Near Infra-red Spectroscopy (NIRS) versus Wet Chemistry: Accuracy and Precision in Forage Analysis

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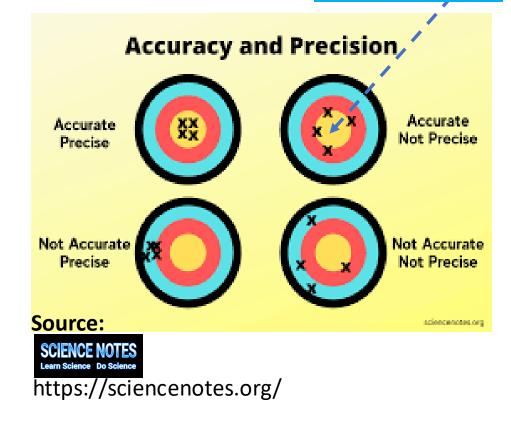
Quality Assurance in Analytical Chemistry

What are two pillars of Quality Assurance?

True or Known Concentration

- Precision of repeated measures
 - Repeatability
 - Reproducibility

Accuracy







Precision and Accuracy in Interlaboratory Collaborative Proficiency Testing

Horwitz, Kamps, and Boyer (1980)

- Examination of the results of over 50 interlaboratory collaborative studies
- Conducted by AOAC
- Various commodities
- Numerous analytes
- Mean coefficient of variation (CV or RSD), expressed as powers of 2
- Mean concentration measured (C), expressed as powers of 10 (i.e., Log₁₀ C)
- Independent of the determinative method





Precision and Accuracy in Interlaboratory Collaborative Proficiency Testing

Statistician Jung Keun Lee expressed Horwitz Equation as:

$$RSD_R$$
, % = $2^{(1-0.5 \log C)}$

C = Concentration of analyte expressed as **mass fraction**

 RSD_R = Relative Standard Deviation or CV under reproducibility conditions.

Michael Thompson (1999) transformed the equation into:

$$RSD_R$$
, % = $2C^{-0.15}$

Or

$$S_R = 2C^{0.85}$$

- HSD, in % on the NFTA Proficiency Test Report = S_R x 100
- C is the RMA (On NFTA PT Reports) expressed as <u>mass fraction</u> (e.g., For 95%, DM; C = 0.95)



Horwitz Equation

 RSD_R , % = 2 (1 - 0.5 Log C) or 2C^{-0.15}

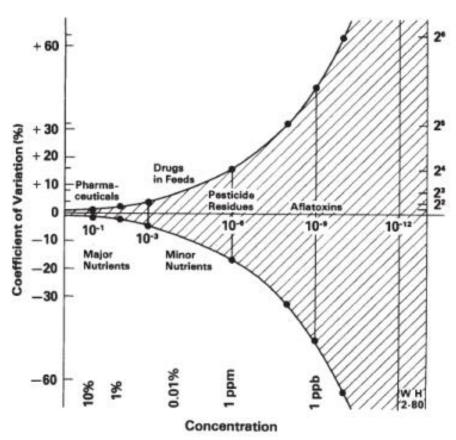


Figure 1. Horwitz Horn, the original curve

Values of Horwitz RSD_R at different concentrations

Analyte concentration		RSD_R
10	%	2.8%
1	%	4.0%
0.1	%	5.7%
0.01	%	8.0%
1	Ppm	16%
1	Ppb	45%
0.1	Ppb	64%







Adjustment of Horwitz equation

 Later, in 2000, Michael Thompson found that precision was overestimated at the extreme low values of C.

 As a result, the Horwitz Equation was further adjusted as:

$$>$$
 S_R = 0.22C; if C is <1.2 x 10⁻⁷(1)

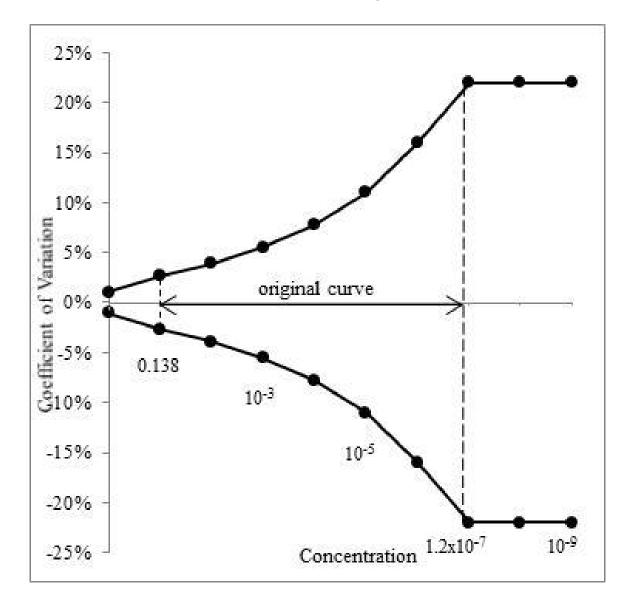
$$>$$
 $S_R = 0.02C^{0.8495}$; if 1.2 x $10^{-7} \le C \le 0.138...$ (2)

