average Aquifer Hydraulic Conductivity
Average Hydraulic Gradient
Average Groundwater Seepage Velocity
Average Groundwater Seepage Velocity
Effective Treatment Zone Pore Volume
Groundwater Flux (per year)
Total Groundwater Volume Treated
(over entire design period)

Values
0.36
0.27
5
0.0161
0.30
109
72,725
158,285
547,579

Hydrogen

617

percent percent ft/day ft/ft ft/day ft/yr gallons gallons/year gallons total

Effective

Units

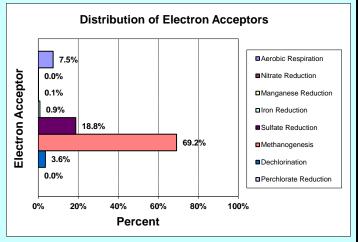
Values	Units
0.36	percent
0.27	percent
1.8E-03	cm/sec
0.0161	m/m
9.1E+00	cm/day
33.2	m/yr
275,287	liters
599,156	liters/yea
2,072,755	liters tota

3. Distribution of Electron Acceptor Demand

Aerobic Respiration
Nitrate Reduction
Sulfate Reduction
Manganese Reduction
Iron Reduction
Methanogenesis
Dechlorination
Perchlorate Reduction

	riyarogen
Percent of Total	Demand (lb)
7.5%	2.475
0.0%	0.011
18.8%	6.234
0.1%	0.037
0.9%	0.285
69.2%	22.961
3.6%	1.184
0.0%	0.000
100.00%	33.19

_	
Hydrogen demand in pounds/gallon:	6.06E-05
Hydrogen demand in grams per liter:	7.26F-03



4. Substrate Equivalents: Design Factor = 10.0

Totals:

Product	Quantity (lb)	Quantity (gallons)
Sodium Lactate Product	15,383	1,398
2. Molasses Product	11,740	978
3. Fructose Product	9,270	828
4. Ethanol Product	4,740	687
5. Sweet Dry Whey (lactose)	7,312	sold by pound
6. HRC®	5,621	sold by pound
7. Linoleic Acid (Sovbean Oil)	2.886	370

Concentration	Effective concentration is for total
(mg/L)	volume of groundwater treated.
1,623	as lactic acid
1,541	as sucrose
1,623	as fructose
830	as ethanol
1,120	as lactose
984	as 40% lactic acid/40% glycerol
632	as soybean oil
632	as soybean oil

Notes:

8. Emulsified Vegetable Oil

- 1. Quantity assumes product is 60% sodium lactate by weight.
- 2. Quantity assumes product is 60% sucrose by weight and weighs 12 pounds per gallon.

4,810

- 3. Quantity assumes product is 80% fructose by weight and weighs 11.2 pounds per gallon.
- 4. Quantity assumes product is 80% ethanol by weight and weighs 6.9 pounds per gallon.
- 5. Quantity assumes product is 70% lactose by weight.
- 6. Quantity assumes HRC $\!\!\!\! ^{\odot}$ is 40% lactic acid and 40% glycerol by weight.
- 7. Quantity of neat soybean oil, corn oil, or canola oil.
- 8. Quantity assumes commercial product is 60% soybean oil by weight.

B.2 TA-03 Treatment Area Design

Site Name: OU7 Pit 1 TA-0	3 CMT-21 Hot 9	Spot and MW-	68	RETURN TO COVER PAGE
Treatment Zana Dharical Dimensions	NOTE: Unshade		•	Hoor Notes
Treatment Zone Physical Dimensions	Values	Range	Units	User Notes
Width (Perpendicular to predominant groundwater flow direction) Length (Parallel to predominant groundwater flow)	45 40	1-10,000 1-1,000	feet	Hotspot CMT-21 area and MW-68, 10 DPT Points
Saturated Thickness	15	1-100	feet	9-24
Freatment Zone Cross Sectional Area	675		ft ²	02.
reatment Zone Volume	27,000		ft ³	
reatment Zone Total Pore Volume (total volume x total porosity)	72,725		gallons	
reatment Zone Effective Pore Volume (total volume x effective porosity)	54,544		gallons	
Design Period of Performance Design Factor (times the electron acceptor hydrogen demand)	3.0 10.0	.5 to 5 2 to 20	year	
Design Factor (times the electron acceptor hydrogen demand)	10.0	2 10 20	unitless	
Treatment Zone Hydrogeologic Properties	_	_		
Fotal Porosity	36%	.05-50	percent	PDI Eastover OU 7
Effective Porosity	27%	.05-50	percent	PDI Eastover OU 7, average based on TP, AP, clay conter
Average Aquifer Hydraulic Conductivity	5	.01-1000	ft/day	OU7 PDI
Average Hydraulic Gradient Average Groundwater Seepage Velocity through the Treatment Zone	1.61E-02 0.30	0.0001-0.1	ft/ft ft/day	2024 Annual Report
Average Groundwater Seepage Velocity through the Treatment Zone	108.8		ft/yr	
Average Groundwater Discharge through the Treatment Zone	148,392		gallons/year	
Soil Bulk Density	1.63	1.4-2.0	gm/cm ³	Average OU- PDI for Eastover and Aquia
Soil Fraction Organic Carbon (foc)	0.28%	0.01-10	percent	PDI-38 Eastover Saturated Zone OU-7 x 2
Native Electron Acceptors				
A. Aqueous-Phase Native Electron Acceptors				
Oxygen	0.2	0.01 to 10	mg/L	CMT-21-2, MW-68 (average) 03/2024
Nitrate	0.02	0.1 to- 20	mg/L	CMT-21-2, MW-68 (average) 03/2024
Sulfate	5	10 to 5,000	mg/L	MW-68 03/2024
Carbon Dioxide (estimated as the amount of Methane produced)	10.0	0.1 to 20	mg/L	Estimate based on previous EISB injections
D. Oall'd Phase Nather Electron Assesstant				
B. Solid-Phase Native Electron Acceptors Manganese (IV) (estimated as the amount of Mn (II) produced)	1	0.1 to 20	mg/L	CMT-21-2, MW-68 (average) 03/2024
Iron (III) (estimated as the amount of Fe (II) produced)	3	0.1 to 20	mg/L	CMT-21-2, MW-68 (average) 03/2024 CMT-21-2, MW-68 (average) 03/2024
. , , , ,	•			,, (,,,,,,
Contaminant Electron Acceptors		_		
Tetrachloroethene (PCE)	0.500		mg/L	CMT-21-2 03/2024
Trichloroethene (TCE)	0.450 14.490		mg/L	CMT-21-2 03/2024
Dichloroethene (cis-DCE, trans-DCE, and 1,1-DCE) Vinyl Chloride (VC)	0.140		mg/L mg/L	CMT-21-2 03/2024 CMT-21-2 03/2024
Carbon Tetrachloride (CT)	0.000		mg/L	Not detected
Trichloromethane (or chloroform) (CF)	0.042		mg/L	CMT-21-2 03/2024
Dichloromethane (or methylene chloride) (MC)	0.000		mg/L	Not detected
Chloromethane	0.000		mg/L	Not detected
Tetrachloroethane (1,1,1,2-PCA and 1,1,2,2-PCA)	0.000		mg/L	Not detected
Trichloroethane (1,1,1-TCA and 1,1,2-TCA)	1.000		mg/L	CMT-24-2-03/2024
Dichloroethane (1,1-DCA and 1,2-DCA) Chloroethane	0.130 0.000		mg/L mg/L	CMT-21-2 03/2024 Not detected
Perchlorate	0.000		mg/L	No data
			<u>. </u>	
Aquifer Geochemistry (Optional Screening Parameters)				
A. Aqueous Geochemistry Ovidetics Reduction Retartial (ORR)	45	400 += : 500	m\/	CMT 24.2 MW 69 (correspond 02/2004
Oxidation-Reduction Potential (ORP) Temperature	-45 13	-400 to +500 5.0 to 30	mV °C	CMT-21-2, MW-68 (average) 03/2024 CMT-21-2, MW-68 (average) 03/2024
oH	6.0	4.0 to 10.0	su	CMT-21-2, MW-68 (average) 03/2024 CMT-21-2, MW-68 (average) 03/2024
Alkalinity	225	10 to 1,000	mg/L	CMT-21-2, MW-68 (average) 03/2024
Total Dissolved Solids (TDS, or salinity)	100	10 to 1,000	mg/L	No data
Specific Conductivity	918	100 to 10,000	μs/cm	CMT-21-2, MW-68 (average) 03/2024
Chloride	46	10 to 10,000	mg/L	CMT-21-2, MW-68 (average) 03/2024
Sulfide - Pre injection	0.1	0.1 to 100	mg/L	Estimated
Sulfide - Post injection	2.5	0.1 to 100	mg/L	Estimated
B. Aquifer Matrix				
Total Iron	18000	200 to 20,000	mg/kg	CSM 2006 Soil Eastover
Cation Exchange Capacity	1	1.0 to 10	meq/100 g	CSM 2006 Soil Eastover
Neutralization Potential	1.0%	1.0 to 100	Percent as CaCO ₃	CSM 2006 Soil Eastover
NOTES:				
Design ROI = 12 ft., Injection Point Spacing = 20 ft., Row Spacing -=	10.0			

Table S.2 S					RETURN TO	COVER PAGE
Site Name: 0	U7 Pit 1 TA-0	3 CMT-21 Hot	Spot and MW-	68	RETURN TO	COVER PAGE
				NOTE: Open cells	•	
. Treatment Zone Physical Dimensions				Values	Range	Units
Width (Perpendicular to predominant groundwater flow	v direction)			45	1-10,000	feet
Length (Parallel to predominant groundwater flow)				40	1-1,000	feet
Saturated Thickness				15	1-100	feet
Treatment Zone Cross Sectional Area				675		ft ²
Treatment Zone Volume				27,000		ft ³
Treatment Zone Effective Pore Volume (total volume)	effective porosit	y)		54,544		gallons
Design Period of Performance				3.0	.5 to 5	year
Treatment Zana Ukuluanaalania Buanantia	_					
. Treatment Zone Hydrogeologic Propertie	S					
Total Porosity				0.36	.05-50	
Effective Porosity				0.27	.05-50	ft/ala
Average Aquifer Hydraulic Conductivity				5	.01-1000	ft/day
Average Craundwater Seepage Valents through the T	rootmont Zono			0.0161	0.1-0.0001	ft/ft ft/dov
Average Groundwater Seepage Velocity through the T				0.30		ft/day
Average Groundwater Seepage Velocity through the T		`		108.8		ft/yr
Average Groundwater Flux through the Treatment Zor	(,		148,392		gallons/year
Soil Bulk Density				1.63	1.4-2.0	gm/cm ³
Soil Fraction Organic Carbon (foc)				0.0028	0.0001-0.1	
Initial Treatment Cell Electron-Acceptor I	Demand (one	total pore volu	ime)			
•				Stoichiometric	Hydrogen	Electron
A. Aqueous-Phase Native Electron Acceptors		Concentration	Mass	demand	Demand	Equivalents
questa i nace native Electron Acceptors			(lb)	(wt/wt h ₂)	(lb)	Mole
Ovugon		(mg/L) 0,2	0.10	7.94	0.01	4
Oxygen						
Nitrate (denitrification) Sulfate		0.0 4.7	0.01 2.14	12.30 11.91	0.00	5
Sunate Carbon Dioxide (estimated as the amount of methane	produced)	10.0	4.55	1.99	0.18 2.29	8
Calbon Dioxide (estimated as the amount of methane	produced)			eptor Demand (lb.)	2.48	0
		Soluble Compet	ing Electron Acc			
				Stoichiometric	Hydrogen	Electron
B. Solid-Phase Native Electron Acceptors		Concentration	Mass	demand	Demand	Equivalents
(Based on manganese and iron produced)	(mg/L)	(lb)	(wt/wt h ₂)	(lb)	Mole	
Manganese (IV) (estimated as the amount of Mn (II) p	,	1.3	5.42	27.25	0.20	2
Iron (III) (estimated as the amount of Fe (II) produced)		3.4	14.30	55.41	0.26	1
	Sol	id-Phase Compet	ing Electron Acc	eptor Demand (lb.)	0.46	
				Stoichiometric	Hydrogen	Electron
C. Soluble Contaminant Electron Acceptors		Concentration	Mass	demand	Demand	Equivalents
		(mg/L)	(lb)	(wt/wt h ₂)	(lb)	Mole
Tetrachloroethene (PCE)		0.500	0.23	20.57	0.01	8
Trichloroethene (TCE)		0.450	0.20	21.73	0.01	6
Dichloroethene (cis-DCE, trans-DCE, and 1,1-DCE)		14.490	6.60	24.05	0.27	4
Vinyl Chloride (VC)		0.140	0.06	31.00	0.00	2
Carbon Tetrachloride (CT)		0.000	0.00	19.08	0.00	8
Trichloromethane (or chloroform) (CF)		0.042	0.02	19.74	0.00	6
Dichloromethane (or methylene chloride) (MC)		0.000	0.00	21.06	0.00	4
Chloromethane		0.000	0.00	25.04	0.00	2
Tetrachloroethane (1,1,1,2-PCA and 1,1,2,2-PCA)		0.000	0.00	20.82	0.00	8
Trichloroethane (1,1,1-TCA and 1,1,2-TCA)		1.000	0.46	22.06	0.02	6
Dichloroethane (1,1-DCA and 1,2-DCA)		0.130	0.06	24.55	0.00	4
Chloroethane		0.000	0.00	32.00	0.00	2
Perchlorate		0.000	0.00	12.33	0.00	6
	Total S			eptor Demand (lb.)	0.32	
				Stoichiometric	Hydrogen	Electror
D. Sorbed Contaminant Electron Acceptors	Koc	Soil Conc.	Mass	demand	Demand	Equivalents
(Soil Concentration = Koc x foc x Cgw)				(wt/wt h ₂)		Mole
·	(mL/g)	(mg/kg)	(lb)		(lb)	
Tetrachloroethene (PCE)	263	0.37	1.01	20.57	0.05	8
Trichloroethene (TCE)	107	0.13	0.37	21.73	0.02	6
Dichloroethene (cis-DCE, trans-DCE, and 1,1-DCE)	45	1.83	5.02	24.05	0.21	4
Vinyl Chloride (VC)	3.0	0.00	0.00	31.00	0.00	2
Carbon Tetrachloride (CT)	224	0.00	0.00	19.08	0.00	8
Trichloromethane (or chloroform) (CF)	63	0.01	0.02	19.74	0.00	6
Dichloromethane (or methylene chloride) (MC)	28	0.00	0.00	21.06	0.00	4
Chloromethane	25	0.00	0.00	25.04	0.00	2
Tetrachloroethane (1,1,1,2-PCA and 1,1,2,2-PCA)	117	0.00	0.00	20.82	0.00	8
Trichloroethane (1,1,1-TCA and 1,1,2-TCA)	105	0.29	0.81	22.06	0.04	6
Dichloroethane (1,1-DCA and 1,2-DCA)	30	0.01	0.03	24.55	0.00	4
			0.00	00.00	0.00	2
Chloroethane	3	0.00	0.00	32.00		
Chloroethane Perchlorate	0.0	0.00	0.00	12.33 eptor Demand (lb.)	0.00 0.00 0.31	6

T-bl- 0.0 0.1	otroto Colembiliano I	. I la columna are are				
	strate Calculations in	Hydrogen I	quivalents			
4. Treatment Cell Electron-Acceptor Flux (per	year)		01-1-1-1	I I de la companya de		
A Oakala Nadar Electron Assess	0		Stoichiometric	Hydrogen	Electron	
A. Soluble Native Electron Acceptors	Concentration	Mass	demand	Demand	Equivalents per	
	(mg/L)	(lb)	(wt/wt h ₂)	(lb)	Mole	
Oxygen	0.2	0.28	7.94	0.04	4	
Nitrate (denitrification)	0.0	0.03	10.25	0.00	5	
Sulfate	4.7	5.82	11.91	0.49	8	
Carbon Dioxide (estimated as the amount of Methane pro-		12.38	1.99	6.22	8	
	Total Competing Ele	ctron Acceptor D	emand Flux (lb/yr)	6.7		
			Stoichiometric	Hydrogen	Electron	
B. Soluble Contaminant Electron Acceptors	Concentration	Mass	demand	Demand	Equivalents per	
	(mg/L)	(lb)	(wt/wt h ₂)	(lb)	Mole	
Tetrachloroethene (PCE)	0.500	0.62	20.57	0.03	8	
Trichloroethene (TCE)	0.450	0.56	21.73	0.03	6	
Dichloroethene (cis-DCE, trans-DCE, and 1,1-DCE)	14.490	17.94	24.05	0.75	4	
Vinyl Chloride (VC)	0.140	0.17	31.00	0.01	2	
Carbon Tetrachloride (CT)	0.000	0.00	19.08	0.00	8	
Trichloromethane (or chloroform) (CF)	0.042	0.05	19.74	0.00	6	
Dichloromethane (or methylene chloride) (MC)	0.000	0.00	21.06	0.00	4	
Chloromethane	0.000	0.00	25.04	0.00	2	
Tetrachloroethane (1,1,1,2-PCA and 1,1,2,2-PCA)	0.000	0.00	20.82	0.00	8	
Trichloroethane (1,1,1-TCA and 1,1,2-TCA)	1.000	1.24	22.06	0.06	6	
Dichloroethane (1,1-DCA and 1,2-DCA)	0.130	0.16	24.55	0.01	4	
Chloroethane	0.000	0.00	32.00	0.00	2	
Perchlorate	0.000	0.00	12.33	0.00	6	
'	otal Soluble Contaminant Ele	ctron Acceptor D	emand Flux (lb/yr)	0.87	<u></u>	
	Initial Hydroge	n Requiremen	t First Year (lb)	11.2		
	Total Life-Cycl	e Hydrogen Re	equirement (lb)	26.4		
5. Design Factors	•				_	
Microbial Efficiency Uncertainty Factor				2X - 4X		
Methane and Solid-Phase Electron Acceptor Uncertainty				2X - 4X		
Remedial Design Factor (e.g., Substrate Leaving Reaction 2	Zone)			1X - 3X		
	•		Design Factor	10.0	1	
Total Life	-Cycle Hydrogen Require	ment with De	U			
	-Cycle Hydrogen Require	ment with De	sign i actor (ib)	204.4	_	
6. Acronyns and Abbreviations						
°C =degrees celsius me	g/100 g = milliequivalents per 10	O grama				
	d/100 g = millequivalents per 10 /kg = milligrams per kilogram	o grams				
μαςτι = imicrosterieris per certimeter ing κg = imingranis per intograni cm/day = centimeters per day mg/L = milligrams per litter						
	cm/sec = centimeters per second m/m = meters per meters					
	ft^2 = square feet $mV = millivolts$					
· · · · · · · · · · · · · · · · · · ·	r = meters per year					
ft/ft = foot per foot su = standard pH units						
	wt H2 = concetration molecular h	nydrogen, weight p	er weight			
gm/cm ³ = grams per cubic centimeter		, , , , , , , , , ,	J			
kg of CaCO3 per mg = kilograms of calcium carbonate pe	r milligram					
lb = pounds	<u> </u>					

Table S.3

Hydrogen Produced by Fermentation Reactions of Common Substrates

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Substrate	Molecular Formula	Substrate Molecular Weight (gm/mole)	Moles of Hydrogen Produced per Mole of Substrate	Ratio of Hydrogen Produced to Substrate (gm/gm)	Range of Moles H ₂ /Mole Substrate
Lactic Acid	C ₃ H ₆ O ₃	90.1	2	0.0448	2 to 3
Molasses (assuming 100% sucrose)	C ₁₂ H ₂₂ O ₁₁	342	8	0.0471	8 to 11
High Fructose Corn Syrup (assuming 50% fructose and 50% glucose)	C ₆ H ₁₂ O ₆	180	4	0.0448	4 to 6
Ethanol	C ₂ H ₆ O	46.1	2	0.0875	2 to 6
Whey (assuming 100% lactose)	C ₁₂ H ₂₂ O ₁₁	342	11	0.0648	11
HRC® (assumes 40% lactic acid and 40% glycerol by weight)	C ₃₉ H ₅₆ O ₃₉	956	28	0.0590	28
Linoleic Acid (Soybean Oil, Corn Oil, Cotton Oil)	C ₁₈ H ₃₂ O ₂	281	16	0.1150	16

Table S.4 Estimated Substrate Requirements for Hydrogen Demand in Table S.3

Design Life (years): 3

Substrate	Design Factor	Pure Substrate Mass Required to Fulfill Hydrogen Demand (pounds)	Substrate Product Required to Fulfill Hydrogen Demand (pounds)	Substrate Mass Required to Fulfill Hydrogen Demand (milligrams)	Effective Substrate Concentration (mg/L)
Lactic Acid	10.0	5,907	5,907	2.68E+09	1,416
Sodium Lactate Product (60 percent solution)	10.0	5,907	12,255	2.68E+09	1,416
Molasses (assuming 6 0	10.0	5,612	9,353	2.55E+09	1,346
HFCS (assuming 40% fructose and 40% glucose by weight)	10.0	5,908	7,385	2.68E+09	1,417
Ethanol Product (assuming 80% ethanol by weight)	10.0	3,021	3,776	1.37E+09	724
Whey (assuming 100% lactose)	10.0	4,078	5,825	1.85E+09	978
HRC® (assumes 40% lactic acid and 40% glycerol by weight)	10.0	4,478	4,478	2.03E+09	859
Linoleic Acid (Soybean Oil, Corn Oil, Cotton Oil)	10.0	2,299	2,299	1.04E+09	551
Commercial Vegetable Oil Emulsion Product (60% oil by weight)	10.0	2,299	3,832	1.04E+09	551

NOTES: Sodium Lactate Product

- 1. Assumes sodium lactate product is 60 percent sodium lactate by weight.
- 2. Molecular weight of sodium lactate (CH_3 -CHOH-COONa) = 112.06.
- 3. Molecular weight of lactic Acid $(C_6H_6O_3) = 90.08$.
- 4. Therefore, sodium lactate product yields 48.4 (0.60 x (90.08/112.06)) percent by weight lactic acid.
- 5. Weight of sodium lactate product = 11.0 pounds per gallon.
- 6. Pounds per gallon of lactic acid in product = 1.323×8.33 lb/gal H2O x $0.60 \times (90.08/112.06) = 5.31$ lb/gal.

