HTTP:PHET.COLORADO.EDU

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NCTM ANNUAL CONFERENCE APRIL 8, 2017 Actively Engage Students in Content and Practices with Interactive Simulations



Goals

What are interactive simulations?

How can I use sims in my classroom?

What does a sim-based lesson look like?

What new sims can I look forward to?

Interactive simulations...

Are engaging

Are dynamic

Are intuitive

Link representations

Connect to the real world



130+ sims for Physics, Chemistry,

Math, Biology, Earth Science

K-12 and college

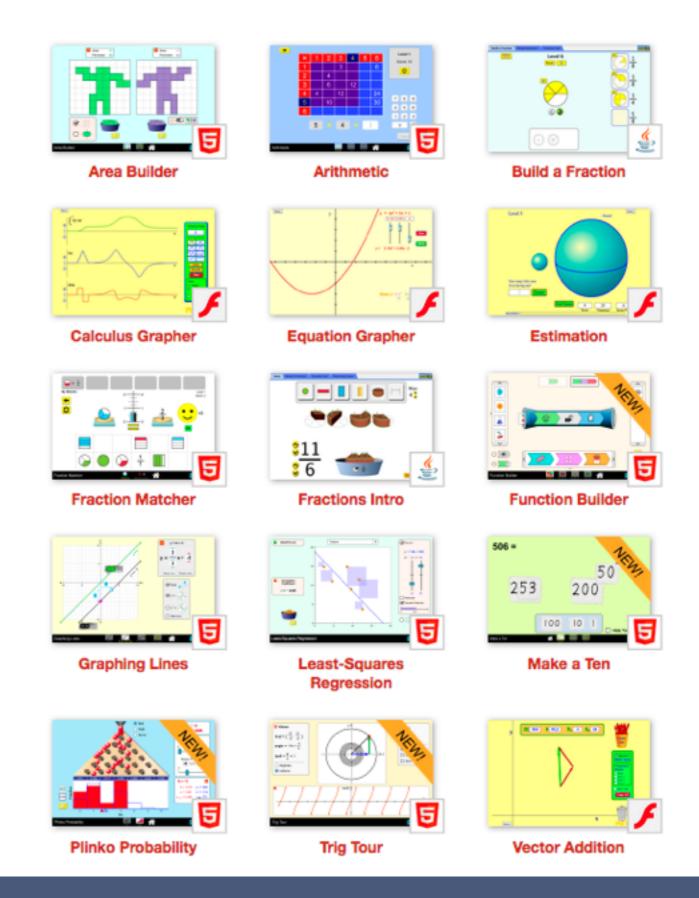
Grant-funded



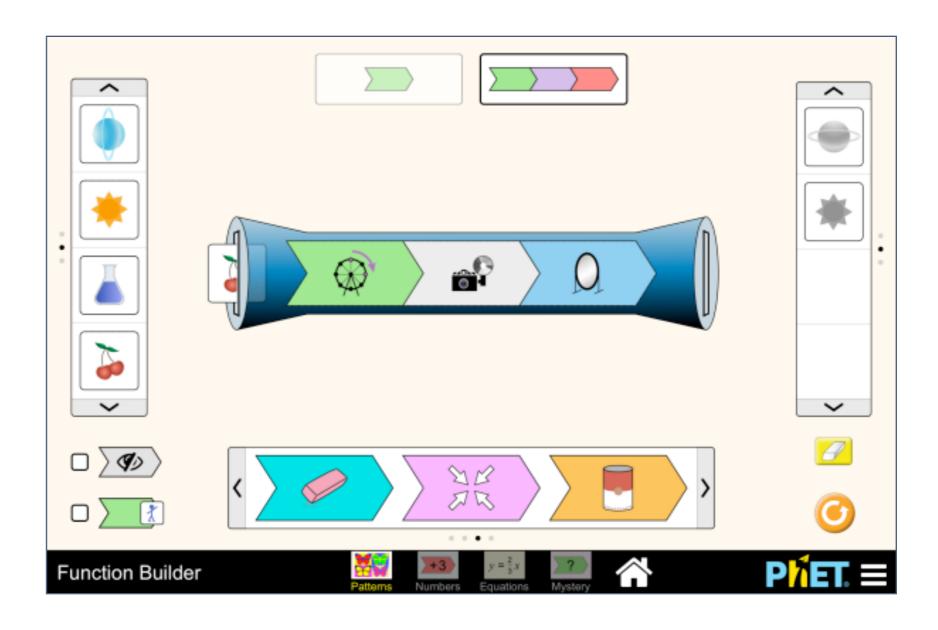


Translated into 78 languages

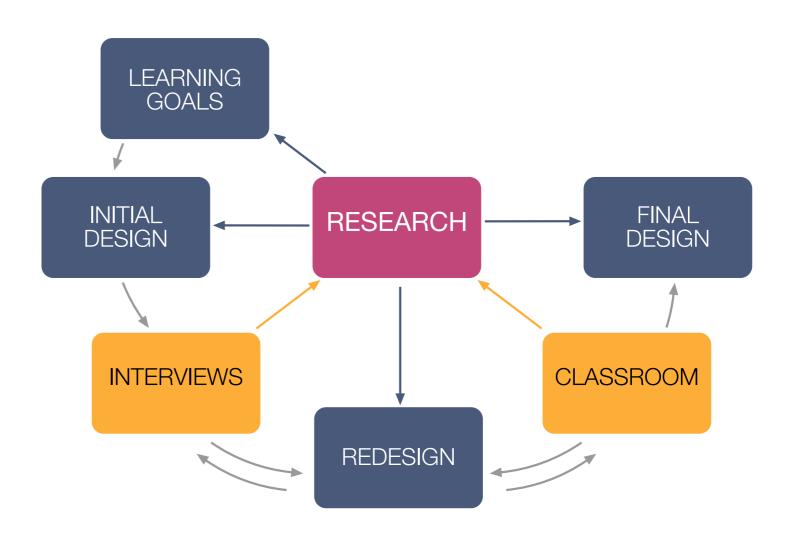