$$> S_R = 0.01C^{0.5}; C > 0.138....(3)$$





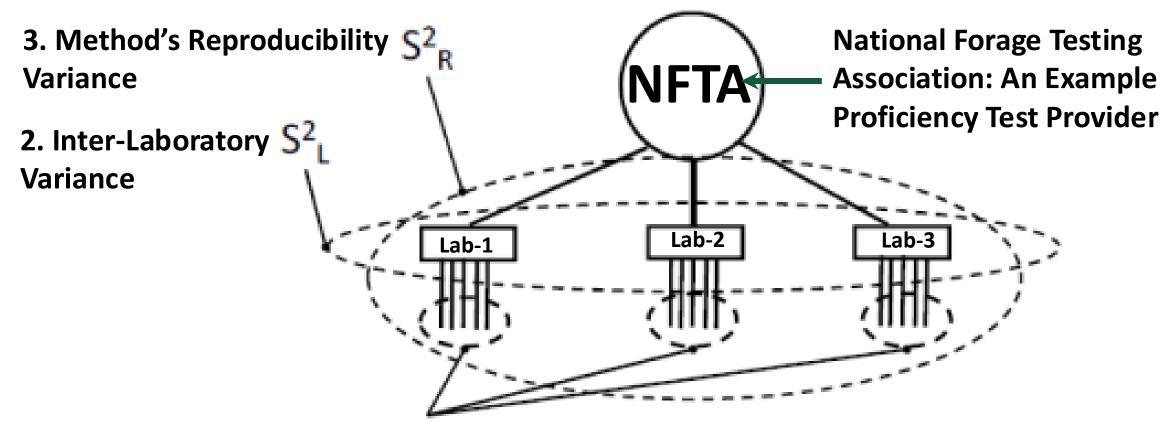
The Horwitz Curve Adjusted for Concentrations







Precision: Repeatability and Reproducibility



1. Intra-Laboratory Variance S²

$$S_R^2 = S_r^2 + S_L^2$$

$$S_{R} = \sqrt{S_{r}^{2} + S_{L}^{2}}$$





The Equations

1)
$$S_r = \sqrt{\frac{(n_1 - 1)s_1^2 + (n_2 - 1)s_2^2 + \dots + (n_k - 1)s_k^2}{n_1 + n_2 + \dots + n_k - k}}$$

2)
$$RSD_r = \sqrt{\frac{(n_1 - 1)RSD_1^2 + (n_2 - 1)RSD_2^2 + ... + (n_k - 1)RSD_k^2}{n_1 + n_2 + ... + n_k - k}}$$

 $RSD_i = (S_i \div Mean_i) \times 100$

3)
$$\mathbf{S_L} = \sqrt{\frac{1}{k-1}} \sum_{i=1}^{k} (Xmean_i - Grand Mean)^2$$

4)
$$S_R = \sqrt{S_r^2 + S_L^2}$$
 5) $RSD_R = \frac{S_R}{Grand Mean} \times 100$



Horwitz Ratios or "HorRat" for Precision

Repeatability
$$HorRat_r = \frac{RSD_r}{PRSD_r}$$

• *Acceptable HorRat*_r: ≤1.3

Reproducibility
$$HorRat_R = \frac{RSD_R}{PRSD_R}$$

- Acceptable **HorRat**_R: ≤2.0
- RSD is calculated from the results reported by the participating labs in replicates
- PRSD is the RSD predicted from Horwitz Equation





Accuracy Using "Z-Score" Based on **Horwitz Function**

$$Z = \frac{\left| \frac{\text{XmeanL} - \mu \left(\text{or RMA} \right)}{S_{\text{Horwitz}} \left(\text{or HSD} \right)} \right|}{S_{\text{Horwitz}} \left(\text{or HSD} \right)}$$

- **RMA** is the Reference Method (i.e., wet chem.) Average
- $ho Xmean_I$ is the mean value of replicated measurements reported by a given laboratory
- \rightarrow HSD = 0.02C^{0.8495}; if 1.2 x 10⁻⁷ \leq C \leq 0.138 (or 13.8%)
- \rightarrow HSD = 0.01C^{0.5}; C > 0.138







