

Homework Problems for the Complexity Explorer Course on Random Walks

1. **Displacement of a random walk.** Consider the Pearson random walk in any spatial dimension in which the length of each step has the fixed value a , but the direction is arbitrary. Compute the fourth moment of the displacement after N steps

$$M_4 \equiv \left\langle \left(\sum_{i=1}^N \mathbf{x}_i \right)^4 \right\rangle.$$

2. **Probability distribution of a biased random walk.** Consider a random walk in one dimension in which a step to the right occurs with probability p and a step to the left occurs with probability $q = 1 - p$.

- (a) Determine the probability $P(r, \ell, t)$ that a walk of t total steps has taken r steps to the right and ℓ steps to the left.
- (b) Transform the above expression for $P(r, \ell, t)$ to $P(x, t)$, the probability that the walk is at x at time t .
- (c) Use Stirling's approximation to compute the long-time limit of the probability distribution $P(x, t)$.

3. **Diffusion equation.** Consider a random walk that steps to the right with probability $1/3$, to the left with probability $1/3$, and remains in place with probability $1/3$.

- (a) Write the Master equation that determines the evolution of $P(x, t)$.
- (b) Taylor expand the master equation and determine the partial differential equation that is satisfied by $P(x, t)$. Determine the diffusion coefficient D of this process and compare it with the diffusion coefficient of the nearest-neighbor random walk.

4. **Kinetic theory.** The density of a typical room-temperature gas is $\rho \approx 10^{20}$ molecules/cc and each molecule has a typical speed that is roughly 30000 cm/sec. Estimate the number of collisions that the ambient air makes with your body per second.

5. **Extreme value statistics.** The distribution of velocities of an ideal gas of molecules of mass m at temperature T is given by the Maxwell-Boltzmann distribution

$$P(\mathbf{v})d\mathbf{v} = \left(\frac{m}{2\pi kT} \right)^{3/2} e^{-mv^2/2kT} d\mathbf{v},$$

where kT , the thermal energy of each molecule is roughly $\frac{1}{40}$ eV at room temperature (≈ 300 K). Thus $P(\mathbf{v})d\mathbf{v}$ is the probability that a molecule has a velocity that is in a range $d\mathbf{v}$ about \mathbf{v} . Using the reasoning given in the discussion of the failure of the central limit theorem, estimate the energy of the most energetic molecule in a gas at room temperature in a room of volume of 1000 cubic meters.

6. **First-passage in a finite interval.** Consider a biased random walk in the finite interval $[0, L]$ in which the walk hops to the right with probability p and to the left with probability $q = 1 - p$. The walk is immediately absorbed when it reaches either 0 or L .

- (a) Determine the probability E_n that the walk is absorbed at L when it starts at site n .
 - (b) Determine the average time t_n for the walk to exit the interval either at 0 or at L when it starts at site n . Find the starting location that maximizes the exit time.
7. **First-passage on the semi-infinite interval.** Consider a biased nearest-neighbor random walk with hopping probabilities p and q to the right and left, respectively, on the semi-infinite interval $[0, \infty]$. The walk is absorbed if it reaches $x = 0$.
- (a) Using the backward Kolmogorov approach, determine the probability E_n that the walk is eventually absorbed at the origin when it starts at site n . Separately consider the cases $p > q$, $p < q$, and $p = q$ (symmetric walk).
 - (b) For the case where $p < q$ (bias to the left), compute the average time t_n for the walk to reach the origin when it starts at site n . What is the limiting behavior of this exit time in the limit $p = q$?