Can be run offline



Sim: Function Builder

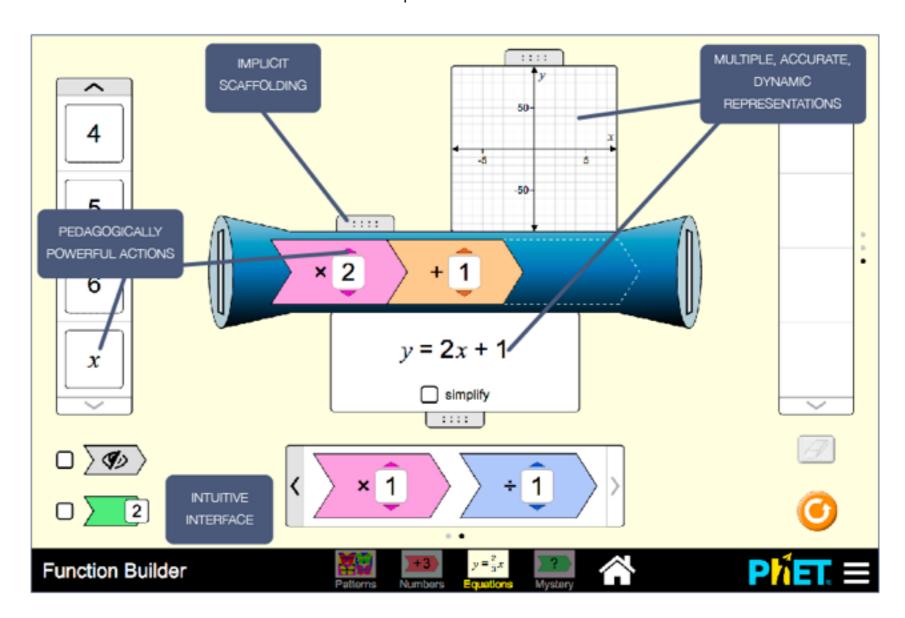


Research-based design

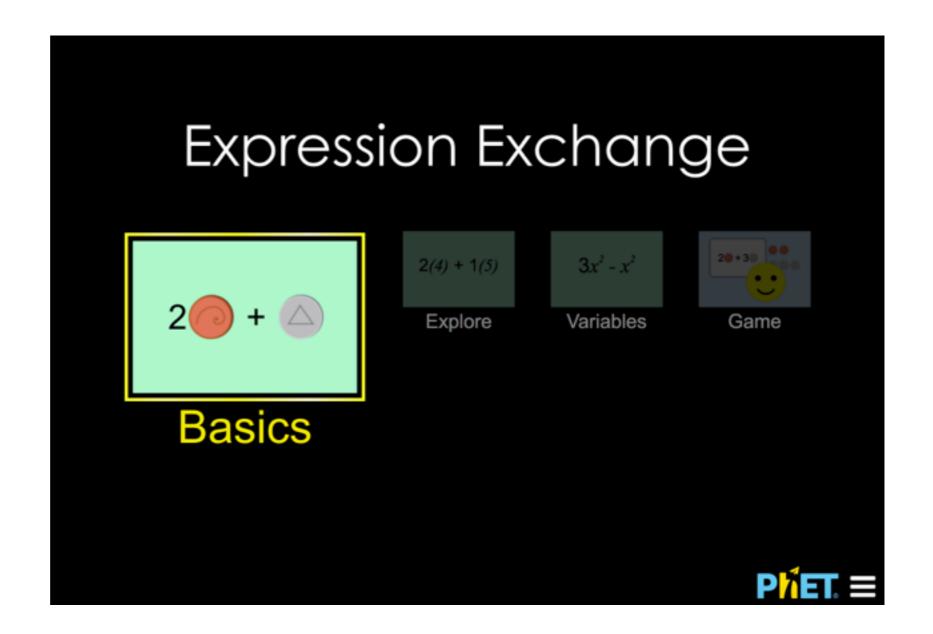


http://phet.colorado.edu/publications/phet_design_process.pdf

Productive Exploration



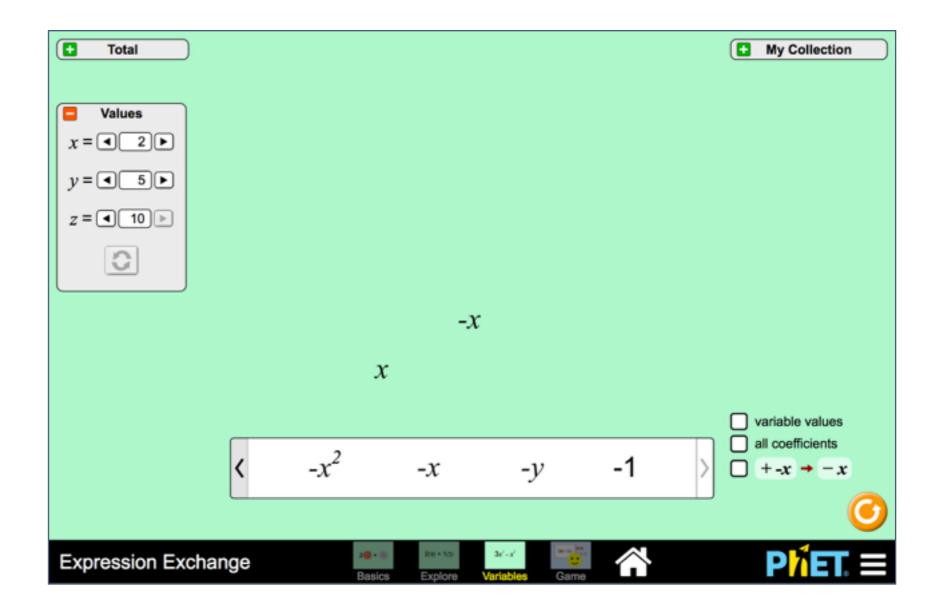
Sim: Expression Exchange



CCSS.Math.Practice.MP2 Reason abstractly and quantitatively.