NOTES: Standard HRC Product

- 1. Assumes HRC product is 40 percent lactic acid and 40 percent glycerol by weight.
- 2. HRC® weighs approximately 9.18 pounds per gallon.

NOTES: Vegetable Oil Emulsion Product

- 1. Assumes emulsion product is 60 percent soybean oil by weight.
- 2. Soybean oil is 7.8 pounds per gallon.
- 3. Assumes specific gravity of emulsion product is 0.96.

Table S.5 Output for Substrate Requirements in Hydrogen Equivalents

Site Name: OU7 Pit 1 TA-03 CMT-21 Hot Spot and MW-68 RETURN TO COVER PAGE

1. Treatment Zone Physical Dimensions

Width (perpendicular to groundwater flow) Length (parallel to groundwater flow) Saturated Thickness Design Period of Performance

Values	Units
45	feet
40	feet
15	feet
3	years

its	Values
t	14
t	12.2
t	4.6
ars	3

Units meters meters meters years

2. Treatment Zone Hydrogeologic Properties

Total Porosity Effective Porosity Average Aquifer Hydraulic Conductivity Average Hydraulic Gradient Average Groundwater Seepage Velocity Average Groundwater Seepage Velocity Effective Treatment Zone Pore Volume Groundwater Flux (per year) Total Groundwater Volume Treated (over entire design period)

Values
0.36
0.27
5
0.0161
0.30
109
54,544
148,392
499,720

Hydrogen

10.0

Units percent percent ft/day ft/ft ft/day ft/yr gallons gallons/year gallons total

Values	Units
0.36	percent
0.27	percent
1.8E-03	cm/sec
0.0161	m/m
9.1E+00	cm/day
33.2	m/yr
206,465	liters
561,709	liters/year
1,891,591	liters total

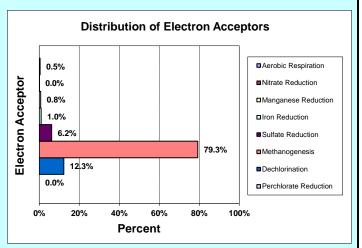
3. Distribution of Electron Acceptor Demand

Aerobic Respiration Nitrate Reduction Sulfate Reduction Manganese Reduction Iron Reduction Methanogenesis Dechlorination Perchlorate Reduction

	riyurogen
Percent of Total	Demand (lb)
0.5%	0.121
0.0%	0.009
6.2%	1.646
0.8%	0.199
1.0%	0.258
79.3%	20.954
12.3%	3.253
0.0%	0.000

Totals: 100.00% 26.44

Hydrogen demand in pounds/gallon:	5.29E-05
Hydrogen demand in grams per liter:	6.34E-03



4. Substrate Equivalents: Design Factor =

	Quantity	Qu
Product	(lb)	(ga

1.	Sodium Lactate Product
2	Molasses Product

- 3. Fructose Product
- 4. Ethanol Product
- 5. Sweet Dry Whey (lactose)
- 6. HRC®

7. I	inoleic Acid (Soybean O
8. I	mulsified Vegetable Oil

Quantity (lb)	Quantity (gallons)
12,255	1,114
9,353	779
7,385	659
3,776	547
5,825	sold by pound
4,478	sold by pound
2,299	295
3,832	491

Effective	
Concentration	Effective concentration is for total
(mg/L)	volume of groundwater treated.
1,416	as lactic acid
1,346	as sucrose
1,417	as fructose
724	as ethanol
978	as lactose
859	as 40% lactic acid/40% glycerol
551	as soybean oil
551	as soybean oil

Notes:

- 1. Quantity assumes product is 60% sodium lactate by weight.
- 2. Quantity assumes product is 60% sucrose by weight and weighs 12 pounds per gallon.
- 3. Quantity assumes product is 80% fructose by weight and weighs 11.2 pounds per gallon.
- 4. Quantity assumes product is 80% ethanol by weight and weighs 6.9 pounds per gallon.
- 5. Quantity assumes product is 70% lactose by weight.
- 6. Quantity assumes HRC® is 40% lactic acid and 40% glycerol by weight.
- 7. Quantity of neat soybean oil, corn oil, or canola oil.
- 8. Quantity assumes commercial product is 60% soybean oil by weight.

B.3 TA-05 Treatment Area Design

Site Name:	OU7 P	it 1 TA-05 DM	W-29A		RETURN TO COVER PAGE
		NOTE: Unshaded boxes are user input.			Harri Natara
Treatment Zone Physical Dimensions		Values	Range	Units	User Notes
Nidth (Perpendicular to predominant groundwater flow direction))	40	1-10,000	feet	Pit 1 Area, Target DMW-29A Area 6 DPT Points
Length (Parallel to predominant groundwater flow) Saturated Thickness		30 18	1-1,000 1-100	feet feet	10-28
Freatment Zone Cross Sectional Area		720	1-100	ft ²	10-20
reatment Zone Volume		21,600		ft ³	
reatment Zone Total Pore Volume (total volume x total porosity)	58,180		gallons	
Treatment Zone Effective Pore Volume (total volume x effective		43,635		gallons	
Design Period of Performance	, ,,	3.0	.5 to 5	year	
Design Factor (times the electron acceptor hydrogen demand)		10.0	2 to 20	unitless	
Treatment Zone Hydrogeologic Properties		000/	05.50		BBI Factoria OLI Z
Total Porosity Effective Porosity		36% 27%	.05-50 .05-50	percent percent	PDI Eastover OU 7 PDI Eastover OU 7, average based on TP, AP, clay conte
Average Aquifer Hydraulic Conductivity		5	.01-1000	ft/day	OU7 PDI
Average Hydraulic Gradient		1.61E-02	0.0001-0.1	ft/ft	2024 Annual Report
Average Groundwater Seepage Velocity through the Treatment	Zone	0.30		ft/day	202. / timudi report
Average Groundwater Seepage Velocity through the Treatment		108.8		ft/yr	
Average Groundwater Discharge through the Treatment Zone		158,285		gallons/year	
Soil Bulk Density		1.63	1.4-2.0	gm/cm ³	Average OU- PDI for Eastover and Aquia
Soil Fraction Organic Carbon (foc)		0.28%	0.01-10	percent	PDI-38 Eastover Saturated Zone OU-7 x 2
Native Electron Accounters					
Native Electron Acceptors A. Aqueous-Phase Native Electron Acceptors					
Oxygen		1.1	0.01 to 10	mg/L	DMW-29A 03/2024
Nitrate		0.05	0.1 to- 20	mg/L	DMW-29A 03/2024
Sulfate		1	10 to 5,000	mg/L	DMW-29A 03/2024
Carbon Dioxide (estimated as the amount of Methane produced))	10.0	0.1 to 20	mg/L	Estimate based on previous EISB injections
					·
B. Solid-Phase Native Electron Acceptors			1		
Manganese (IV) (estimated as the amount of Mn (II) produced)		0	0.1 to 20	mg/L	DMW-29A 03/2024
Iron (III) (estimated as the amount of Fe (II) produced)		9	0.1 to 20	mg/L	DMW-29A 03/2024
Contaminant Electron Acceptors					
Tetrachloroethene (PCE)		0.006		mg/L	DMW-29A 03/2024
Trichloroethene (TCE)		0.100		mg/L	DMW-29A 03/2024
Dichloroethene (cis-DCE, trans-DCE, and 1,1-DCE)		0.126		mg/L	DMW-29A 03/2024
Vinyl Chloride (VC)		0.007		mg/L	DMW-29A 03/2024
Carbon Tetrachloride (CT)		0.000		mg/L	Not detected
Trichloromethane (or chloroform) (CF)		0.000		mg/L	Not detected
Dichloromethane (or methylene chloride) (MC)		0.000		mg/L	Not detected
Chloromethane		0.000		mg/L	Not detected
Tetrachloroethane (1,1,1,2-PCA and 1,1,2,2-PCA)		0.000		mg/L	Not detected
Trichloroethane (1,1,1-TCA and 1,1,2-TCA)		0.000		mg/L	Not detected
Dichloroethane (1,1-DCA and 1,2-DCA)		0.000		mg/L	Not detected
Chloroethane		0.010		mg/L	DMW-29A 03/2024
Perchlorate		0.000		mg/L	No data
Aguifer Geochemistry (Optional Screening Paran	neters)				
A. Aqueous Geochemistry	,				
Oxidation-Reduction Potential (ORP)		310	-400 to +500	mV	DMW-29A 03/2024
Temperature		14	5.0 to 30	°C	DMW-29A 03/2024
рН		5.5	4.0 to 10.0	su	DMW-29A 03/2024
Alkalinity		78	10 to 1,000	mg/L	DMW-29A 03/2024
Total Dissolved Solids (TDS, or salinity)		100	10 to 1,000	mg/L	No data
Specific Conductivity		750	100 to 10,000	•	DMW-29A 03/2024
Chloride		19	10 to 10,000	mg/L	DMW-29A 03/2024
Sulfide - Pre injection		0.1	0.1 to 100	mg/L	Estimate Not Detected
Sulfide - Post injection		1.2	0.1 to 100	mg/L	DMW-29A 03/2024
B. Aquifer Matrix					
Total Iron		18000	200 to 20,000	mg/kg	CSM 2006 Soil Eastover
Cation Exchange Capacity		1	1.0 to 10	meq/100 g	CSM 2006 Soil Eastover
		1.0%	1.0 to 100	Percent as CaCO ₃	CSM 2006 Soil Eastover
Neutralization Potential					
NOTES:					

1 able 5.2 S	ubstrate Ca	alculations in	Hydrogen	Equivalents		
Site Name:	OU7 F	Pit 1 TA-05 DM	W-29A		RETURN TO	COVER PAGE
				NOTE: Open cells	are user input.	
. Treatment Zone Physical Dimensions				Values	Range	Units
Width (Perpendicular to predominant groundwater flow direction) 40						feet
Length (Parallel to predominant groundwater flow)				30	1-1,000	feet
Saturated Thickness				18	1-100	feet
Treatment Zone Cross Sectional Area				720		ft ²
Treatment Zone Volume				21,600		ft ³
Treatment Zone Effective Pore Volume (total volume)	x effective porosit	v)		43,635		gallons
Design Period of Performance		,,		3.0	.5 to 5	year
						ĺ
. Treatment Zone Hydrogeologic Propertie	s					
Total Porosity				0.36	.05-50	
Effective Porosity				0.27	.05-50	£4/-l
Average Aquifer Hydraulic Conductivity				5	.01-1000	ft/day
Average Hydraulic Gradient	Frankmant Zana			0.0161	0.1-0.0001	ft/ft
Average Groundwater Seepage Velocity through the T				0.30		ft/day
Average Groundwater Seepage Velocity through the T		`		108.8		ft/yr
Average Groundwater Flux through the Treatment Zor	. (,		158,285		gallons/year
Soil Bulk Density				1.63	1.4-2.0	gm/cm ³
Soil Fraction Organic Carbon (foc)				0.0028	0.0001-0.1	
. Initial Treatment Cell Electron-Acceptor D	Demand (one	total pore volu	me)			
				Stoichiometric	Hydrogen	Electron
A. Aqueous-Phase Native Electron Acceptors		Concentration	Mass	demand	Demand	Equivalents p
		(mg/L)	(lb)	(wt/wt h ₂)	(lb)	Mole
Oxygen		1.1	0.40	7.94	0.05	4
Nitrate (denitrification)		0.1	0.02	12.30	0.00	5
Sulfate		0.5	0.18	11.91	0.02	8
Carbon Dioxide (estimated as the amount of methane	produced)	10.0	3.64	1.99	1.83	8
	,,	Soluble Compet	ing Electron Acc	eptor Demand (lb.)	1.90	
				Stoichiometric	Hydrogen	Electron
B. Solid-Phase Native Electron Acceptors		Concentration	Mass	demand	Demand	Equivalents
(Based on manganese and iron produced)	(mg/L)	(lb)	(wt/wt h ₂)	(lb)	Mole	
Manganese (IV) (estimated as the amount of Mn (II) p.	0.5	1.99	27.25	0.07	2	
Iron (III) (estimated as the amount of Fe (II) produced)	9.0	38.94	55.41	0.70	1	
non (iii) (estimated as the amount of re (ii) produced)				eptor Demand (lb.)	0.78	'
			• • • • • • • • • • • • • • • • • • • •	Stoichiometric		
C. Soluble Contaminant Electron Acceptors		Concentration	Mass	demand	Hydrogen Demand	Electron
C. Soluble Containinant Electron Acceptors						Equivalents
		(mg/L)	(lb)	(wt/wt h ₂)	(lb)	Mole
Tetrachloroethene (PCE)		0.006	0.00	20.57	0.00	8
Trichloroethene (TCE)		0.100	0.04	21.73	0.00	6
Dichloroethene (cis-DCE, trans-DCE, and 1,1-DCE)		0.126	0.05	24.05	0.00	4
Vinyl Chloride (VC)		0.007	0.00	31.00	0.00	2
Carbon Tetrachloride (CT)		0.000	0.00	19.08	0.00	8
Trichloromethane (or chloroform) (CF)		0.000	0.00	19.74	0.00	6
Dichloromethane (or methylene chloride) (MC)		0.000	0.00	21.06	0.00	4
Chloromethane		0.000	0.00	25.04	0.00	2
Tetrachloroethane (1,1,1,2-PCA and 1,1,2,2-PCA)		0.000	0.00	20.82	0.00	8
Trichloroethane (1,1,1-TCA and 1,1,2-TCA)		0.000	0.00	22.06	0.00	6
Dichloroethane (1,1-DCA and 1,2-DCA)		0.000	0.00	24.55	0.00	4
Chloroethane		0.010	0.00	32.00	0.00	2
Perchlorate	Takel	0.000	0.00	12.33	0.00	6
	lotal S	Soluble Contamin	AIIL EIECTON ACC	eptor Demand (lb.)	0.00	
D. Oarthad Contambrant Ethics		0.11.0		Stoichiometric	Hydrogen	Electron
D. Sorbed Contaminant Electron Acceptors	Koc	Soil Conc.	Mass	demand	Demand	Equivalents
(Soil Concentration = Koc x foc x Cgw)	(mL/g)	(mg/kg)	(lb)	(wt/wt h ₂)	(lb)	Mole
Tetrachloroethene (PCE)	263	0.00	0.01	20.57	0.00	8
Trichloroethene (TCE)	107	0.03	0.07	21.73	0.00	6
Dichloroethene (cis-DCE, trans-DCE, and 1,1-DCE)	45	0.02	0.03	24.05	0.00	4
Vinyl Chloride (VC)	3.0	0.00	0.00	31.00	0.00	2
Carbon Tetrachloride (CT)	224	0.00	0.00	19.08	0.00	8
Trichloromethane (or chloroform) (CF)	63	0.00	0.00	19.74	0.00	6
Dichloromethane (or methylene chloride) (MC)	28	0.00	0.00	21.06	0.00	4
Chloromethane	25	0.00	0.00	25.04	0.00	2
Tetrachloroethane (1,1,1,2-PCA and 1,1,2,2-PCA)	117	0.00	0.00	20.82	0.00	8
Trichloroethane (1,1,1-TCA and 1,1,2-TCA)	105	0.00	0.00	22.06	0.00	6
	30	0.00	0.00	24.55	0.00	4
Dichloroethane (1.1-DCA and 1.2-DCA)		00				
Dichloroethane (1,1-DCA and 1,2-DCA) Chloroethane	3	0.00	0.00	32.00	0.00	2
	3 0.0	0.00	0.00	32.00 12.33	0.00	6

	substrate Calculatio	ns in Hydrog	en Equivalents						
4. Treatment Cell Electron-Acceptor Flux (p	er year)								
			Stoichiometric	Hydrogen	Electron				
A. Soluble Native Electron Acceptors	Concentr	ration Mass	demand	Demand	Equivalents per				
	(mg/l	_) (lb)	(wt/wt h ₂)	(lb)	Mole				
Oxygen	1.1	1.45	7.94	0.18	4				
Nitrate (denitrification)	0.1	0.07	10.25	0.01	5				
Sulfate	0.5	0.66	11.91	0.06	8				
Carbon Dioxide (estimated as the amount of Methane		13.21	1.99	6.64	8				
Total Competing Electron Acceptor Demand Flux (lb/yr) 6.9									
			Stoichiometric	Hydrogen	Electron				
B. Soluble Contaminant Electron Acceptors	Concentr	ation Mass	demand	Demand	Equivalents per				
	(mg/l	_) (lb)	(wt/wt h ₂)	(lb)	Mole				
Tetrachloroethene (PCE)	0.000	6 0.01	20.57	0.00	8				
Trichloroethene (TCE)	0.10	0.13	21.73	0.01	6				
Dichloroethene (cis-DCE, trans-DCE, and 1,1-DCE)	0.12	6 0.17	24.05	0.01	4				
Vinyl Chloride (VC)	0.00		31.00	0.00	2				
Carbon Tetrachloride (CT)	0.00		19.08	0.00	8				
Trichloromethane (or chloroform) (CF)	0.000		19.74	0.00	6				
Dichloromethane (or methylene chloride) (MC)	0.000		21.06	0.00	4				
Chloromethane	0.000		25.04	0.00	2				
Tetrachloroethane (1,1,1,2-PCA and 1,1,2,2-PCA)	0.00		20.82	0.00	8				
Trichloroethane (1,1,1-TCA and 1,1,2-TCA)	0.00		22.06	0.00	6				
Dichloroethane (1,1-DCA and 1,2-DCA)	0.00		24.55	0.00	4				
Chloroethane	0.010		32.00	0.00	2				
Perchlorate	0.00		12.33	0.00	6				
Total Soluble Contaminant Electron Acceptor Demand Flux (lb/yr) 0.01									
			ment First Year (Ib n Requirement (Ib						
5. Design Factors	TOTAL LITE	-cycle nyuroge	ii Kequireillelli (ib	23.4					
				0V 4V					
Microbial Efficiency Uncertainty Factor Methane and Solid-Phase Electron Acceptor Uncertainty				2X - 4X 2X - 4X					
Remedial Design Factor (e.g., Substrate Leaving React				1X - 3X					
Remedial Design Factor (e.g., Substrate Leaving React	on zone)		Decima Foote		7				
			Design Facto						
	₋ife-Cycle Hydrogen Re	equirement with	Design Factor (lb) 233.7					
6. Acronyns and Abbreviations									
°C =degrees celsius	meg/100 g = milliequivalents	ner 100 grams							
μs/cm = microsiemens per centimeter	mg/kg = milligrams per kilog								
cm/day = centimeters per day									
cm/sec = centimeters per second m/m = meters per meters									
ft ² = square feet mV = millivolts									
ft/day = feet per day m/yr = meters per year									
ft/ft = foot per foot su = standard pH units									
ft/yr = feet per year	wt/wt H2 = concetration mole	ecular hydrogen, we	ight per weight						
gm/cm ³ = grams per cubic centimeter									
kg of CaCO3 per mg = kilograms of calcium carbonate	e per milligram								
lb = pounds									

Table S.3

Hydrogen Produced by Fermentation Reactions of Common Substrates

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Substrate	Molecular Formula	Substrate Molecular Weight (gm/mole)	Moles of Hydrogen Produced per Mole of Substrate	Ratio of Hydrogen Produced to Substrate (gm/gm)	Range of Moles H ₂ /Mole Substrate
Lactic Acid	C ₃ H ₆ O ₃	90.1	2	0.0448	2 to 3
Molasses (assuming 100% sucrose)	C ₁₂ H ₂₂ O ₁₁	342	8	0.0471	8 to 11
High Fructose Corn Syrup (assuming 50% fructose and 50% glucose)	C ₆ H ₁₂ O ₆	180	4	0.0448	4 to 6
Ethanol	C ₂ H ₆ O	46.1	2	0.0875	2 to 6
Whey (assuming 100% lactose)	C ₁₂ H ₂₂ O ₁₁	342	11	0.0648	11
HRC [®] (assumes 40% lactic acid and 40% glycerol by weight)	C ₃₉ H ₅₆ O ₃₉	956	28	0.0590	28
Linoleic Acid (Soybean Oil, Corn Oil, Cotton Oil)	C ₁₈ H ₃₂ O ₂	281	16	0.1150	16

Table S.4 Estimated Substrate Requirements for Hydrogen Demand in Table S.3

Design Life (years): 3

Substrate	Design Factor	Pure Substrate Mass Required to Fulfill Hydrogen Demand (pounds)	Substrate Product Required to Fulfill Hydrogen Demand (pounds)	Substrate Mass Required to Fulfill Hydrogen Demand (milligrams)	Effective Substrate Concentration (mg/L)
Lactic Acid	10.0	5,221	5,221	2.37E+09	1,207
Sodium Lactate Product (60 percent solution)	10.0	5,221	10,832	2.37E+09	1,207
Molasses (assuming 6 0	10.0	4,960	8,267	2.25E+09	1,146
HFCS (assuming 40% fructose and 40% glucose by weight)	10.0	5,222	6,528	2.37E+09	1,207
Ethanol Product (assuming 80% ethanol by weight)	10.0	2,670	3,338	1.21E+09	617
Whey (assuming 100% lactose)	10.0	3,604	5,149	1.63E+09	833
HRC® (assumes 40% lactic acid and 40% glycerol by weight)	10.0	3,958	3,958	1.80E+09	732
Linoleic Acid (Soybean Oil, Corn Oil, Cotton Oil)	10.0	2,032	2,032	9.22E+08	470
Commercial Vegetable Oil Emulsion Product (60% oil by weight)	10.0	2,032	3,387	9.22E+08	470

NOTES: Sodium Lactate Product

- 1. Assumes sodium lactate product is 60 percent sodium lactate by weight.
- 2. Molecular weight of sodium lactate (CH_3 -CHOH-COONa) = 112.06.
- 3. Molecular weight of lactic Acid ($C_6H_6O_3$) = 90.08 .
- 4. Therefore, sodium lactate product yields 48.4 (0.60 x (90.08/112.06)) percent by weight lactic acid.
- 5. Weight of sodium lactate product = 11.0 pounds per gallon.
- 6. Pounds per gallon of lactic acid in product = 1.323×8.33 lb/gal H2O x $0.60 \times (90.08/112.06) = 5.31$ lb/gal.