This Study

NFTA Proficiency Test Reports from 6 Laboratories



All 6 laboratories provided NIRS results

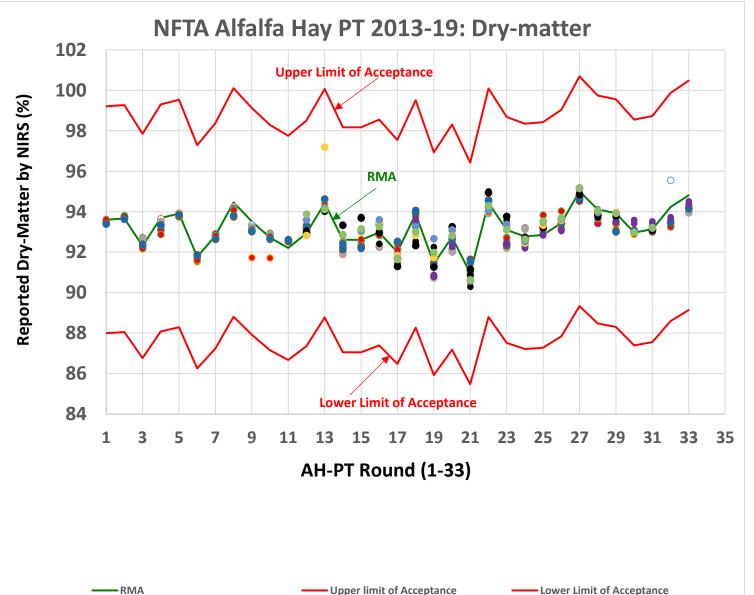


4 laboratories provided both <u>NIRS</u> and <u>Wet Chemistry</u> results

33 NFTA <u>Alfalfa Hay PT rounds during 2013-2019</u>







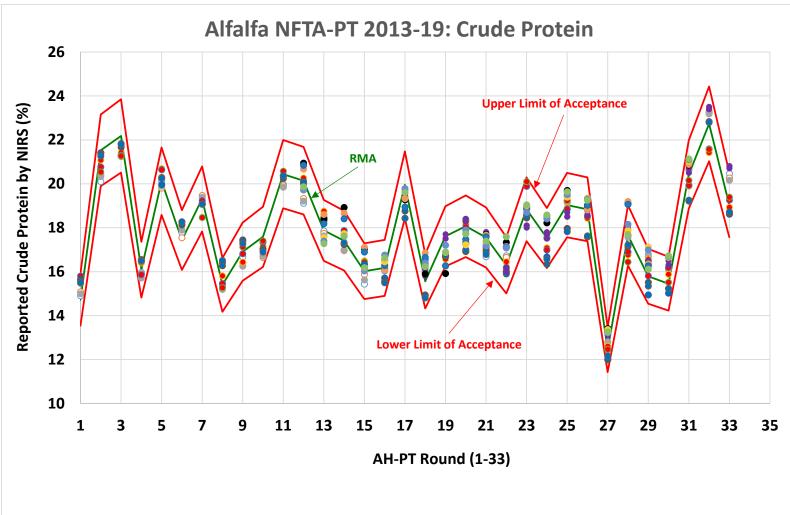
- **6 Laboratories**
- **Triplicate analysis**
- 33 PT rounds

Acceptance **Envelope:**

RMA ± 3HSD







- 6 Laboratories
- Triplicate analysis
- 33 PT rounds

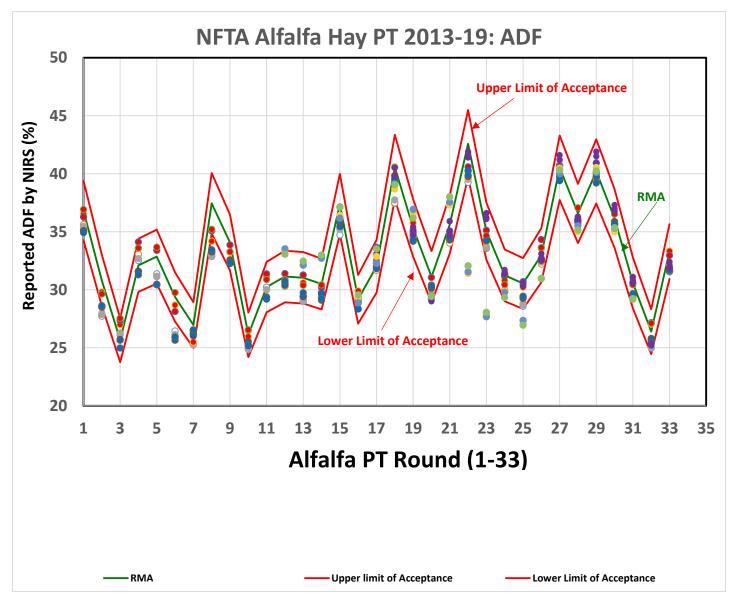
Acceptance Envelope:

RMA ± 3HSD





Lower Limit of Acceptance



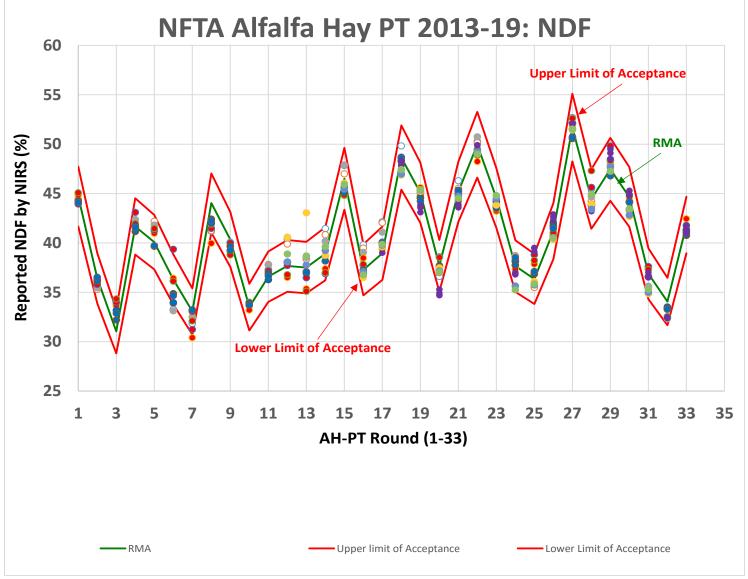
- 6 Laboratories
- Triplicate analysis
- 33 PT rounds