CCSS.Math.Content. 7.NS.A.1.b

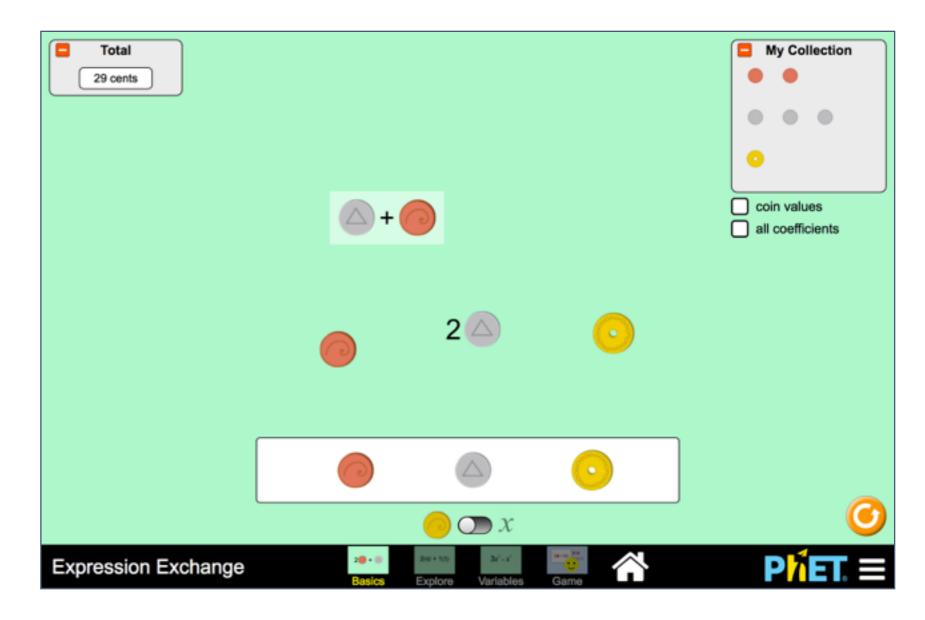
Demonstration



What will happen when we combine x and -x?

CCSS.Math.Practice.MP1 Make sense of problems and persevere in solving them.

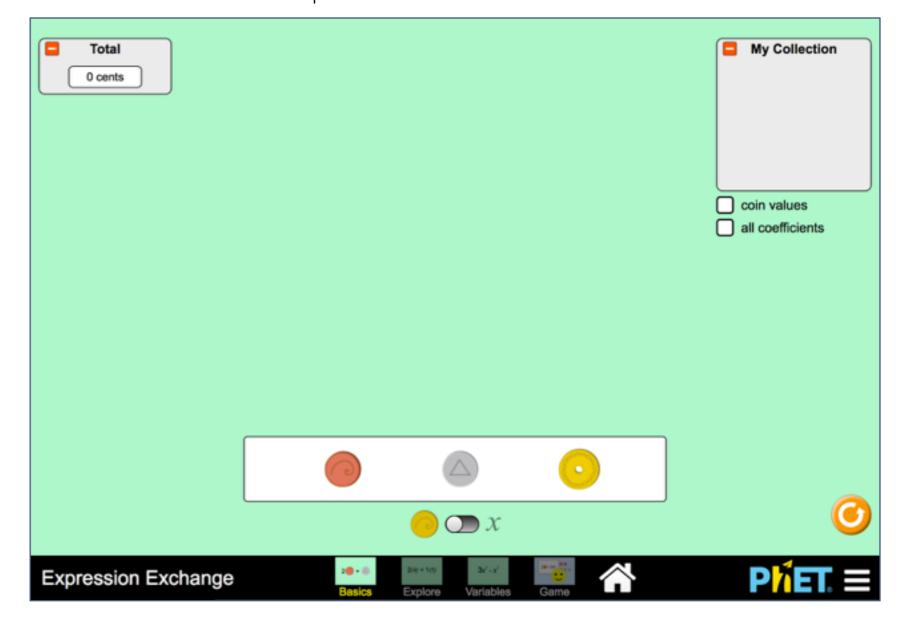
Introductory Exploration



What are the different ways you can combine coins?

CCSS.Math.Practice.MP8 Look for and express regularity in repeated reasoning.

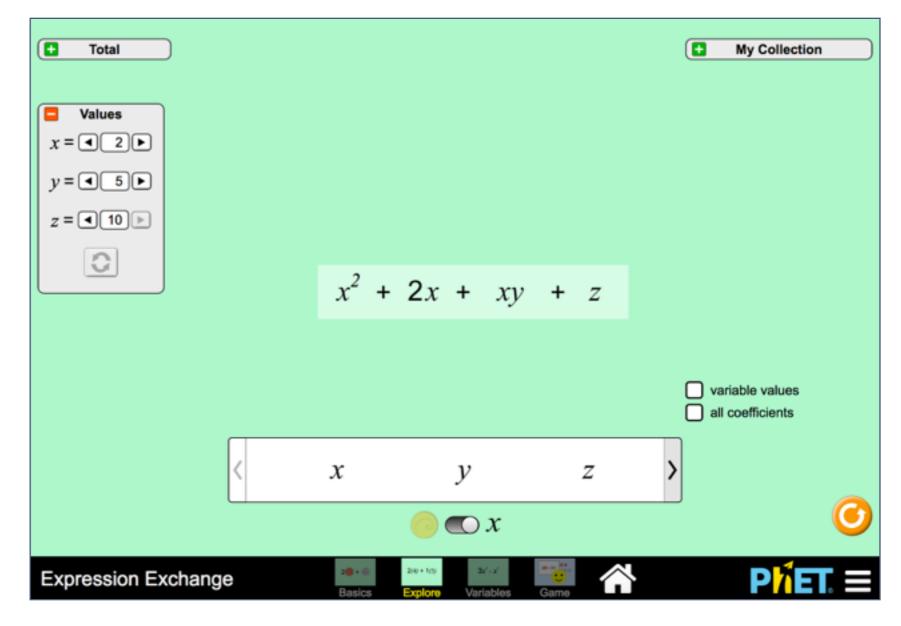
Guided Exploration



Build an expression that equals 97 cents.