NOTES: Standard HRC Product

- 1. Assumes HRC product is 40 percent lactic acid and 40 percent glycerol by weight.
- 2. HRC® weighs approximately 9.18 pounds per gallon.

NOTES: Vegetable Oil Emulsion Product

- 1. Assumes emulsion product is 60 percent soybean oil by weight.
- 2. Soybean oil is 7.8 pounds per gallon.
- 3. Assumes specific gravity of emulsion product is 0.96.

Table S.5 Output for Substrate Requirements in Hydrogen Equivalents

Site Name: OU7 Pit 1 TA-05 DMW-29A

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1. Treatment Zone Physical Dimensions

Width (perpendicular to groundwater flow) Length (parallel to groundwater flow) Saturated Thickness Design Period of Performance

Values	Units
40	feet
30	feet
18	feet
3	years

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3

2. Treatment Zone Hydrogeologic Properties

Total Porosity
Effective Porosity
Average Aquifer Hydraulic Conductivity
Average Hydraulic Gradient
Average Groundwater Seepage Velocity
Average Groundwater Seepage Velocity
Effective Treatment Zone Pore Volume
Groundwater Flux (per year)
Total Groundwater Volume Treated
(over entire design period)

Values	
0.36	1
0.27	I
5	I
0.0161	I
0.30	I
109	I
43,635	I
158,285	I
518,489	Ī

Hydrogen

261

Units
percent
percent
ft/day
ft/ft
ft/day
ft/yr
gallons
gallons/yea
gallons tota

Effective

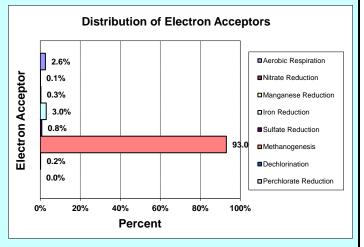
Values	Units
0.36	percent
0.27	percent
1.8E-03	cm/sec
0.0161	m/m
9.1E+00	cm/day
33.2	m/yr
165,172	liters
599,156	liters/yea
1,962,640	liters tota
	•

3. Distribution of Electron Acceptor Demand

Aerobic Respiration
Nitrate Reduction
Sulfate Reduction
Manganese Reduction
Iron Reduction
Methanogenesis
Dechlorination
Perchlorate Reduction

	i iyai ogcii
Percent of Total	Demand (lb)
2.6%	0.599
0.1%	0.021
0.8%	0.182
0.3%	0.073
3.0%	0.703
93.0%	21.741
0.2%	0.051
0.0%	0.000
100.00%	23.37

<u>-</u>	
Hydrogen demand in pounds/gallon:	4.51E-05
Hydrogen demand in grams per liter:	5.40F-03



4. Substrate Equivalents: Design Factor = 10.0

Totals:

Product	Quantity (lb)	Quantity (gallons)
Sodium Lactate Product	10,832	985
2. Molasses Product	8,267	689
3. Fructose Product	6,528	583
4. Ethanol Product	3,338	484
5. Sweet Dry Whey (lactose)	5,149	sold by pound
6. HRC®	3,958	sold by pound

Concentration	Effective concentration is for total
(mg/L)	volume of groundwater treated.
1,207	as lactic acid
1,146	as sucrose
1,207	as fructose
617	as ethanol
833	as lactose
732	as 40% lactic acid/40% glycerol
470	as soybean oil
470	as soybean oil

Notes

7. Linoleic Acid (Soybean Oil)

8. Emulsified Vegetable Oil

- 1. Quantity assumes product is 60% sodium lactate by weight.
- 2. Quantity assumes product is 60% sucrose by weight and weighs 12 pounds per gallon.

2,032

- 3. Quantity assumes product is 80% fructose by weight and weighs 11.2 pounds per gallon.
- 4. Quantity assumes product is 80% ethanol by weight and weighs 6.9 pounds per gallon.
- 5. Quantity assumes product is 70% lactose by weight.
- 6. Quantity assumes HRC $\!\!\!\! ^{\odot}$ is 40% lactic acid and 40% glycerol by weight.
- 7. Quantity of neat soybean oil, corn oil, or canola oil.
- 8. Quantity assumes commercial product is 60% soybean oil by weight.

B.4 TA-06 Treatment Area Design

Site Name: 0	U7 Pit 1	TA-06 CMT-20) (INJ-211)		RETURN TO COVER PAGE
Total Control Total Blood of Biographic	ı	NOTE: Unshade		•	Harri Natara
Treatment Zone Physical Dimensions	Г	Values	Range	Units	User Notes
Nidth (Perpendicular to predominant groundwater flow direction) ength (Parallel to predominant groundwater flow)		25 50	1-10,000 1-1,000	feet	Pit 1 Area, INJ-211
Saturated Thickness		20	1-100	feet	10-30 ft
Freatment Zone Cross Sectional Area		500		ft ²	10 00 N
Freatment Zone Volume		25,000		ft ³	
Treatment Zone Total Pore Volume (total volume x total porosity))	67,338		gallons	
Treatment Zone Effective Pore Volume (total volume x effective)	porosity)	50,504		gallons	
Design Period of Performance		1.5	.5 to 5	year	
Design Factor (times the electron acceptor hydrogen demand)		10.0	2 to 20	unitless	
Treatment Zone Hydrogeologic Properties					
Total Porosity		36%	.05-50	percent	PDI Eastover OU 7
Effective Porosity		27%	.05-50	percent	PDI Eastover OU 7, average based on TP, AP, clay conter
Average Aquifer Hydraulic Conductivity		5	.01-1000	ft/day	OU7 PDI
Average Hydraulic Gradient		1.61E-02	0.0001-0.1	ft/ft	2024 Annual Report
Average Groundwater Seepage Velocity through the Treatment 2		0.30		ft/day	
Average Groundwater Seepage Velocity through the Treatment 2 Average Groundwater Discharge through the Treatment Zone	LUITE	108.8		ft/yr gallons/year	
Soil Bulk Density		1.63	1.4-2.0	gm/cm ³	Average OU- PDI for Eastover and Aquia
Soil Fraction Organic Carbon (foc)		0.28%	0.01-10	percent	PDI-38 Eastover Saturated Zone OU-7 x 2
<u> </u>					
Native Electron Acceptors					
A. Aqueous-Phase Native Electron Acceptors	г				OMT 00 0 (00 (000 I)
Oxygen		0.1	0.01 to 10 0.1 to- 20	mg/L	CMT-20-3 (03/2024) CMT-20-3 (03/2024)
Nitrate Sulfate		0.03	10 to 5,000	mg/L mg/L	CMT-20-3 (03/2024) CMT-20-3 (03/2024)
Carbon Dioxide (estimated as the amount of Methane produced)		10.0	0.1 to 20	mg/L	Estimate based on previous EISB injections
	<u>l</u>			<u>.</u>	
B. Solid-Phase Native Electron Acceptors			_		
Manganese (IV) (estimated as the amount of Mn (II) produced)		1	0.1 to 20	mg/L	Data from 11/2020 (CMT20)
ron (III) (estimated as the amount of Fe (II) produced)		10	0.1 to 20	mg/L	Estimated
Contaminant Electron Acceptors					
Tetrachloroethene (PCE)		0.000		mg/L	CMT-20-3 03/2024
Trichloroethene (TCE)		0.001		mg/L	CMT-20-3 03/2024
Dichloroethene (cis-DCE, trans-DCE, and 1,1-DCE)		0.221		mg/L	CMT-20-3 03/2024
Vinyl Chloride (VC)		0.120		mg/L	CMT-20-3 03/2024
Carbon Tetrachloride (CT)		0.000		mg/L	Not detected
Trichloromethane (or chloroform) (CF)		0.000		mg/L	Not detected
Dichloromethane (or methylene chloride) (MC) Chloromethane		0.000		mg/L mg/L	Not detected Not detected
Tetrachloroethane (1,1,1,2-PCA and 1,1,2,2-PCA)		0.000		mg/L	Not detected
Trichloroethane (1,1,1-TCA and 1,1,2-TCA)		0.000		mg/L	Not detected
Dichloroethane (1,1-DCA and 1,2-DCA)		0.024		mg/L	CMT-20-3 03/2024
Chloroethane		0.000		mg/L	Not detected
Perchlorate		0.000		mg/L	No data
Amaifea Coooli amietra (Ontional Concesion Beaut					
Aquifer Geochemistry (Optional Screening Param A. Aqueous Geochemistry	ieters)				
Oxidation-Reduction Potential (ORP)		-64	-400 to +500	mV	CMT-20-3 03/2024
Temperature		16	5.0 to 30	°C	CMT-20-3 03/2024
pH		6.4	4.0 to 10.0	su	CMT-20 03/2024
Alkalinity		153	10 to 1,000	mg/L	CMT-20 03/2024
Total Dissolved Solids (TDS, or salinity)		100	10 to 1,000	mg/L	No data
Specific Conductivity		479	100 to 10,000	•	CMT-20-3 03/2024
Chloride Sulfide - Pre injection		125 0.1	10 to 10,000 0.1 to 100	mg/L mg/L	CMT-20 03/2024 Estimate
Sulfide - Pre injection Sulfide - Post injection		0.1	0.1 to 100	mg/L	Estimate
	ļ	0.2	0.110 100	y-	
B. Aquifer Matrix					
Total Iron		18000	200 to 20,000	mg/kg	CSM 2006 Soil Eastover
Cation Exchange Capacity		1	1.0 to 10	meq/100 g	CSM 2006 Soil Eastover
Neutralization Potential		1.0%	1.0 to 100	reicent as CaCO ₃	CSM 2006 Soil Eastover
NOTES:					

Table 3.2 3	Substrate Ca	alculations in	Hydrogen	Equivalents		
Site Name:	OU7 Pit 1	TA-06 CMT-20) (INJ-211)		RETURN TO	COVER PAGE
·				NOTE: Open cells	are user input.	
. Treatment Zone Physical Dimensions				Values	Range	Units
Width (Perpendicular to predominant groundwater flow	w direction)			25	1-10,000	feet
Length (Parallel to predominant groundwater flow)	,			50	1-1,000	feet
Saturated Thickness				20	1-100	feet
Treatment Zone Cross Sectional Area				500		ft ²
Treatment Zone Volume				25,000		ft ³
Treatment Zone Effective Pore Volume (total volume :	x effective porosit	\v)		50,504		gallons
Design Period of Performance	po	• • • • • • • • • • • • • • • • • • • •		1.5	.5 to 5	year
· ·					.0 .0 0	,
2. Treatment Zone Hydrogeologic Propertie	s					
Total Porosity				0.36	.05-50	
Effective Porosity				0.27	.05-50	
Average Aquifer Hydraulic Conductivity				5	.01-1000	ft/day
Average Hydraulic Gradient				0.0161	0.1-0.0001	ft/ft
Average Groundwater Seepage Velocity through the 1				0.30		ft/day
Average Groundwater Seepage Velocity through the 1				108.8		ft/yr
Average Groundwater Flux through the Treatment Zor	. ()		109,920		gallons/year
Soil Bulk Density				1.63	1.4-2.0	gm/cm ³
Soil Fraction Organic Carbon (foc)				0.0028	0.0001-0.1	
. Initial Treatment Cell Electron-Acceptor I	Demand (one	total pore volu	me)			
,	(3		•	Stoichiometric	Hydrogen	Electron
A. Aqueous-Phase Native Electron Acceptors		Concentration	Mass	demand	Demand	Equivalents p
, account the control and a co		(mg/L)	(lb)	(wt/wt h ₂)	(lb)	Mole
Oxygen		0.1	0.03	7.94	0.00	4
Oxygen		0.0	0.03	12.30	0.00	5
Nitrate (denitrification) Sulfate		2.2	0.93	12.30	0.00	8
Carbon Dioxide (estimated as the amount of methane	produced)	10.0	4.21	1.99	2.12	8
Carbon bioxide (estimated as the amount of methane	produced)			eptor Demand (lb.)	2.20	0
		Colubic Compet	ing Electron Acc	. ,,,		
D. Oall J. Dharas Nect.		0		Stoichiometric	Hydrogen	Electron
B. Solid-Phase Native Electron Acceptors		Concentration	Mass	demand	Demand	Equivalents p
(Based on manganese and iron produced)		(mg/L)	(lb)	(wt/wt h ₂)	(lb)	Mole
Manganese (IV) (estimated as the amount of Mn (II) p		1.3	2.36	27.25	0.09	2
Iron (III) (estimated as the amount of Fe (II) produced)		10.0	17.97	55.41	0.32	1
	Sol	id-Phase Compet	ing Electron Acc	eptor Demand (lb.)	0.41	
				Stoichiometric	Hydrogen	Electron
C. Soluble Contaminant Electron Acceptors		Concentration	Mass	demand	Demand	Equivalents p
		(mg/L)	(lb)	(wt/wt h ₂)	(lb)	Mole
Tetrachloroethene (PCE)		0.000	0.00	20.57	0.00	8
Trichloroethene (TCE)		0.001	0.00	21.73	0.00	6
Dichloroethene (cis-DCE, trans-DCE, and 1,1-DCE)		0.221	0.09	24.05	0.00	4
Vinyl Chloride (VC)		0.120	0.05	31.00	0.00	2
Carbon Tetrachloride (CT)		0.000	0.00	19.08	0.00	8
Trichloromethane (or chloroform) (CF)		0.000	0.00	19.74	0.00	6
Dichloromethane (or methylene chloride) (MC)		0.000	0.00	21.06	0.00	4
Chloromethane		0.000	0.00	25.04	0.00	2
Tetrachloroethane (1,1,1,2-PCA and 1,1,2,2-PCA)		0.000	0.00	20.82	0.00	8
Trichloroethane (1,1,1-TCA and 1,1,2-TCA)		0.000	0.00	22.06	0.00	6
Dichloroethane (1,1-DCA and 1,2-DCA)		0.024	0.01	24.55	0.00	4
Chloroethane		0.000	0.00	32.00	0.00	2
Perchlorate		0.000	0.00	12.33	0.00	6
	Total S			eptor Demand (lb.)	0.01	j
				Stoichiometric	Hydrogen	Electron
D. Sorbed Contaminant Electron Acceptors	Koc	Soil Conc.	Mass	demand	Demand	Equivalents
(Soil Concentration = Koc x foc x Cgw)	(mL/g)	(mg/kg)	(lb)	(wt/wt h ₂)	(lb)	Mole
Tetrachloroethene (PCE)	263	0.00	0.00	20.57	0.00	8
Trichloroethene (TCE)	107	0.00	0.00	21.73	0.00	6
Dichloroethene (cis-DCE, trans-DCE, and 1,1-DCE)	45	0.03	0.07	24.05	0.00	4
Vinyl Chloride (VC)	3.0	0.03	0.00	31.00	0.00	2
Carbon Tetrachloride (CT)	224	0.00	0.00	19.08	0.00	8
		0.00	0.00	19.08	0.00	
Trichloromethane (or chloroform) (CF)	63					6
Dichloromethane (or methylene chloride) (MC)	28	0.00	0.00	21.06	0.00	4
Chloromethane	25	0.00	0.00	25.04	0.00	2
Tetrachloroethane (1,1,1,2-PCA and 1,1,2,2-PCA)	117	0.00	0.00	20.82	0.00	8
Trichloroethane (1,1,1-TCA and 1,1,2-TCA)	105	0.00	0.00	22.06	0.00	6
Dichloroethane (1,1-DCA and 1,2-DCA)	30	0.00	0.01	24.55	0.00	4
Chloroethane	3	0.00	0.00	32.00	0.00	2
		0.00	0.00	12.33	0.00	6
Perchlorate	0.0	0.00		eptor Demand (lb.)	0.00	

Table S.2 Substrate Calculations in Hydrogen Equivalents						
4. Treatment Cell Electron-Acceptor Flux (per ye			_			
. " ,	,		Stoichiometric	Hydrogen	Electron	
A. Soluble Native Electron Acceptors	Concentration	Mass	demand	Demand	Equivalents per	
	(mg/L)	(lb)	(wt/wt h ₂)	(lb)	Mole	
Oxygen	0.1	0.07	7.94	0.01	4	
Nitrate (denitrification)	0.0	0.02	10.25	0.00	5	
Sulfate	2.2	2.02	11.91	0.17	8	
Carbon Dioxide (estimated as the amount of Methane produ		9.17	1.99	4.61	8	
Total Competing Electron Acceptor Demand Flux (lb/yr) 4.8						
Stoichiometric Hydrog					Electron	
B. Soluble Contaminant Electron Acceptors	Concentration	Mass	demand	Demand	Equivalents per	
	(mg/L)	(lb)	(wt/wt h ₂)	(lb)	Mole	
Tetrachloroethene (PCE)	0.000	0.00	20.57	0.00	8	
Trichloroethene (TCE)	0.001	0.00	21.73	0.00	6	
Dichloroethene (cis-DCE, trans-DCE, and 1,1-DCE)	0.221	0.20	24.05	0.01	4	
Vinyl Chloride (VC)	0.120	0.11	31.00	0.00	2	
Carbon Tetrachloride (CT)	0.000	0.00	19.08	0.00	8	
Trichloromethane (or chloroform) (CF)	0.000	0.00	19.74	0.00	6	
Dichloromethane (or methylene chloride) (MC)	0.000	0.00	21.06	0.00	4	
Chloromethane	0.000	0.00	25.04	0.00	2	
Tetrachloroethane (1,1,1,2-PCA and 1,1,2,2-PCA)	0.000	0.00	20.82	0.00	8	
Trichloroethane (1,1,1-TCA and 1,1,2-TCA)	0.000	0.00	22.06	0.00	6	
Dichloroethane (1,1-DCA and 1,2-DCA)	0.024	0.02	24.55	0.00	4	
Chloroethane	0.000	0.00	32.00	0.00	2	
Perchlorate	0.000	0.00	12.33	0.00	6	
Total Soluble Contaminant Electron Acceptor Demand Flux (lb/yr) 0.01						
Initial Hydrogen Requirement First Year (Ib) 7.4						
	Total Life-Cycl	e Hydrogen R	equirement (lb)	9.8		
5. Design Factors					_	
Microbial Efficiency Uncertainty Factor 2X - 4X						
Methane and Solid-Phase Electron Acceptor Uncertainty 2X				2X - 4X		
Remedial Design Factor (e.g., Substrate Leaving Reaction Zone) 1X - 3X 1X - 3X						
Design Factor 10.0						
Total Life-Cycle Hydrogen Requirement with Design Factor (lb) 98.3						
6. Acronyns and Abbreviations	e, e.e , a. e ge e qu e		o.g ao.o. ()	55.5	_	
and Abbreviations						
°C =degrees celsius meg	/100 g = milliequivalents per 10	00 grams				
	mg/kg = milligrams per kilogram					
	mg/L = milligrams per liter					
cm/sec = centimeters per second m/m						
ft ² = square feet mV =						
ft/yr = feet per year wt/wt H2 = concetration molecular hydrogen, weight per weight						
gm/cm ³ = grams per cubic centimeter						
kg of CaCO3 per mg = kilograms of calcium carbonate per	milligram					
lb = pounds						

