



# Why Integrate STEM?

"Scientists use technological tools to conduct experiments and mathematics and statistics to interpret the data produced by those experiments; engineers draw on scientific knowledge and mathematical reasoning to develop and model potential design inventions and solutions; technologists who build and maintain the products and systems designed by engineers must understand the scientific and mathematical principles governing their operation. And these professionals interact with one another in increasingly diverse and multidisciplinary teams."

 STEM Integration in K-12 Education: Status, Prospects, and an Agenda for Research
 National Academy for Engineering and National Research Council



# STEM Integration



- STEM integration does not encourage teaching the four disciplines as independent silos.
- All four STEM content areas will not be integrated into all lessons, all the time.
- Look for meaningful connections and mathematical topics which can be explored using natural phenomena or design challenges.

# Why focus on the "M" in STEM?

STEM learning reinforces that mathematics isn't about "one answer".

16% of American high school seniors are proficient in mathematics and interested in a STEM career.

29% of
Americans rate
this country's
K-12 education in
STEM subjects
as above
average.

U.S Dept. of Edu.

U.S Dept. of Edu.

STEM jobs in the U.S will increase 14% from 2010-2020.
3 million of these jobs will go unfulfilled.

STEM learning broadens students' perspective of mathematics to be more than computational approaches.

College students
interested in
STEM-related
careers often
abandon this path
within their first
two years of study.

Bill Nye





- Identify a location on the coordinate plane by using an ordered pair of numbers, or representative values, in reference to perpendicular number lines, called axes.
- Perform biotechnology skills in a simulated application using fixed volume micropipettes and a coordinate grid system.

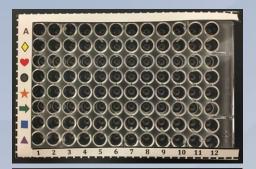


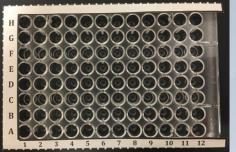
Students will begin by investigating the 10 µl micropipette and micropipette tips.

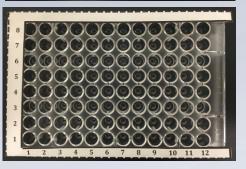
- Practice handling the micropipette and micropipette tips.
- Predict: How much water will a 10 µl micropipette dispense?
- How did your prediction compare to the actual amount of water dispensed?
- How much smaller is a microliter compared to a liter? How much smaller are 10 μl compared to a liter?
- Why do you think scientists would work with a sample of liquid that is this small?



Students will then observe and explore the 96 well-plate or Well Plate Grid.







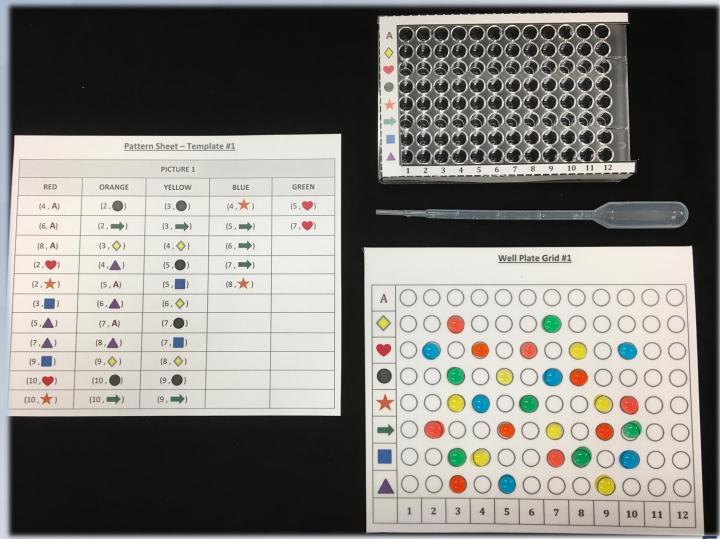
- What do you notice about the symbols, or values, that stretch across the **top** of your well plate (or grid)?
- What do you notice about he symbols, or values, that stretch across the **side** of your well plate (or grid)?
- These lines are called "axes". How might you use axes to locate a well?
- How might you write the location of a well?
- What might we call the point at which both axes meet?
- Well plates use a coordinate system to locate each well. Have you seen a grid, or coordinate system, before? Did it look the same, or different, from this system? Explain.