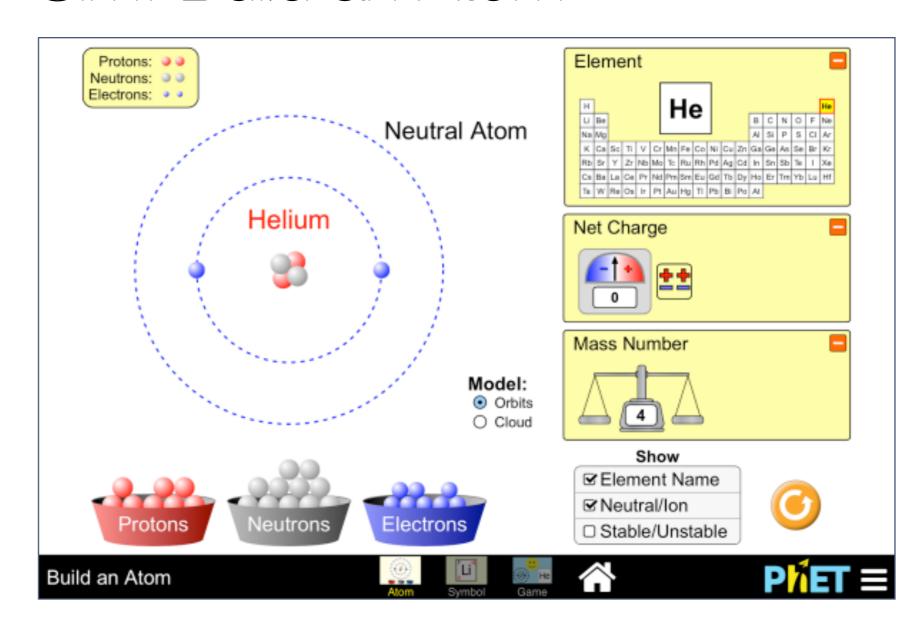
CCSS.Math.Content.
6.EE.A.1
Write and evaluate
numerical expressions
involving whole-number
exponents.

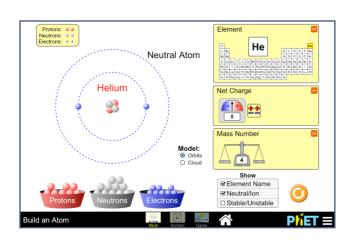
Clicker Questions



What is the total value of this expression?

Sim: Build an Atom





Teacher facilitation strategies

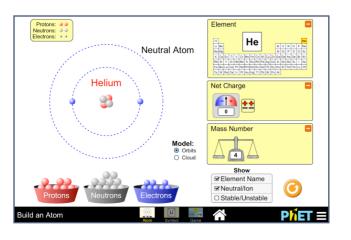
Make learning objectives explicit and meaningful

Start with open play

Avoid explicit instruction

Monitor and measure student learning

Reflect and refine facilitation techniques



Lesson: Atomic Addition

Overview

Prerequisite Skills:

- Identify integers as positive and negative numbers.
- Graph integers on a number line.
- Understand absolute value as the distance a number is from zero on a numberline.

Learning Goals:

- Identify additive inverses (zero pairs).
- · Model addition of integers using protons and electrons, and extend this to a number line.
- Create a rule for adding integers.

Common Core Standards:

CCSS.Math.Content.7.NS.A.1.a Describe situations in which opposite quantities combine to make 0.

CCSS.Math.Content.7.NS.A.1.b Understand p + q as the number located a distance |q| from p, in the positive or negative direction depending on whether q is positive or negative. Show that a number and its opposite have a sum of 0 (are additive inverses). Interpret sums of rational numbers by describing real-world contexts.

CCSS.Math.Content.7.NS.A.1.d Apply properties of operations as strategies to add and subtract rational numbers.