Table S.3

Hydrogen Produced by Fermentation Reactions of Common Substrates

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Substrate	Molecular Formula	Substrate Molecular Weight (gm/mole)	Moles of Hydrogen Produced per Mole of Substrate	Ratio of Hydrogen Produced to Substrate (gm/gm)	Range of Moles H ₂ /Mole Substrate
Lactic Acid	C ₃ H ₆ O ₃	90.1	2	0.0448	2 to 3
Molasses (assuming 100% sucrose)	C ₁₂ H ₂₂ O ₁₁	342	8	0.0471	8 to 11
High Fructose Corn Syrup (assuming 50% fructose and 50% glucose)	C ₆ H ₁₂ O ₆	180	4	0.0448	4 to 6
Ethanol	C ₂ H ₆ O	46.1	2	0.0875	2 to 6
Whey (assuming 100% lactose)	C ₁₂ H ₂₂ O ₁₁	342	11	0.0648	11
HRC [®] (assumes 40% lactic acid and 40% glycerol by weight)	C ₃₉ H ₅₆ O ₃₉	956	28	0.0590	28
Linoleic Acid (Soybean Oil, Corn Oil, Cotton Oil)	C ₁₈ H ₃₂ O ₂	281	16	0.1150	16

Table S.4 Estimated Substrate Requirements for Hydrogen Demand in Table S.3

Design Life (years): 1.5

Substrate	Design Factor	Pure Substrate Mass Required to Fulfill Hydrogen Demand (pounds)	Substrate Product Required to Fulfill Hydrogen Demand (pounds)	Substrate Mass Required to Fulfill Hydrogen Demand (milligrams)	Effective Substrate Concentration (mg/L)
Lactic Acid	10.0	2,195	2,195	9.96E+08	1,221
Sodium Lactate Product (60 percent solution)	10.0	2,195	4,554	9.96E+08	1,221
Molasses (assuming 6 0	10.0	2,085	3,476	9.46E+08	1,160
HFCS (assuming 40% fructose and 40% glucose by weight)	10.0	2,196	2,745	9.96E+08	1,222
Ethanol Product (assuming 80% ethanol by weight)	10.0	1,123	1,403	5.09E+08	625
Whey (assuming 100% lactose)	10.0	1,515	2,165	6.87E+08	843
HRC® (assumes 40% lactic acid and 40% glycerol by weight)	10.0	1,664	1,664	7.55E+08	741
Linoleic Acid (Soybean Oil, Corn Oil, Cotton Oil)	10.0	854	854	3.88E+08	475
Commercial Vegetable Oil Emulsion Product (60% oil by weight)	10.0	854	1,424	3.88E+08	475

NOTES: Sodium Lactate Product

- 1. Assumes sodium lactate product is 60 percent sodium lactate by weight.
- 2. Molecular weight of sodium lactate (CH_3 -CHOH-COONa) = 112.06.
- 3. Molecular weight of lactic Acid $(C_6H_6O_3) = 90.08$.
- 4. Therefore, sodium lactate product yields 48.4 (0.60 x (90.08/112.06)) percent by weight lactic acid.
- 5. Weight of sodium lactate product = 11.0 pounds per gallon.
- 6. Pounds per gallon of lactic acid in product = 1.323×8.33 lb/gal H2O x $0.60 \times (90.08/112.06) = 5.31$ lb/gal.

NOTES: Standard HRC Product

- 1. Assumes HRC product is 40 percent lactic acid and 40 percent glycerol by weight.
- 2. HRC® weighs approximately 9.18 pounds per gallon.

NOTES: Vegetable Oil Emulsion Product

- 1. Assumes emulsion product is 60 percent soybean oil by weight.
- 2. Soybean oil is 7.8 pounds per gallon.
- 3. Assumes specific gravity of emulsion product is 0.96.

Table S.5 Output for Substrate Requirements in Hydrogen Equivalents

Site Name: OU7 Pit 1 TA-06 CMT-20 (INJ-211)

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1. Treatment Zone Physical Dimensions

Width (perpendicular to groundwater flow) Length (parallel to groundwater flow) Saturated Thickness Design Period of Performance

Values	Unit
25	feet
50	feet
20	feet
1.5	year

its	Values
t	8
t	15.2
t	6.1
irs	1.5

Units meters meters meters years

2. Treatment Zone Hydrogeologic Properties

Total Porosity
Effective Porosity
Average Aquifer Hydraulic Conductivity
Average Hydraulic Gradient
Average Groundwater Seepage Velocity
Average Groundwater Seepage Velocity
Effective Treatment Zone Pore Volume
Groundwater Flux (per year)
Total Groundwater Volume Treated
(over entire design period)

Values
0.36
0.27
5
0.0161
0.30
109
50,504
109,920
215,383

Hydrogen

110

183

Units
percent
percent
ft/day
ft/ft
ft/day
ft/yr
gallons
gallons/year
gallons total

Effective

Values	Units
0.36	perce
0.27	perce
1.8E-03	cm/s
0.0161	m/m
9.1E+00	cm/d
33.2	m/yr
191,171	liters
416,081	liters
815,292	liters

Units percent percent cm/sec m/m cm/day m/yr liters liters/year

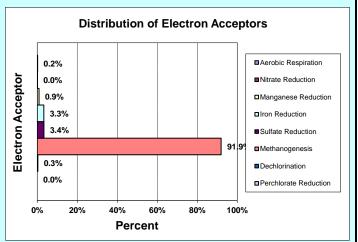
total

3. Distribution of Electron Acceptor Demand

Aerobic Respiration
Nitrate Reduction
Sulfate Reduction
Manganese Reduction
Iron Reduction
Methanogenesis
Dechlorination
Perchlorate Reduction

	riyarogen
Percent of Total	Demand (lb)
0.2%	0.018
0.0%	0.004
3.4%	0.332
0.9%	0.087
3.3%	0.324
91.9%	9.032
0.3%	0.029
0.0%	0.000
100.00%	9.83

Hydrogen demand in pounds/gallon: 4.56E-05
Hydrogen demand in grams per liter: 5.47E-03



4. Substrate Equivalents: Design Factor = 10.0

Totals:

Product	Quantity (lb)	Quantity (gallons)
Sodium Lactate Product	4,554	414
2. Molasses Product	3,476	290
3. Fructose Product	2,745	245
4. Ethanol Product	1,403	203
5. Sweet Dry Whey (lactose)	2,165	sold by pound
6. HRC®	1,664	sold by pound

Concentration	Effective concentration is for total
(mg/L)	volume of groundwater treated.
1,221	as lactic acid
1,160	as sucrose
1,222	as fructose
625	as ethanol
843	as lactose
741	as 40% lactic acid/40% glycerol
475	as soybean oil
475	as soybean oil

Notes:

7. Linoleic Acid (Soybean Oil)

8. Emulsified Vegetable Oil

- 1. Quantity assumes product is 60% sodium lactate by weight.
- 2. Quantity assumes product is 60% sucrose by weight and weighs 12 pounds per gallon.

854

1,424

- 3. Quantity assumes product is 80% fructose by weight and weighs 11.2 pounds per gallon.
- 4. Quantity assumes product is 80% ethanol by weight and weighs 6.9 pounds per gallon.
- 5. Quantity assumes product is 70% lactose by weight.
- 6. Quantity assumes HRC $\!\!\!\! ^{\odot}$ is 40% lactic acid and 40% glycerol by weight.
- 7. Quantity of neat soybean oil, corn oil, or canola oil.
- 8. Quantity assumes commercial product is 60% soybean oil by weight.

B.5 TA-07 Treatment Area Design

Site Name: OU7 Pit	2 TA-07 MW-85,	and MW-48		RETURN TO COVER PAGE
Treatment Zone Physical Dimensions	NOTE: Unshade Values	d boxes are use Range	r input. Units	User Notes
Width (Perpendicular to predominant groundwater flow direction)	40	1-10,000	feet	Pit 2 Area. 7 DPT Points
ength (Parallel to predominant groundwater flow)	34	1-1,000	feet	The Zamod, a Differential
Saturated Thickness	12	1-100	feet	20-32
reatment Zone Cross Sectional Area	480		ft ²	
reatment Zone Volume	16,320		ft ³	
Treatment Zone Total Pore Volume (total volume x total porosity)	43,958		gallons	
reatment Zone Effective Pore Volume (total volume x effective porosity Design Period of Performance		 E to E	gallons	
Design Factor (times the electron acceptor hydrogen demand)	3.0 10.0	.5 to 5 2 to 20	year unitless	
Treatment Zone Hydrogeologic Properties Total Porosity	36%	05.50	naraant	PDI Eastover OU 7
Effective Porosity	27%	.05-50 .05-50	percent percent	PDI Eastover OU 7, average based on TP, AP, clay conten
Average Aquifer Hydraulic Conductivity	5	.01-1000	ft/day	OU7 PDI
Average Hydraulic Gradient	1.20E-02	0.0001-0.1	ft/ft	2024 Annual, Pit 2 Upper Area
Average Groundwater Seepage Velocity through the Treatment Zone	0.22		ft/day	
Average Groundwater Seepage Velocity through the Treatment Zone	81.1		ft/yr	
Average Groundwater Discharge through the Treatment Zone	78,651		gallons/year	
Soil Bulk Density	1.63	1.4-2.0	gm/cm ³	Average OU- PDI for Eastover and Aquia PDI-38 Eastover Saturated Zone OU-7 x 2
Soil Fraction Organic Carbon (foc)	0.28%	0.01-10	percent	F DI-30 Eastover Saturated Zone OU-7 X Z
Native Electron Acceptors				
A. Aqueous-Phase Native Electron Acceptors		7		
Oxygen	3.1	0.01 to 10	mg/L	MW-85, MW-48 (average), 3/2024
Nitrate	0.05	0.1 to- 20	mg/L	MW-85, MW-48 (average), 3/2024
Sulfate	10.0	10 to 5,000	mg/L	MW-85, MW-48 (average), 3/2024 Estimate based on previous EISB injections
Carbon Dioxide (estimated as the amount of Methane produced)	10.0	0.1 to 20	mg/L	Estimate based on previous EISB injections
B. Solid-Phase Native Electron Acceptors		_		
Manganese (IV) (estimated as the amount of Mn (II) produced)	0	0.1 to 20	mg/L	MW-85, MW-48 (average), 3/2024
ron (III) (estimated as the amount of Fe (II) produced)	4	0.1 to 20	mg/L	MW-85, MW-48 (average), 3/2024
Contaminant Electron Acceptors				
Tetrachloroethene (PCE)	0.007		mg/L	MW-85 3/2024
Trichloroethene (TCE)	0.007		mg/L	MW-85 3/2024
Dichloroethene (cis-DCE, trans-DCE, and 1,1-DCE)	9.318		mg/L	MW-85 3/2024
Vinyl Chloride (VC)	0.570		mg/L	MW-85 3/2024
Carbon Tetrachloride (CT)	0.000		mg/L	Not detected
Trichloromethane (or chloroform) (CF) Dichloromethane (or methylene chloride) (MC)	0.000		mg/L mg/L	Not detected Not detected
Chloromethane	0.000		mg/L	Not detected
Tetrachloroethane (1,1,1,2-PCA and 1,1,2,2-PCA)	0.000		mg/L	Not detected
Trichloroethane (1,1,1-TCA and 1,1,2-TCA)	0.023		mg/L	MW-85 3/2024
Dichloroethane (1,1-DCA and 1,2-DCA)	0.018		mg/L	MW-85 3/2024
Chloroethane	0.000		mg/L	Not detected
Perchlorate	0.000		mg/L	No data
Aquifer Geochemistry (Optional Screening Parameters)			
A. Aqueous Geochemistry				
Oxidation-Reduction Potential (ORP)	49	-400 to +500	mV	MW-85, MW-48 (average), 3/2024
Temperature	16	5.0 to 30	°C	MW-85, MW-48 (average), 3/2024
oH Alkalinity	5.8 160	4.0 to 10.0 10 to 1,000	su mg/L	MW-85, MW-48 (average), 3/2024 MW-85, MW-48 (average), 3/2024
Fotal Dissolved Solids (TDS, or salinity)	100	10 to 1,000	mg/L	No data
Specific Conductivity	547	100 to 10,000		MW-85, MW-48 (average), 3/2024
Chloride	110	10 to 10,000	mg/L	MW-85, MW-48 (average), 3/2024
Sulfide - Pre injection	0.1	0.1 to 100	mg/L	Estimate Not Detected
Sulfide - Post injection	2.9	0.1 to 100	mg/L	Data from MW-85 3/2024
3. Aquifer Matrix				
Total Iron	18000	200 to 20,000	mg/kg	CSM 2006 Soil Eastover
Cation Exchange Capacity	1	1.0 to 10	meq/100 g	CSM 2006 Soil Eastover
Neutralization Potential	1.0%	1.0 to 100	Percent as CaCO ₃	CSM 2006 Soil Eastover
NOTES:				
OILO.				

- Tubic	S.2 Substrate C	alculations ii	i nyurogeni	Equivalents		
Site Name:	OU7 Pit 2	2 TA-07 MW-85,	and MW-48		RETURN TO	COVER PAGE
				NOTE: Open cells	are user input.	
. Treatment Zone Physical Dimensi	ons			Values	Range	Units
Width (Perpendicular to predominant ground	lwater flow direction)			40	1-10,000	feet
Length (Parallel to predominant groundwater	r flow)			34	1-1,000	feet
Saturated Thickness				12	1-100	feet
Treatment Zone Cross Sectional Area				480		ft ²
Treatment Zone Volume				16,320		ft ³
Treatment Zone Effective Pore Volume (tota	I volume x effective poros	sitv)		32,969		gallons
Design Period of Performance				3.0	.5 to 5	year
· ·	_			0.0	.0 .0 0	you.
. Treatment Zone Hydrogeologic Pr	roperties					
Total Porosity				0.36	.05-50	
Effective Porosity				0.27	.05-50	
Average Aquifer Hydraulic Conductivity				5	.01-1000	ft/day
Average Hydraulic Gradient				0.012	0.1-0.0001	ft/ft
Average Groundwater Seepage Velocity thro				0.22		ft/day
Average Groundwater Seepage Velocity thro				81.1		ft/yr
Average Groundwater Flux through the Treat	tment Zor	0		78,651		gallons/year
Soil Bulk Density				1.63	1.4-2.0	gm/cm ³
Soil Fraction Organic Carbon (foc)				0.0028	0.0001-0.1	
Initial Treatment Cell Electron-Acc	ceptor Demand (on	e total pore volu	ıme)			
				Stoichiometric	Hydrogen	Electron
A. Aqueous-Phase Native Electron Accept	tors	Concentration	Mass	demand	Demand	Equivalents
		(mg/L)	(lb)	(wt/wt h ₂)	(lb)	Mole
Oxygen		3.1	0.85	7.94	0.11	4
Nitrate (denitrification)		0.1	0.01	12.30	0.00	5
Sulfate		3.95	1.09	11.91	0.09	8
Carbon Dioxide (estimated as the amount of	methane produced)	10.0	2.75	1.99	1.38	8
(, , , , , , , , , , , , , , , , , , , ,			eptor Demand (lb.)	1.58	
			•	Stoichiometric	Hydrogen	
P. Calid Dhace Native Fleetren Assentare		Concentration	Mass	demand	Demand	Electron
B. Solid-Phase Native Electron Acceptors	•					Equivalents
(Based on manganese and iron produced)		(mg/L)	(lb)	(wt/wt h ₂)	(lb)	Mole
Manganese (IV) (estimated as the amount of		0.4	0.99	27.25	0.04	2
Iron (III) (estimated as the amount of Fe (II) p		3.7	8.33	55.41	0.15	1
	S	olid-Phase Compet	ing Electron Acc	eptor Demand (lb.)	0.19	
				Stoichiometric	Hydrogen	Electron
C. Soluble Contaminant Electron Accepto	ors	Concentration	Mass	demand	Demand	Equivalents
		(mg/L)	(lb)	(wt/wt h ₂)	(lb)	Mole
Tetrachloroethene (PCE)		0.007	0.00	20.57	0.00	8
Trichloroethene (TCE)		0.007	0.00	21.73	0.00	6
Dichloroethene (cis-DCE, trans-DCE, and 1,	1-DCE)	9.318	2.56	24.05	0.11	4
Vinyl Chloride (VC)	/	0.570	0.16	31.00	0.01	2
Carbon Tetrachloride (CT)		0.000	0.00	19.08	0.00	8
Trichloromethane (or chloroform) (CF)		0.000	0.00	19.74	0.00	6
Dichloromethane (or methylene chloride) (M	C)	0.000	0.00	21.06	0.00	4
Chloromethane	-,	0.000	0.00	25.04	0.00	2
Tetrachloroethane (1,1,1,2-PCA and 1,1,2,2-	-PCA)	0.000	0.00	20.82	0.00	8
Trichloroethane (1,1,1-TCA and 1,1,2-TCA)	,	0.000	0.01	22.06	0.00	6
Dichloroethane (1,1,1-TCA and 1,1,2-TCA)		0.023	0.00	24.55	0.00	4
Chloroethane		0.000	0.00	32.00	0.00	2
Perchlorate		0.000	0.00	12.33	0.00	6
	Tota			eptor Demand (lb.)	0.11	
				Stoichiometric	Hydrogen	Electron
D. Sorbed Contaminant Electron Accepto	rs Koc	Soil Conc.	Mass	demand	Demand	Equivalents
(Soil Concentration = Koc x foc x Cgw)	(mL/g)	(mg/kg)	(lb)	(wt/wt h ₂)	(lb)	Mole
Tetrachloroethene (PCE)	263	0.00	0.01	20.57	0.00	8
Trichloroethene (TCE)	107	0.00	0.00	21.73	0.00	6
Dichloroethene (cis-DCE, trans-DCE, and 1,		1.17	1.95	24.05	0.08	4
Vinyl Chloride (VC)	3.0	0.00	0.01	31.00	0.00	2
Carbon Tetrachloride (CT)	224	0.00	0.00	19.08	0.00	8
Trichloromethane (or chloroform) (CF)	63	0.00	0.00	19.74	0.00	6
Dichloromethane (or methylene chloride) (Mi		0.00	0.00	21.06	0.00	4
` ,	,					2
Chloromethane Tetrachloroethane (1.1.1.2 BCA and 1.1.2.2	25 117	0.00	0.00	25.04	0.00	
Tetrachloroethane (1,1,1,2-PCA and 1,1,2,2-		0.00	0.00	20.82	0.00	8
Trichloroethane (1,1,1-TCA and 1,1,2-TCA)	105	0.01	0.01	22.06	0.00	6
Dichloroethane (1,1-DCA and 1,2-DCA)	30	0.00	0.00	24.55	0.00	4
Chloroethane	3	0.00	0.00	32.00	0.00	2
	0.0	0.00	0.00	12.33	0.00	6
Perchlorate		I Sorbed Contamin			0.08	U

i—————————————————————————————————————						
Table S.2 Substr	ate Calculations in	Hydrogen I	Equivalents			
4. Treatment Cell Electron-Acceptor Flux (per year						
, ,,	<i>'</i>		Stoichiometric	Hydrogen	Electron	
A. Soluble Native Electron Acceptors	Concentration	Mass	demand	Demand	Equivalents per	
	(mg/L)	(lb)	(wt/wt h ₂)	(lb)	Mole	
Oxygen	3.1	2.03	7.94	0.26	4	
Nitrate (denitrification)	0.1	0.03	10.25	0.00	5	
Sulfate	3.95	2.59	11.91	0.22	8	
Carbon Dioxide (estimated as the amount of Methane produce	ed) 10	6.56	1.99	3.30	8	
Total Competing Electron Acceptor Demand Flux (lb/yr) 3.8						
			Stoichiometric	Hydrogen	Electron	
B. Soluble Contaminant Electron Acceptors	Concentration	Mass	demand	Demand	Equivalents per	
	(mg/L)	(lb)	(wt/wt h ₂)	(lb)	Mole	
Tetrachloroethene (PCE)	0.007	0.00	20.57	0.00	8	
Trichloroethene (TCE)	0.007	0.00	21.73	0.00	6	
Dichloroethene (cis-DCE, trans-DCE, and 1,1-DCE)	9.318	6.12	24.05	0.25	4	
Vinyl Chloride (VC)	0.570	0.37	31.00	0.01	2	
Carbon Tetrachloride (CT)	0.000	0.00	19.08	0.00	8	
Trichloromethane (or chloroform) (CF)	0.000	0.00	19.74	0.00	6	
Dichloromethane (or methylene chloride) (MC)	0.000	0.00	21.06	0.00	4	
Chloromethane	0.000	0.00	25.04	0.00	2	
Tetrachloroethane (1,1,1,2-PCA and 1,1,2,2-PCA)	0.000	0.00	20.82	0.00	8	
Trichloroethane (1,1,1-TCA and 1,1,2-TCA)	0.023	0.02	22.06	0.00	6	
Dichloroethane (1,1-DCA and 1,2-DCA)	0.018	0.01	24.55	0.00	4	
Chloroethane	0.000	0.00	32.00	0.00	2	
Perchlorate	0.000	0.00	12.33	0.00	6	
Total	Soluble Contaminant Elec	ctron Acceptor D	emand Flux (lb/yr)	0.27		
	Initial Hydroge	n Requiremen	t First Year (lb)	6.0		
	Total Life-Cycl	e Hydrogen R	equirement (lb)	14.1		
5. Design Factors					_	
Microbial Efficiency Uncertainty Factor				2X - 4X		
Methane and Solid-Phase Electron Acceptor Uncertainty				2X - 4X		
Remedial Design Factor (e.g., Substrate Leaving Reaction Zone	e)			1X - 3X		
			Design Factor	10.0		
Total Life-Cv	cle Hydrogen Reguire	ment with De	_	140.9		
6. Acronyns and Abbreviations	ole riyaregen nequire	ment with be	oigii i dotoi (ib)	140.5		
o. Actoriyiis and Abbreviations						
°C =degrees celsius meg/10	0 g = milliequivalents per 10	00 grams				
	= milligrams per kilogram					
cm/day = centimeters per day mg/L = milligrams per liter						
cm/sec = centimeters per second m/m = meters per meters						
ft ² = square feet mV = millivolts						
ft/day = feet per day m/yr = meters per year						
ft/ft = foot per foot su = sta	ft/ft = foot per foot su = standard pH units					
ft/yr = feet per year wt/wt H2 = concetration molecular hydrogen, weight per weight						
gm/cm ³ = grams per cubic centimeter		-	-			
kg of CaCO3 per mg = kilograms of calcium carbonate per mil	lligram					
lb = pounds						