Using the materials provided on your table, choose a Pattern Sheet and complete the image by pipetting the appropriate colors into the corresponding wells.

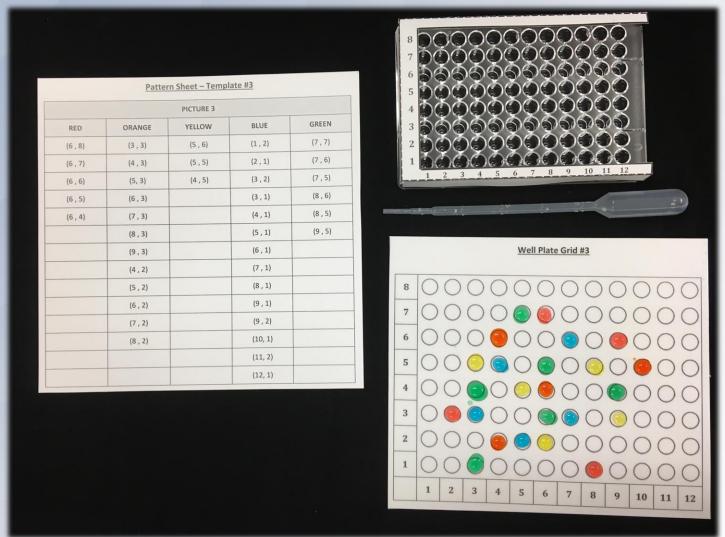


- ✓ Select a new tip for each color.
- ✓ Dispose of used tips in the paper cup.

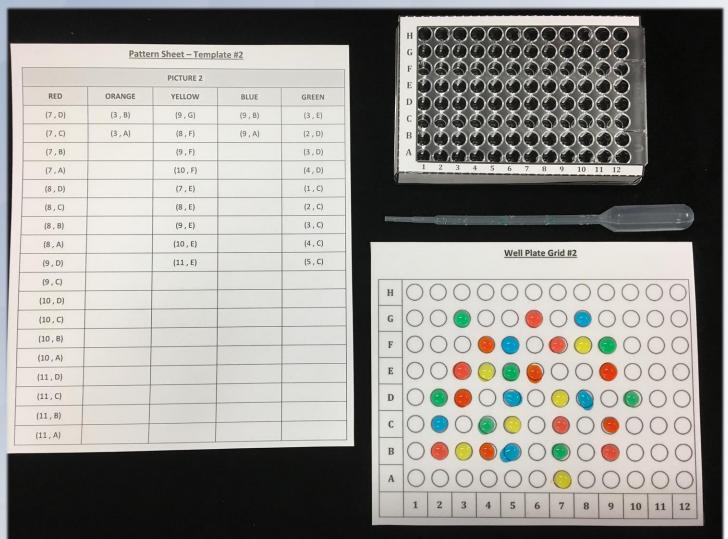














#### **Biology and Geometry: A Telling Image – DEBRIEF**

- Was pipetting easy? Challenging? Explain.
- How did you determine the location in which to pipette each color solution?
- What surprised you about the volume, or amount of liquid, that was dispensed from the micropipette?
- What other activities might a well plate be used for?
- What materials might a scientist use a well plate to explore?



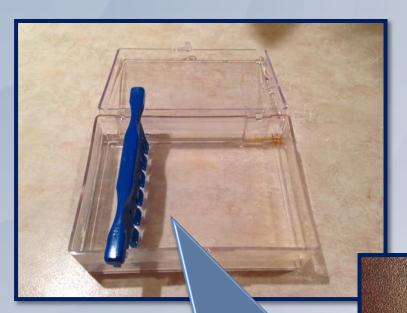


### **Extension: Gel Electrophoresis**

- Apply knowledge of biotechnology skills and practices by performing electrophoresis to resolve the colored dye molecules within a sample.
- Conduct a scientific experiment by precisely following a multistep procedure and performing technical tasks.