Acceptance Envelope:

RMA ± 3HSD







- 6 Laboratories
- Triplicate analysis
- 33 PT rounds

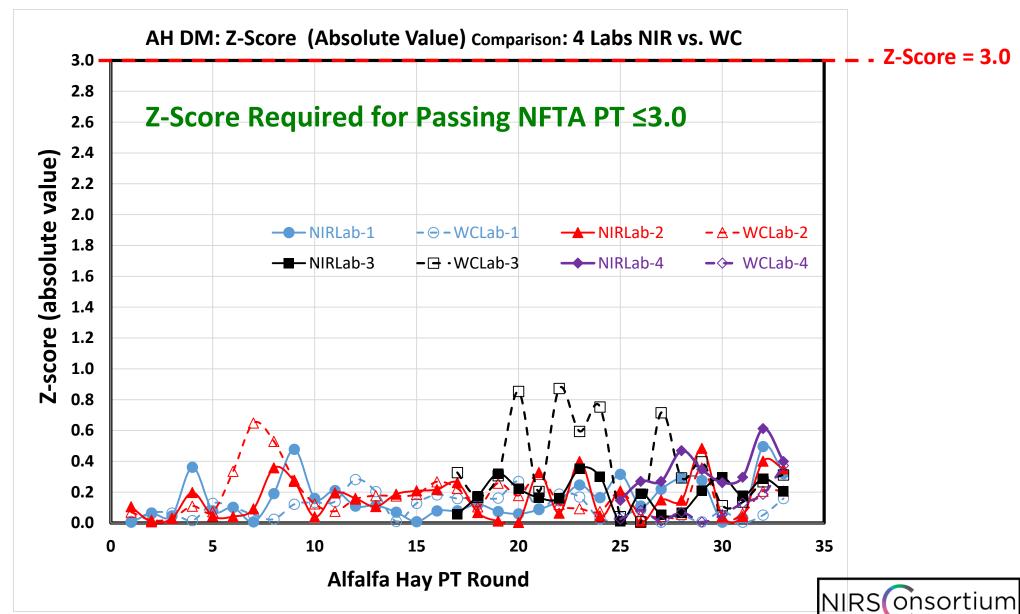
Acceptance Envelope:

RMA ± 3HSD





Forage DM Testing Accuracy: NIRS vs. WC

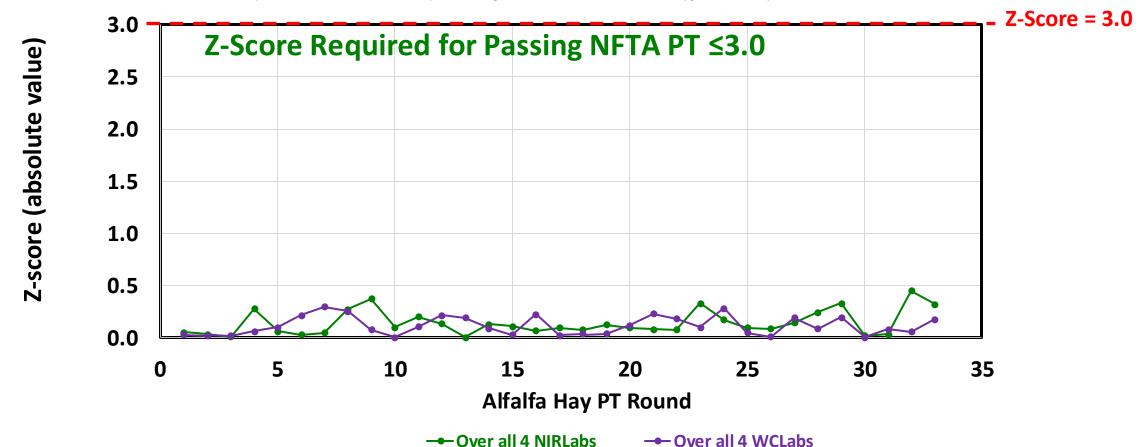


Forage and Feed



Forage DM Testing Accuracy: NIRS vs. WC

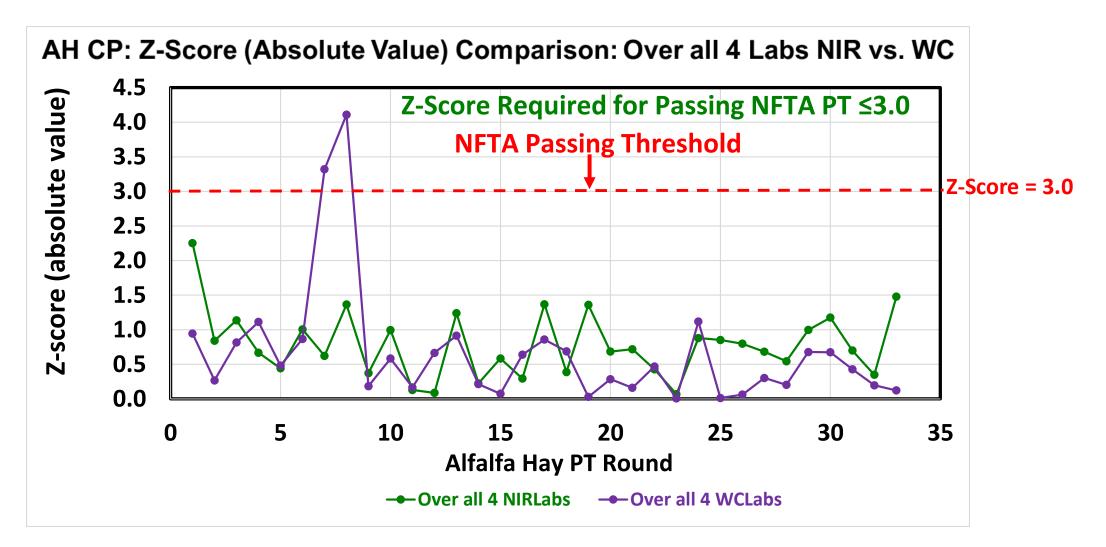
AH DM, Z-Score (Absolute Value) Comparison: Over all (pooled) 4 Labs NIR vs. WC





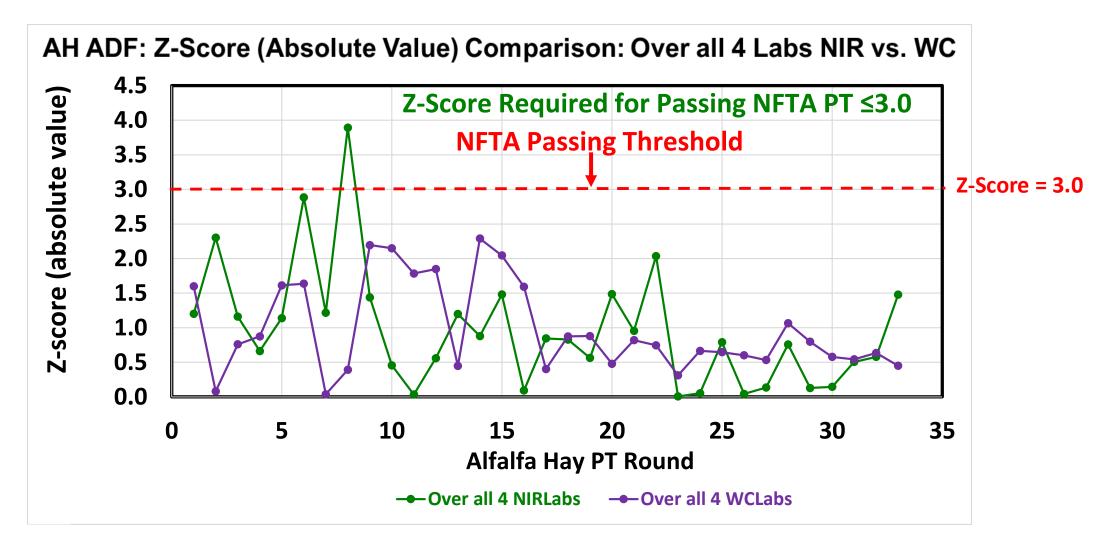


Forage CP Testing Accuracy: NIRS vs. WC





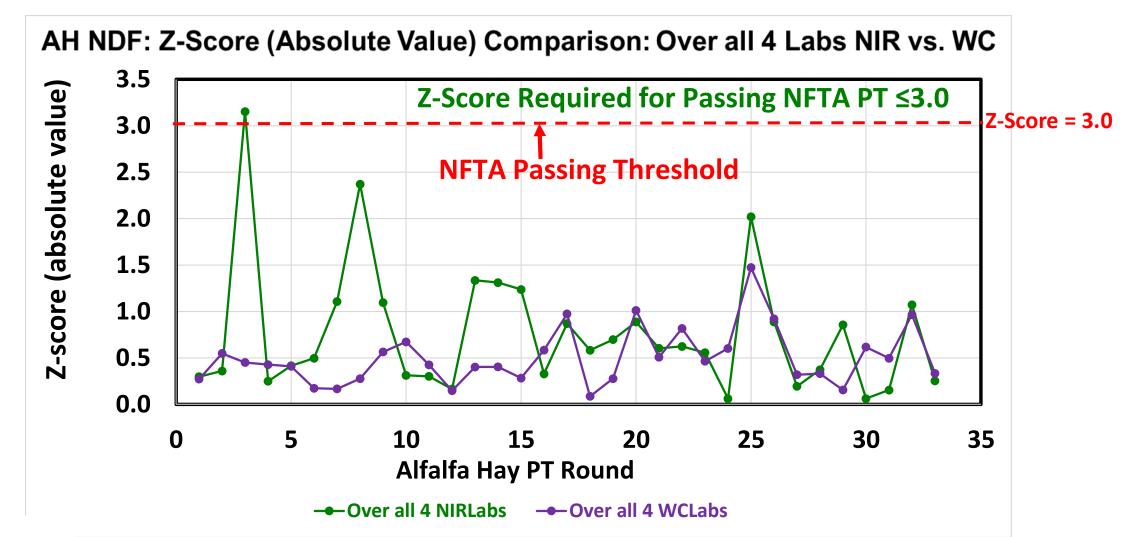
Forage ADF Testing Accuracy: NIRS vs. WC