Table S.3

Hydrogen Produced by Fermentation Reactions of Common Substrates

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Substrate	Molecular Formula	Substrate Molecular Weight (gm/mole)	Moles of Hydrogen Produced per Mole of Substrate	Ratio of Hydrogen Produced to Substrate (gm/gm)	Range of Moles H ₂ /Mole Substrate
Lactic Acid	C ₃ H ₆ O ₃	90.1	2	0.0448	2 to 3
Molasses (assuming 100% sucrose)	C ₁₂ H ₂₂ O ₁₁	342	8	0.0471	8 to 11
High Fructose Corn Syrup (assuming 50% fructose and 50% glucose)	C ₆ H ₁₂ O ₆	180	4	0.0448	4 to 6
Ethanol	C ₂ H ₆ O	46.1	2	0.0875	2 to 6
Whey (assuming 100% lactose)	C ₁₂ H ₂₂ O ₁₁	342	11	0.0648	11
HRC [®] (assumes 40% lactic acid and 40% glycerol by weight)	C ₃₉ H ₅₆ O ₃₉	956	28	0.0590	28
Linoleic Acid (Soybean Oil, Corn Oil, Cotton Oil)	C ₁₈ H ₃₂ O ₂	281	16	0.1150	16

Table S.4
Estimated Substrate Requirements for Hydrogen Demand in Table S.3

Design Life (years): 3

Substrate	Design Factor	Pure Substrate Mass Required to Fulfill Hydrogen Demand (pounds)	Substrate Product Required to Fulfill Hydrogen Demand (pounds)	Substrate Mass Required to Fulfill Hydrogen Demand (milligrams)	Effective Substrate Concentration (mg/L)
Lactic Acid	10.0	3,148	3,148	1.43E+09	1,403
Sodium Lactate Product (60 percent solution)	10.0	3,148	6,532	1.43E+09	1,403
Molasses (assuming 6 0	10.0	2,991	4,985	1.36E+09	1,333
HFCS (assuming 40% fructose and 40% glucose by weight)	10.0	3,149	3,936	1.43E+09	1,403
Ethanol Product (assuming 80% ethanol by weight)	10.0	1,610	2,013	7.30E+08	718
Whey (assuming 100% lactose)	10.0	2,173	3,105	9.86E+08	968
HRC® (assumes 40% lactic acid and 40% glycerol by weight)	10.0	2,387	2,387	1.08E+09	851
Linoleic Acid (Soybean Oil, Corn Oil, Cotton Oil)	10.0	1,226	1,226	5.56E+08	546
Commercial Vegetable Oil Emulsion Product (60% oil by weight)	10.0	1,226	2,043	5.56E+08	546

NOTES: Sodium Lactate Product

- 1. Assumes sodium lactate product is 60 percent sodium lactate by weight.
- 2. Molecular weight of sodium lactate (CH_3 -CHOH-COONa) = 112.06.
- 3. Molecular weight of lactic Acid $(C_6H_6O_3) = 90.08$.
- 4. Therefore, sodium lactate product yields 48.4 (0.60 x (90.08/112.06)) percent by weight lactic acid.
- 5. Weight of sodium lactate product = 11.0 pounds per gallon.
- 6. Pounds per gallon of lactic acid in product = 1.323×8.33 lb/gal H2O x $0.60 \times (90.08/112.06) = 5.31$ lb/gal.

NOTES: Standard HRC Product

- 1. Assumes HRC product is 40 percent lactic acid and 40 percent glycerol by weight.
- 2. HRC® weighs approximately 9.18 pounds per gallon.

NOTES: Vegetable Oil Emulsion Product

- 1. Assumes emulsion product is 60 percent soybean oil by weight.
- 2. Soybean oil is 7.8 pounds per gallon.
- 3. Assumes specific gravity of emulsion product is 0.96.

Table S.5 Output for Substrate Requirements in Hydrogen Equivalents

Site Name: OU7 Pit 2 TA-07 MW-85, and MW-48 RETURN TO COVER PAGE

1. Treatment Zone Physical Dimensions

Width (perpendicular to groundwater flow) Length (parallel to groundwater flow) Saturated Thickness Design Period of Performance

Values	Unit
40	feet
34	feet
12	feet
3	year

values	
12	
10.4	
3.7	
3	

Units meters meters meters years

2. Treatment Zone Hydrogeologic Properties

Total Porosity Effective Porosity Average Aquifer Hydraulic Conductivity Average Hydraulic Gradient Average Groundwater Seepage Velocity Average Groundwater Seepage Velocity Effective Treatment Zone Pore Volume Groundwater Flux (per year) Total Groundwater Volume Treated (over entire design period)

Values
0.36
0.27
5
0.012
0.22
81
32,969
78,651
268,921

Hydrogen

10.0

percent percent ft/day ft/ft ft/day ft/yr gallons gallons/year gallons total

Effective

Units

Values	Units
0.36	percent
0.27	percent
1.8E-03	cm/sec
0.012	m/m
6.8E+00	cm/day
24.7	m/yr
124,797	liters
297,717	liters/year
1,017,948	liters total

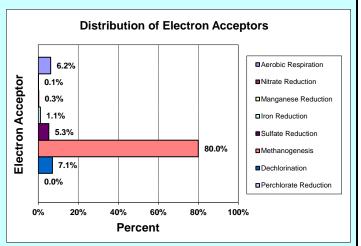
3. Distribution of Electron Acceptor Demand

Aerobic Respiration Nitrate Reduction Sulfate Reduction Manganese Reduction Iron Reduction Methanogenesis Dechlorination Perchlorate Reduction

	riyurogen
Percent of Total	Demand (lb)
6.2%	0.876
0.1%	0.011
5.3%	0.744
0.3%	0.036
1.1%	0.150
80.0%	11.276
7.1%	0.999
0.0%	0.000
100.00%	14.09

Totals:

Hydrogen demand in pounds/gallon: 5.24E-05 Hydrogen demand in grams per liter: 6.28E-03



4. Substrate Equivalents: Design Factor =

Product	Quantity (lb)	Quantity (gallons)
Sodium Lactate Product	6,532	594
2. Molasses Product	4,985	415
3. Fructose Product	3,936	351
4. Ethanol Product	2,013	292
5. Sweet Dry Whey (lactose)	3,105	sold by pound

Sodium Lactate Product	6,532	594
Molasses Product	4,985	415
Fructose Product	3,936	351
Ethanol Product	2,013	292
Sweet Dry Whey (lactose)	3,105	sold by pound
HRC [®]	2,387	sold by pound
Linoleic Acid (Soybean Oil)	1,226	157
Emulsified Vegetable Oil	2,043	262

LITECTIVE						
Concentration	Effective concentration is for total					
(mg/L)	volume of groundwater treated.					
1,403	as lactic acid					
1,333	as sucrose					
1,403	as fructose					
718	as ethanol					
968	as lactose					
851	as 40% lactic acid/40% glycerol					
546	as soybean oil					
546	as soybean oil					

Notes:

6. | 7. 8.

- 1. Quantity assumes product is 60% sodium lactate by weight.
- 2. Quantity assumes product is 60% sucrose by weight and weighs 12 pounds per gallon.
- 3. Quantity assumes product is 80% fructose by weight and weighs 11.2 pounds per gallon.
- 4. Quantity assumes product is 80% ethanol by weight and weighs 6.9 pounds per gallon.
- 5. Quantity assumes product is 70% lactose by weight.
- 6. Quantity assumes HRC® is 40% lactic acid and 40% glycerol by weight.
- 7. Quantity of neat soybean oil, corn oil, or canola oil.
- 8. Quantity assumes commercial product is 60% soybean oil by weight.

B.6 TA-08 Treatment Area Design

Site Name:	OU7 Pit 2	2 TA-08 CMT-18	8, MW-144		RETURN TO COVER PAGE
<u> </u>		NOTE: Unshaded	·	r input.	
Treatment Zone Physical Dimensions		Values	Range	Units	User Notes
Nidth (Perpendicular to predominant groundwater flow directi	on)	45	1-10,000	feet	Pit 2 Area, 7 DPT points
Length (Parallel to predominant groundwater flow)		34	1-1,000	feet	47.00
Saturated Thickness Freatment Zone Cross Sectional Area		11 495	1-100	feet ft ²	17-28
Treatment Zone Volume		16,830		ft ³	
Treatment Zone Total Pore Volume (total volume x total poro:	sitv)	45,332		gallons	
Freatment Zone Effective Pore Volume (total volume x effecti		33,999		gallons	
Design Period of Performance	, ,,	3.0	.5 to 5	year	
Design Factor (times the electron acceptor hydrogen demand)	10.0	2 to 20	unitless	
To a decide to the late of the Board of the					
Treatment Zone Hydrogeologic Properties		200/	05.50		DDI Festerra OLL 7
Total Porosity Effective Porosity		36% 27%	.05-50 .05-50	percent percent	PDI Eastover OU 7 PDI Eastover OU 7, average based on TP, AP, clay conten
Average Aquifer Hydraulic Conductivity		5	.01-1000	ft/day	OU7 PDI
Average Hydraulic Gradient		1.20E-02	0.0001-0.1	ft/ft	2024 Annual, Pit 2 Upper Area
Average Groundwater Seepage Velocity through the Treatme	nt Zone	0.22		ft/day	20217 Mindall, Tit 2 Opportuod
Average Groundwater Seepage Velocity through the Treatme		81.1		ft/yr	
Average Groundwater Discharge through the Treatment Zone		81,109		gallons/year	
Soil Bulk Density		1.63	1.4-2.0	gm/cm ³	Average OU- PDI for Eastover and Aquia
Soil Fraction Organic Carbon (foc)		0.28%	0.01-10	percent	PDI-38 Eastover Saturated Zone OU-7 x 2
Native Electron Acceptors					
A. Aqueous-Phase Native Electron Acceptors Oxygen		0.8	0.01 to 10	mg/L	CMT-18 03/2024
Vitrate		0.16	0.01 to 10	mg/L	CMT-18 03/2024 CMT-18 03/2024
Sulfate		1	10 to 5,000	mg/L	CMT-18 03/2024
Carbon Dioxide (estimated as the amount of Methane produc	ed)	10.0	0.1 to 20	mg/L	Estimate based on previous EISB injections
			_		•
B. Solid-Phase Native Electron Acceptors					
Manganese (IV) (estimated as the amount of Mn (II) produce	d)	1	0.1 to 20	mg/L	CMT-18 03/2024
ron (III) (estimated as the amount of Fe (II) produced)		3	0.1 to 20	mg/L	CMT-18 03/2024
Contaminant Electron Acceptors					
Tetrachloroethene (PCE)		0.007		mg/L	CMT-18 03/2024
Trichloroethene (TCE)		0.006		mg/L	CMT-18 03/2024
Dichloroethene (cis-DCE, trans-DCE, and 1,1-DCE)		2.609		mg/L	CMT-18 03/2024
Vinyl Chloride (VC)		0.920		mg/L	CMT-18 03/2024
Carbon Tetrachloride (CT)		0.000		mg/L	Not detected
Trichloromethane (or chloroform) (CF)		0.010		mg/L	CMT-18 03/2024
Dichloromethane (or methylene chloride) (MC)		0.000		mg/L	Not detected
Chloromethane		0.000		mg/L	Not detected
Tetrachloroethane (1,1,1,2-PCA and 1,1,2,2-PCA)		0.000		mg/L	Not detected
Trichloroethane (1,1,1-TCA and 1,1,2-TCA)		0.000		mg/L	Not detected
Dichloroethane (1,1-DCA and 1,2-DCA) Chloroethane		0.027 0.000		mg/L mg/L	CMT-18 03/2024 Not detected
Perchlorate		0.000		mg/L	Not detected No data
-		2.500		y –	
Aquifer Geochemistry (Optional Screening Par	ameters)				
A. Aqueous Geochemistry		0.10	400 : 555		OMT 40, 00/0004
Oxidation-Reduction Potential (ORP)		249	-400 to +500	mV °C	CMT-18 03/2024
Femperature DH		15 5.2	5.0 to 30 4.0 to 10.0	°C su	CMT-18 03/2024 CMT-18 03/2024
on Alkalinity		93	10 to 1,000	mg/L	CMT-18 03/2024 CMT-18 03/2024
Total Dissolved Solids (TDS, or salinity)		100	10 to 1,000	mg/L	No data
Specific Conductivity		339	100 to 10,000		CMT-18 03/2024
Chloride		121	10 to 10,000	mg/L	CMT-18 03/2024
Sulfide - Pre injection		0.1	0.1 to 100	mg/L	Estimate Not Detected
Sulfide - Post injection		1.6	0.1 to 100	mg/L	CMT-18 03/2024
A sould a Market					
B. Aquifer Matrix		40000	200 + 20 022		CCM 2000 Cail Factories
Total Iron		18000	200 to 20,000		CSM 2006 Soil Eastover
Cation Exchange Capacity Neutralization Potential		1 1.0%	1.0 to 10 1.0 to 100	meq/100 g Percent as CaCO ₃	CSM 2006 Soil Eastover CSM 2006 Soil Eastover
NOTES:					

Table S.2 S	Substrate Ca	alculations in	n Hydrogen	Equivalents			
Site Name:	OU7 Pit 2	2 TA-08 CMT-1	8, MW-144		RETURN TO	COVER PAGE	
1. Treatment Zone Physical Dimensions			•	NOTE: Open cells Values	•	Units	
Width (Perpendicular to predominant groundwater flo	w direction)			45	Range 1-10,000	feet	
	ngth (Parallel to predominant groundwater flow)						
Saturated Thickness				34 11	1-1,000 1-100	feet feet	
Treatment Zone Cross Sectional Area				495		ft ²	
Treatment Zone Volume				16,830		ft ³	
Treatment Zone Effective Pore Volume (total volume	x effective porosit	y)		33,999		gallons	
Design Period of Performance				3.0	.5 to 5	year	
2. Treatment Zone Hydrogeologic Propertie	s						
Total Porosity				0.36	.05-50		
Effective Porosity				0.27	.05-50	6.7.1	
Average Aquifer Hydraulic Conductivity Average Hydraulic Gradient				5 0.012	.01-1000 0.1-0.0001	ft/day ft/ft	
Average Groundwater Seepage Velocity through the	Freatment Zone			0.012	0.1-0.0001	ft/day	
Average Groundwater Seepage Velocity through the				81.1		ft/yr	
Average Groundwater Flux through the Treatment Zor)		81,109		gallons/year	
Soil Bulk Density				1.63	1.4-2.0	gm/cm ³	
Soil Fraction Organic Carbon (foc)				0.0028	0.0001-0.1		
3. Initial Treatment Cell Electron-Acceptor I	Demand (one	total pore volu	ıme)				
				Stoichiometric	Hydrogen	Electron	
A. Aqueous-Phase Native Electron Acceptors		Concentration	Mass	demand	Demand	Equivalents pe	
		(mg/L)	(lb)	(wt/wt h ₂)	(lb)	Mole	
Oxygen		0.8	0.23	7.94	0.03	4	
Nitrate (denitrification) Sulfate		0.2	0.05 0.14	12.30 11.91	0.00 0.01	5 8	
Carbon Dioxide (estimated as the amount of methane	produced)	10.0	2.84	1.99	1.43	8	
Carson 2 ionide (commuted de me amedin en memane	p.oddood)			eptor Demand (lb.)	1.47		
				Stoichiometric	Hydrogen	Electron	
B. Solid-Phase Native Electron Acceptors		Concentration	Mass	demand	Demand	Equivalents pe	
(Based on manganese and iron produced)		(mg/L)	(lb)	(wt/wt h ₂)	(lb)	Mole	
Manganese (IV) (estimated as the amount of Mn (II) p	roduced)	0.6	1.43	27.25	0.05	2	
Iron (III) (estimated as the amount of Fe (II) produced		3.2	7.43	55.41	0.13	1	
	Sol	id-Phase Compet	ing Electron Acc	eptor Demand (lb.)	0.19		
				Stoichiometric	Hydrogen	Electron	
C. Soluble Contaminant Electron Acceptors		Concentration	Mass	demand	Demand	Equivalents pe	
T (DOF)		(mg/L)	(lb)	(wt/wt h ₂)	(lb)	Mole	
Tetrachloroethene (PCE) Trichloroethene (TCE)		0.007 0.006	0.00	20.57 21.73	0.00	8	
Dichloroethene (cis-DCE, trans-DCE, and 1,1-DCE)		2.609	0.74	24.05	0.03	4	
Vinyl Chloride (VC)		0.920	0.26	31.00	0.01	2	
Carbon Tetrachloride (CT)		0.000	0.00	19.08	0.00	8	
Trichloromethane (or chloroform) (CF)		0.010	0.00	19.74	0.00	6	
Dichloromethane (or methylene chloride) (MC)		0.000	0.00	21.06	0.00	4	
Chloromethane		0.000	0.00	25.04 20.82	0.00	8	
Tetrachloroethane (1,1,1,2-PCA and 1,1,2,2-PCA) Trichloroethane (1,1,1-TCA and 1,1,2-TCA)		0.000	0.00	22.06	0.00	6	
Dichloroethane (1,1,1-1CA and 1,2-DCA)		0.027	0.00	24.55	0.00	4	
Chloroethane		0.000	0.00	32.00	0.00	2	
Perchlorate		0.000	0.00	12.33	0.00	6	
	Total S	Soluble Contamin	ant Electron Acc	eptor Demand (lb.)	0.04		
D. Oarthad Oarthan I		0. ". 0		Stoichiometric	Hydrogen	Electron	
D. Sorbed Contaminant Electron Acceptors	Koc	Soil Conc.	Mass	demand	Demand	Equivalents p	
(Soil Concentration = Koc x foc x Cgw)	(mL/g)	(mg/kg)	(lb)	(wt/wt h ₂)	(lb)	Mole	
Tetrachloroethene (PCE) Trichloroethene (TCE)	263 107	0.01	0.01 0.00	20.57 21.73	0.00	8	
Dichloroethene (cis-DCE, trans-DCE, and 1,1-DCE)	45	0.00	0.56	24.05	0.02	4	
Vinyl Chloride (VC)	3.0	0.01	0.01	31.00	0.00	2	
Carbon Tetrachloride (CT)	224	0.00	0.00	19.08	0.00	8	
Trichloromethane (or chloroform) (CF)	63	0.00	0.00	19.74	0.00	6	
Dichloromethane (or methylene chloride) (MC)	28	0.00	0.00	21.06	0.00	4	
Chloromethane Tetrachloroethane (1,1,1,2-PCA and 1,1,2,2-PCA)	25 117	0.00	0.00	25.04 20.82	0.00	8	
Trichloroethane (1,1,1,2-PCA and 1,1,2,2-PCA)	105	0.00	0.00	22.06	0.00	6	
Dichloroethane (1,1,1-1CA and 1,1,2-1CA)	30	0.00	0.00	24.55	0.00	4	
Chloroethane	3	0.00	0.00	32.00	0.00	2	
Perchlorate	0.0	0.00	0.00	12.33 eptor Demand (lb.)	0.00 0.02	6	