#### **Extension: Gel Electrophoresis**



Directions & Checklist for Preparing Gels for Run							
CHECK	STEP	INSTRUCTIONS					
	1	Put 1 agarose tablet in 50 mL of water.					
	2	Let it sit for 2 minutes or until completely dissolved.					
	3	Heat in microwave until it boils. Roughly 45-65 seconds. Allow to boil for about 10 seconds.					
	4	Stop. Swirl contents. Heat again for 20-30 seconds.					
	5	Stop. Swirl contents. Heat again for 20-30 seconds.					
	6	Let the agarose cool for 3-5 minutes. Do NOT let it solidify.					
7		Pour it into a plastic box.					
8		Insert a comb so it rests at a hinge.					
9 10 11		Allow the gel to set up. It will turn cloudy when it is setup.					
		Remove comb once gel is set up. Clean comb in soap and water.					
		Close box lid and store in refrigerator until ready to use.					
	12	Repeat for the other boxes.					

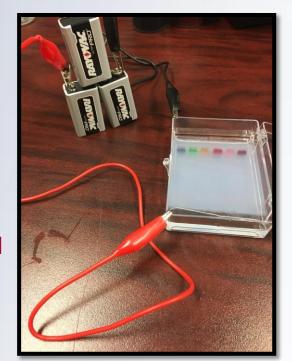
Optional
Mathematics
Challenge:
Construct a Comb
pg. 28



#### **Extension: Gel Electrophoresis**

#### **Procedure:**

- Remove all materials from the bag.
- Prepare the power source by connecting three 9V batteries together by inserting the positive terminal of one into the negative terminal of another.
- Attach the **red** alligator clip to the positive terminal and the **black** clip to the negative terminal.
- Open the gel box and carefully place one unfolded paperclip at each end of the gel. Adjust the paperclips to fit in the box, if necessary.
- Pour enough TAE buffer into the gel box to slightly submerge the gel.
- Using the micropipette, load approximately 20 µl of Kool-Aid into each well.
- Attach the other end of the black alligator clip to the paperclip above the wells, and red clip to the paperclip below the wells.





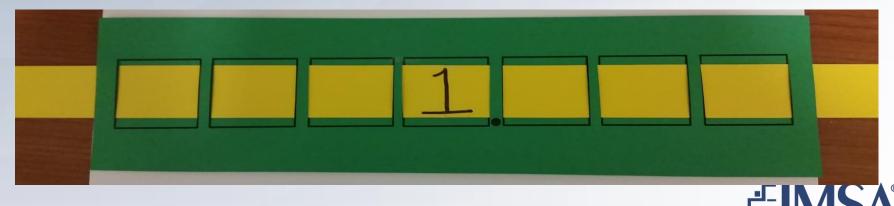
#### **Extension: Gel Electrophoresis – DEBRIEF**

- What happened to your Kool-Aid samples?
- Gel electrophoresis needs a power source. That power source may be batteries or a direct current from an electrical outlet. Why do you think power is needed?
- Did all of the Kool-Aid samples separate into individual color dyes? Provide evidence to support your thoughts.
- How did electrophoresis separate the Kool-Aid samples?
- What other items could we separate with electrophoresis?



### Come Scale Away - ACTIVITY 1

- Introduce Powers of Ten through a hands-on exploration using a paper "tool".
- Begin by sliding the number 1 until it is in the ones column.
- Multiply by 100. Represent this number using the slider to move the 1 and adding zeroes.
- Go back to 1. Now, divide by 100. Represent this number using the slider to move the 1 and adding zeroes.



Part 2: Powers of 10 Observation Table

With your partner, observe the patterns in the table below:

Standard Form	Base 10 Form	Exponential Form	Decimal
1,000,000	10 x 10 x 10 x 10 x 10 x 10	10 <sup>6</sup>	
100,000	10 x 10 x 10 x 10 x 10	10 <sup>5</sup>	
10,000	10 x 10 x 10 x 10	10 <sup>4</sup>	
1,000	10 x 10 x 10	10 <sup>3</sup>	
100	10 x 10	10 <sup>2</sup>	
10	10	$10^{1}$	
1	1	$10^{0}$	
$\frac{1}{10}$	$\frac{1}{10}$	$10^{-1}$	0.1
$\frac{1}{100}$	$\frac{1}{10 \times 10}$	$10^{-2}$	0.01
$\frac{1}{1,000}$	$\frac{1}{10 \times 10 \times 10}$	$10^{-3}$	0.001
1 10,000	$\frac{1}{10 \times 10 \times 10 \times 10}$	$10^{-4}$	0.0001
1 100,000	$\frac{1}{10 \times 10 \times 10 \times 10 \times 10}$	$10^{-5}$	0.00001
1 1,000,000	$\frac{1}{10 \times 10 \times 10 \times 10 \times 10 \times 10}$	$10^{-6}$	0.000001

What patterns do you observe in the table?