Mathematical Practices:

- 1. Make sense of problems and persevere in solving them
- Reason abstractly and quantitatively
- 4. Model with mathematics.
- 5. Use appropriate tools strategically

Florida Science Standards:

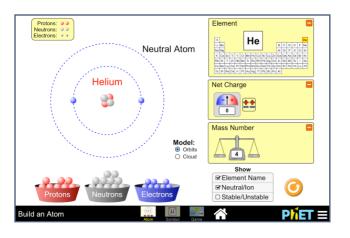
SC.912.P.8.4: Explore the scientific theory of atoms (also known as atomic theory) by describing the structure of atoms in terms of protons, neutrons and electrons, and differentiate among these particles in terms of their mass, electrical charges and locations within the atom.

Materials:

- PhET Build an Atom simulation:
- http://phet.colorado.edu/sims/html/build-an-atom/latest/build-an-atom_en.html
- Computers/tablets for each student or pair of students
- Atomic Addition Activity Sheet (1 per student)

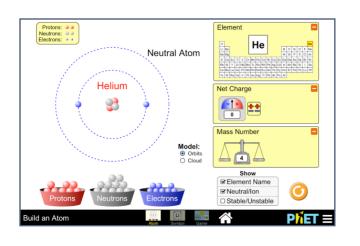
Estimated Time:

Approximately 45 minutes



Lesson: Atomic Addition

 the net charge of an element for today's lesson. Circulate the room and ask students: What do you think net charge means? What happens to the net charge when you add protons? Neutrons? Electrons? What does neutral mean? When does the mass number change? 	Jot down discoveries as #1 on the activity worksheet.
 5. What could represent positive/negative integers? What could represent zero? 6. Why do you think sometimes the net charge is circled? 7. What does the arrow indicate with the net charge? Ask students to briefly share what they wrote down for #1 on the activity sheet, and discuss any of the questions above. 	
Guided Exploration	15-20 minutes
Teacher will	Students will
 Encourage students to begin working on #2-12 in pairs. Try to give them at least 5 minutes where the teacher is silent before probing/aiding. Circulate the room to be available for questions and ask probing/pushing questions, such as: Do the neutrons affect the net charge? If you want a negative net charge, do you need more protons or electrons? What happens if you only use electrons? 	 Complete #2-12 on the activity sheet. Respond to teacher questions. Ask questions or ask for help as needed.
 What happens if you only use protons? What happens if you only use neutrons? What happens if you make an atom with the same number of protons, neutrons, and electrons? Can you make an atom with a net charge of + 3? -4? -5? Can you make a Carbon atom/ion with a net charge of -2? Can you make a Hydrogen atom with a net charge of + 2? Can you make a Neon atom that is neutral or has a net charge of 0? 	



Activity Guidelines

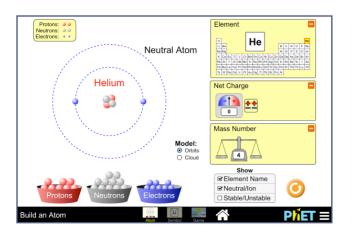
Start with open play

Avoid explicit instruction

Leverage affordances of the sim

Use open, investigative questions and challenge prompts

http://phet.colorado.edu/en/for-teachers/activity-guide



Activity: Atomic Addition

Explore the Build an Atom simulation for a few minutes, building whatever atoms you choose. Write down 1-3
observations you have about building an atom.

2. Build an atom with a **positive net charge**, then **complete the table** below, and **draw the diagram** of its net charge using + and – symbols for protons and electrons.

Protons	
Electrons	
Neutrons	
Net Charge	



 Build a new atom with a different positive net charge, then complete the table below, and draw the diagram of its net charge using + and – symbols for protons and electrons.