Table S.2 Substrate Calculations in Hydrogen Equivalents 4. Treatment Cell Electron-Acceptor Flux (per year) Stoichiometric Hydrogen Electron A. Soluble Native Electron Acceptors Concentration Mass demand Demand Equivalents pe (mg/L) (lb) (wt/wt h₂) (lb) Mole 8.0 0.54 7.94 0.07 4 Nitrate (denitrification) 0.2 0.11 10.25 0.01 5 Sulfate 0.5 0.34 11.91 0.03 8 Carbon Dioxide (estimated as the amount of Methane produced) 10 6.77 1.99 3.40 8 Total Competing Electron Acceptor Demand Flux (lb/yr) 3.5 Stoichiometric Hydrogen Electron **B. Soluble Contaminant Electron Acceptors** Concentration Mass Demand Equivalents per demand (mg/L) (lb) (wt/wt h₂) (lb) Mole Tetrachloroethene (PCE) 0.007 0.00 20.57 0.00 Trichloroethene (TCE) 0.006 0.00 21.73 0.00 6 Dichloroethene (cis-DCE, trans-DCE, and 1,1-DCE) 2.609 1.77 24.05 0.07 4 Vinvl Chloride (VC) 0.920 0.62 31.00 0.02 2 Carbon Tetrachloride (CT) 0.000 0.00 19.08 0.00 8 Trichloromethane (or chloroform) (CF) 0.010 0.01 19.74 0.00 6 Dichloromethane (or methylene chloride) (MC) 0.000 0.00 21.06 0.00 Chloromethane 0.000 0.00 25.04 0.00 2 Tetrachloroethane (1,1,1,2-PCA and 1,1,2,2-PCA) 0.000 0.00 20.82 0.00 8 Trichloroethane (1,1,1-TCA and 1,1,2-TCA) 0.000 0.00 22.06 0.00 6 Dichloroethane (1,1-DCA and 1,2-DCA) 0.027 0.02 24.55 0.00 4 Chloroethane 0.000 0.00 32.00 0.00 Perchlorate 0.000 0.00 0.00 Total Soluble Contaminant Electron Acceptor Demand Flux (lb/yr) 0.10 5.3 Initial Hydrogen Requirement First Year (Ib) Total Life-Cycle Hydrogen Requirement (lb) 12.5 5. Design Factors Microbial Efficiency Uncertainty Factor 2X - 4X Methane and Solid-Phase Electron Acceptor Uncertainty 2X - 4X Remedial Design Factor (e.g., Substrate Leaving Reaction Zone) 1X - 3X **Design Factor** 10.0 Total Life-Cycle Hydrogen Requirement with Design Factor (lb) 125.3 6. Acronyns and Abbreviations °C =degrees celsius meq/100 g = milliequivalents per 100 grams μs/cm = microsiemens per centimeter mg/kg = milligrams per kilogram cm/day = centimeters per day mg/L = milligrams per liter cm/sec = centimeters per second m/m = meters per meters ft2 = square feet mV = millivolts ft/day = feet per day m/yr = meters per year ft/ft = foot per foot su = standard pH units ft/yr = feet per year wt/wt H2 = concetration molecular hydrogen, weight per weight gm/cm³ = grams per cubic centimeter kg of CaCO3 per mg = kilograms of calcium carbonate per milligram lb = pounds

Table S.3

Hydrogen Produced by Fermentation Reactions of Common Substrates

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Substrate	Molecular Formula	Substrate Molecular Weight (gm/mole)	Moles of Hydrogen Produced per Mole of Substrate	Ratio of Hydrogen Produced to Substrate (gm/gm)	Range of Moles H ₂ /Mole Substrate
Lactic Acid	C ₃ H ₆ O ₃	90.1	2	0.0448	2 to 3
Molasses (assuming 100% sucrose)	C ₁₂ H ₂₂ O ₁₁	342	8	0.0471	8 to 11
High Fructose Corn Syrup (assuming 50% fructose and 50% glucose)	C ₆ H ₁₂ O ₆	180	4	0.0448	4 to 6
Ethanol	C ₂ H ₆ O	46.1	2	0.0875	2 to 6
Whey (assuming 100% lactose)	C ₁₂ H ₂₂ O ₁₁	342	11	0.0648	11
HRC® (assumes 40% lactic acid and 40% glycerol by weight)	C ₃₉ H ₅₆ O ₃₉	956	28	0.0590	28
Linoleic Acid (Soybean Oil, Corn Oil, Cotton Oil)	C ₁₈ H ₃₂ O ₂	281	16	0.1150	16

Table S.4 Estimated Substrate Requirements for Hydrogen Demand in Table S.3

Design Life (years): 3

Substrate	Design Factor	Pure Substrate Mass Required to Fulfill Hydrogen Demand (pounds)	Substrate Product Required to Fulfill Hydrogen Demand (pounds)	Substrate Mass Required to Fulfill Hydrogen Demand (milligrams)	Effective Substrate Concentration (mg/L)
Lactic Acid	10.0	2,800	2,800	1.27E+09	1,210
Sodium Lactate Product (60 percent solution)	10.0	2,800	5,808	1.27E+09	1,210
Molasses (assuming 6 0	10.0	2,660	4,433	1.21E+09	1,149
HFCS (assuming 40% fructose and 40% glucose by weight)	10.0	2,800	3,500	1.27E+09	1,210
Ethanol Product (assuming 80% ethanol by weight)	10.0	1,432	1,790	6.49E+08	619
Whey (assuming 100% lactose)	10.0	1,933	2,761	8.77E+08	835
HRC® (assumes 40% lactic acid and 40% glycerol by weight)	10.0	2,122	2,122	9.63E+08	734
Linoleic Acid (Soybean Oil, Corn Oil, Cotton Oil)	10.0	1,090	1,090	4.94E+08	471
Commercial Vegetable Oil Emulsion Product (60% oil by weight)	10.0	1,090	1,816	4.94E+08	471

NOTES: Sodium Lactate Product

- 1. Assumes sodium lactate product is 60 percent sodium lactate by weight.
- 2. Molecular weight of sodium lactate (CH_3 -CHOH-COONa) = 112.06.
- 3. Molecular weight of lactic Acid $(C_6H_6O_3) = 90.08$.
- 4. Therefore, sodium lactate product yields 48.4 (0.60 x (90.08/112.06)) percent by weight lactic acid.
- 5. Weight of sodium lactate product = 11.0 pounds per gallon.
- 6. Pounds per gallon of lactic acid in product = 1.323×8.33 lb/gal H2O x $0.60 \times (90.08/112.06) = 5.31$ lb/gal.

NOTES: Standard HRC Product

- 1. Assumes HRC product is 40 percent lactic acid and 40 percent glycerol by weight.
- 2. HRC® weighs approximately 9.18 pounds per gallon.

NOTES: Vegetable Oil Emulsion Product

- 1. Assumes emulsion product is 60 percent soybean oil by weight.
- 2. Soybean oil is 7.8 pounds per gallon.
- 3. Assumes specific gravity of emulsion product is 0.96.

Table S.5 Output for Substrate Requirements in Hydrogen Equivalents

Site Name: OU7 Pit 2 TA-08 CMT-18, MW-144

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1. Treatment Zone Physical Dimensions

Width (perpendicular to groundwater flow) Length (parallel to groundwater flow) Saturated Thickness Design Period of Performance

Values	Unit
45	feet
34	feet
11	feet
3	year

Values	Units
14	meters
10.4	meters
3.4	meters
3	years

2. Treatment Zone Hydrogeologic Properties

Total Porosity
Effective Porosity
Average Aquifer Hydraulic Conductivity
Average Hydraulic Gradient
Average Groundwater Seepage Velocity
Average Groundwater Seepage Velocity
Effective Treatment Zone Pore Volume
Groundwater Flux (per year)
Total Groundwater Volume Treated
(over entire design period)

Values
0.36
0.27
5
0.012
0.22
81
33,999
81,109
277,325

Hydrogen

percent percent ft/day ft/ft ft/day ft/yr gallons gallons/year gallons total

Effective

Units

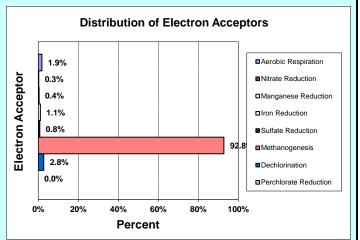
Values	Units
0.36	percent
0.27	percent
1.8E-03	cm/sec
0.012	m/m
6.8E+00	cm/day
24.7	m/yr
128,696	liters
307,021	liters/year
1,049,759	liters total

3. Distribution of Electron Acceptor Demand

Aerobic Respiration
Nitrate Reduction
Sulfate Reduction
Manganese Reduction
Iron Reduction
Methanogenesis
Dechlorination
Perchlorate Reduction

		riyurogen
	Percent of Total	Demand (lb)
	1.9%	0.233
	0.3%	0.035
	0.8%	0.097
	0.4%	0.053
	1.1%	0.134
	92.8%	11.629
	2.8%	0.350
	0.0%	0.000
Totals:	100.00%	12.53

Hydrogen demand in pounds/gallon: 4.52E-05
Hydrogen demand in grams per liter: 5.41E-03



4. Substrate Equivalents: Design Factor = 10.0

	Quantity	Quantity
Product	(lb)	(gallons)
Sodium Lactate Product	5,808	528
2. Molasses Product	4,433	369
3. Fructose Product	3,500	313
4. Ethanol Product	1,790	259
5. Sweet Dry Whey (lactose)	2,761	sold by pound
6. HRC [®]	2,122	sold by pound
7. Linoleic Acid (Soybean Oil)	1,090	140

Concentration	Effective concentration is for total
(mg/L)	volume of groundwater treated.
1,210	as lactic acid
1,149	as sucrose
1,210	as fructose
619	as ethanol
835	as lactose
734	as 40% lactic acid/40% glycerol
471	as soybean oil
471	as soybean oil

Notes

8. Emulsified Vegetable Oil

- 1. Quantity assumes product is 60% sodium lactate by weight.
- 2. Quantity assumes product is 60% sucrose by weight and weighs 12 pounds per gallon.

1,816

- 3. Quantity assumes product is 80% fructose by weight and weighs 11.2 pounds per gallon.
- 4. Quantity assumes product is 80% ethanol by weight and weighs 6.9 pounds per gallon.
- 5. Quantity assumes product is 70% lactose by weight.
- 6. Quantity assumes HRC $\!\!\!\! ^{\rm B}$ is 40% lactic acid and 40% glycerol by weight.
- 7. Quantity of neat soybean oil, corn oil, or canola oil.
- 8. Quantity assumes commercial product is 60% soybean oil by weight.

B.7 TA-09 Treatment Area Design

Site Name: OU7 Pit 2 TA-09 II	NJ-163, INJ-164	4, INJ-165, INJ	-207	RETURN TO COVER PAGE
Freatment Zone Physical Dimensions	NOTE: Unshade	d boxes are user Range	input. Units	User Notes
ridth (Perpendicular to predominant groundwater flow direction)	40		feet	Pit 2 Area
ength (Parallel to predominant groundwater flow)	30		feet	111271104
aturated Thickness	10		feet	16-26
reatment Zone Cross Sectional Area	400		ft ²	
reatment Zone Volume	12,000		ft ³	
reatment Zone Total Pore Volume (total volume x total porosity)	32,322		gallons	
reatment Zone Effective Pore Volume (total volume x effective porosity) esign Period of Performance	24,242 1.0		gallons	
esign Feriod of Feriormance esign Factor (times the electron acceptor hydrogen demand)	10.0		year unitless	
Freatment Zone Hydrogeologic Properties				
otal Porosity	36%	.05-50	percent	PDI Eastover OU 7
ffective Porosity	27%	.05-50	percent	PDI Eastover OU 7, average based on TP, AP, clay conte
verage Aquifer Hydraulic Conductivity	5		ft/day	OU7 PDI
verage Hydraulic Gradient	2.23E-02		ft/ft	Between injection transect and MW-149 03/2024
verage Groundwater Seepage Velocity through the Treatment Zone	0.41 150.7		ft/day	
verage Groundwater Seepage Velocity through the Treatment Zone verage Groundwater Discharge through the Treatment Zone	150.7		ft/yr gallons/year	
bil Bulk Density	1.63		gm/cm ³	Average OU- PDI for Eastover and Aquia
bil Fraction Organic Carbon (foc)	0.28%	0.01-10	percent	PDI-38 Eastover Saturated Zone OU-7 x 2
Native Electron Acceptors				
. Aqueous-Phase Native Electron Acceptors xygen	0.5	0.01 to 10	mg/L	MW-80, MW-83, MW-148 (average) 03/2024
itrate	0.05		mg/L	MW-80, MW-83, MW-148 (average) 03/2024 MW-80, MW-83, MW-148 (average) 03/2024
ulfate	15		mg/L	MW-80, MW-83, MW-148 (average) 03/2024
arbon Dioxide (estimated as the amount of Methane produced)	10.0		mg/L	Estimate based on previous EISB injections
Solid-Phase Native Electron Acceptors		0.4 : 00		NAM 00 NAM 00 NAM 440 /- > > 00/0004
anganese (IV) (estimated as the amount of Mn (II) produced) on (III) (estimated as the amount of Fe (II) produced)	10		mg/L mg/L	MW-80, MW-83, MW-148 (average) 03/2024 Estimated
on (iii) (estimated as the amount of re (ii) produced)	10	0.1 10 20	mg/L	Estinated
Contaminant Electron Acceptors		_		
etrachloroethene (PCE)	0.041		mg/L	MW-80 03/2024
richloroethene (TCE)	0.020		mg/L	MW-80 03/2024
ichloroethene (cis-DCE, trans-DCE, and 1,1-DCE) inyl Chloride (VC)	0.224 0.054		mg/L	MW-80 03/2024
arbon Tetrachloride (CT)	0.000		mg/L mg/L	MW-80 03/2024 Not detected
richloromethane (or chloroform) (CF)	0.000	-	mg/L	Not detected
ichloromethane (or methylene chloride) (MC)	0.000		mg/L	Not detected
hloromethane	0.000		mg/L	Not detected
etrachloroethane (1,1,1,2-PCA and 1,1,2,2-PCA)	0.000		mg/L	Not detected
richloroethane (1,1,1-TCA and 1,1,2-TCA)	0.003		mg/L	MW-148 03/2024
ichloroethane (1,1-DCA and 1,2-DCA)	0.003		mg/L	Average for plume area
hloroethane erchlorate	0.000		mg/L mg/L	Not detected No data
	0.000		mg/L	NO data
Aquifer Geochemistry (Optional Screening Parameters) Aqueous Geochemistry				
xidation-Reduction Potential (ORP)	71	_	mV	MW-80, MW-83, MW-148 (average) 03/2024
emperature	13		°C	MW-80, MW-83, MW-148 (average) 03/2024
1	5.9		su	MW-80, MW-83, MW-148 (average) 03/2024
kalinity otal Dissolved Solids (TDS, or salinity)	54 100		mg/L mg/L	MW-80, MW-83, MW-148 (average) 03/2024 Estimate
pecific Conductivity	533		μs/cm	MW-80, MW-83, MW-148 (average) 03/2024
hloride	178		mg/L	MW-80, MW-83, MW-148 (average) 03/2024
ulfide - Pre injection	0.1		mg/L	Estimate
ulfide - Post injection	0.2	0.1 to 100	mg/L	Estimate
. Aquifer Matrix				
otal Iron	18000	200 to 20,000	mg/kg	CSM 2006 Soil Eastover
ation Exchange Capacity	1	1.0 to 10	meq/100 g	CSM 2006 Soil Eastover
eutralization Potential	1.0%		Percent as CaCO ₃	

Table 5.2 5	Substrate Ca	alculations in	Hydrogen I	Equivalents		
Site Name: OU7	7 Pit 2 TA-09 I	NJ-163, INJ-16	4, INJ-165, INJ	J-207	RETURN TO	COVER PAGE
		ĺ	•	NOTE: Open cells	are user input.	
. Treatment Zone Physical Dimensions				Values	Range	Units
Width (Perpendicular to predominant groundwater flow		40	1-10,000	feet		
Length (Parallel to predominant groundwater flow)	,			30	1-1,000	feet
Saturated Thickness					1-100	feet
Treatment Zone Cross Sectional Area				400		ft ²
Treatment Zone Volume				12,000		ft ³
Treatment Zone Effective Pore Volume (total volume :	x effective porosit	tv)		24,242		gallons
Design Period of Performance	. 000t0 po.00	· 3 /		1.0	.5 to 5	year
· ·					.0 .0 0	you.
2. Treatment Zone Hydrogeologic Propertie	es .					
Total Porosity				0.36	.05-50	
Effective Porosity				0.27	.05-50	
Average Aquifer Hydraulic Conductivity				5	.01-1000	ft/day
Average Hydraulic Gradient				0.0223	0.1-0.0001	ft/ft
Average Groundwater Seepage Velocity through the 1				0.41		ft/day
Average Groundwater Seepage Velocity through the 1				150.7		ft/yr
Average Groundwater Flux through the Treatment Zor	. ()		121,799		gallons/year
Soil Bulk Density				1.63	1.4-2.0	gm/cm ³
Soil Fraction Organic Carbon (foc)				0.0028	0.0001-0.1	
. Initial Treatment Cell Electron-Acceptor I	Demand (one	total pore volu	me)			
•	•		•	Stoichiometric	Hydrogen	Electron
A. Aqueous-Phase Native Electron Acceptors		Concentration	Mass	demand	Demand	Equivalents p
		(mg/L)	(lb)	(wt/wt h ₂)	(lb)	Mole
Oxygen		0.5	0.10	7.94	0.01	4
Nitrate (denitrification)		0.1	0.01	12.30	0.00	5
Sulfate		14.7	2.97	11.91	0.25	8
Carbon Dioxide (estimated as the amount of methane	produced)	10.0	2.02	1.99	1.02	8
Carbon Brosado (commutos do trio amount of motivario	p.oaaooa)			eptor Demand (lb.)	1.28	
				Stoichiometric	Hydrogen	
P. Calid Bhase Native Electron Assentare		Concentration	Mass	demand	Demand	Electron
B. Solid-Phase Native Electron Acceptors						Equivalents p
(Based on manganese and iron produced)	(mg/L)	(lb)	(wt/wt h ₂)	(lb)	Mole	
Manganese (IV) (estimated as the amount of Mn (II) p		0.7	0.89	27.25	0.03	2
Iron (III) (estimated as the amount of Fe (II) produced)		10.0	12.19	55.41	0.22	1
	Sol	id-Phase Compet	ing Electron Acc	eptor Demand (lb.)	0.25	
				Stoichiometric	Hydrogen	Electron
C. Soluble Contaminant Electron Acceptors		Concentration	Mass	demand	Demand	Equivalents p
		(mg/L)	(lb)	(wt/wt h ₂)	(lb)	Mole
Tetrachloroethene (PCE)		0.041	0.01	20.57	0.00	8
Trichloroethene (TCE)		0.020	0.00	21.73	0.00	6
Dichloroethene (cis-DCE, trans-DCE, and 1,1-DCE)		0.224	0.05	24.05	0.00	4
Vinyl Chloride (VC)		0.054	0.01	31.00	0.00	2
Carbon Tetrachloride (CT)		0.000	0.00	19.08	0.00	8
Trichloromethane (or chloroform) (CF)		0.000	0.00	19.74	0.00	6
Dichloromethane (or methylene chloride) (MC)		0.000	0.00	21.06	0.00	4
Chloromethane		0.000	0.00	25.04	0.00	2
Tetrachloroethane (1,1,1,2-PCA and 1,1,2,2-PCA)		0.000	0.00	20.82	0.00	8
Trichloroethane (1,1,1-TCA and 1,1,2-TCA)		0.003	0.00	22.06	0.00	6
Dichloroethane (1,1-DCA and 1,2-DCA)		0.003	0.00	24.55	0.00	4
Chloroethane		0.000	0.00	32.00	0.00	2
Perchlorate		0.000	0.00	12.33	0.00	6
	Total S			eptor Demand (lb.)	0.00	
				Stoichiometric	Hydrogen	Electron
D. Sorbed Contaminant Electron Acceptors	Koc	Soil Conc.	Mass	demand	Demand	Equivalents
(Soil Concentration = Koc x foc x Cgw)	(mL/g)	(mg/kg)	(lb)	(wt/wt h ₂)	(lb)	Mole
Tetrachloroethene (PCE)	263	0.03	0.04	20.57	0.00	8
Trichloroethene (TCE)	107	0.03	0.01	21.73	0.00	6
Dichloroethene (cis-DCE, trans-DCE, and 1,1-DCE)	45	0.03	0.03	24.05	0.00	4
Vinyl Chloride (VC)	3.0	0.00	0.00	31.00	0.00	2
Carbon Tetrachloride (CT)	224	0.00	0.00	19.08	0.00	8
Trichloromethane (or chloroform) (CF)	63	0.00	0.00	19.74	0.00	6
	28	0.00	0.00	21.06		
Dichloromethane (or methylene chloride) (MC)					0.00	4
Chloromethane	25	0.00	0.00	25.04	0.00	2
Tetrachloroethane (1,1,1,2-PCA and 1,1,2,2-PCA)	117	0.00	0.00	20.82	0.00	8
Trichloroethane (1,1,1-TCA and 1,1,2-TCA)	105	0.00	0.00	22.06	0.00	6
Dichloroethane (1,1-DCA and 1,2-DCA)	30	0.00	0.00	24.55	0.00	4
Chloroethane	3	0.00	0.00	32.00	0.00	2
		0.00	0.00	12.33	0.00	6
Perchlorate	0.0			eptor Demand (lb.)	0.00	