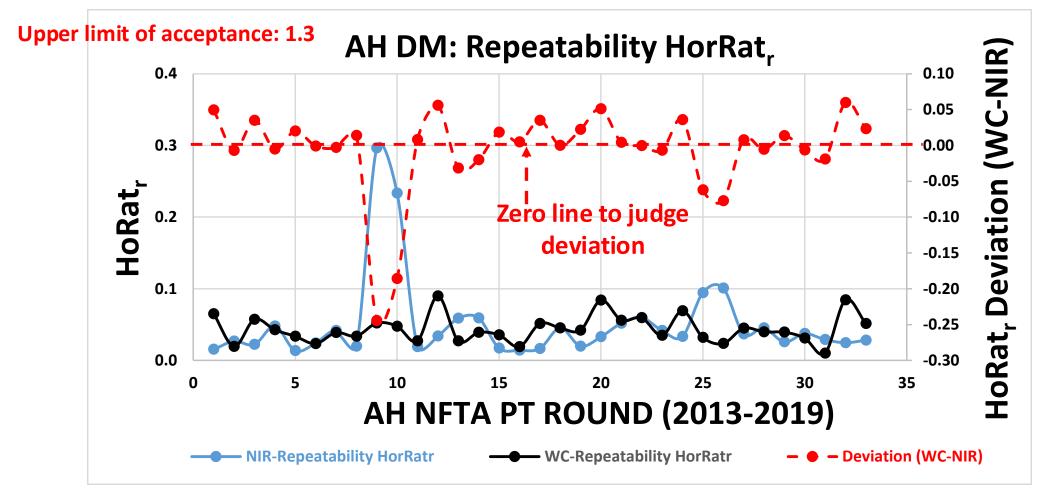
Forage NDF Testing Accuracy: NIRS vs. WC







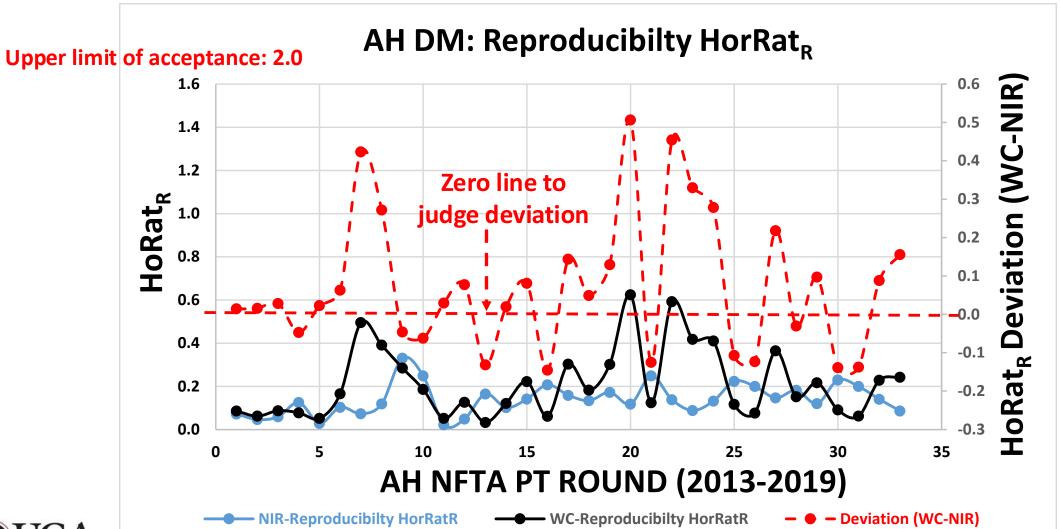
Forage DM Testing Repeatability: NIRS is Better Than WC