Protons	
Electrons	
Neutrons	
Net Charge	

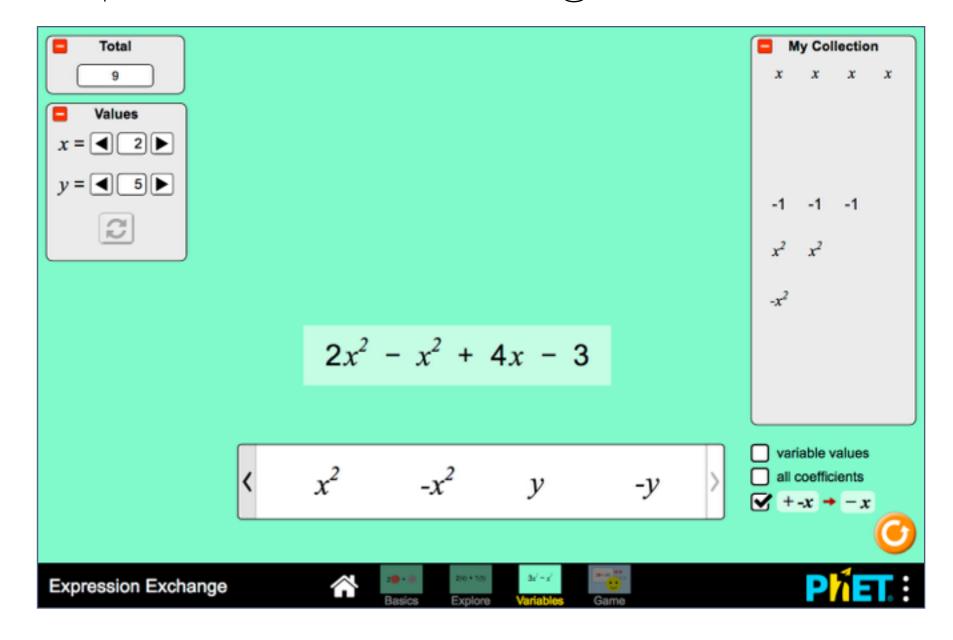


4. Build an atom with a **negative net charge**, then **complete the table** below, and **draw the diagram** of its net charge using + and – symbols for protons and electrons.

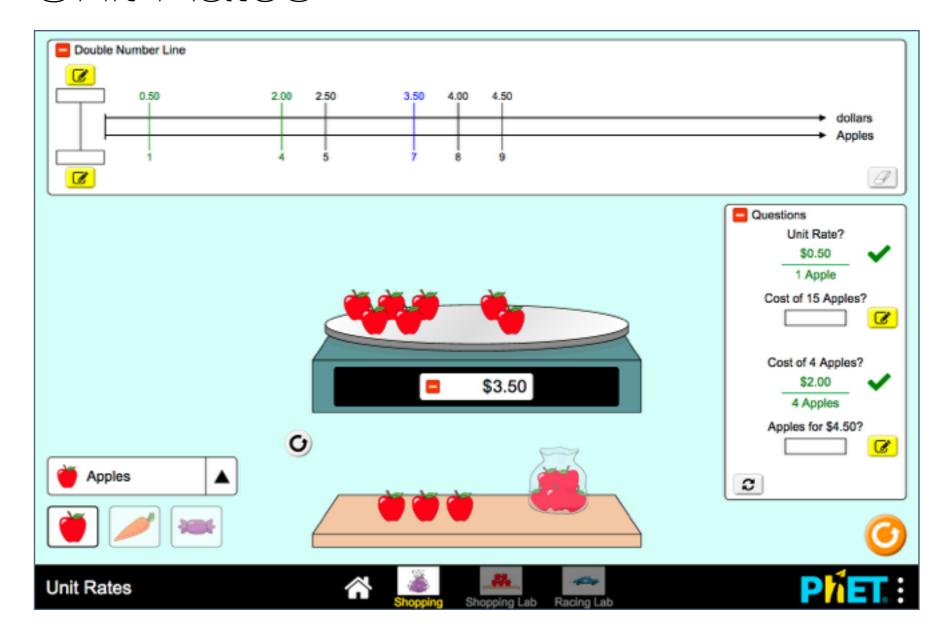
Protons	
Electrons	
Neutrons	
Net Charge	