Table S.2 Substrate	Calculations in	Hydrogen I	Equivalents					
4. Treatment Cell Electron-Acceptor Flux (per year)								
			Stoichiometric	Hydrogen	Electron			
A. Soluble Native Electron Acceptors	Concentration	Mass	demand	Demand	Equivalents per			
	(mg/L)	(lb)	(wt/wt h ₂)	(lb)	Mole			
Oxygen	0.5	0.49	7.94	0.06	4			
Nitrate (denitrification)	0.1	0.05	10.25	0.00	5			
Sulfate	14.7	14.94	11.91	1.25	8			
Carbon Dioxide (estimated as the amount of Methane produced)	10	10.16	1.99	5.11	8			
	Total Competing Elec	ctron Acceptor D	emand Flux (lb/yr)	6.4				
			Stoichiometric	Hydrogen	Electron			
B. Soluble Contaminant Electron Acceptors	Concentration	Mass	demand	Demand	Equivalents per			
'	(mg/L)	(lb)	(wt/wt h ₂)	(lb)	Mole			
Tetrachloroethene (PCE)	0.041	0.04	20.57	0.00	8			
Trichloroethene (TCE)	0.020	0.04	21.73	0.00	6			
Dichloroethene (cis-DCE, trans-DCE, and 1,1-DCE)	0.224	0.23	24.05	0.01	4			
Vinyl Chloride (VC)	0.054	0.05	31.00	0.00	2			
Carbon Tetrachloride (CT)	0.000	0.00	19.08	0.00	8			
Trichloromethane (or chloroform) (CF)	0.000	0.00	19.74	0.00	6			
Dichloromethane (or methylene chloride) (MC)	0.000	0.00	21.06	0.00	4			
Chloromethane	0.000	0.00	25.04	0.00	2			
Tetrachloroethane (1,1,1,2-PCA and 1,1,2,2-PCA)	0.000	0.00	20.82	0.00	8			
Trichloroethane (1,1,1-TCA and 1,1,2-TCA)	0.003	0.00	22.06	0.00	6			
Dichloroethane (1,1-DCA and 1,2-DCA)	0.003	0.00	24.55	0.00	4			
Chloroethane	0.000	0.00	32.00	0.00	2			
Perchlorate	0.000	0.00	12.33	0.00	6			
Total Sol	uble Contaminant Elec	ctron Acceptor D	emand Flux (lb/yr)	0.01				
	Initial Hydroge	n Poquiromon	t First Voor (lb)	8.0	Ī			
			equirement (lb)		_			
F. Dooign Footors	Total Life-Cycli	e riyarogen ix	equirement (ib)	0.0	_			
5. Design Factors				01/ 41/				
Microbial Efficiency Uncertainty Factor				2X - 4X 2X - 4X				
Methane and Solid-Phase Electron Acceptor Uncertainty								
Remedial Design Factor (e.g., Substrate Leaving Reaction Zone)				1X - 3X	1			
			Design Factor	10.0				
· · · · · · · · · · · · · · · · · · ·	Hydrogen Require	ment with De	sign Factor (lb)	79.8				
6. Acronyns and Abbreviations								
0-								
	= milliequivalents per 10	00 grams						
μs/cm = microsiemens per centimeter mg/kg = milligrams per kilogram								
cm/day = centimeters per day mg/L = milligrams per liter								
cm/sec = centimeters per second m/m = meters per meters								
ft ² = square feet mV = millivolts								
	ft/day = feet per day m/yr = meters per year							
ft/ft = foot per foot su = standa								
	concetration molecular h	nydrogen, weight i	per weight					
gm/cm³ = grams per cubic centimeter								
kg of CaCO3 per mg = kilograms of calcium carbonate per milligra	am							
lb = pounds								

Table S.3

Hydrogen Produced by Fermentation Reactions of Common Substrates

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Substrate	Molecular Formula	Substrate Molecular Weight (gm/mole)	Moles of Hydrogen Produced per Mole of Substrate	Ratio of Hydrogen Produced to Substrate (gm/gm)	Range of Moles H ₂ /Mole Substrate
Lactic Acid	C ₃ H ₆ O ₃	90.1	2	0.0448	2 to 3
Molasses (assuming 100% sucrose)	C ₁₂ H ₂₂ O ₁₁	342	8	0.0471	8 to 11
High Fructose Corn Syrup (assuming 50% fructose and 50% glucose)	C ₆ H ₁₂ O ₆	180	4	0.0448	4 to 6
Ethanol	C ₂ H ₆ O	46.1	2	0.0875	2 to 6
Whey (assuming 100% lactose)	C ₁₂ H ₂₂ O ₁₁	342	11	0.0648	11
HRC [®] (assumes 40% lactic acid and 40% glycerol by weight)	C ₃₉ H ₅₆ O ₃₉	956	28	0.0590	28
Linoleic Acid (Soybean Oil, Corn Oil, Cotton Oil)	C ₁₈ H ₃₂ O ₂	281	16	0.1150	16

Table S.4 Estimated Substrate Requirements for Hydrogen Demand in Table S.3

Design Life (years): 1

Substrate	Design Factor	Pure Substrate Mass Required to Fulfill Hydrogen Demand (pounds)	Substrate Product Required to Fulfill Hydrogen Demand (pounds)	Substrate Mass Required to Fulfill Hydrogen Demand (milligrams)	Effective Substrate Concentration (mg/L)
Lactic Acid	10.0	1,783	1,783	8.09E+08	1,463
Sodium Lactate Product (60 percent solution)	10.0	1,783	3,699	8.09E+08	1,463
Molasses (assuming 6 0	10.0	1,694	2,823	7.68E+08	1,390
HFCS (assuming 40% fructose and 40% glucose by weight)	10.0	1,783	2,229	8.09E+08	1,463
Ethanol Product (assuming 80% ethanol by weight)	10.0	912	1,140	4.14E+08	748
Whey (assuming 100% lactose)	10.0	1,231	1,758	5.58E+08	1,010
HRC® (assumes 40% lactic acid and 40% glycerol by weight)	10.0	1,352	1,352	6.13E+08	887
Linoleic Acid (Soybean Oil, Corn Oil, Cotton Oil)	10.0	694	694	3.15E+08	569
Commercial Vegetable Oil Emulsion Product (60% oil by weight)	10.0	694	1,157	3.15E+08	569

NOTES: Sodium Lactate Product

- 1. Assumes sodium lactate product is 60 percent sodium lactate by weight.
- 2. Molecular weight of sodium lactate (CH_3 -CHOH-COONa) = 112.06.
- 3. Molecular weight of lactic Acid $(C_6H_6O_3) = 90.08$.
- 4. Therefore, sodium lactate product yields 48.4 (0.60 x (90.08/112.06)) percent by weight lactic acid.
- 5. Weight of sodium lactate product = 11.0 pounds per gallon.
- 6. Pounds per gallon of lactic acid in product = 1.323×8.33 lb/gal H2O x $0.60 \times (90.08/112.06) = 5.31$ lb/gal.

NOTES: Standard HRC Product

- 1. Assumes HRC product is 40 percent lactic acid and 40 percent glycerol by weight.
- 2. HRC® weighs approximately 9.18 pounds per gallon.

NOTES: Vegetable Oil Emulsion Product

- 1. Assumes emulsion product is 60 percent soybean oil by weight.
- 2. Soybean oil is 7.8 pounds per gallon.
- 3. Assumes specific gravity of emulsion product is 0.96.

Table S.5 Output for Substrate Requirements in Hydrogen Equivalents

Site Name: OU7 Pit 2 TA-09 INJ-163, INJ-164, INJ-165, INJ-207

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1. Treatment Zone Physical Dimensions

Width (perpendicular to groundwater flow) Length (parallel to groundwater flow) Saturated Thickness Design Period of Performance

Values	Un
40	fee
30	fee
10	fee
1	yea

its	Values
t	12
t	9.1
t	3.0
irs	1

Units meters meters meters years

2. Treatment Zone Hydrogeologic Properties

Total Porosity
Effective Porosity
Average Aquifer Hydraulic Conductivity
Average Hydraulic Gradient
Average Groundwater Seepage Velocity
Average Groundwater Seepage Velocity
Effective Treatment Zone Pore Volume
Groundwater Flux (per year)
Total Groundwater Volume Treated
(over entire design period)

Values
0.36
0.27
5
0.0223
0.41
151
24,242
121,799
146,041

Hydrogen

89

148

percent
percent
ft/day
ft/ft
ft/day
ft/yr
gallons
gallons/yea
gallons tota

Effective

Units

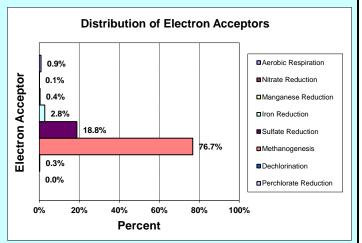
Values	Units
0.36	percent
0.27	percent
1.8E-03	cm/sec
0.0223	m/m
1.3E+01	cm/day
45.9	m/yr
91,762	liters
461,048	liters/year
552,810	liters total

3. Distribution of Electron Acceptor Demand

Aerobic Respiration
Nitrate Reduction
Sulfate Reduction
Manganese Reduction
Iron Reduction
Methanogenesis
Dechlorination
Perchlorate Reduction

	riyarogen
Percent of Total	Demand (lb)
0.9%	0.074
0.1%	0.006
18.8%	1.504
0.4%	0.032
2.8%	0.220
76.7%	6.124
0.3%	0.021
0.0%	0.000
100 00%	7 98

Hydrogen demand in pounds/gallon:	5.46E-05
Hydrogen demand in grams per liter:	6.55E-03



4. Substrate Equivalents: Design Factor = 10.0

Totals:

Product	Quantity (lb)	Quantity (gallons)
Sodium Lactate Product	3,699	336
2. Molasses Product	2,823	235
3. Fructose Product	2,229	199
4. Ethanol Product	1,140	165
5. Sweet Dry Whey (lactose)	1,758	sold by pound
6. HRC®	1,352	sold by pound

LifeCtive	
Concentration	Effective concentration is for total
(mg/L)	volume of groundwater treated.
1,463	as lactic acid
1,390	as sucrose
1,463	as fructose
748	as ethanol
1,010	as lactose
887	as 40% lactic acid/40% glycerol
569	as soybean oil
569	as soybean oil

Notes

7. Linoleic Acid (Soybean Oil)

8. Emulsified Vegetable Oil

- 1. Quantity assumes product is 60% sodium lactate by weight.
- 2. Quantity assumes product is 60% sucrose by weight and weighs 12 pounds per gallon.

694

1,157

- 3. Quantity assumes product is 80% fructose by weight and weighs 11.2 pounds per gallon.
- 4. Quantity assumes product is 80% ethanol by weight and weighs 6.9 pounds per gallon.
- 5. Quantity assumes product is 70% lactose by weight.
- 6. Quantity assumes HRC $\!\!\!\! ^{\odot}$ is 40% lactic acid and 40% glycerol by weight.
- 7. Quantity of neat soybean oil, corn oil, or canola oil.
- 8. Quantity assumes commercial product is 60% soybean oil by weight.

B.8 TA-11 Treatment Area Design

Site Name: OU7 Pit 3 TA	\-11 MW-109, M			RETURN TO COVER PAGE
Treatment Zone Physical Dimensions	NOTE: Unshade Values	d boxes are use Range	r input. Units	User Notes
Vidth (Perpendicular to predominant groundwater flow direction)	40	1-10,000	feet	Pit 3, 9 DPT Points
ength (Parallel to predominant groundwater flow)	34	1-1,000	feet	
aturated Thickness	14	1-100	feet	11-25
reatment Zone Cross Sectional Area	560		ft ²	
Treatment Zone Volume	19,040		ft ³	
Treatment Zone Total Pore Volume (total volume x total porosity)	51,285		gallons	
Treatment Zone Effective Pore Volume (total volume x effective porosity)	38,463		gallons	
Design Period of Performance Design Factor (times the electron acceptor hydrogen demand)	3.0 10.0	.5 to 5 2 to 20	year unitless	
Design Factor (times the electron acceptor hydrogen demand)	10.0	2 10 20	uniness	_
Treatment Zone Hydrogeologic Properties				
Total Porosity	36%	.05-50	percent	PDI Eastover OU 7
Effective Porosity	27%	.05-50	percent	PDI Eastover OU 7, average based on TP, AP, clay conter
Average Aquifer Hydraulic Conductivity	5	.01-1000	ft/day	OU7 PDI
Average Hydraulic Gradient	1.80E-02	0.0001-0.1	ft/ft	Flow path in injection area 03/2024
Average Groundwater Seepage Velocity through the Treatment Zone	0.33		ft/day	
Average Groundwater Seepage Velocity through the Treatment Zone Average Groundwater Discharge through the Treatment Zone	121.8 137,792		ft/yr gallons/year	
Soil Bulk Density	1.63	1.4-2.0	galloris/year gm/cm ³	Average OU- PDI for Eastover and Aquia
Soil Fraction Organic Carbon (foc)	0.28%	0.01-10	percent	PDI-38 Eastover Saturated Zone OU-7 x 2
Native Electron Acceptors				
A. Aqueous-Phase Native Electron Acceptors Oxygen	0.4	0.01 to 10	mg/L	MW-97 03/2024
Vitrate	0.05	0.01 to 10	mg/L	MW-97 03/2024 MW-97 03/2024
Sulfate	2	10 to 5,000	mg/L	MW-95 03/2024
Carbon Dioxide (estimated as the amount of Methane produced)	10.0	0.1 to 20	mg/L	Estimate based on previous EISB injections
· · · · · · · · · · · · · · · · · · ·				
B. Solid-Phase Native Electron Acceptors	_			
Manganese (IV) (estimated as the amount of Mn (II) produced)	1	0.1 to 20	mg/L	MW-95 03/2024
ron (III) (estimated as the amount of Fe (II) produced)	10	0.1 to 20	mg/L	Estimated
Contaminant Electron Acceptors				
Tetrachloroethene (PCE)	0.000		mg/L	Not detected
Trichloroethene (TCE)	0.001		mg/L	MW-109 03/2024
Dichloroethene (cis-DCE, trans-DCE, and 1,1-DCE)	2.404		mg/L	MW-109 03/2024
Vinyl Chloride (VC)	0.560		mg/L	MW-109 03/2024
Carbon Tetrachloride (CT)	0.000		mg/L	Not detected
Trichloromethane (or chloroform) (CF)	0.000		mg/L	Not detected
Dichloromethane (or methylene chloride) (MC)	0.000		mg/L	Not detected
Chloromethane	0.000		mg/L	Not detected
Tetrachloroethane (1,1,1,2-PCA and 1,1,2,2-PCA)	0.000		mg/L	Not detected
Frichloroethane (1,1,1-TCA and 1,1,2-TCA) Dichloroethane (1,1-DCA and 1,2-DCA)	0.000		mg/L	Not detected
Chloroethane (1,1-DCA and 1,2-DCA)	0.000		mg/L mg/L	Not detected Not detected
Perchlorate	0.000		mg/L	No data
Cioniorate	0.000	J	mg/L	110 data
Aquifer Geochemistry (Optional Screening Parameters)				
A. Aqueous Geochemistry	1			
Oxidation-Reduction Potential (ORP)	-56	-400 to +500	mV	MW-98 03/2024
Temperature	19	5.0 to 30	°C	MW-98 03/2024
DH Allestinites	6.4	4.0 to 10.0	su	MW-98 03/2024
Alkalinity Fotal Dissolved Solids (TDS, or salinity)	72	10 to 1,000 10 to 1,000	mg/L mg/L	MW-98 03/2024
Specific Conductivity	100 269	10 to 1,000 100 to 10,000	•	Estimated MW-98 03/2024
Specific Conductivity Chloride	269	10 to 10,000	mg/L	MW-98 03/2024 MW-98 03/2024
Sulfide - Pre injection	0.1	0.1 to 100	mg/L	Estimate
Sulfide - Post injection	0.2	0.1 to 100	mg/L	Estimate
,			Ŭ	
B. Aquifer Matrix	1			
Total Iron	18000	200 to 20,000		CSM 2006 Soil Eastover
Cation Exchange Capacity Neutralization Potential	1.0%	1.0 to 10 1.0 to 100	meq/100 g	CSM 2006 Soil Eastover CSM 2006 Soil Eastover
YOUR GILLARIOT FURTILIAI	1.070	1.0 10 100	. 5.55 45 54503	OCIVI 2000 SUII LASIUVEI