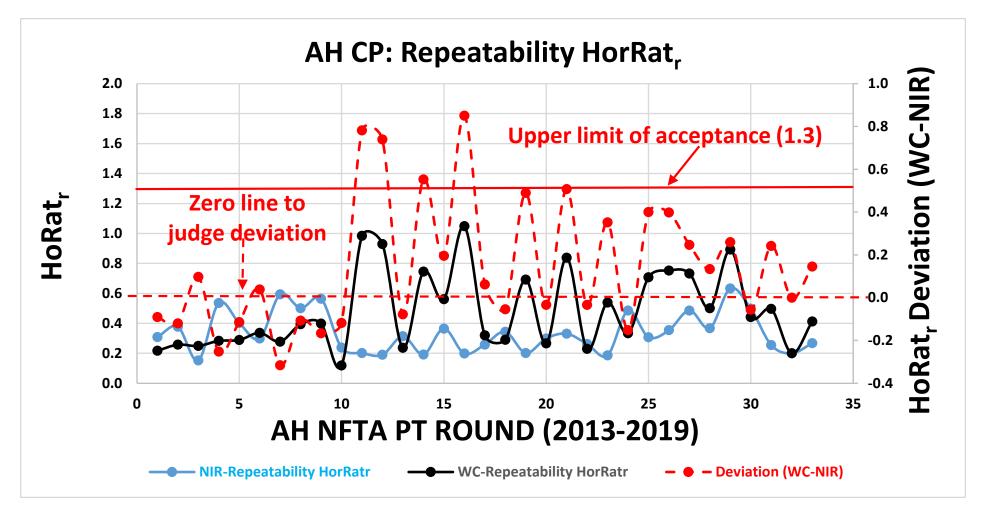
Forage DM Testing Reproducibility: NIRS is Better Than WC







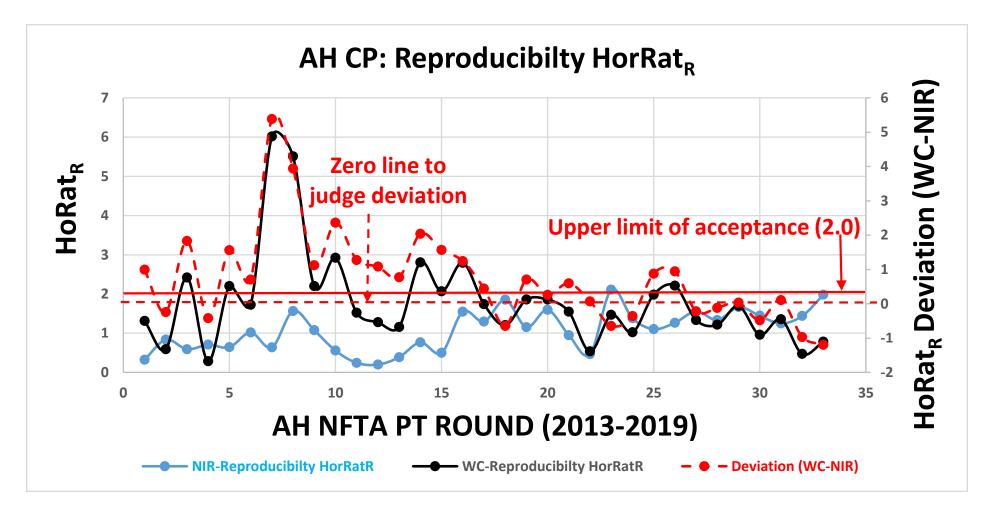
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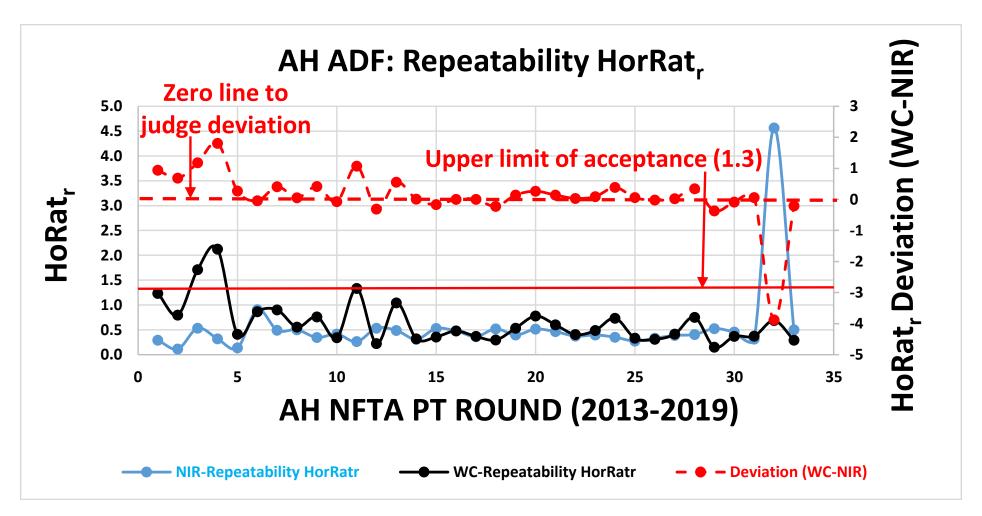
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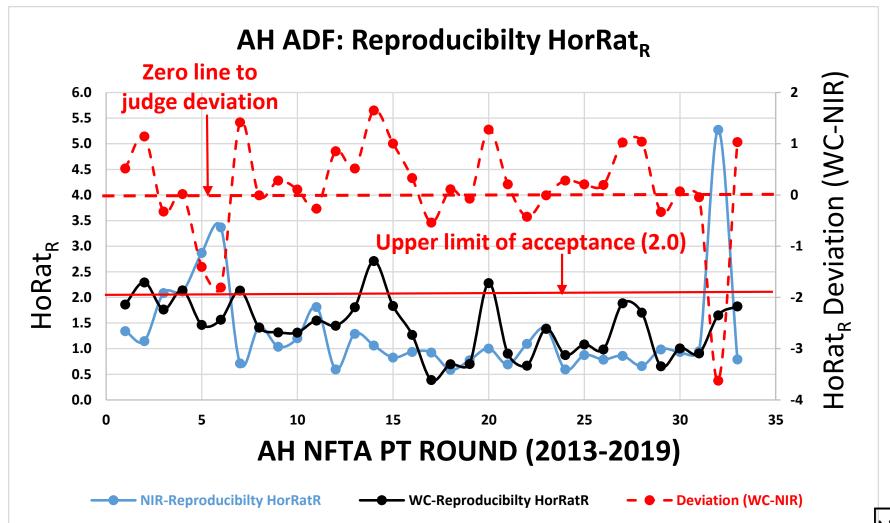
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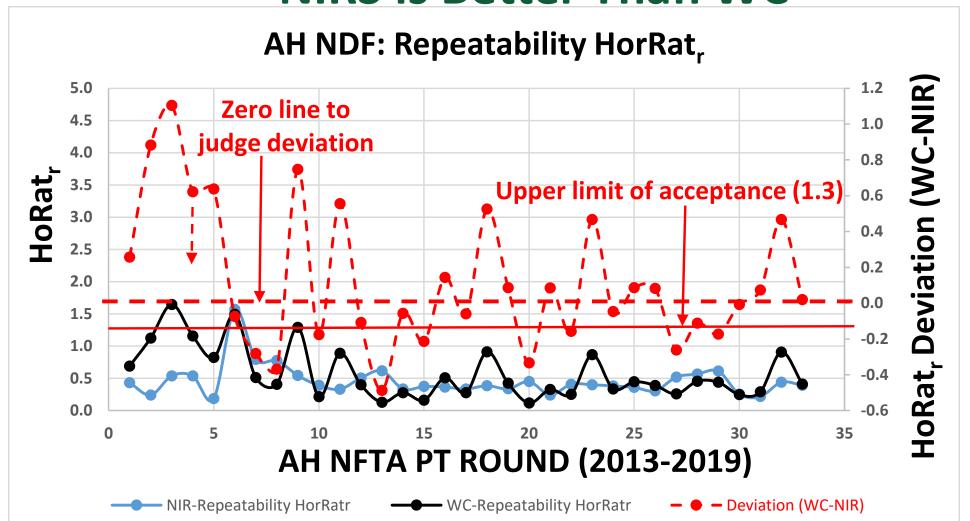
Forage ADF Testing Reproducibility: NIRS is Better Than WC