How does the Exponential Form compare to Base 10 Form?

What do you think the "–" means in the Exponential Forms?



ultiply

## Come Scale Away – ACTIVITY 2

 $10^{-5}$ 

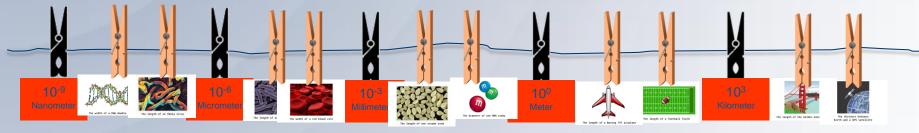
#### Relate Metric Table to Powers of Ten Table

	Standard Form	Base 10 Form	Exponential Form	Decimal				
	1,000,000	10 x 10 x 10 x 10 x 10 x 10	$10^{6}$		N	Actric Table		
	100,000	10 x 10 x 10 x 10 x 10	10 <sup>5</sup>		Metric Table			
	10,000	10 x 10 x 10 x 10	10 <sup>4</sup>	Prefix	Abbreviation	Power of 10	× "bigger" than base	
				giga	G	<b>10</b> <sup>9</sup>	1,000,000,000	
	1,000	10 x 10 x 10	10 <sup>3</sup>	mega	M	<b>10</b> <sup>6</sup>	1,000,000	
	100	10 x 10	10 <sup>2</sup>	kilo	k	10 <sup>3</sup>	1,000	
			1	hecto	h	10 <sup>2</sup>	100	
	10	10	$10^{1}$	deka	da	10 <sup>1</sup>	10	
	1	1	$10^{0}$	meter	m	Base 10 <sup>0</sup>	1	
	1	1	$10^{-1}$	deci	d	<b>10</b> <sup>-1</sup>	0.1	
	10	$\overline{10}$		centi	С	10 <sup>-2</sup>	0.01	
	1	1		milli	m	<b>10</b> <sup>-3</sup>	0.001	
	$\frac{1}{100}$	$\frac{1}{10 \times 10}$	$10^{-2}$	micro	μ	<b>10</b> <sup>-6</sup>	0.000001	
	1	1		nano	n	<b>10</b> <sup>-9</sup>	0.00000001	
	1,000	$\frac{1}{10 \times 10 \times 10}$	$10^{-3}$	0.001				
	1 10,000	$\frac{1}{10 \times 10 \times 10 \times 10}$	$10^{-4}$	0.0001			#IMSA®	

0.00001

#### Come Scale Away – ACTIVITY 2

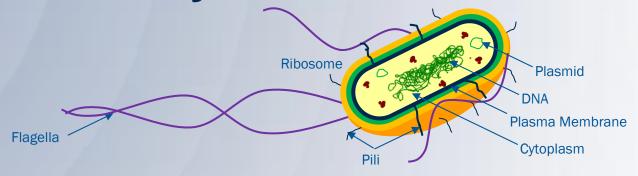
At your tables, order the orange Metric Measure cards along the string from smallest to largest and secure with clothespins.



- Now sort the Sorting Cards from smallest to largest along the "number line."
- The Clue Cards can help with the sorting.



### Come Scale Away - ACTIVITY 3



- Student teams will be provided with sticky note sheets to share their ideas on the function of each labeled part of the *E. coli* bacterium.
- Use student words as much as possible to reach a consensus about each function.
- Then, create a model of the chromosomal DNA contained within the cell using snack bags, meter sticks, and thread.

# Come Scale Away - DEBRIEF

- How difficult do you think it must be for biologists to work with organisms that are so very small?
- Why might biologists be interested in the relative size of DNA contained within a singlecelled organism?
- What other "things" can we use powers of 10 to measure?





## **THANK YOU!**

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- Download lessons from: goo.gl/V3mOJn