10	Oubstrate Ot	alculations ii	Hydrogen I	_quivalents		
Site Name: OU7 Pit 3 TA-11 MW-109, MW-108, MW-97 RETURN TO COVER PAGE						
				NOTE: Open cells	are user input.	
. Treatment Zone Physical Dimensions				Values	Range	Units
Width (Perpendicular to predominant groundwater to	flow direction)			40	1-10,000	feet
Length (Parallel to predominant groundwater flow)				34	1-1,000	feet
Saturated Thickness				14	1-100	feet
Treatment Zone Cross Sectional Area				560		ft ²
Treatment Zone Volume				19.040		ft ³
Treatment Zone Effective Pore Volume (total volume x effective porosity)			38,463		gallons	
Design Period of Performance	,	<i>37</i>		3.0	.5 to 5	year
· ·						,
. Treatment Zone Hydrogeologic Proper	ties					
Total Porosity				0.36	.05-50	
Effective Porosity				0.27	.05-50	
Average Aquifer Hydraulic Conductivity				5	.01-1000	ft/day
Average Hydraulic Gradient				0.01802	0.1-0.0001	ft/ft
Average Groundwater Seepage Velocity through th				0.33		ft/day
Average Groundwater Seepage Velocity through th				121.8		ft/yr
Average Groundwater Flux through the Treatment 2	Zor (0		137,792		gallons/year
Soil Bulk Density				1.63	1.4-2.0	gm/cm ³
Soil Fraction Organic Carbon (foc)				0.0028	0.0001-0.1	
. Initial Treatment Cell Electron-Accepto	r Demand (one	total pore volu	me)			
	·			Stoichiometric	Hydrogen	Electron
A. Aqueous-Phase Native Electron Acceptors		Concentration	Mass	demand	Demand	Equivalents
,		(mg/L)	(lb)	(wt/wt h ₂)	(lb)	Mole
Ovygon		0.4	0.12	7.94	0.02	4
Oxygen Nitrate (denitrification)		0.4	0.12	12.30	0.02	5
Sulfate		1.7	0.55	11.91	0.05	8
Carbon Dioxide (estimated as the amount of metha	ne produced)	10.0	3.21	1.99	1.61	8
Oarbort bloxide (estimated as the amount of metha	ne produced)			eptor Demand (lb.)	1.68	
		Colubic Compet	ing Electron Acco	Stoichiometric		
P. Calid Dhaga Native Floatron Assentare		Concentration	Mass	demand	Hydrogen	Electron
B. Solid-Phase Native Electron Acceptors			Mass		Demand	Equivalents p
(Based on manganese and iron produced)		(mg/L)	(lb)	(wt/wt h ₂)	(lb)	Mole
Manganese (IV) (estimated as the amount of Mn (II) produced)		0.7	2.71	27.25	0.10	2
Iron (III) (estimated as the amount of Fe (II) produce		10.0	37.70	55.41	0.68	1
	Sol	lid-Phase Compet	ing Electron Acc	eptor Demand (lb.)	0.78	
				Stoichiometric	Hydrogen	Electron
C. Soluble Contaminant Electron Acceptors		Concentration	Mass	demand	Demand	Equivalents p
		(mg/L)	(lb)	(wt/wt h ₂)	(lb)	Mole
Tetrachloroethene (PCE)		0.000	0.00	20.57	0.00	8
Trichloroethene (TCE)		0.001	0.00	21.73	0.00	6
Dichloroethene (cis-DCE, trans-DCE, and 1,1-DCE)	2.404	0.77	24.05	0.03	4
Vinyl Chloride (VC)	,	0.560	0.18	31.00	0.01	2
Carbon Tetrachloride (CT)		0.000	0.00	19.08	0.00	8
Trichloromethane (or chloroform) (CF)		0.000	0.00	19.74	0.00	6
Dichloromethane (or methylene chloride) (MC)		0.000	0.00	21.06	0.00	4
Chloromethane		0.000	0.00	25.04	0.00	2
Tetrachloroethane (1,1,1,2-PCA and 1,1,2,2-PCA)		0.000	0.00	20.82	0.00	8
Trichloroethane (1,1,1-TCA and 1,1,2-TCA)		0.000	0.00	22.06	0.00	6
Dichloroethane (1,1-DCA and 1,2-DCA)		0.000	0.00	24.55	0.00	4
Chloroethane		0.000	0.00	32.00	0.00	2
Perchlorate		0.000	0.00	12.33	0.00	6
	Total S		ant Electron Acc	eptor Demand (lb.)	0.04	
				Stoichiometric	Hydrogen	Electron
D. Sorbed Contaminant Electron Acceptors	Koc	Soil Conc.	Mass	demand	Demand	Equivalents
(Soil Concentration = Koc x foc x Cgw)	(mL/g)	(mg/kg)	(lb)	(wt/wt h ₂)	(lb)	Mole
Tetrachloroethene (PCE)	263	0.00	0.00	20.57	0.00	8
Trichloroethene (TCE)	107	0.00	0.00	21.73	0.00	6
Dichloroethene (cis-DCE, trans-DCE, and 1,1-DCE		0.30	0.59	24.05	0.02	4
Vinyl Chloride (VC)	3.0	0.00	0.01	31.00	0.02	2
Carbon Tetrachloride (CT)	224	0.00	0.00	19.08	0.00	8
Trichloromethane (or chloroform) (CF)	63	0.00	0.00	19.74	0.00	6
Dichloromethane (or methylene chloride) (MC)		0.00	0.00			_
, , ,	28			21.06	0.00	4
Chloromethane Tetraphloroethane (1.1.1.2 PCA and 1.1.2.3 PCA)	25	0.00	0.00	25.04	0.00	2
Tetrachloroethane (1,1,1,2-PCA and 1,1,2,2-PCA)	117	0.00	0.00	20.82	0.00	8
Trichloroethane (1,1,1-TCA and 1,1,2-TCA)	105	0.00	0.00	22.06	0.00	6
Dichloroethane (1,1-DCA and 1,2-DCA)	30	0.00	0.00	24.55	0.00	4
Chloroethane	3	0.00	0.00	32.00	0.00	2
	0.0	0.00	0.00	12.33	0.00	6
Perchlorate				eptor Demand (lb.)	0.02	_

Table S.2	Substrate Ca	Iculations in	Hydrogen I	Equivalents		
4. Treatment Cell Electron-Acceptor Flux (•		
·	. ,			Stoichiometric	Hydrogen	Electron
A. Soluble Native Electron Acceptors		Concentration	Mass	demand	Demand	Equivalents per
		(mg/L)	(lb)	(wt/wt h ₂)	(lb)	Mole
Oxygen		0.4	0.44	7.94	0.06	4
Nitrate (denitrification)		0.1	0.06	10.25	0.01	5
Sulfate		1.7	1.95	11.91	0.16	8
Carbon Dioxide (estimated as the amount of Methan		10	11.50	1.99	5.78	8
Total Competing Electron Acceptor Demand Flux (lb/yr) 6.				6.0		
				Stoichiometric	Hydrogen	Electron
B. Soluble Contaminant Electron Acceptors		Concentration	Mass	demand	Demand	Equivalents per
		(mg/L)	(lb)	(wt/wt h ₂)	(lb)	Mole
Tetrachloroethene (PCE)		0.000	0.00	20.57	0.00	8
Trichloroethene (TCE)		0.001	0.00	21.73	0.00	6
Dichloroethene (cis-DCE, trans-DCE, and 1,1-DCE)		2.404	2.76	24.05	0.11	4
Vinyl Chloride (VC)		0.560	0.64	31.00	0.02	2
Carbon Tetrachloride (CT)		0.000	0.00	19.08	0.00	8
Trichloromethane (or chloroform) (CF)		0.000	0.00	19.74	0.00	6
Dichloromethane (or methylene chloride) (MC)		0.000	0.00	21.06	0.00	4
Chloromethane		0.000	0.00	25.04	0.00	2
Tetrachloroethane (1,1,1,2-PCA and 1,1,2,2-PCA)		0.000	0.00	20.82	0.00	8
Trichloroethane (1,1,1-TCA and 1,1,2-TCA)		0.000	0.00	22.06	0.00	6
Dichloroethane (1,1-DCA and 1,2-DCA)		0.000	0.00	24.55	0.00	4
Chloroethane		0.000	0.00	32.00	0.00	2
Perchlorate	Tatal Calable	0.000	0.00	12.33	0.00	6
Total Soluble Contaminant Electron Acceptor Demand Flux (lb/yr) 0.14						<u>]</u> =
				t First Year (lb)		
	Т	Total Life-Cycle	e Hydrogen R	equirement (lb)	20.9	
5. Design Factors						
Microbial Efficiency Uncertainty Factor					2X - 4X	
Methane and Solid-Phase Electron Acceptor Uncertain	nty				2X - 4X	
Remedial Design Factor (e.g., Substrate Leaving Read	ction Zone)				1X - 3X	_
				Design Factor	10.0	
Total	Life-Cycle Hyd	rogen Require	ment with De	sign Factor (lb)	209.3	
6. Acronyns and Abbreviations						_
°C =degrees celsius		equivalents per 10	00 grams			
μs/cm = microsiemens per centimeter	mg/kg = milligram					
cm/day = centimeters per day mg/L = milligrams per liter						
cm/sec = centimeters per second m/m = meters per meters						
ft ² = square feet mV = millivolts						
	ft/day = feet per day m/yr = meters per year					
	tt/ft = foot per foot su = standard pH units tt/yr = feet per year wt/wt H2 = concetration molecular hydrogen, weight per weight					
ft/yr = feet per year	wt/wt H2 = conce	tration molecular h	iyarogen, weight p	per weight		
gm/cm³ = grams per cubic centimeter						
kg of CaCO3 per mg = kilograms of calcium carbona	ate per milligram					
lb = pounds						

Table S.3

Hydrogen Produced by Fermentation Reactions of Common Substrates

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Substrate	Molecular Formula	Substrate Molecular Weight (gm/mole)	Moles of Hydrogen Produced per Mole of Substrate	Ratio of Hydrogen Produced to Substrate (gm/gm)	Range of Moles H ₂ /Mole Substrate
Lactic Acid	C ₃ H ₆ O ₃	90.1	2	0.0448	2 to 3
Molasses (assuming 100% sucrose)	C ₁₂ H ₂₂ O ₁₁	342	8	0.0471	8 to 11
High Fructose Corn Syrup (assuming 50% fructose and 50% glucose)	C ₆ H ₁₂ O ₆	180	4	0.0448	4 to 6
Ethanol	C ₂ H ₆ O	46.1	2	0.0875	2 to 6
Whey (assuming 100% lactose)	C ₁₂ H ₂₂ O ₁₁	342	11	0.0648	11
HRC [®] (assumes 40% lactic acid and 40% glycerol by weight)	C ₃₉ H ₅₆ O ₃₉	956	28	0.0590	28
Linoleic Acid (Soybean Oil, Corn Oil, Cotton Oil)	C ₁₈ H ₃₂ O ₂	281	16	0.1150	16

Table S.4 Estimated Substrate Requirements for Hydrogen Demand in Table S.3

Design Life (years): 3

Substrate	Design Factor	Pure Substrate Mass Required to Fulfill Hydrogen Demand (pounds)	Substrate Product Required to Fulfill Hydrogen Demand (pounds)	Substrate Mass Required to Fulfill Hydrogen Demand (milligrams)	Effective Substrate Concentration (mg/L)
Lactic Acid	10.0	4,677	4,677	2.12E+09	1,240
Sodium Lactate Product (60 percent solution)	10.0	4,677	9,703	2.12E+09	1,240
Molasses (assuming 6 0	10.0	4,443	7,405	2.02E+09	1,178
HFCS (assuming 40% fructose and 40% glucose by weight)	10.0	4,678	5,847	2.12E+09	1,241
Ethanol Product (assuming 80% ethanol by weight)	10.0	2,392	2,990	1.08E+09	634
Whey (assuming 100% lactose)	10.0	3,228	4,612	1.46E+09	856
HRC® (assumes 40% lactic acid and 40% glycerol by weight)	10.0	3,545	3,545	1.61E+09	752
Linoleic Acid (Soybean Oil, Corn Oil, Cotton Oil)	10.0	1,820	1,820	8.26E+08	483
Commercial Vegetable Oil Emulsion Product (60% oil by weight)	10.0	1,820	3,034	8.26E+08	483

NOTES: Sodium Lactate Product

- 1. Assumes sodium lactate product is 60 percent sodium lactate by weight.
- 2. Molecular weight of sodium lactate (CH_3 -CHOH-COONa) = 112.06.
- 3. Molecular weight of lactic Acid ($C_6H_6O_3$) = 90.08 .
- 4. Therefore, sodium lactate product yields 48.4 (0.60 x (90.08/112.06)) percent by weight lactic acid.
- 5. Weight of sodium lactate product = 11.0 pounds per gallon.
- 6. Pounds per gallon of lactic acid in product = 1.323×8.33 lb/gal H2O x $0.60 \times (90.08/112.06) = 5.31$ lb/gal.

NOTES: Standard HRC Product

- 1. Assumes HRC product is 40 percent lactic acid and 40 percent glycerol by weight.
- 2. HRC® weighs approximately 9.18 pounds per gallon.

NOTES: Vegetable Oil Emulsion Product

- 1. Assumes emulsion product is 60 percent soybean oil by weight.
- 2. Soybean oil is 7.8 pounds per gallon.
- 3. Assumes specific gravity of emulsion product is 0.96.

Table S.5 Output for Substrate Requirements in Hydrogen Equivalents

Site Name: OU7 Pit 3 TA-11 MW-109, MW-108, MW-97

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1. Treatment Zone Physical Dimensions

Width (perpendicular to groundwater flow) Length (parallel to groundwater flow) Saturated Thickness Design Period of Performance

Values	Units
40	feet
34	feet
14	feet
3	years

Values	Units
12	meters
10.4	meters
4.3	meters
3	years

2. Treatment Zone Hydrogeologic Properties

Total Porosity
Effective Porosity
Average Aquifer Hydraulic Conductivity
Average Hydraulic Gradient
Average Groundwater Seepage Velocity
Average Groundwater Seepage Velocity
Effective Treatment Zone Pore Volume
Groundwater Flux (per year)
Total Groundwater Volume Treated
(over entire design period)

Values
0.36
0.27
5
0.01802
0.33
122
38,463
137,792
451,839

Units
percent
percent
ft/day
ft/ft
ft/day
ft/yr
gallons
gallons/year
gallons total

Effective

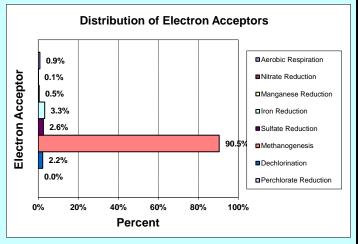
Values	Units
0.36	percent
0.27	percent
1.8E-03	cm/sec
0.01802	m/m
1.0E+01	cm/day
37.1	m/yr
145,596	liters
521,584	liters/year
1,710,348	liters total
	•'

3. Distribution of Electron Acceptor Demand

Aerobic Respiration
Nitrate Reduction
Sulfate Reduction
Manganese Reduction
Iron Reduction
Methanogenesis
Dechlorination
Perchlorate Reduction

пуагоден	
Percent of Total	Demand (lb)
0.9%	0.180
0.1%	0.018
2.6%	0.538
0.5%	0.100
3.3%	0.680
90.5%	18.947
2.2%	0.470
0.0%	0.000
100.00%	20.93

Hydrogen demand in pounds/gallon: 4.63E-05
Hydrogen demand in grams per liter: 5.55E-03



4. Substrate Equivalents: Design Factor = 10.0

Totals:

Product	Quantity (lb)	Quantity (gallons)
Sodium Lactate Product	9,703	882
2. Molasses Product	7,405	617
3. Fructose Product	5,847	522
4. Ethanol Product	2,990	433
5. Sweet Dry Whey (lactose)	4,612	sold by pound
6. HRC®	3,545	sold by pound

Concentration (mg/L)	Effective concentration is for total volume of groundwater treated.
1,240	as lactic acid
1,178	as sucrose
1,241	as fructose
634	as ethanol
856	as lactose
752	as 40% lactic acid/40% glycerol
483	as soybean oil
483	as soybean oil

Notes

7. Linoleic Acid (Soybean Oil)

8. Emulsified Vegetable Oil

- 1. Quantity assumes product is 60% sodium lactate by weight.
- 2. Quantity assumes product is 60% sucrose by weight and weighs 12 pounds per gallon.

1,820

3,034

233

- 3. Quantity assumes product is 80% fructose by weight and weighs 11.2 pounds per gallon.
- 4. Quantity assumes product is 80% ethanol by weight and weighs 6.9 pounds per gallon.
- 5. Quantity assumes product is 70% lactose by weight.
- 6. Quantity assumes HRC $\!\!\!\! ^{\odot}$ is 40% lactic acid and 40% glycerol by weight.
- 7. Quantity of neat soybean oil, corn oil, or canola oil.
- 8. Quantity assumes commercial product is 60% soybean oil by weight.

B.9 Injection Point Volume Calculations

SITE NAME: DSCR Operable Unit 7, TA-02 INJECTION POINT CALCULATIONS Where, h is thickness, r is planned radius, Θ_{M} is mobile porosity, and V_{inj} is injection volume. **POROSITY** 0.35 $V_{inj} = \pi \cdot h \cdot r_{inj}^2 \cdot \theta_M$ 0.02 Θ_{M} per Remediation Engr Design Concepts, 2nd Edition, pp 177 h 16 feet injection interval thickness feet desired radius of influence \mathbf{r}_{inj} 144.7645895 ft³

Delivery volume over the treated interval

Injection volume per foot

gallons

SITE NAME: DSCR Operable Unit 7, TA-03 INJECTION POINT CALCULATIONS Where, h is thickness, r is planned radius, Θ_{M} is mobile porosity, and V_{inj} is injection volume. 0.35 **POROSITY** $V_{inj} = \pi \cdot h \cdot r_{inj}^2 \cdot \theta_M$ 0.02 Θ_{M} per Remediation Engr Design Concepts, 2nd Edition, pp 177 h 15 feet injection interval thickness

desired radius of influence

 \mathbf{r}_{inj} 135.7168026 ft³ gallons Delivery volume over the treated interval 67.7 gallons/ft Injection volume per foot

feet

SITE NAME: DSCR Operable Unit 7, TA-05 INJECTION POINT CALCULATIONS Where, h is thickness, r is planned radius, Θ_{M} is mobile porosity, and V_{inj} is injection volume. 0.35 **POROSITY** $V_{inj} = \pi \cdot h \cdot r_{inj}^2 \cdot \theta_M$ 0.02 Θ_{M} per Remediation Engr Design Concepts, 2nd Edition, pp 177 h 18 feet injection interval thickness feet desired radius of influence \mathbf{r}_{inj} 162.8601632 ft³

Delivery volume over the treated interval

Injection volume per foot

1218.3 gallons

SITE NAME: DSCR Operable Unit 7, TA-06 INJECTION POINT CALCULATIONS Where, h is thickness, r is planned radius, Θ_{M} is mobile porosity, and V_{inj} is injection volume. 0.35 **POROSITY** $V_{inj} = \pi \cdot h \cdot r_{inj}^2 \cdot \theta_M$ 0.02 Θ_{M} per Remediation Engr Design Concepts, 2nd Edition, pp 177 20 h feet injection interval thickness feet desired radius of influence \mathbf{r}_{inj} 180.9557368 ft³ 1353.6 gallons Delivery volume over the treated interval

Injection volume per foot

SITE NAME: DSCR Operable Unit 7, TA-07 INJECTION POINT CALCULATIONS Where, h is thickness, r is planned radius, Θ_{M} is mobile porosity,

and V_{inj} is injection volume.

 $V_{inj} = \pi \cdot h \cdot r_{inj}^2 \cdot \theta_M$ 0.02 Θ_{M}

per Remediation Engr Design Concepts, 2nd Edition, pp 177

0.35

POROSITY

12 h feet injection interval thickness

feet desired radius of influence \mathbf{r}_{inj}

108.5734421 ft³

gallons Delivery volume over the treated interval 67.7 gallons/ft

Injection volume per foot

SITE NAME: DSCR Operable Unit 7, TA-08 INJECTION POINT CALCULATIONS Where, h is thickness, r is planned radius, Θ_{M} is mobile porosity, and V_{inj} is injection volume. 0.35 **POROSITY** $V_{inj} = \pi \cdot h \cdot r_{inj}^2 \cdot \theta_M$ 0.02 Θ_{M} per Remediation Engr Design Concepts, 2nd Edition, pp 177 h 11 feet injection interval thickness feet desired radius of influence \mathbf{r}_{inj} 99.52565527 ft³

Delivery volume over the treated interval

Injection volume per foot

gallons

SITE NAME: DSCR Operable Unit 7, TA-09 INJECTION POINT CALCULATIONS Where, h is thickness, r is planned radius, Θ_{M} is mobile porosity, and V_{inj} is injection volume. 0.35 **POROSITY** $V_{inj} = \pi \cdot h \cdot r_{inj}^2 \cdot \theta_M$ 0.02 Θ_{M} per Remediation Engr Design Concepts, 2nd Edition, pp 177 10 h feet injection interval thickness feet desired radius of influence \mathbf{r}_{inj} 90.47786842 ft³

Delivery volume over the treated interval

Injection volume per foot

gallons

SITE NAME: DSCR Operable Unit 7, TA-11 INJECTION POINT CALCULATIONS Where, h is thickness, r is planned radius, $\Theta_{\rm M}$ is mobile porosity, and V_{inj} is injection volume. POROSITY 0.35 $V_{inj} = \pi \cdot h \cdot r_{inj}^2 \cdot \theta_{\rm M}$ $\Theta_{\rm M} = 0.02$ per Remediation Engr Design Concepts, 2nd Edition, pp 177

injection interval thickness

Injection volume per foot

r_{inj} 12 feet desired radius of influence

/_{inj} 126.6690158 ft³

947.5 gallons Delivery volume over the treated interval

h

14

feet

Appendix C Regulatory Correspondence



UNITED STATES ENVIRONMENTAL PROTECTION AGENCY Wheeling Field Office 1060 Chapline Street Wheeling, West Virginia 26003

March 7, 2013

Ms. Susan Trussell U.S. Army Corps of Engineers, Tulsa District CESWT-EC-EE 1645 S. 101st E. Avenue Tulsa, Oklahoma 74128-4609

Re: Defense Supply Center Richmond Operable Unit 7 Richmond, Chesterfield County, VA

Dear Ms. Trussell:

We have received correspondence from your consultant, AECOM Technical Services, dated February 28, 2013 describing ground water remediation efforts at the above referenced facility. We understand that you intend to utilize enhanced in-situ bioremediation processes, involving shallow injection wells, to reduce volatile organic compound (VOC) contamination. The subsurface emplacement of fluids through injection wells as part of an aquifer cleanup project is subject to the ground water protection requirements of the Underground Injection Control (UIC) program. The UIC program is administered by the Environmental Protection Agency (EPA) in the Commonwealth of Virginia.

Based upon our understanding of aquifer remediation of contaminated soil and ground water in general, and the proposed subsurface emplacement of nutrients to facilitate biodegradation specifically, we believe that the injection wells pose minimal potential to adversely impact ground water. For these reasons you may proceed with plans to construct and operate aquifer remediation related injection wells and you will not be required to obtain an Underground Injection Control (UIC) program permit.

The above referenced facility has been added to our inventory of shallow injection wells. The UIC program prohibits the subsurface emplacement of fluids which have the potential to adversely impact underground sources of drinking water.

EPA approval or "rule authorization" of the injection wells is contingent upon operator compliance with all applicable requirements. We appreciate your cooperation in these matters and the opportunity to address these issues with you. Please contact me at (304) 234-0286 if you have any questions.

Sincerely,

Mark A. Nelson, Hydrologist Water Protection Division

Cc: Manish M. Joshi, AECOM