Forage NDF Testing Repeatability: NIRS is Better Than WC

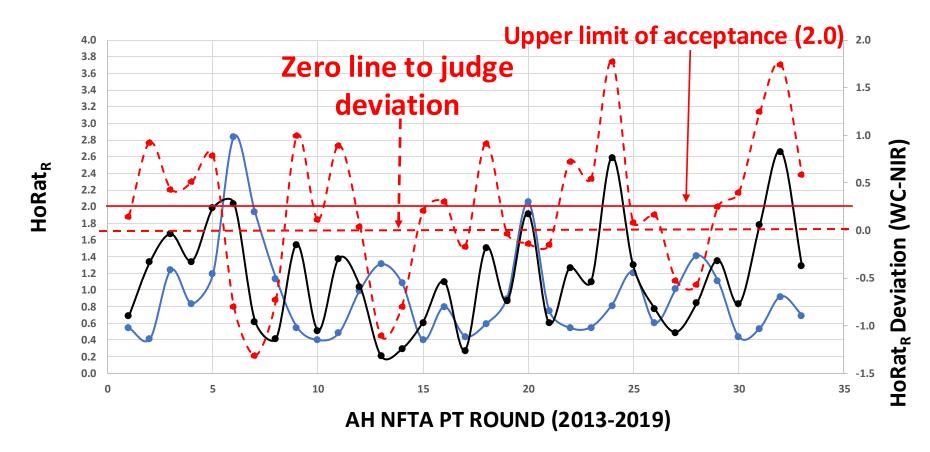






Forage NDF Testing Reproducibility: NIRS is Better Than WC

AH NDF: Reproducibilty HorRat_R







Conclusions

 This presentation is a robust demonstration of the FIRST FACT, which is:

"When NIR calibration models are developed using good science and applied properly, NIR is as accurate as wet chemistry in forage nutritional analysis."

 This presentation is also a robust demonstration of the SECOND FACT, which is:

> "Both intra-laboratory and inter-laboratory precisions of NIR method are better than those of wet chemistry method."





Conclusions

 Many nutritionists who are quick to brush off NIRS, citing its poor accuracy should be pleasantly surprised and change their thoughts by this robust demonstration of similar accuracy of NIRS and WC, and even better precision of NIRS over WC.





Thank You Very Much?

Questions/Comments/Suggestions



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