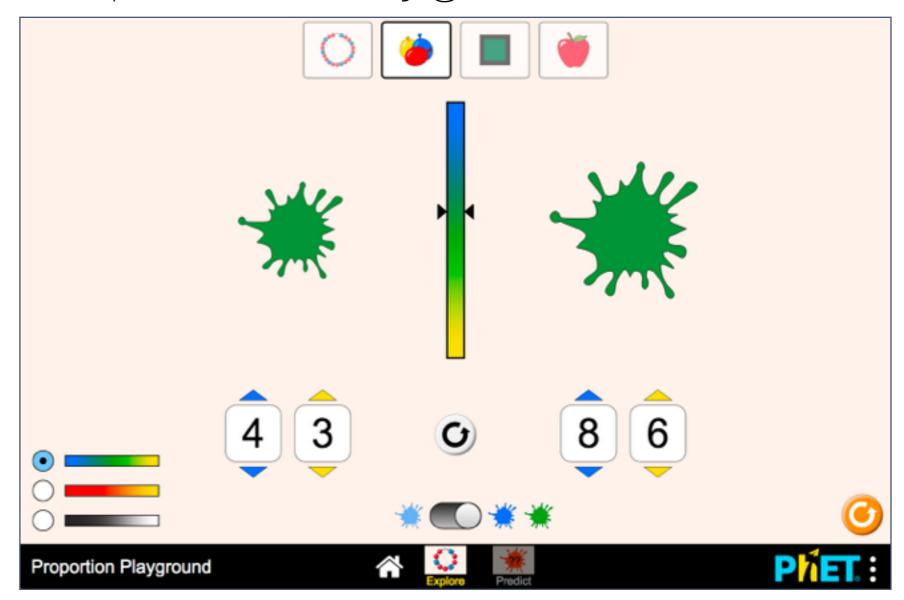
Expression Exchange



Unit Rates



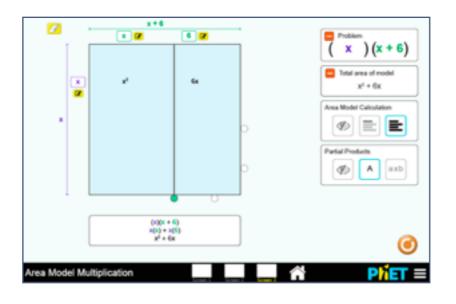
Proportion Playground

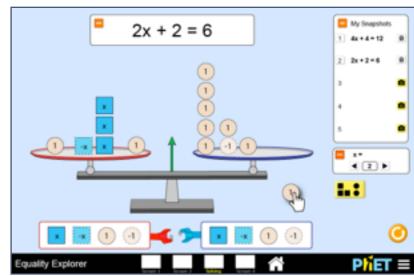


Upcoming Sim Topics

Area model multiplication

Equations and inequalities on a balance





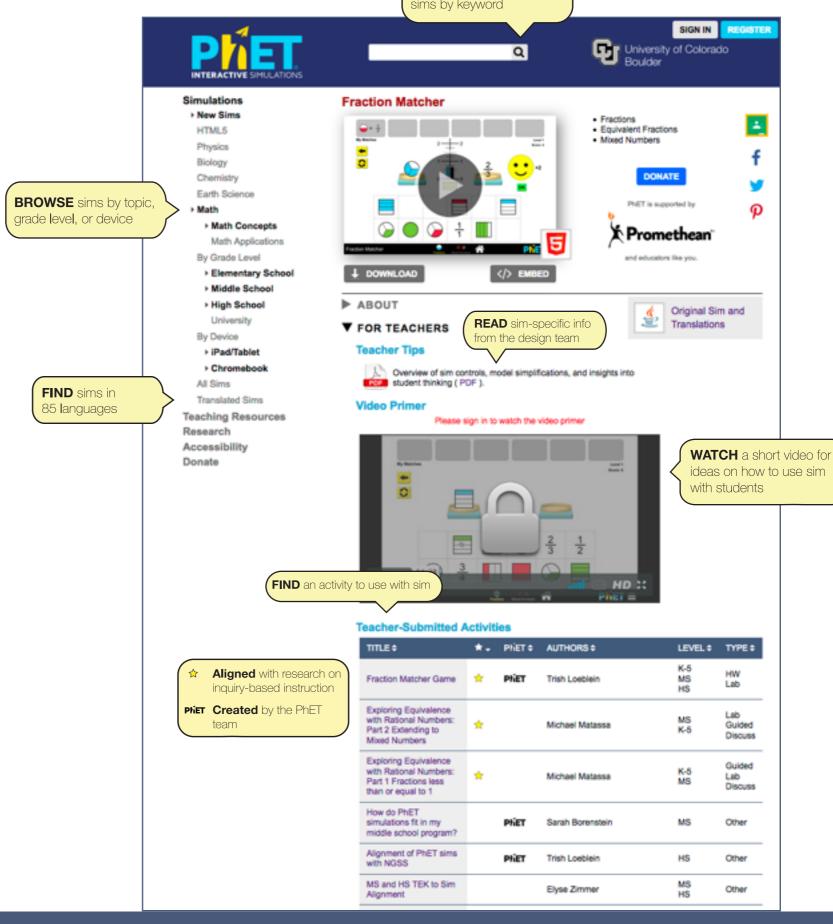
Number lines: comparing integers

Number lines: operations with integers

SEARCH for activities and sims by keyword

Where do I find sims and resources?

phet.colorado.edu



Thank you!

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